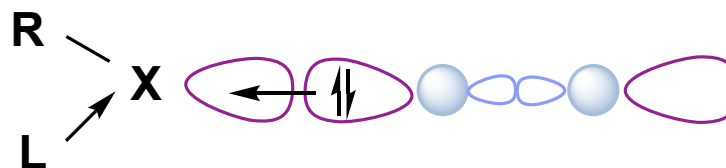
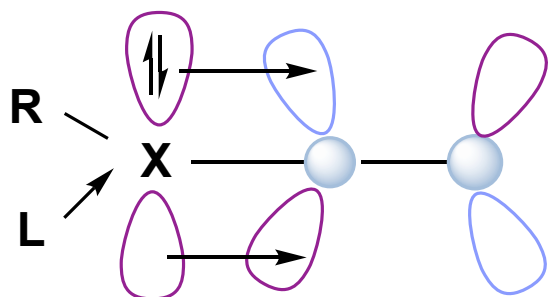
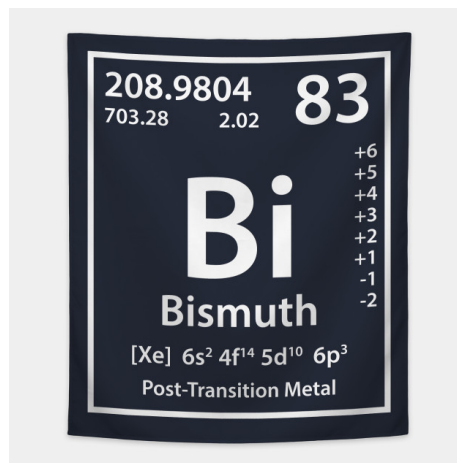


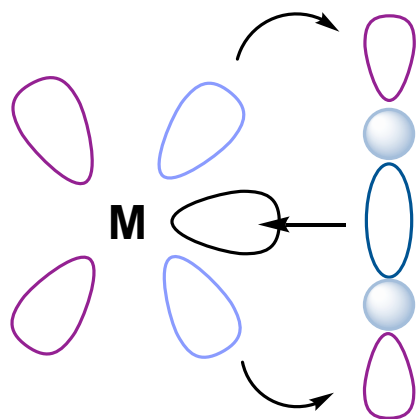
Main-group Elements (Bi) in Catalysis for Organic Synthesis



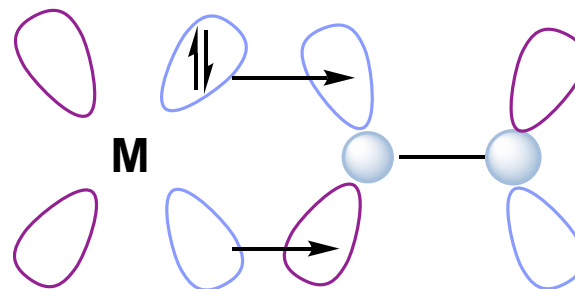
Xiaheng Zhang
MacMillan Group Meeting
March 18th, 2020

Mechanisms of Activation Mode for Main-group Elements

■ Activation mode of transition metal elements

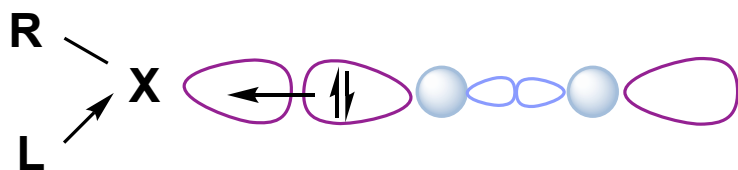


σ donation into empty d -orbital

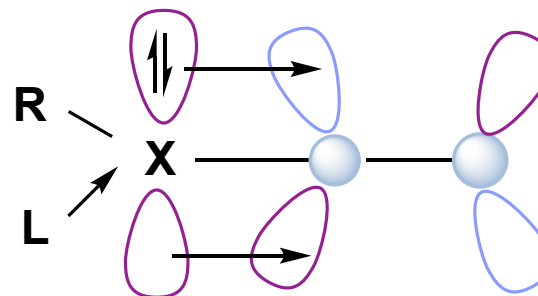


π donation from d -orbital into π^* -orbital

■ Activation mode of main-group elements



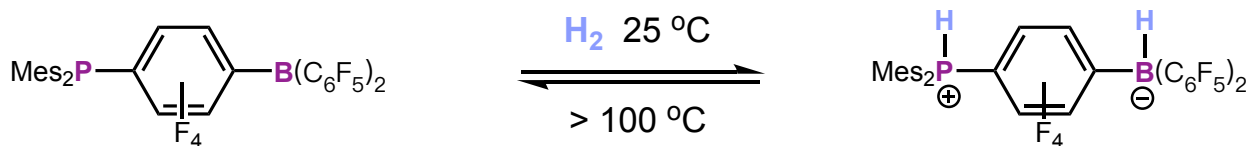
σ donation into empty p -orbital



π donation from p -orbital into π^* -orbital

Activation of Small Molecules by Main-group Elements

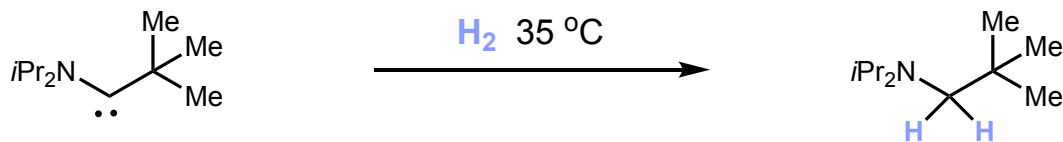
Frustrated Lewis pairs activation



Stephan, D. W. et al. *Science* **2006**, 314, 1124.

Stephan, D. W. et al. *Science* **2016**, 354, aaf7229.

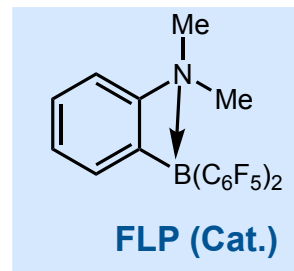
Activation of H_2 by carbenes



Frey, G. D.; Lavallo, V.; Donnadieu, B.; Schoeller, W. W.; Bertrand, G. *Science* **2007**, 316, 439.

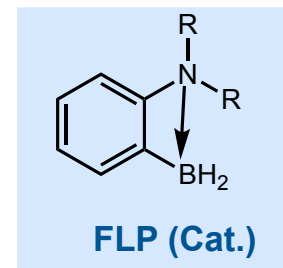
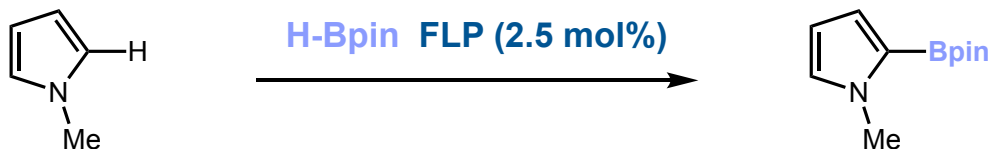
Catalysis Enabled by Main-group Elements

Frustrated Lewis pairs catalyzed hydrogenation



Chernichenko, K.; Madaraś, A.; Papai, I.; Nieger, M.; Leskela, M.; Repo, T. *Nat. Chem.* **2013**, *5*, 718.

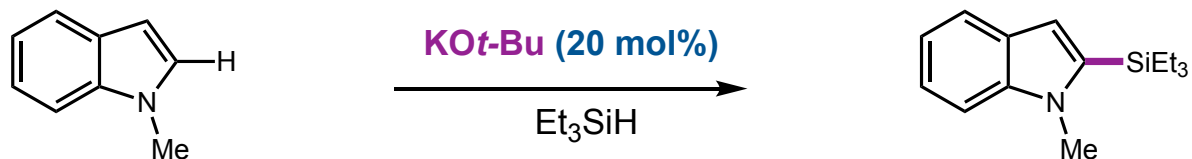
FLP catalyzed C-H activation



Legare, M.-A.; Courtemanche, M.-A.; Rochette, É.; Fontaine, F.-G. *Science* **2015**, *349*, 513.

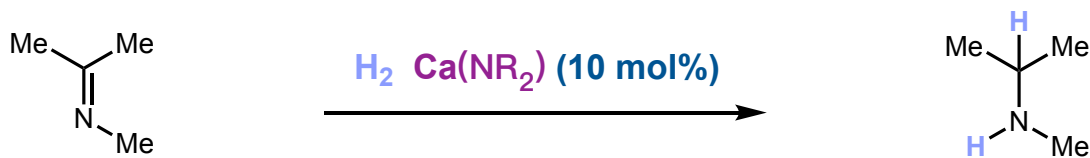
Catalysis Enabled by Main-group Elements

C-H silylation catalyzed by alkali metal catalyst



Toutov, A. A.; Liu, W.-B.; Betz, K. N.; Fedorov, A.; Stoltz, B. M.; Grubbs, R. H. *Nature* **2015**, 518, 80.

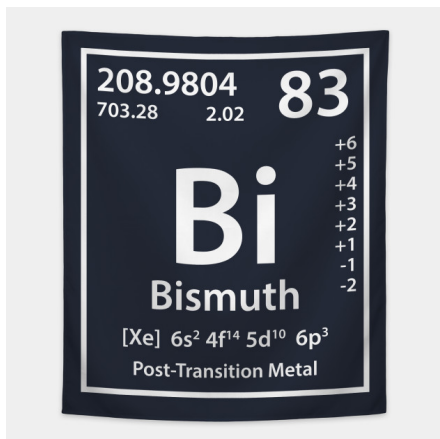
Imine hydrogenation catalyzed by alkaline earth metal catalyst



Harder, S. et. al. *Nat. Catal.* **2018**, 1, 40.

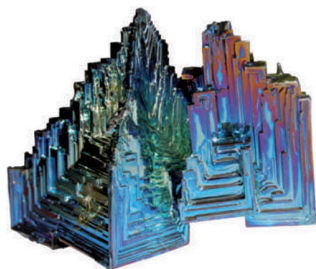
Main-group Elements (Bi) in Catalysis for Organic Synthesis

Outline



- The facts of bismuth
- Synthesis of organobismuthines
- The application of organobismuthines in organic synthesis
- Redox chemistry of bismuth

Why do people care about bismuth catalysis



electron configuration: $4f^{14}5d^{10}6s^26p^3$
common oxidation state: +1, +3, +5



Peptobismol OTC

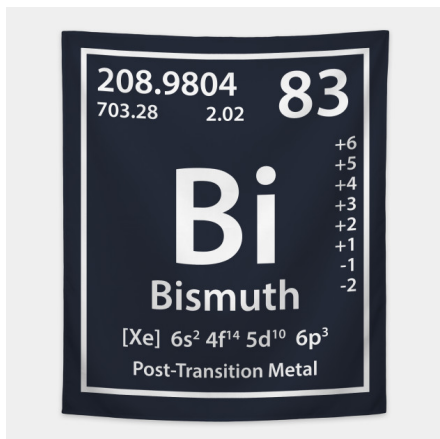
stomach disorders

- naturally abundant
- 'green element' even less toxic than NaCl
- widely used in industry, antibiotics, radiopaque bone cements, polymers

Can we discover new reactivity that is unique to bismuth catalysis?

Main-group Elements (Bi) in Catalysis for Organic Synthesis

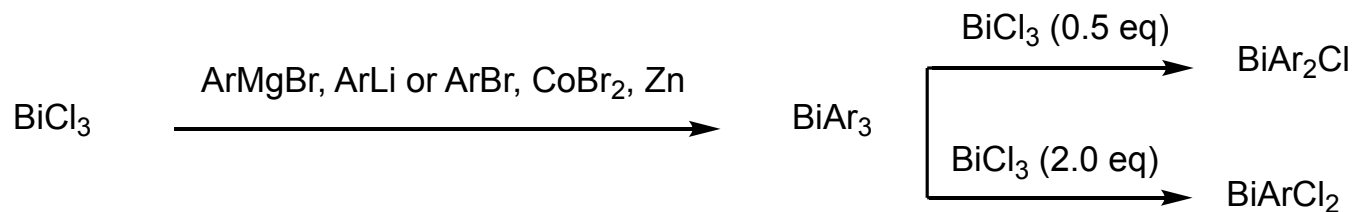
Outline



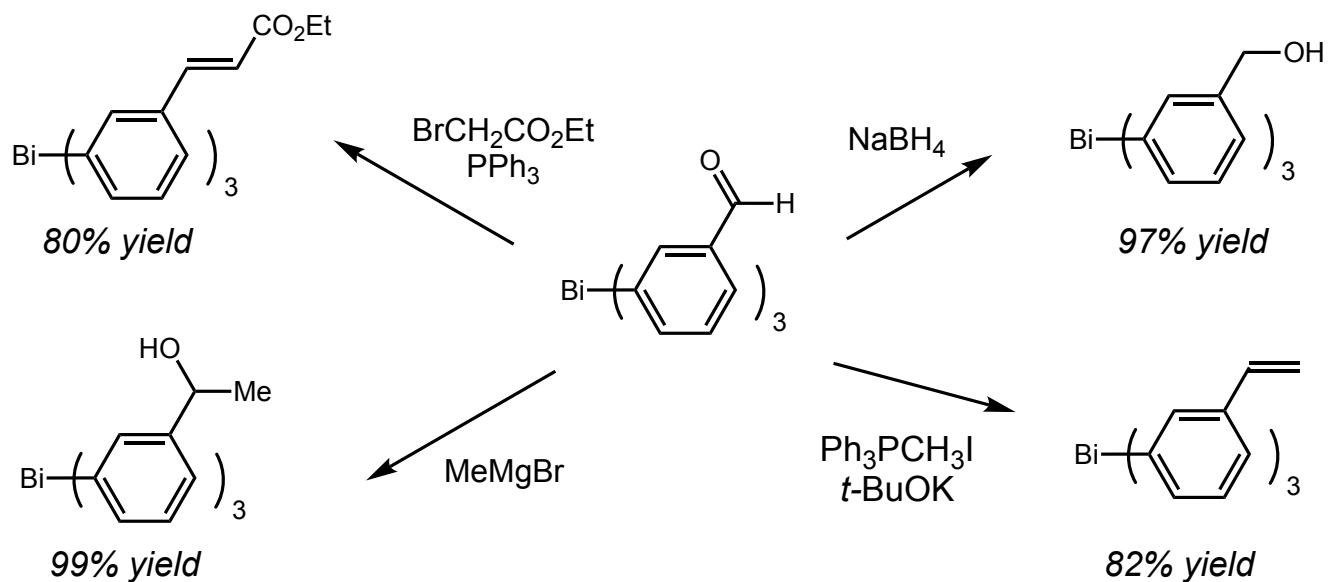
- The facts of bismuth
- Synthesis of organobismuthines
- The application of organobismuthines in organic synthesis
- Redox chemistry of bismuth

Synthesis of organobismuthines

■ Synthesis of trivalent organobismuthines

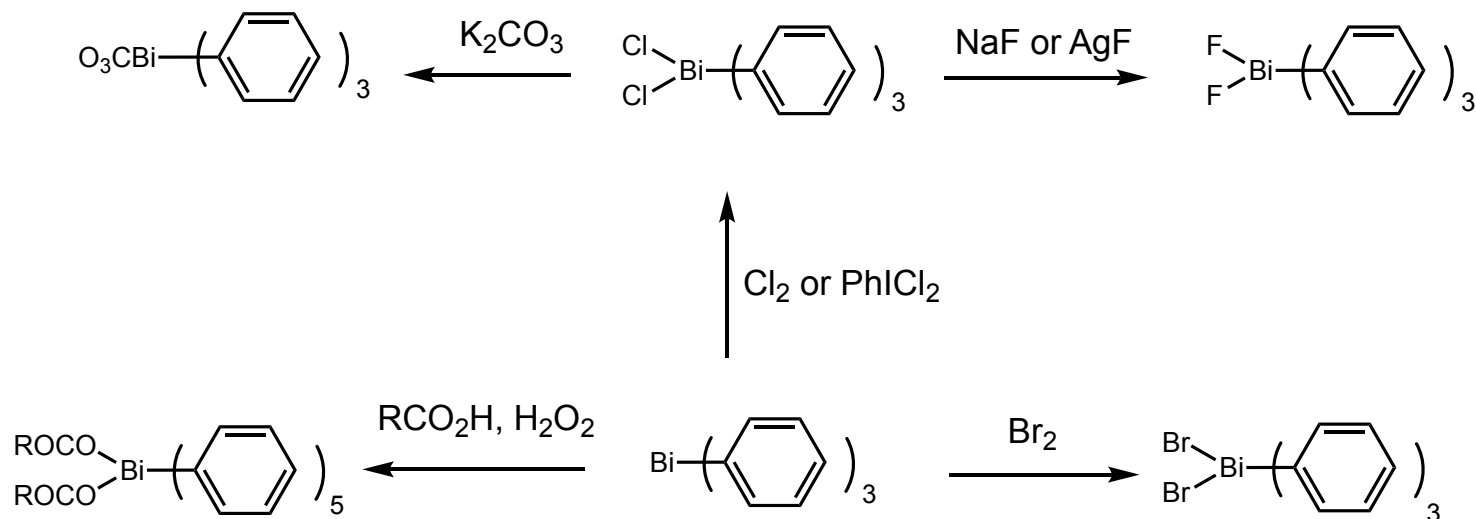


■ Synthesis of highly functionalized trivalent organobismuthines



Synthesis of organobismuthines

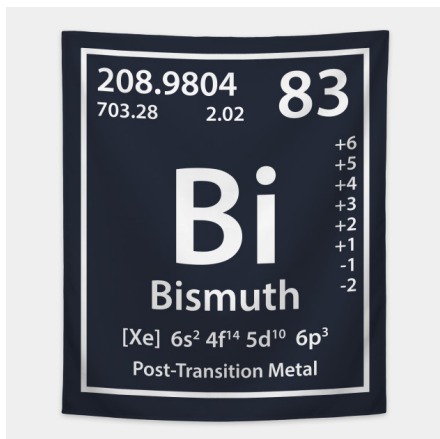
■ Synthesis of pentavalent organobismuthines



Michaelis, A. *Ber. Dtsch. Chem. Ges.* **1887**, 20, 52.
Barton, D. H. R.; Finet, J.-P. *Pure Appl. Chem.* **1987**, 59, 937.

Main-group Elements (Bi) in Catalysis for Organic Synthesis

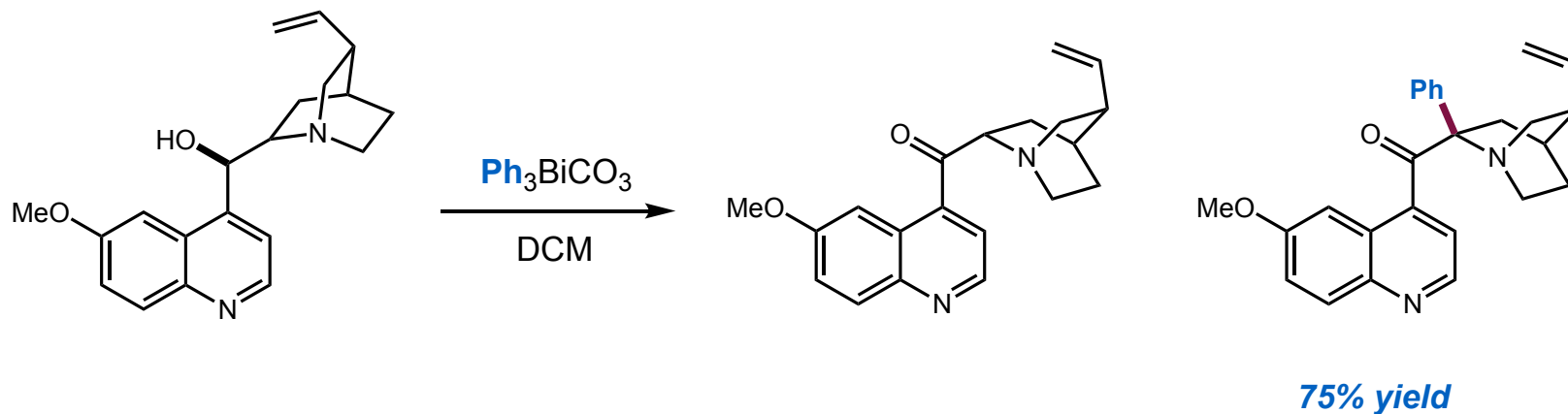
Outline



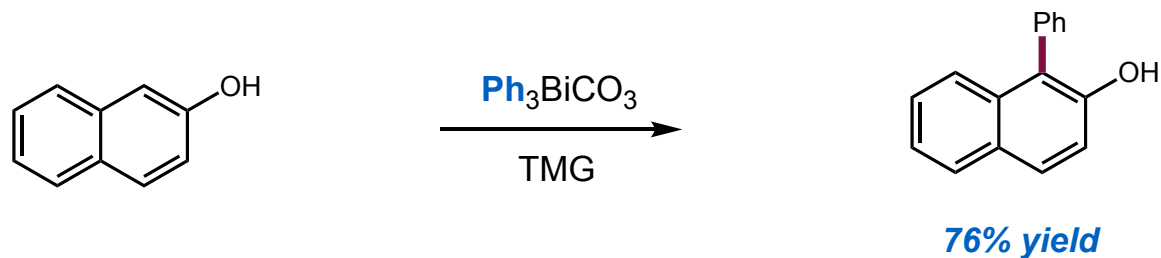
- The facts of bismuth
- Synthesis of organobismuthines
- The application of organobismuthines in organic synthesis
- Redox chemistry of bismuth

Organobismuth as arylation reagents

■ The first arylation reaction using organobismuthines

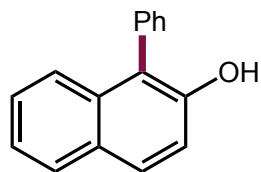
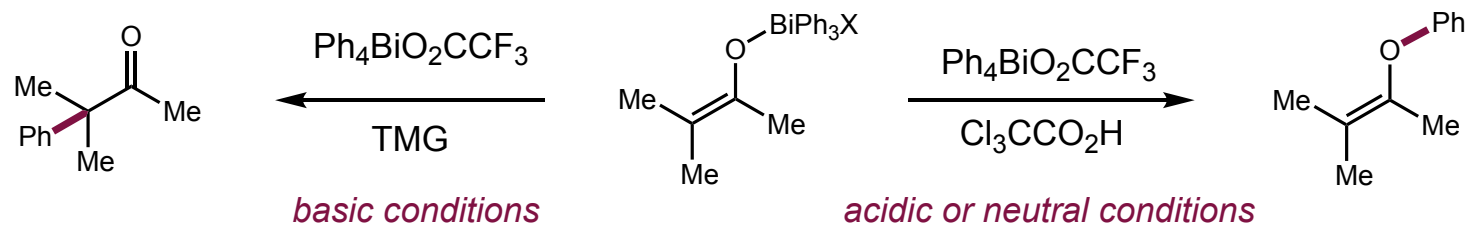


■ The first C-arylation of phenols using organobismuthines

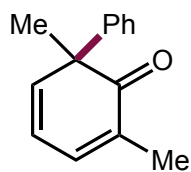


Organobismuth as arylation reagents

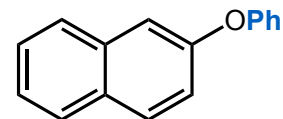
■ O-arylation and C-arylation controlled by condition



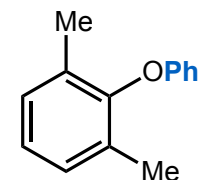
90% yield



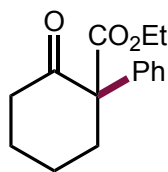
72% yield



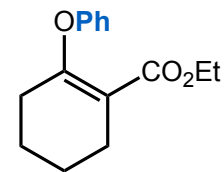
90% yield



58% yield



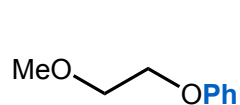
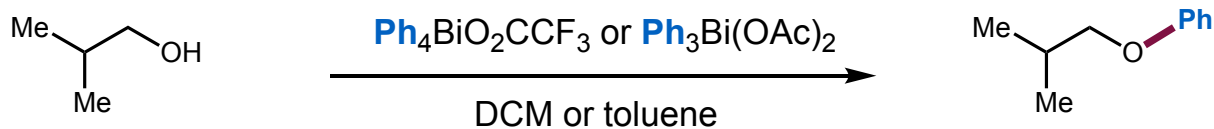
91% yield



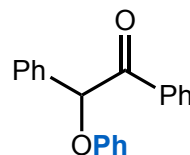
57% yield

Organobismuth as arylation reagents

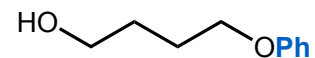
■ O-arylation of alcohols, diols, aminoalcohols



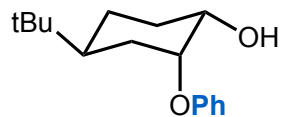
86% yield



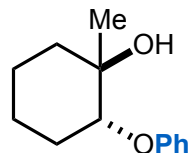
88% yield



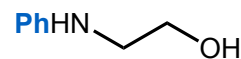
80% yield



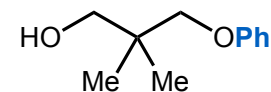
73% yield



88% yield



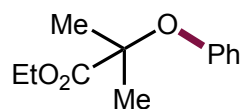
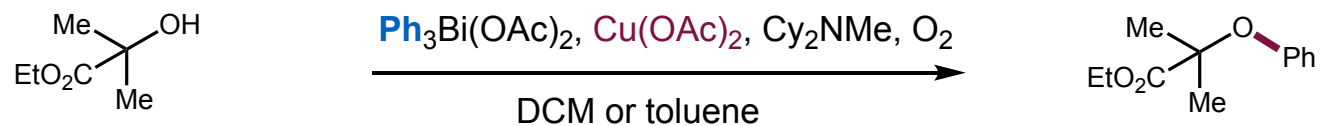
51% yield



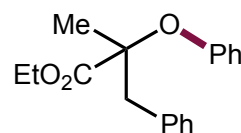
99% yield

Organobismuth as arylation reagents

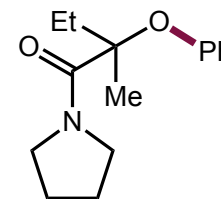
■ O-arylation of tertiary alcohols via Cu catalysis



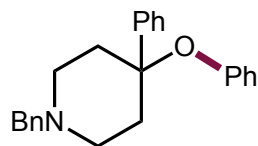
89% yield



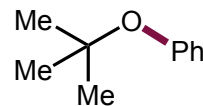
89% yield



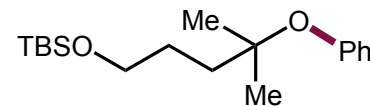
94% yield



88% yield



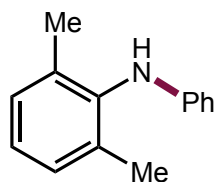
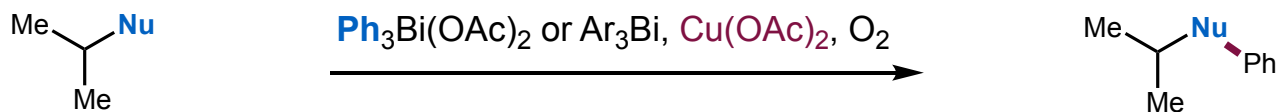
87% yield



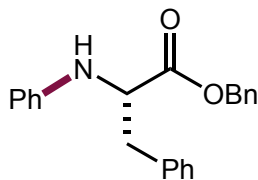
50% yield

Organobismuth as arylation reagents

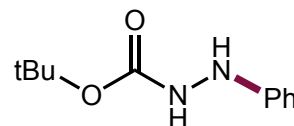
■ Cu catalyzed arylation of *N*, *S*-nucleophiles



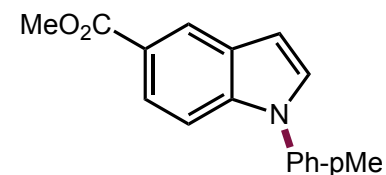
96% yield



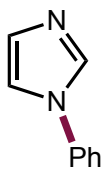
80% yield



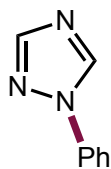
91% yield



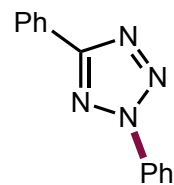
87% yield



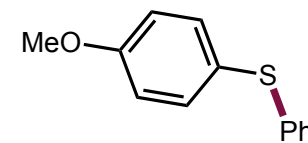
87% yield



60% yield



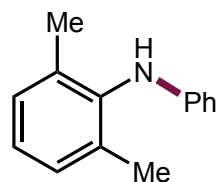
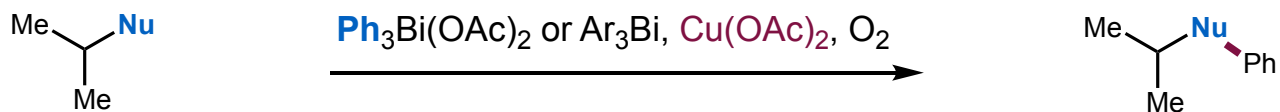
87% yield



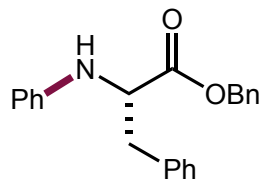
84% yield

Organobismuth as arylation reagents

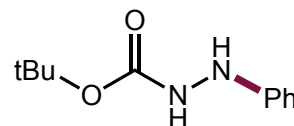
■ Cu catalyzed arylation of *N*, *S*-nucleophiles



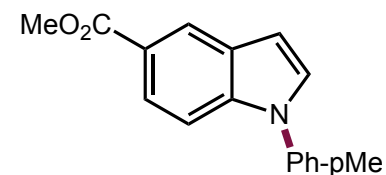
96% yield



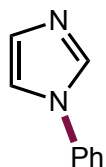
80% yield



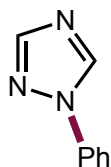
91% yield



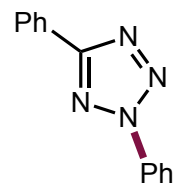
87% yield



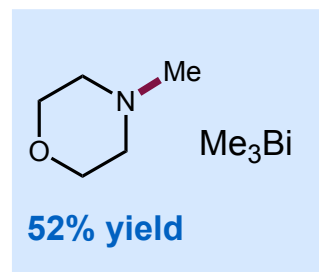
87% yield



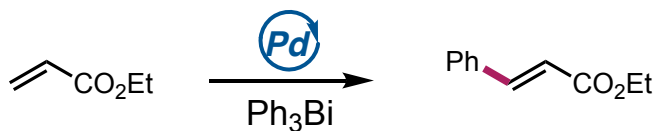
60% yield



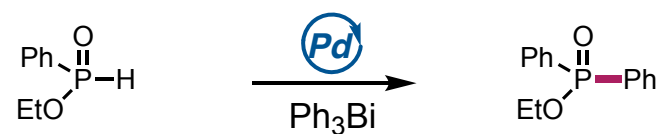
87% yield



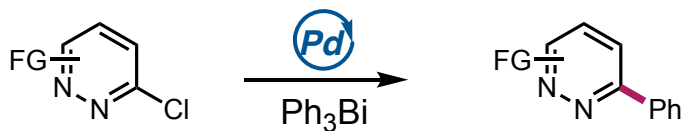
Cross-coupling using organobismuth



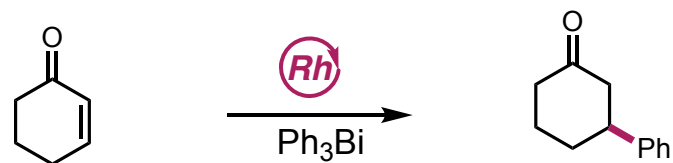
Kawamura, T.; Kikukawa, K.; Takagi, M.; Matsuda, T. *Bull. Chem. Soc. Jpn.* **1977**, *50*, 2021.



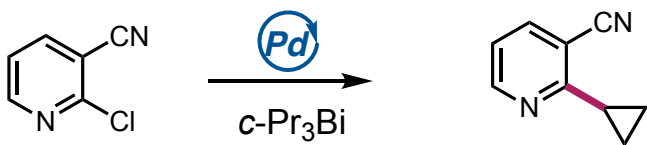
Wang, T.; Sang, S.; Liu, L.; Qiao, H.; Gao, Y.; Zhao, Y. *J. Org. Chem.* **2014**, *79*, 608.



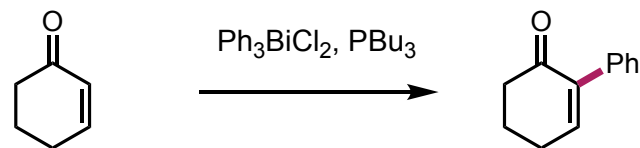
Gagnon, A. *J. Org. Chem.* **2016**, *81*, 5401.



Li, C.-J. et. al. *J. Am. Chem. Soc.* **2001**, *123*, 7451.



Gagnon, A.; Duplessis, M.; Alsabeh, P.; Barabé, F. *J. Org. Chem.* **2008**, *73*, 3604.

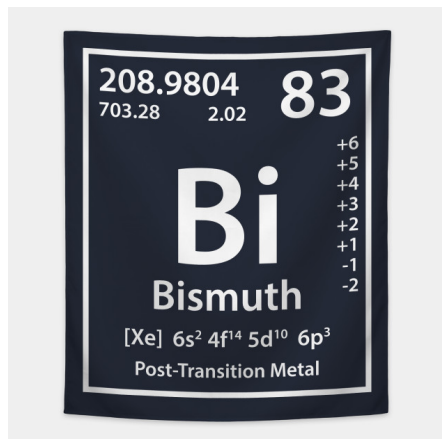


Krische, M. J. *J. Am. Chem. Soc.* **2004**, *126*, 5350.

Bi redox chemistry is unclear with rare examples.

Main-group Elements (Bi) in Catalysis for Organic Synthesis

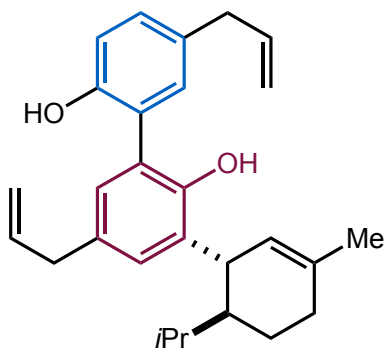
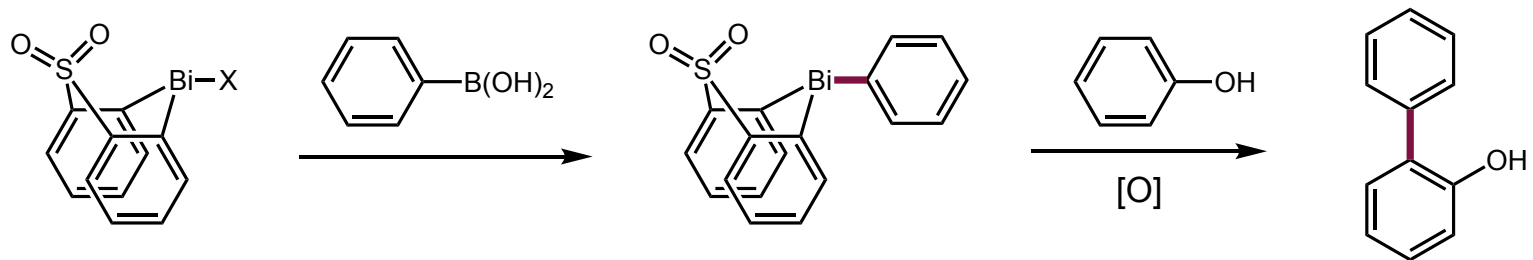
Outline



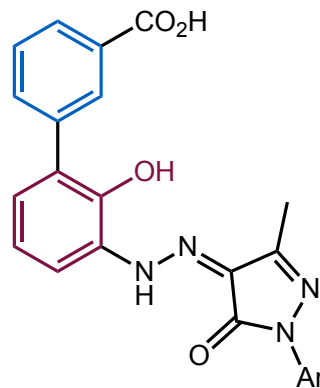
- The facts of bismuth
- Synthesis of organobismuthines
- The application of organobismuthines in organic synthesis
- Redox chemistry of bismuth

Bi(III)/Bi(V) redox activation

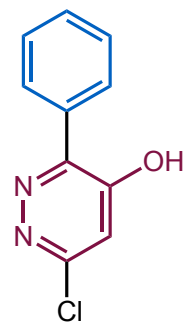
ortho-Arylation of phenols via Bi(III)/Bi(V) redox activation



Liganas
antibacterial, cytotoxic



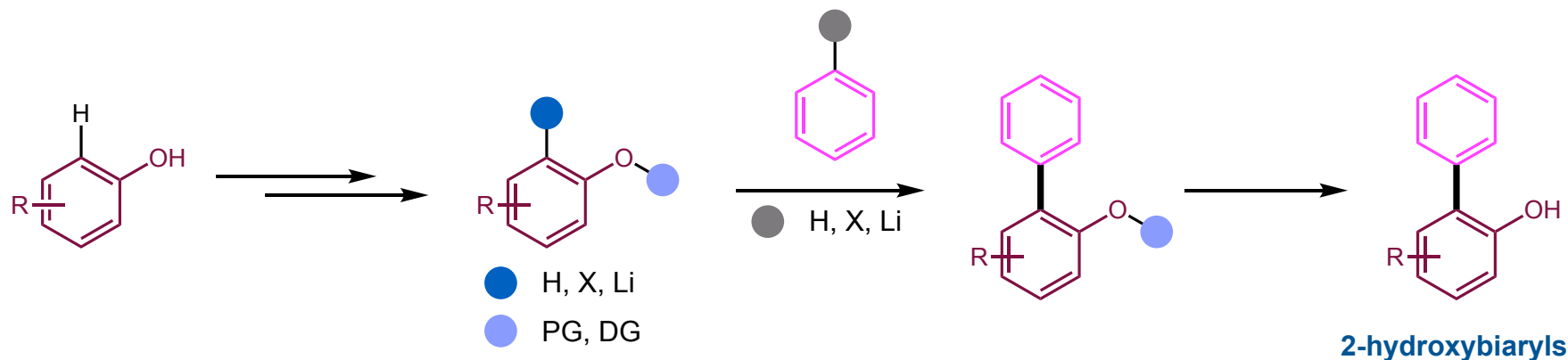
Pharmaceuticals
aplastic anemia



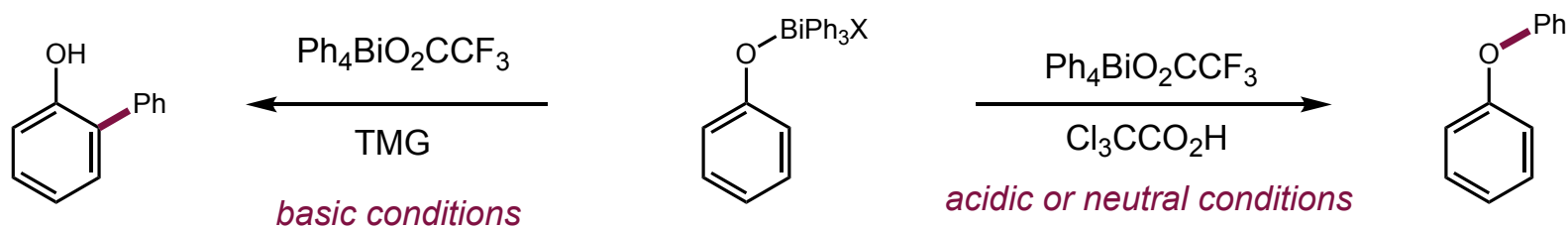
Agrochemicals
herbicide

Bi(III)/Bi(V) redox activation

■ Current approaches for the preparation of 2-hydroxybiaryls



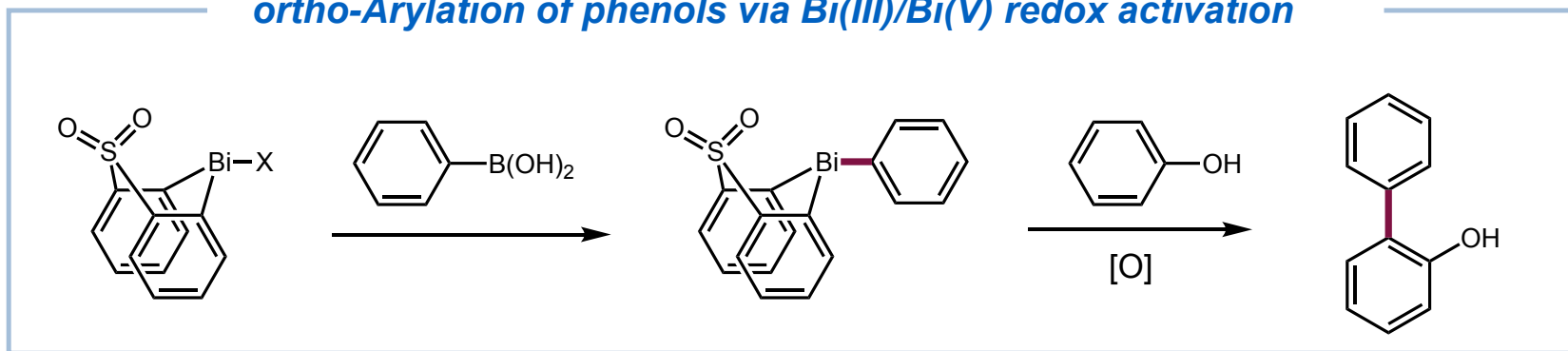
■ Barton's pioneer discovery



- O or C arylation is highly dependent on the nature of the organobismuth and reaction conditions
- multistep synthesis of arylbismuth reagents, transfer one of the aryl groups in the organobismuth
- lack of systematic studies of mechanism which impedes practical exploration of the method

Bi(III)/Bi(V) redox activation

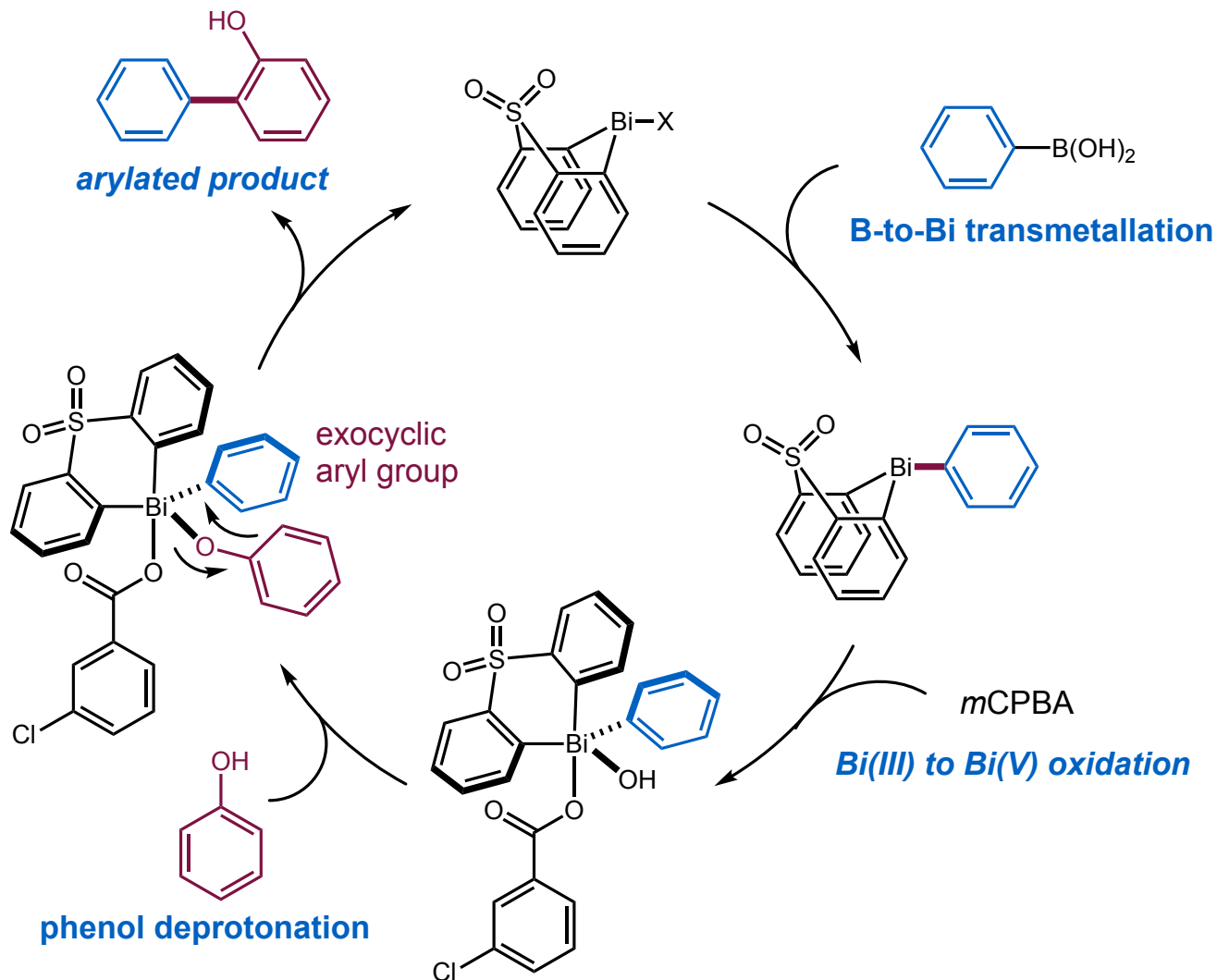
ortho-Arylation of phenols via Bi(III)/Bi(V) redox activation



- *one-pot boron-to-bismuth transmetallation*
- *commercially available starting materials*
- *completely prevents the formation of O-arylated products*
- *operates under air in non-anhydrous conditions*

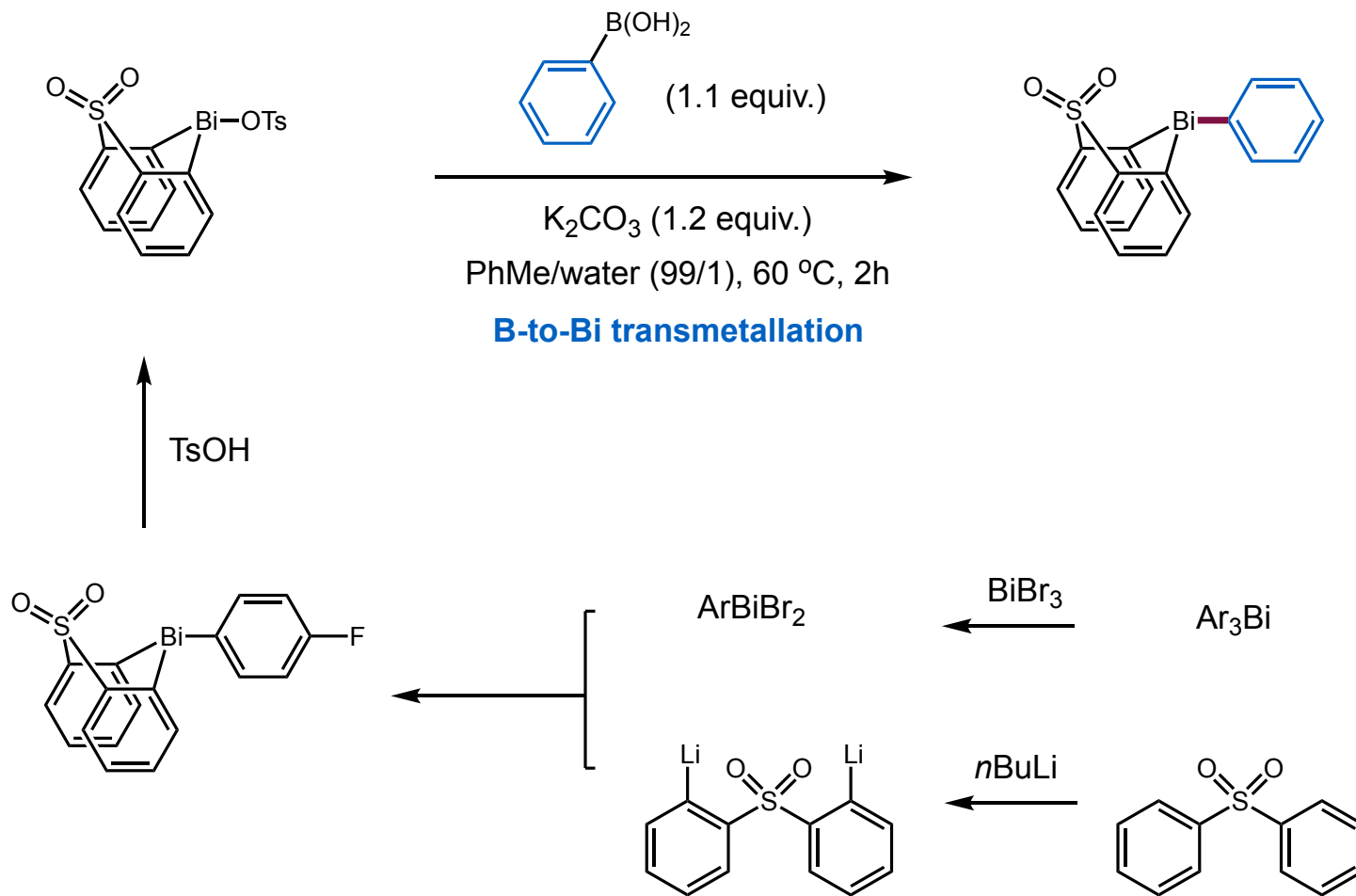
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ Reaction design



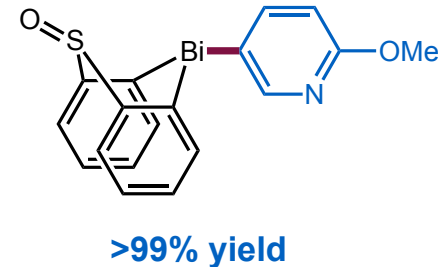
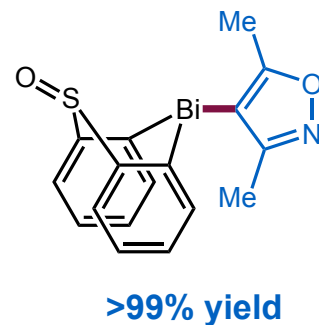
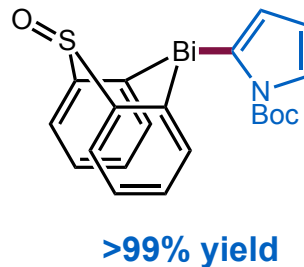
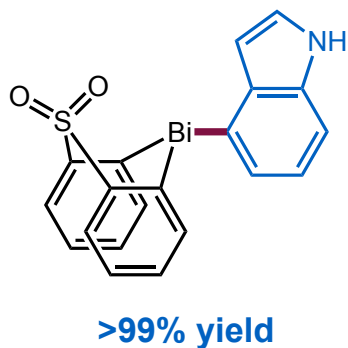
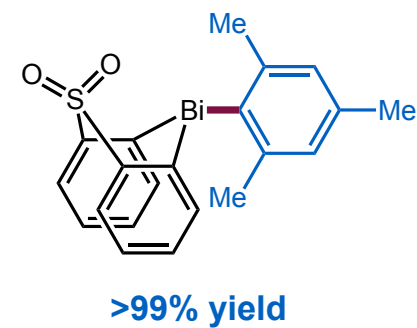
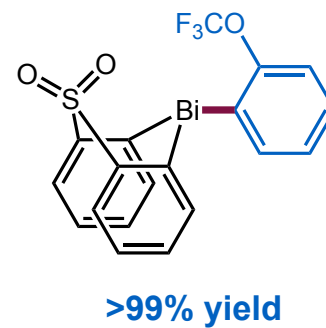
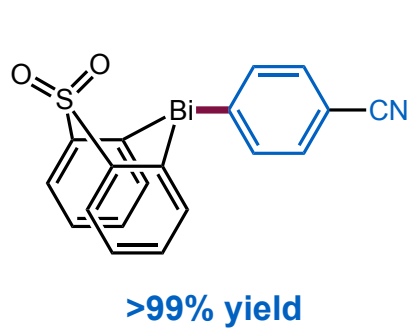
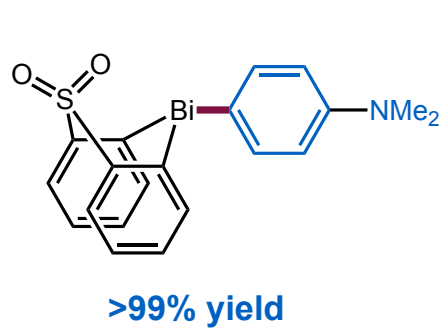
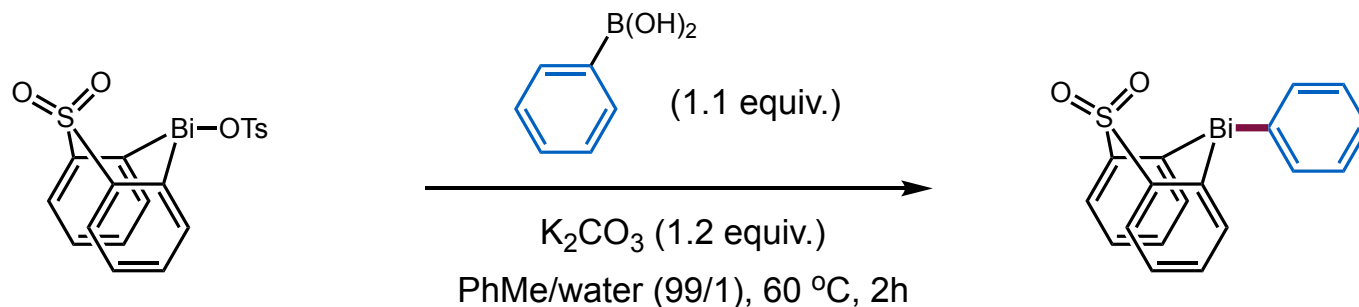
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ Transmetalation to universal bismacyclic precursor



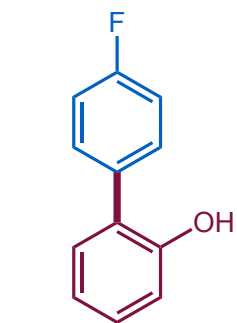
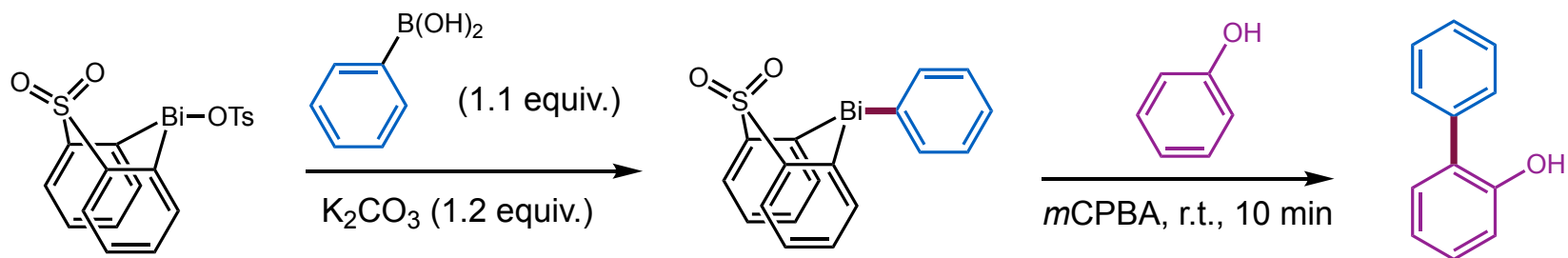
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ Transmetalation to universal bismacyclic precursor

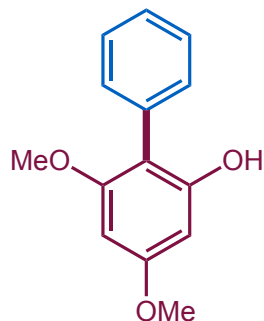


Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

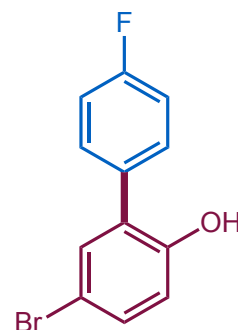
■ One-pot, Bi(V)-mediated arylation of phenols and naphthols



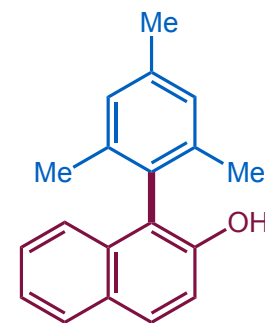
69% yield



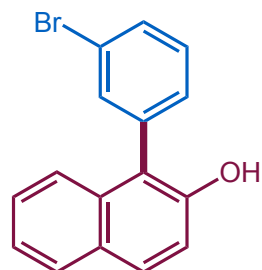
77% yield



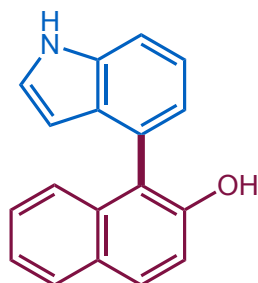
59% yield



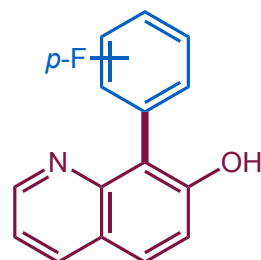
89% yield



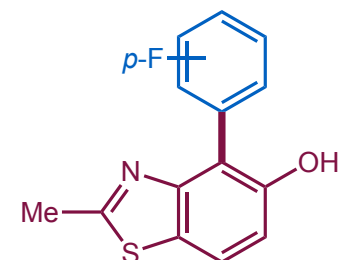
87% yield



59% yield



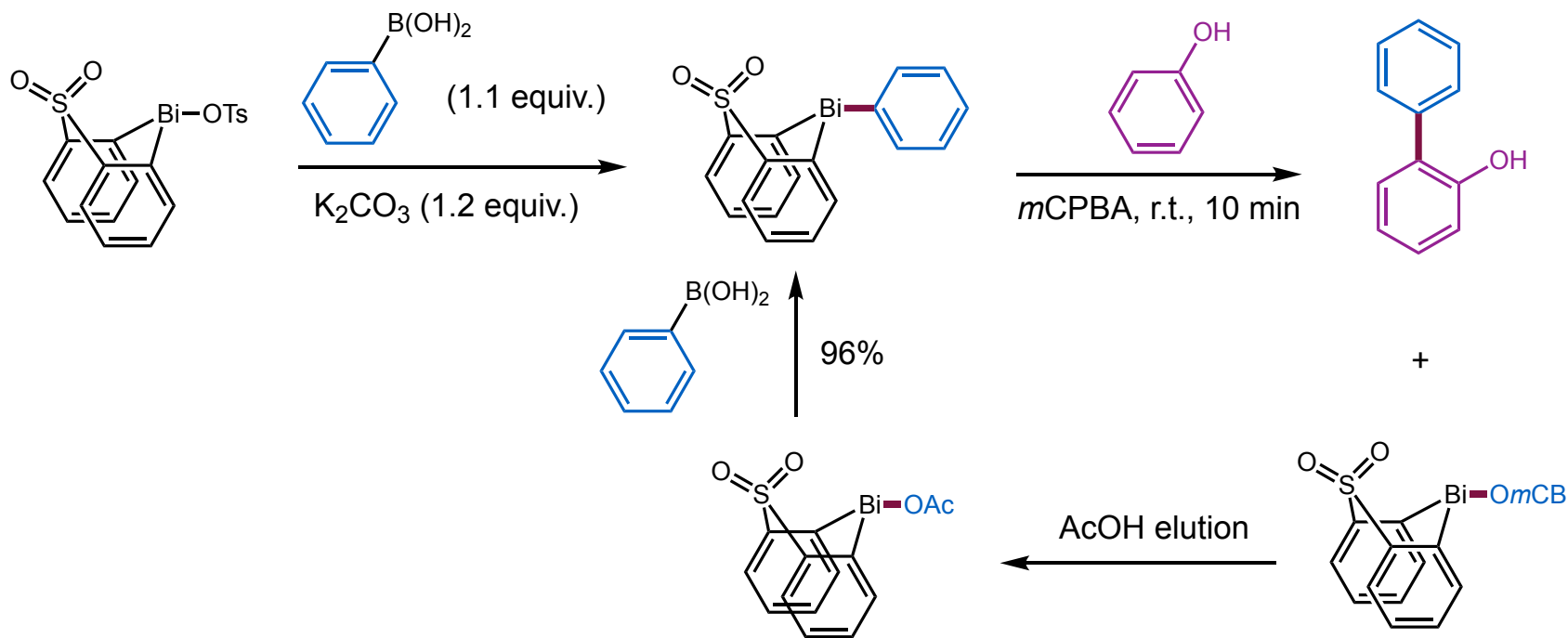
93% yield



90% yield

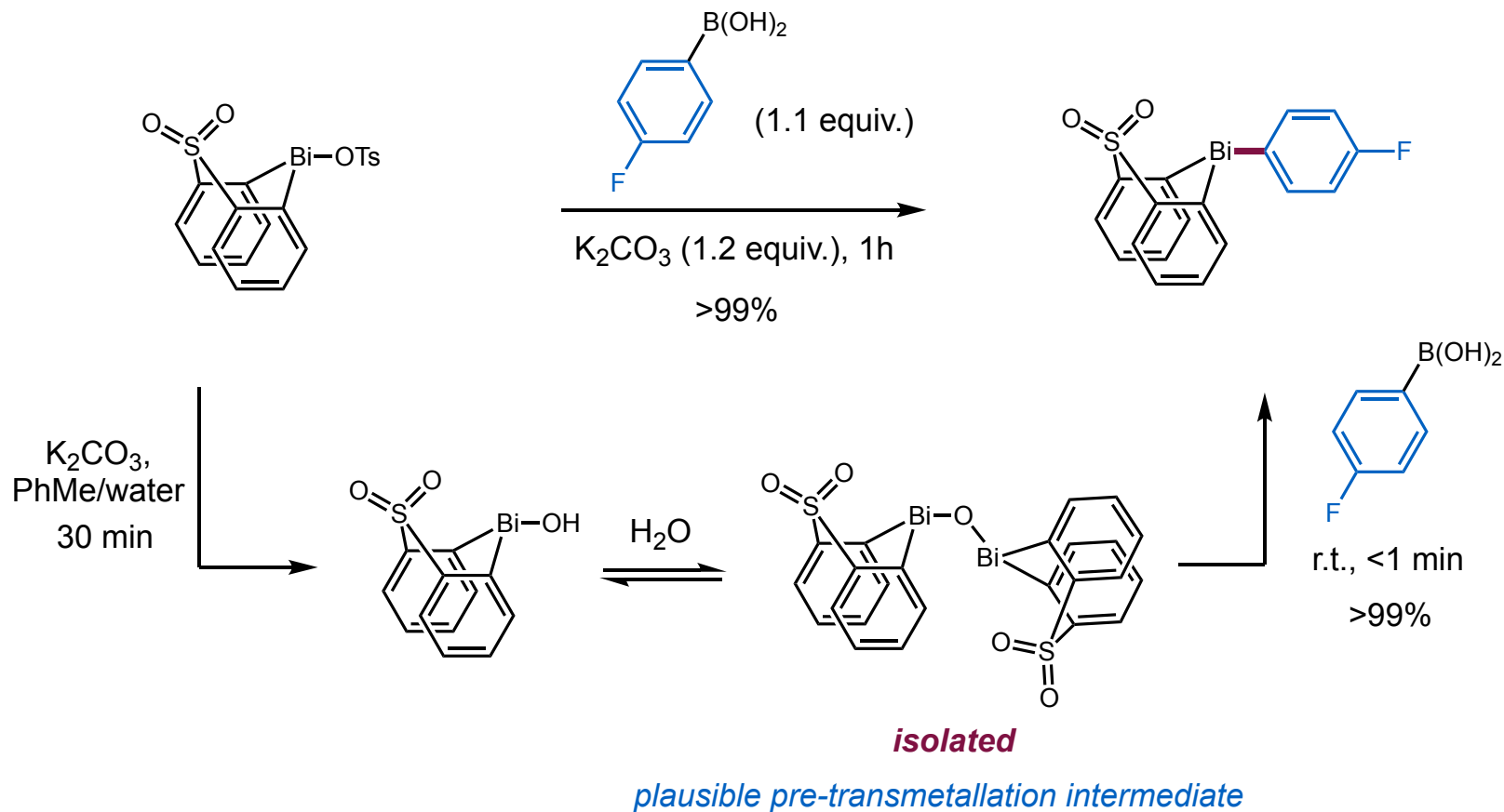
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ One-pot, Bi(V)-mediated arylation of phenols and naphthols



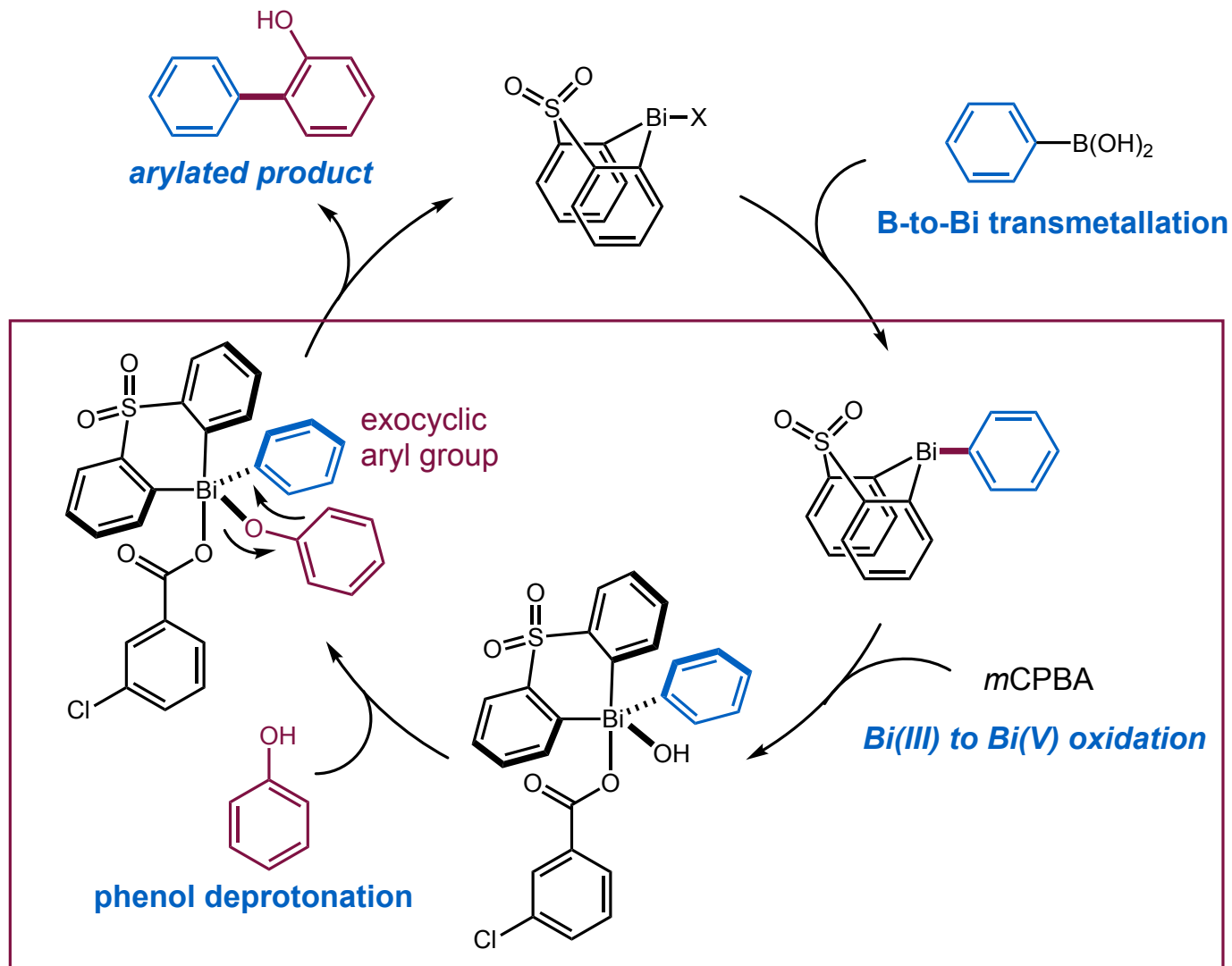
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

Mechanistic studies for the reaction



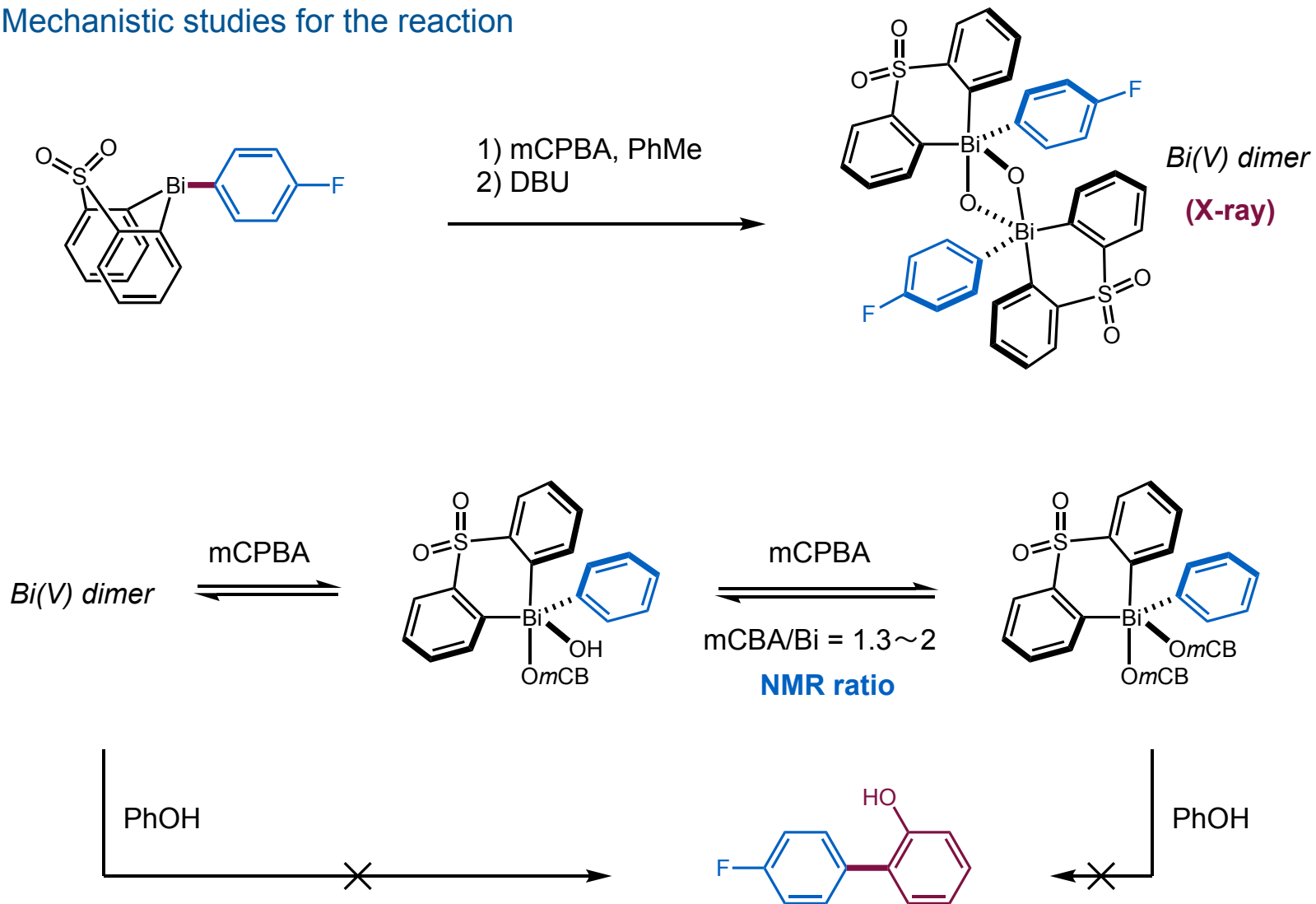
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ Reaction design



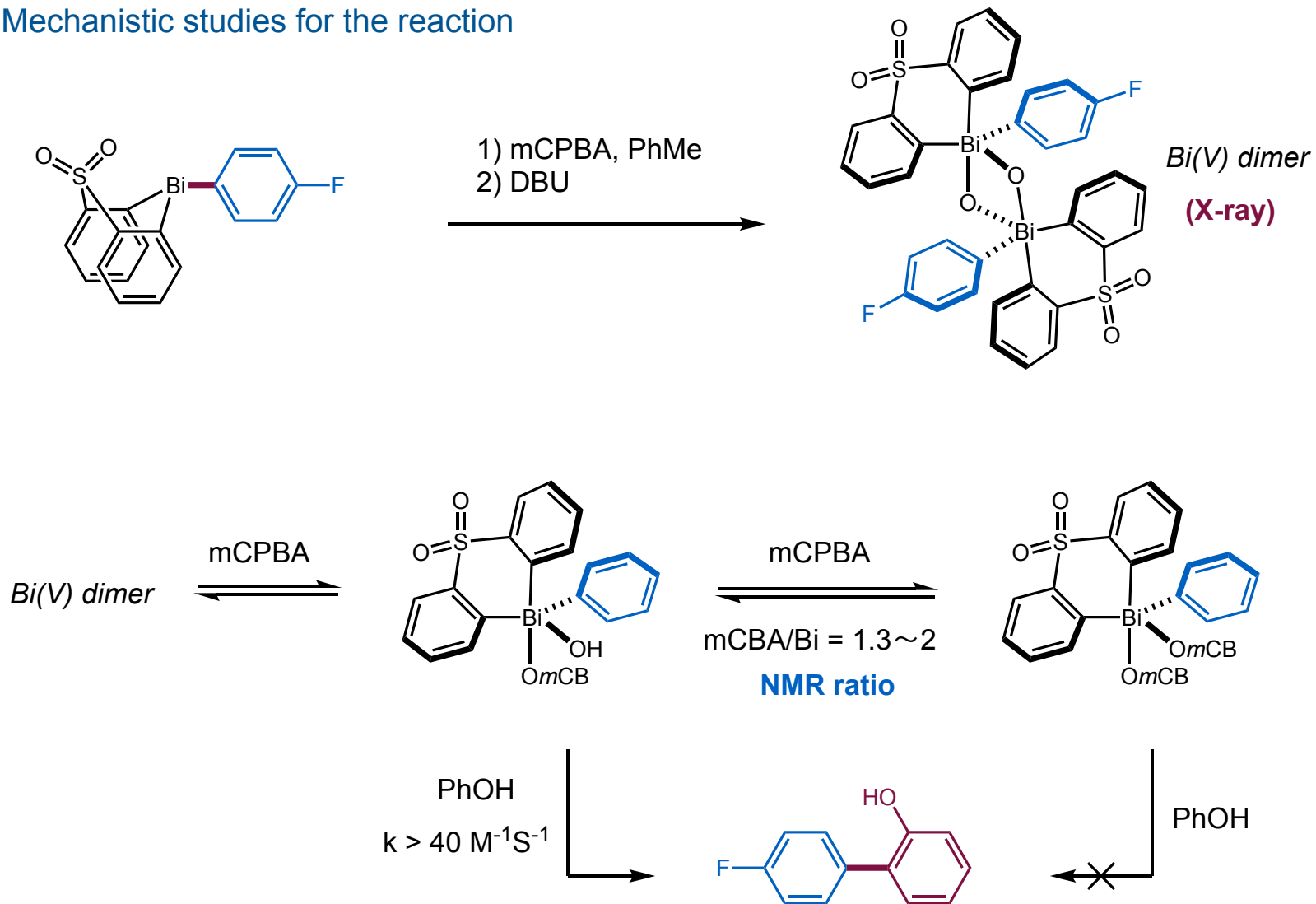
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ Mechanistic studies for the reaction



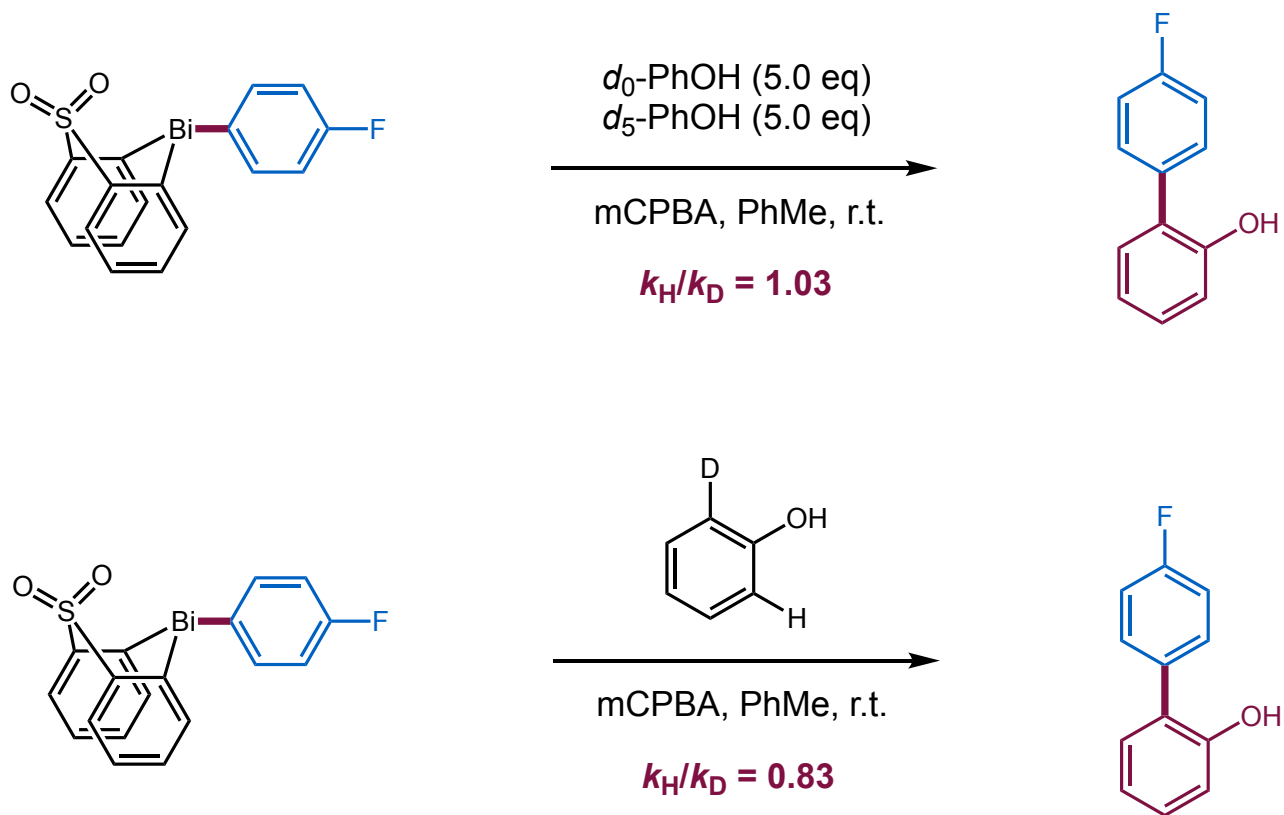
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ Mechanistic studies for the reaction



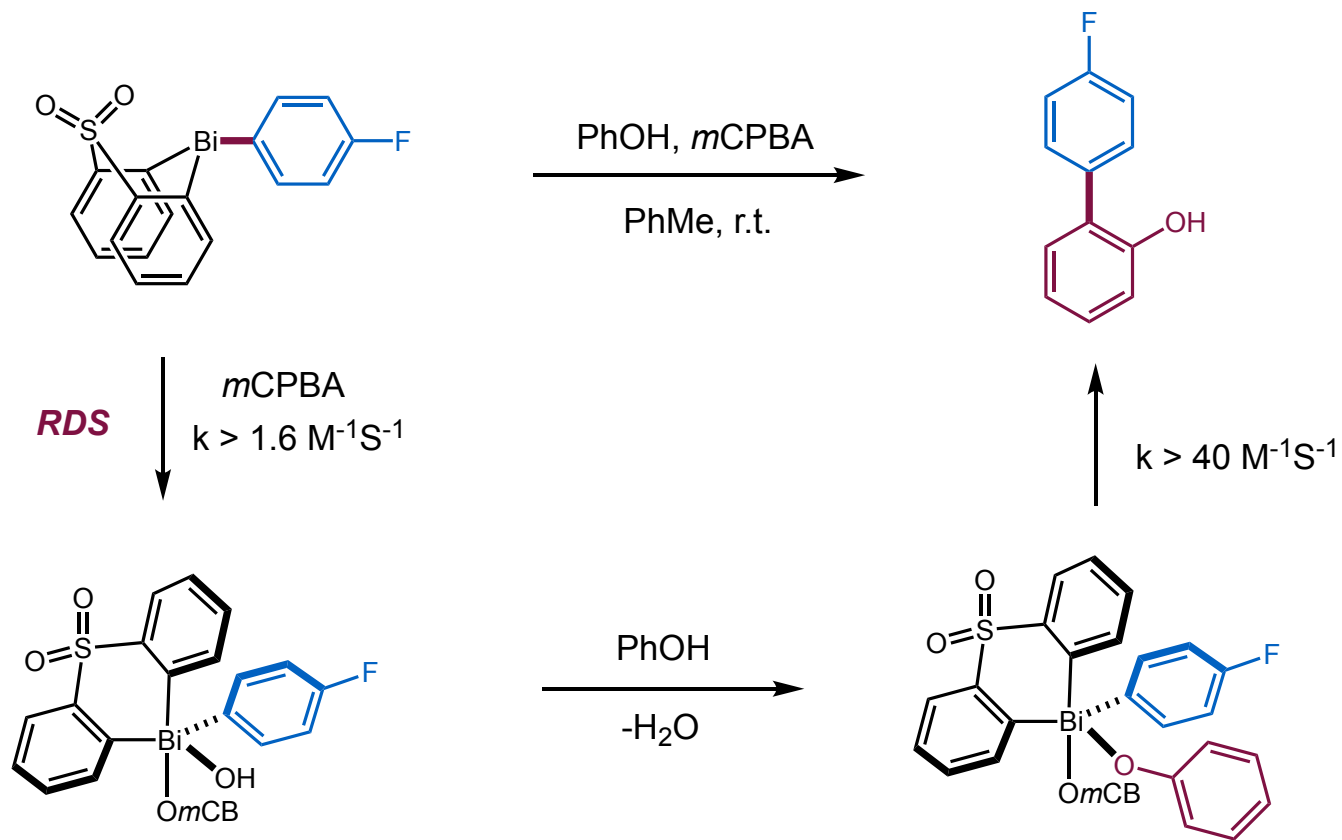
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ Mechanistic studies for the reaction



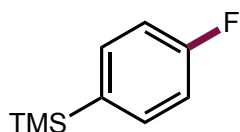
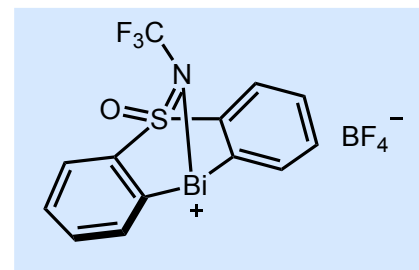
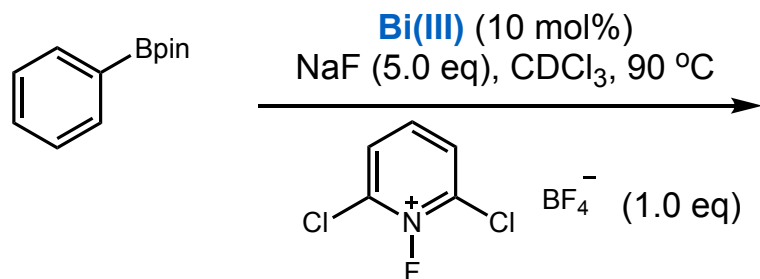
Bi(III)/Bi(V) redox activation for ortho-Arylation of phenols

■ Mechanistic studies for the reaction

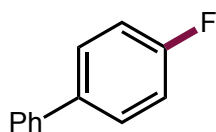


What about Bi(III)/Bi(V) redox catalysis?

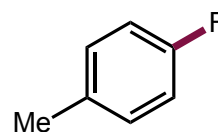
■ Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis



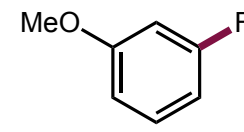
90% yield



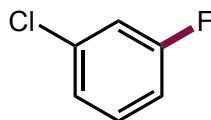
71% yield



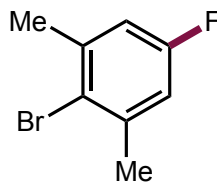
77% yield



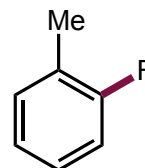
55% yield



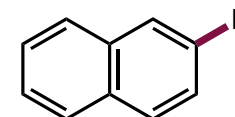
36% yield



84% yield



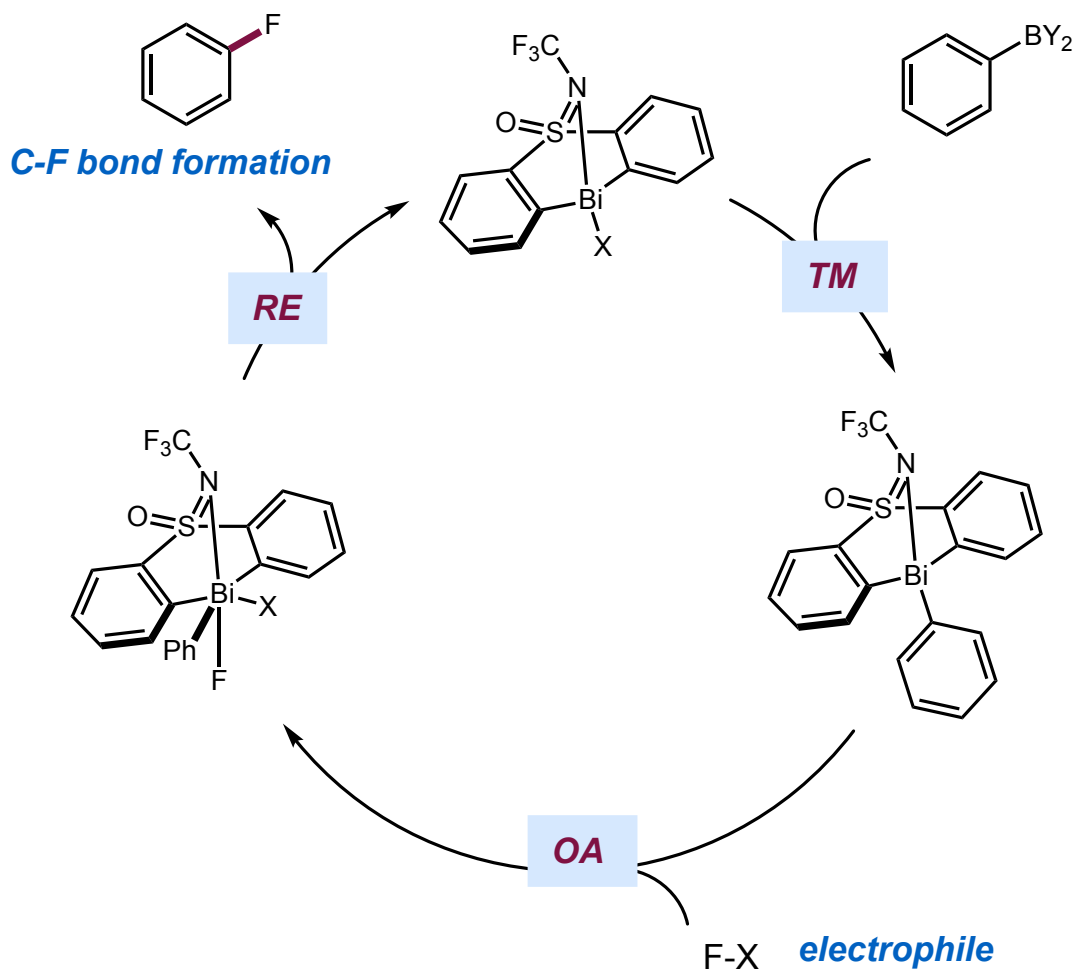
45% yield



49% yield

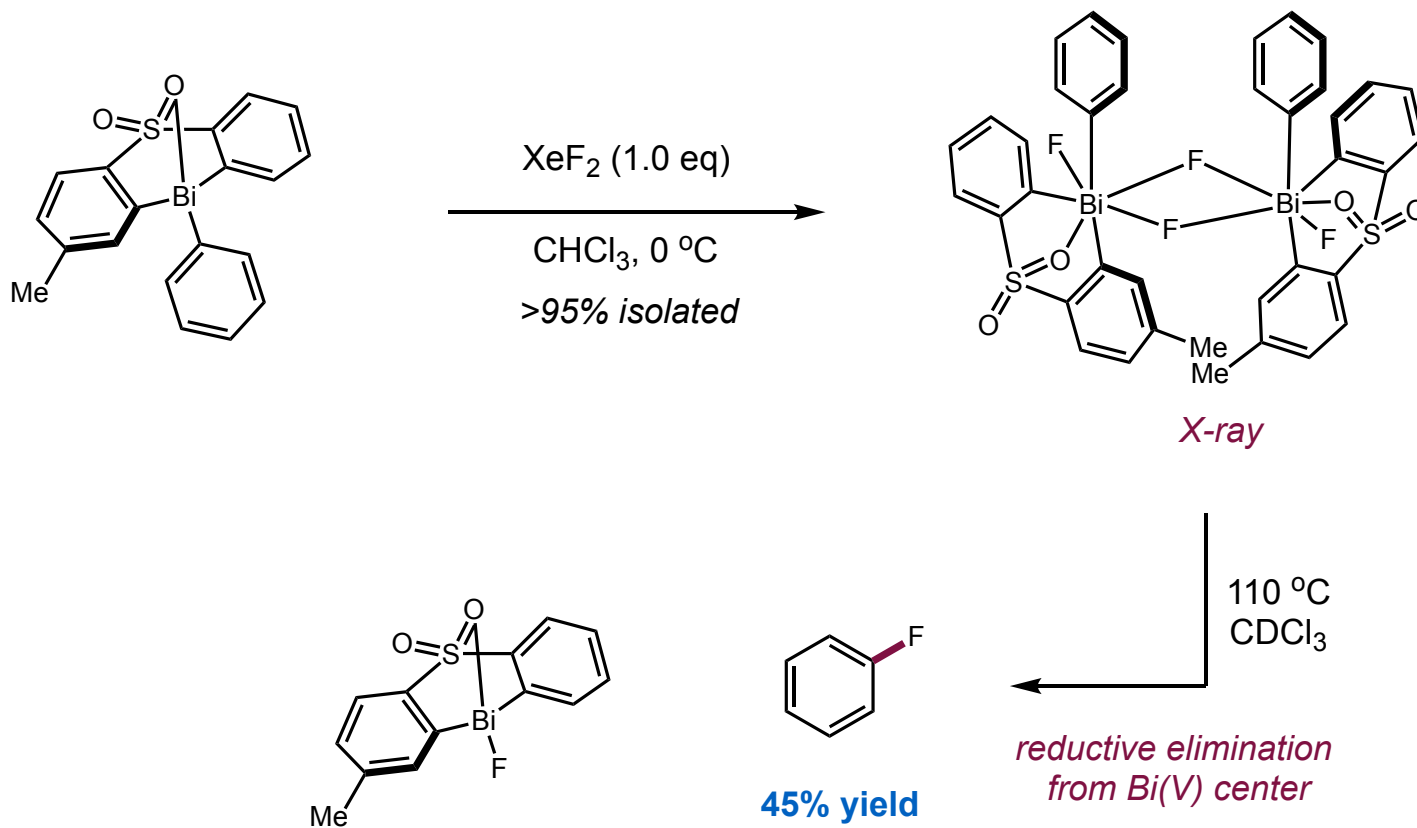
Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis

■ Proposed mechanism for Bi(III)/Bi(V) redox catalysis



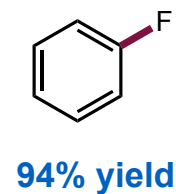
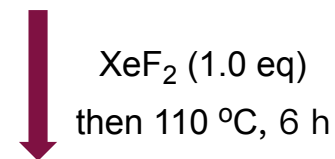
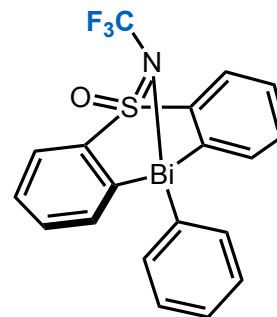
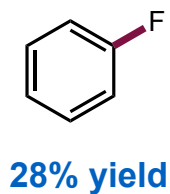
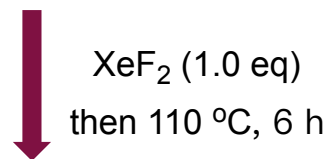
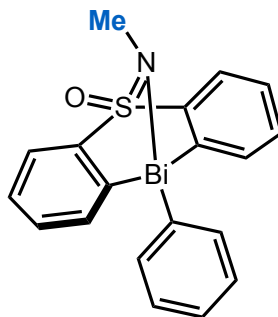
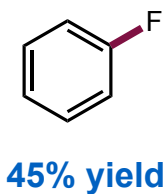
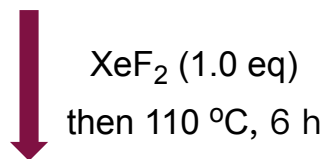
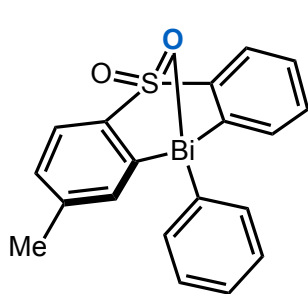
Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis

■ Proof of concept study



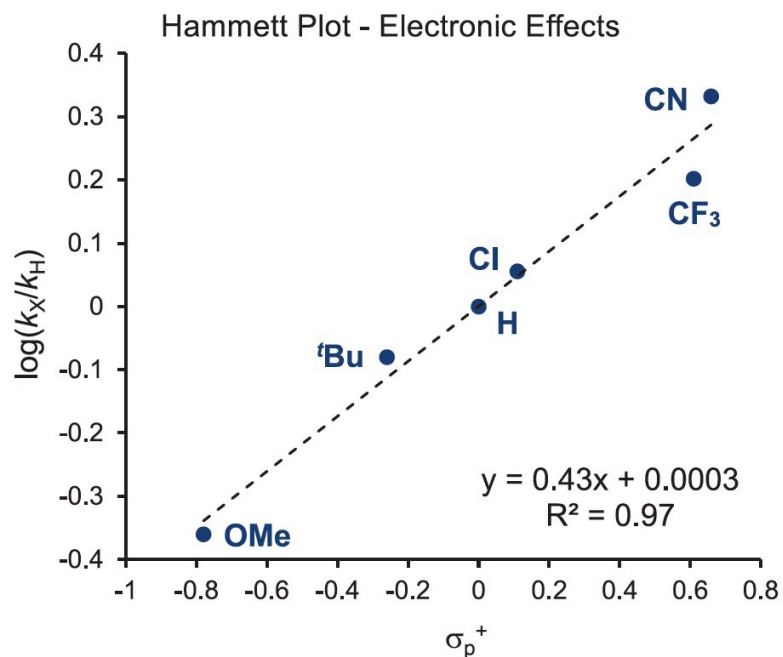
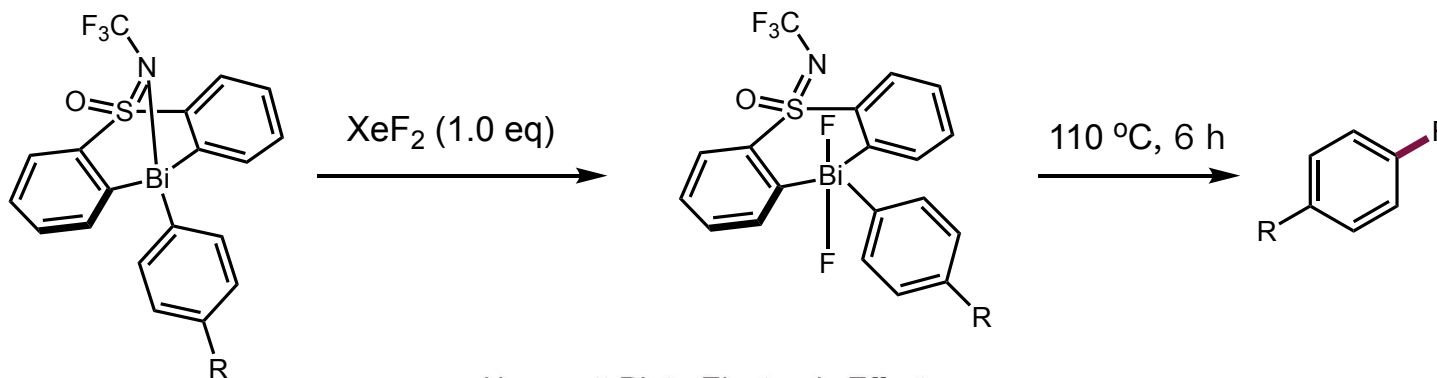
Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis

■ Ligand effect for the reductive elimination



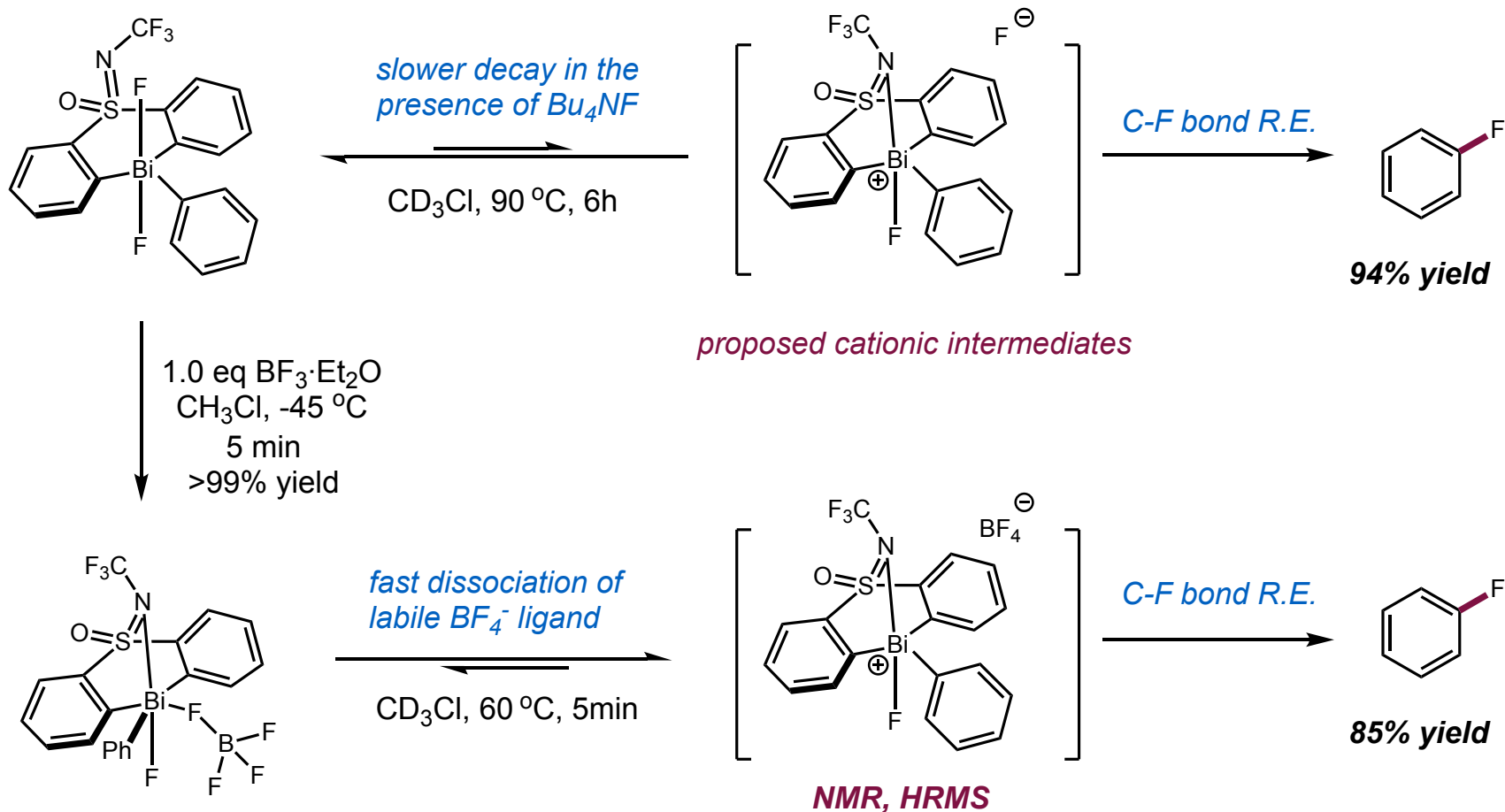
Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis

Kinetic analysis of the reductive elimination



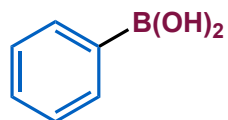
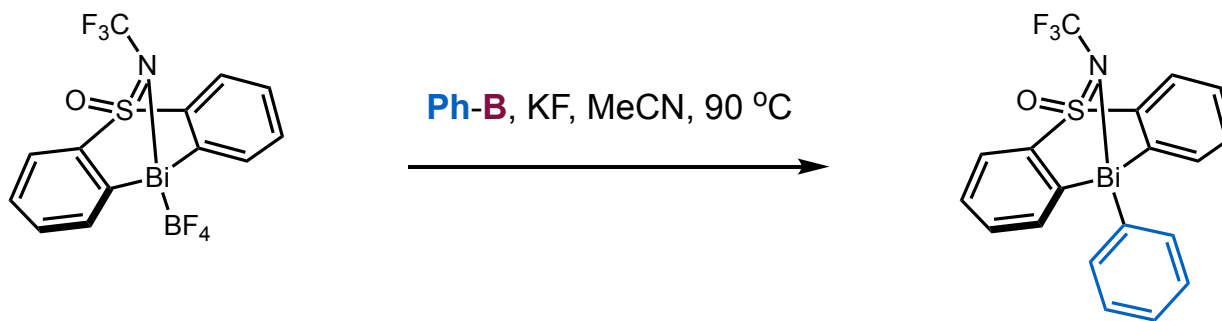
Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis

Mechanism study of the reductive elimination

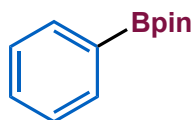


Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis

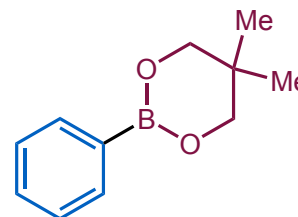
■ Transmetalation of arylborone to Bi(III) complex



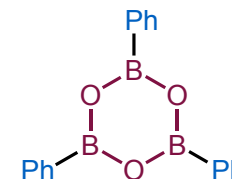
95% yield



67% yield



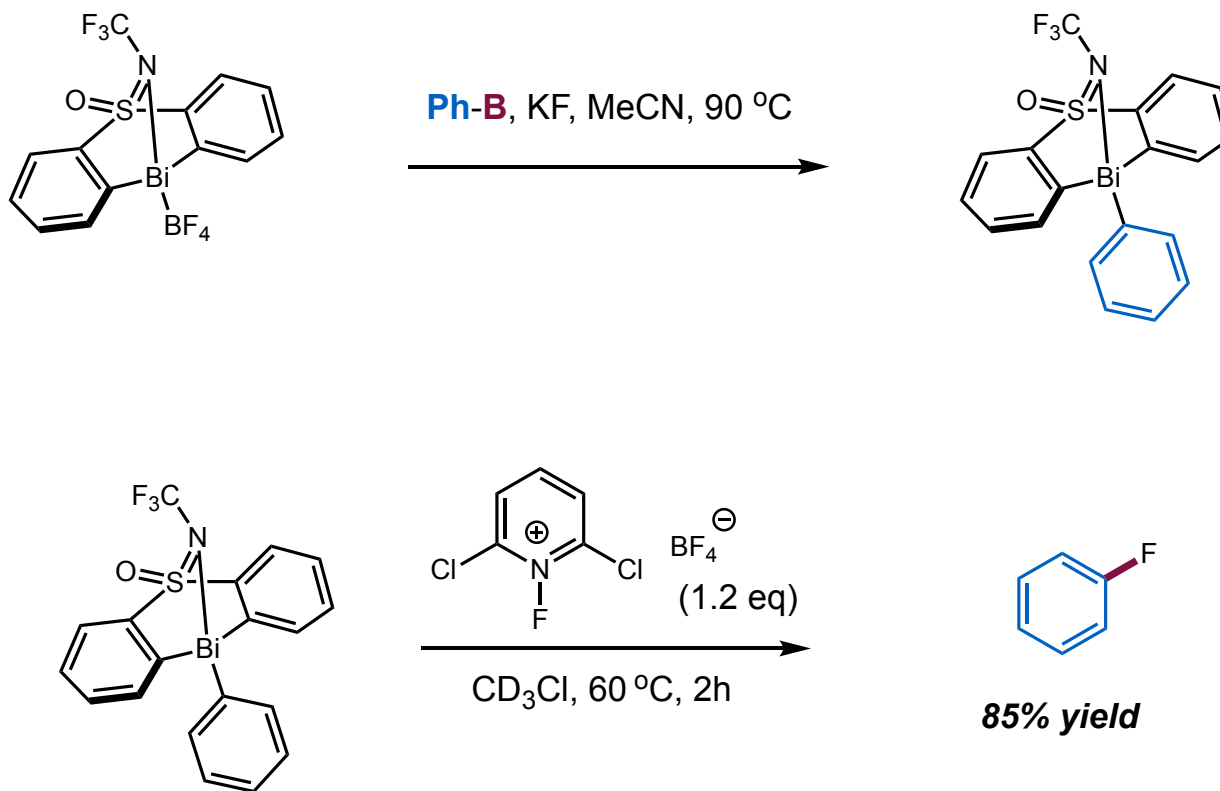
72% yield



80% yield

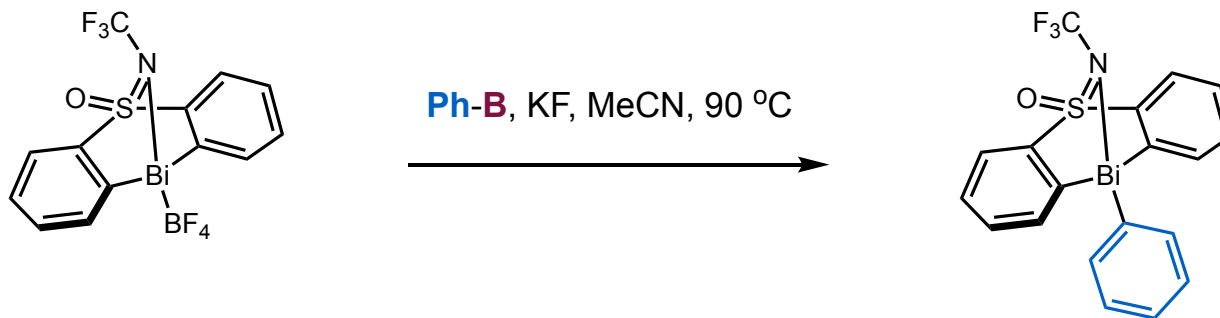
Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis

■ Merging element steps together for Bi(III)/Bi(V) redox catalysis

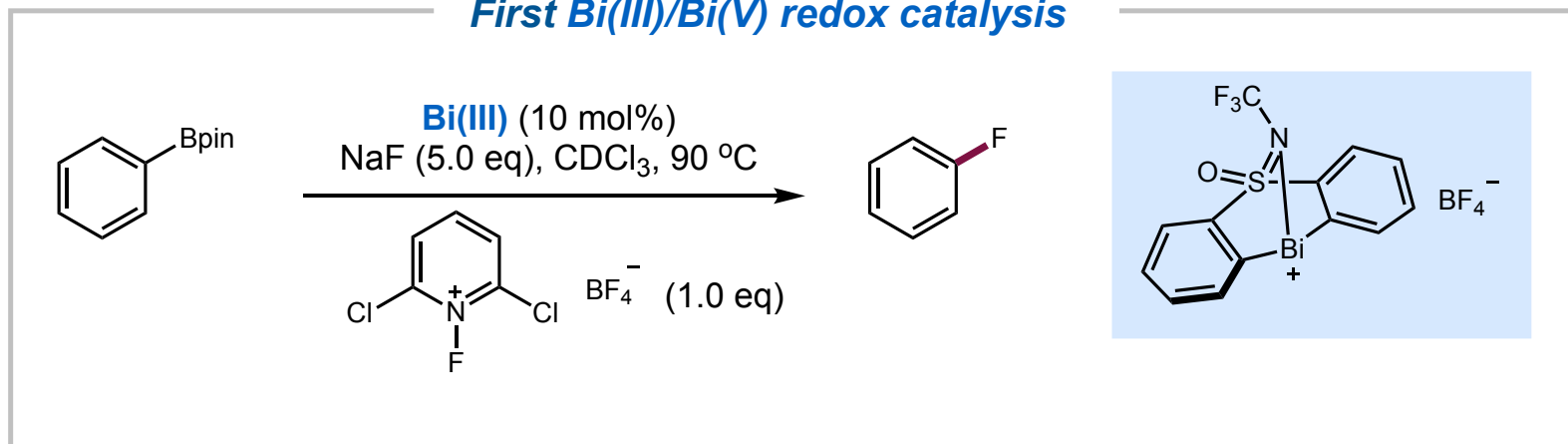


Fluorination of arylboron via Bi(III)/Bi(V) redox catalysis

■ Merging element steps together for Bi(III)/Bi(V) redox catalysis

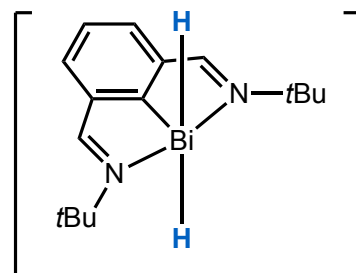
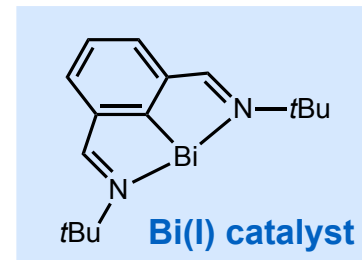
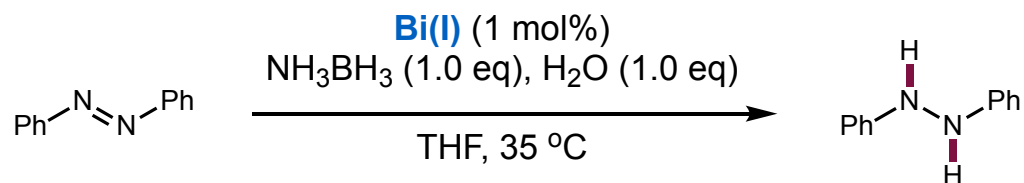


First Bi(III)/Bi(V) redox catalysis



What about Bi(I)/Bi(III) redox catalysis?

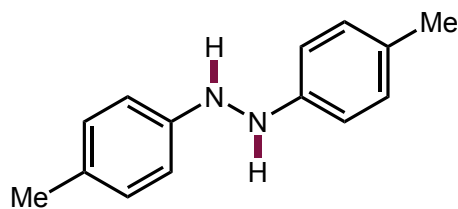
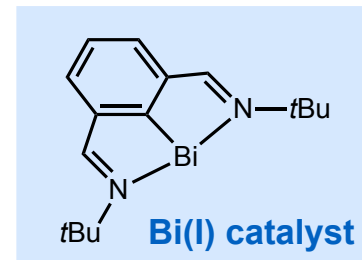
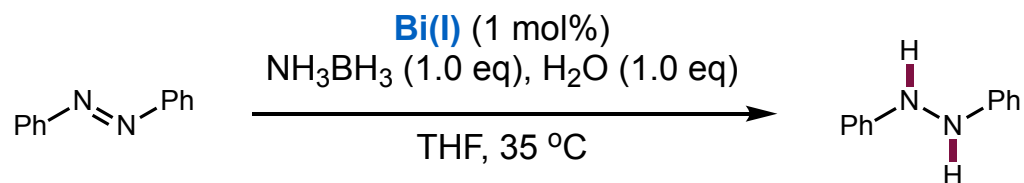
■ Bi(I) catalyzed transfer hydrogenation



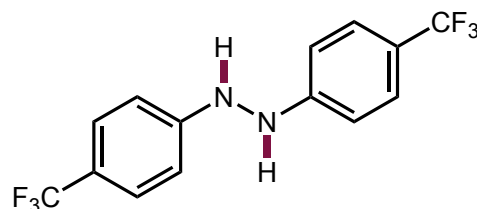
via Bi(III) hydride

What about Bi(I)/Bi(III) redox catalysis?

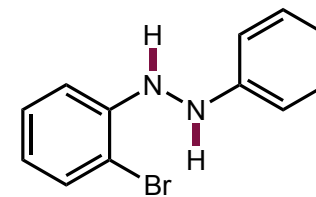
■ Bi(I) catalyzed transfer hydrogenation



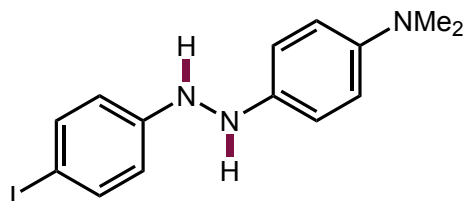
97% yield



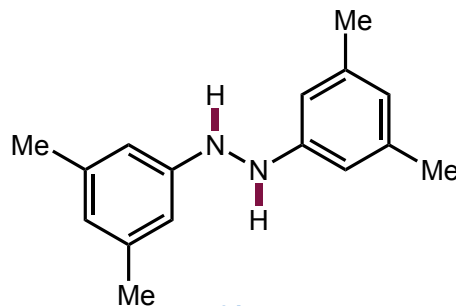
99% yield



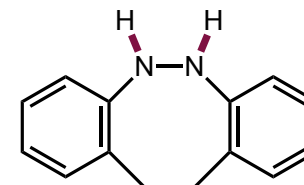
98% yield



99% yield



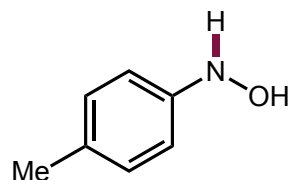
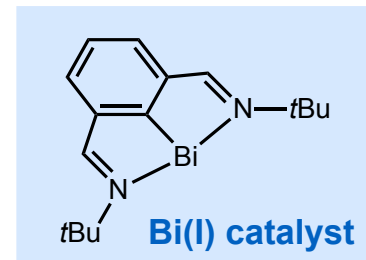
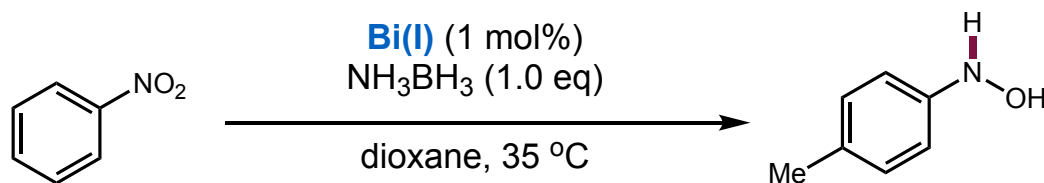
98% yield



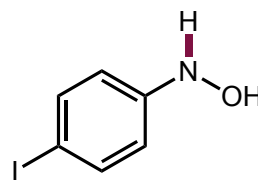
99% yield

Transfer hydrogenation via Bi(I)/Bi(III) redox catalysis

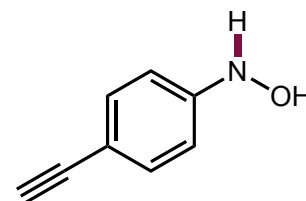
■ Bi(I) catalyzed transfer hydrogenation of nitroarenes



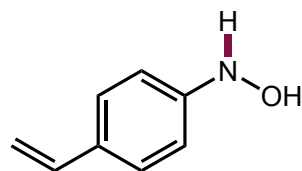
97% yield



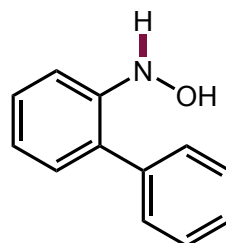
88% yield



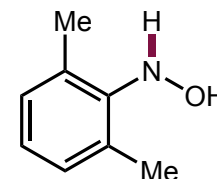
67% yield



82% yield



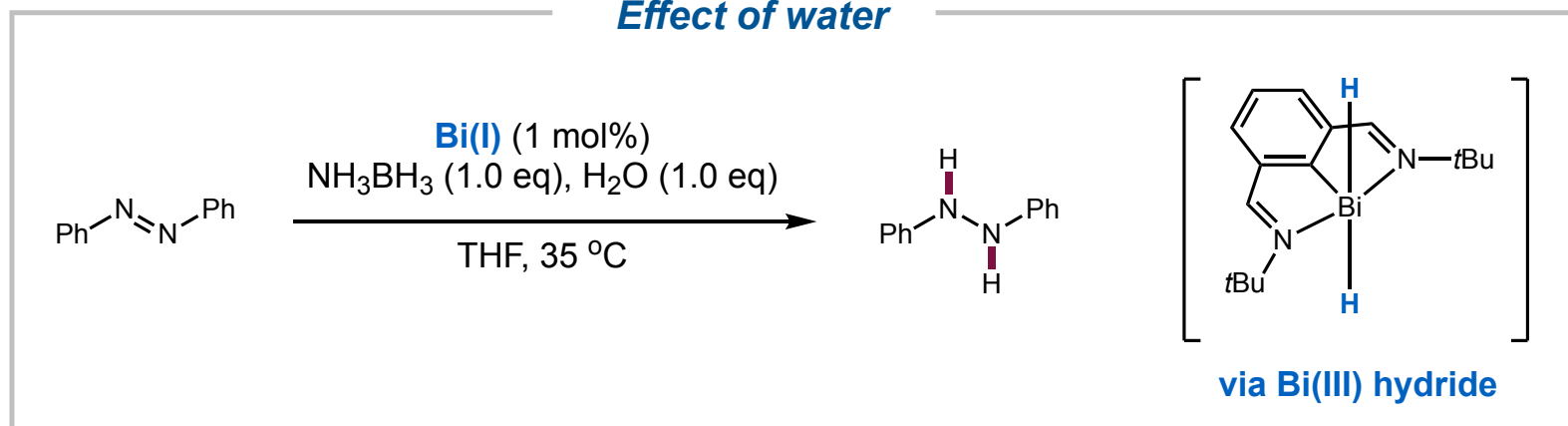
75% yield



22% yield

Transfer hydrogenation via Bi(I)/Bi(III) redox catalysis

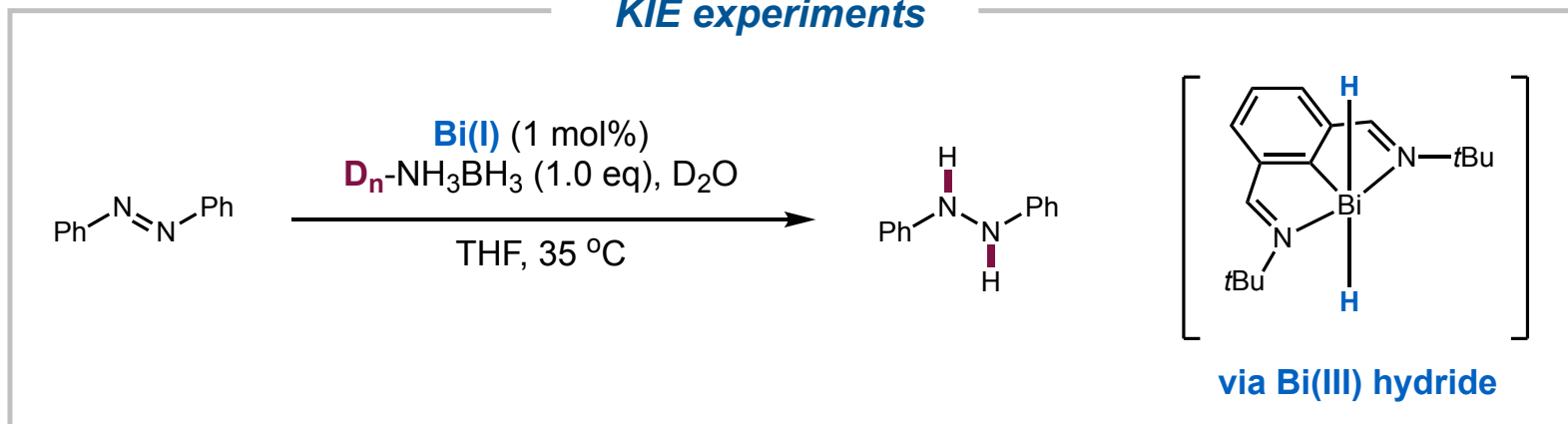
Effect of water



	NH ₃ BH ₃	NH ₂ MeBH ₃	NHMe ₂ BH ₃	NMe ₃ BH ₃	NH ₃ BEt ₃
without H ₂ O	57% (16h)	55% (16h)	36% (16h)	10% (16h)	- (16h)
with 1.0 equiv. H ₂ O	99% (2h)	97% (7h)	90% (16h)	31% (16h)	- (16h)

Transfer hydrogenation via Bi(I)/Bi(III) redox catalysis

KIE experiments



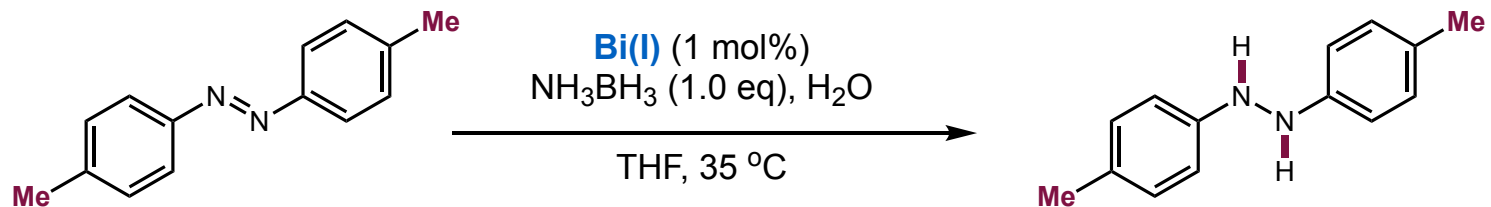
Kinetic Isotope Effect

ND_3BH_3	NH_3BD_3	ND_3BD_3
1.63	3.94	7.05

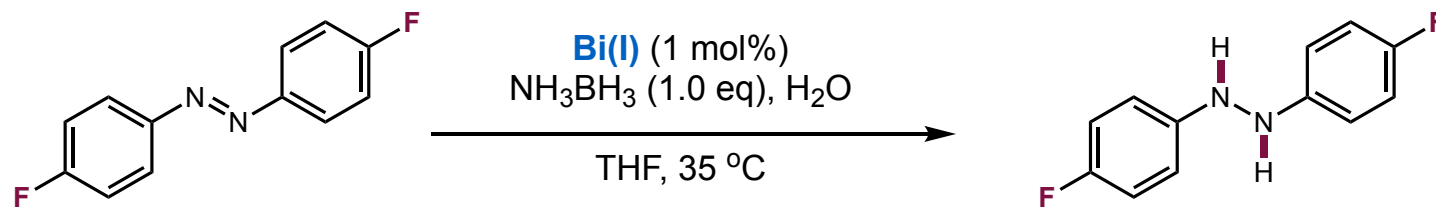
Both N-H and B-H bonds are cleaved in the rate-determining step

Transfer hydrogenation via Bi(I)/Bi(III) redox catalysis

Competition experiment study



Product 1:1 ratio observed

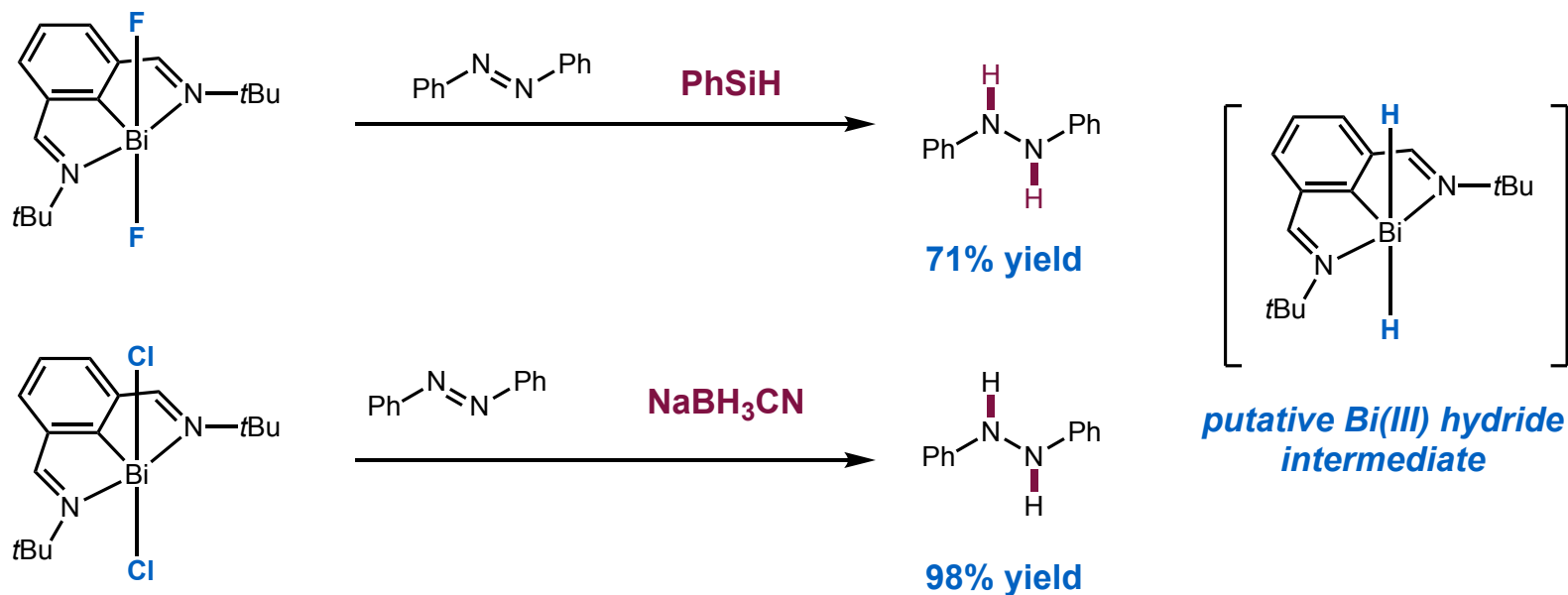


■ azoarenes are not participating in the RDS of this transformation

■ Bi(I) and AB are both involved in the RDS

Transfer hydrogenation via Bi(I)/Bi(III) redox catalysis

■ Stoichiometric study for Bi(III)—H species



■ *Bi(I) and H_2 was detected during the reaction*

■ *proton is from hydridic sources*

Transfer hydrogenation via Bi(I)/Bi(III) redox catalysis

■ Mass-spectrometry study for Bi(III)—H species

