



The Most Important yet Forgotten Organic Transformations

Group Meeting

April 28th, 2010

Benjamin D. Horning

Thank You Contributors!

Thank you very much to everyone who contributed to this group meeting.

A list of contributors and respondees to my email query is given below.

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Tristan Lambert	Pavel Nagorny	Laura Kiessling	Brian Connell
Nicholas Turro	Mike Giuliano	Zhibin Guan	John Montgomery
Robert Knowles	Andrew Phillips		

The Most Important yet Forgotten Organic Transformations

■ Ignoring the important criterion (subjective), where do we find forgotten reactions?

Baran group meetings: <http://www.scripps.edu/chem/baran/html/meetingschedule.html>

<http://www.monomerchem.com/display4.html>

<http://www.organic-chemistry.org/namedreactions/>

The Merck Index contains a named reactions section

1950s Chemische Berichte

<http://orgchem.chem.uconn.edu/namereact/named.html>

<http://www.ecompound.com/>

<http://www.chempensoftware.com/organicreactions.htm>

http://en.wikipedia.org/wiki/List_of_organic_reactions

■ How do we break free of the confines of our "Americanized" view of chemistry?

A French text on named reactions, available online:

Réactions Organiques Classées par Auteurs

<http://www.infotheque.info/cache/9085/www.refer.mg/cours/alcene/index.html>

A German named reactions book from the 1960's (available in our library):

Namenreaktionen der organischen Chemie. English (Organic name reactions; a contribution to the terminology of organic chemistry, biochemistry and theoretical organic chemistry. New York, Wiley, 1964
By Helmut Krauch and Werner Kunz

(Lots of overlap with the Merck Index named reactions section)

The Most Important yet Forgotten Organic Transformations

■ First, some definitions

important - adjective - of much or great significance or consequence

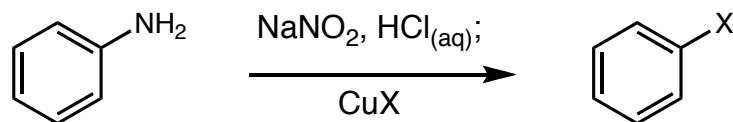
forgotten - past participle of forget - to cease or fail to remember; be unable to recall

- The general consensus is that there are no important yet forgotten organic transformations;
If they are important, they are important because they are being used.
 - The most important yet forgotten problems in organic synthesis
 - Things people should know about that they generally don't
 - The reactions that could or should be more important were they not forgotten
- The control of "relative face selectivity." For example, the exo-selective Diels-Alder reaction
- Direct, catalyzed fluorination, chlorination, and hydroxylation into organic molecules via C-H activation, including diastereo- and enantioselective processes at sp_3 -hybridized stereocenter

The Most Historically Important yet Forgotten Reactions

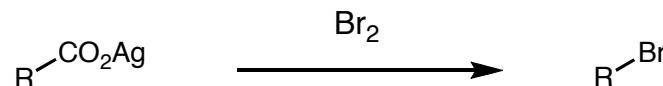
■ The Sandmeyer reaction

Sandmeyer, T. *Ber.* **1884**, 17, 1633-1635.



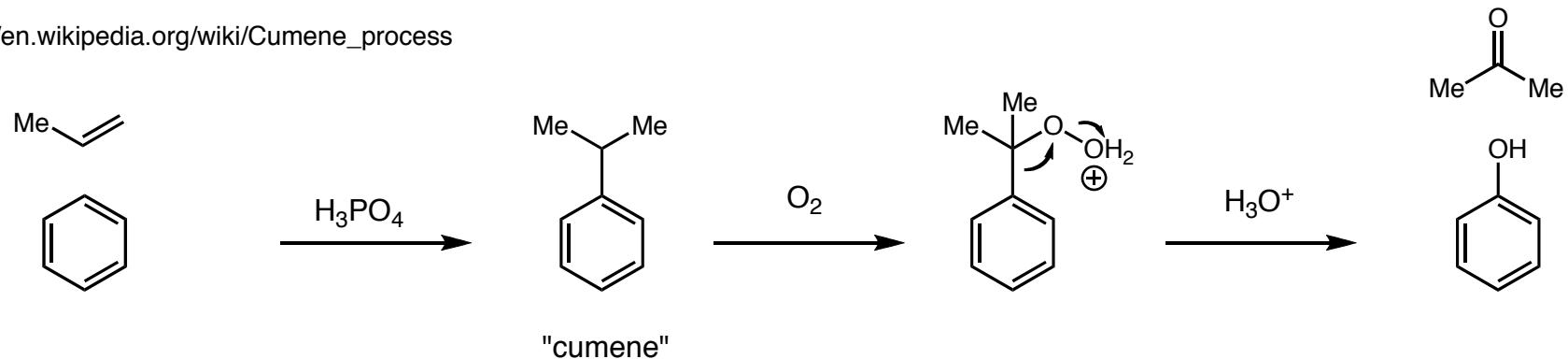
■ The Hunsdiecker reaction

Hunsdiecker, H.; Hunsdiecker, C. *Ber.* **1942**, 75, 291-297.



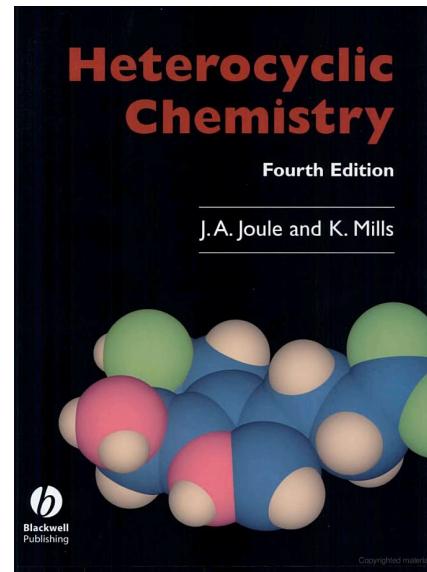
■ The Cumene process

http://en.wikipedia.org/wiki/Cumene_process

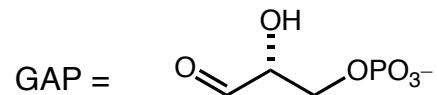


Things People Should Know About that they Generally Don't

■ Heterocycle Chemistry



■ The Calvin Cycle by which mother nature makes all her organic compounds

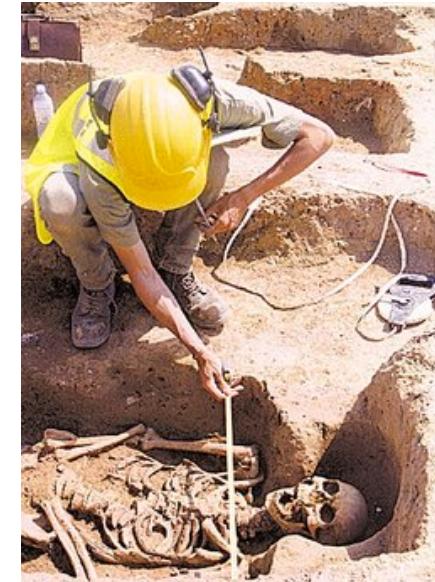


■ Other biochemical reactions, such as reactions of thioesters and thiolate/disulfide interchange

Native Chemical Ligation: Dawson, P. E.; Muir, T. W.; Clark-Lewis, I.; Kent, S. B. *Science*, **1994**, 266, 776-779.

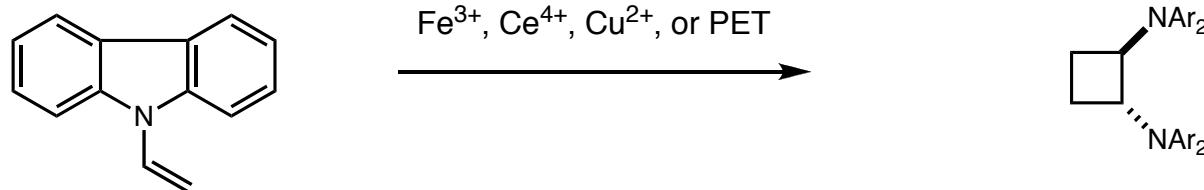
Tonight's Topics

- Radical-cation sigmatropic rearrangements
- Enantioselective photodeconjugation of enoates
- Arene/olefin photocycloaddition
- Trichlorosilane/tertiary amine combination and the Benkeser reaction
- Willgerodt/Kindler reaction
- Burton trifluoromethylation
- Matteson alkylation and the Zweifel Olefin synthesis
- Berman's NO chemistry



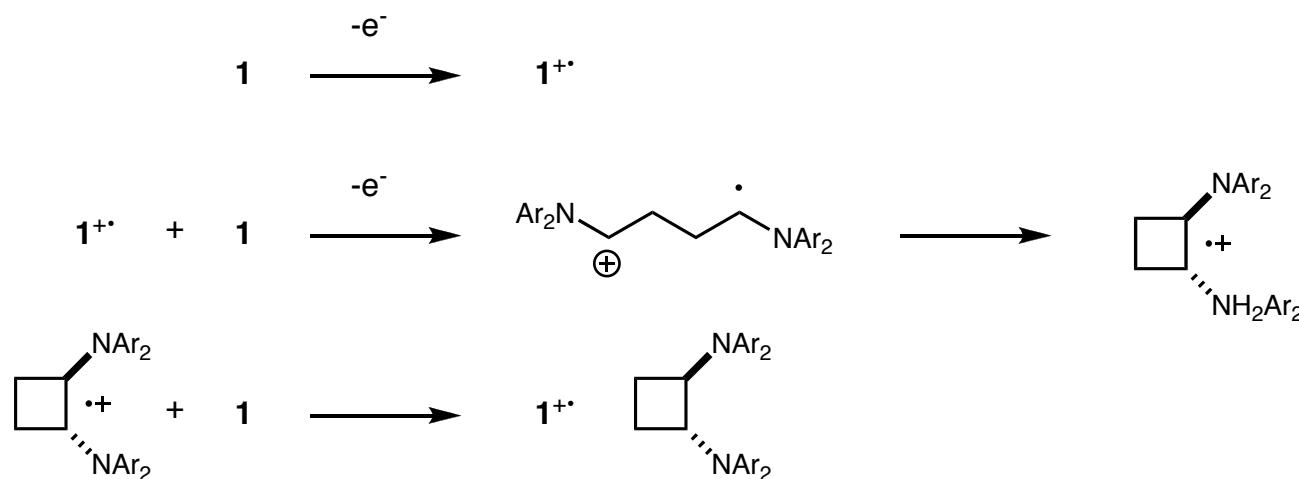
Radical Cation Pericyclic Reactions

- Oxidative conditions were shown to yield the homodimerization of vinyl carbazole **1**



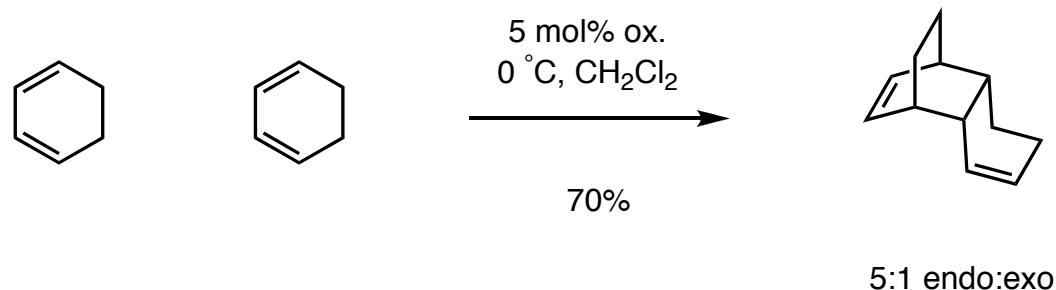
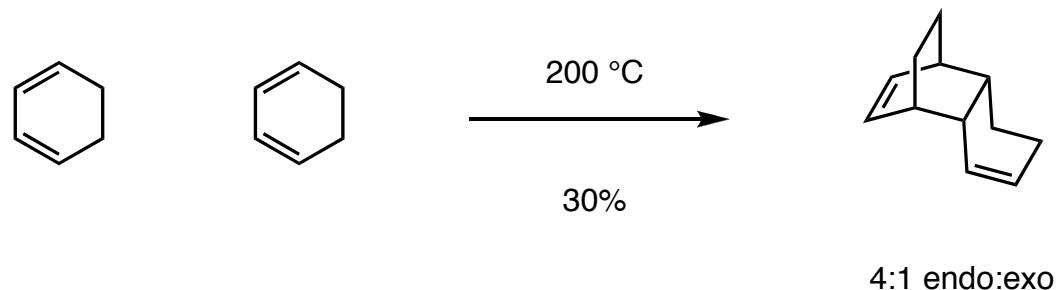
Initial discovery:
Further utilization:

Ledwith, A. *Acc. Chem. Res.* **1972**, 5, 133-139.
Bauld, N. L. *et al.* *Acc. Chem. Res.* **1987**, 20, 371-378.
Bauld, N. L. *Tetrahedron*, **1989**, 45 (17), 5307-5363.



Radical Cation Pericyclic Reactions

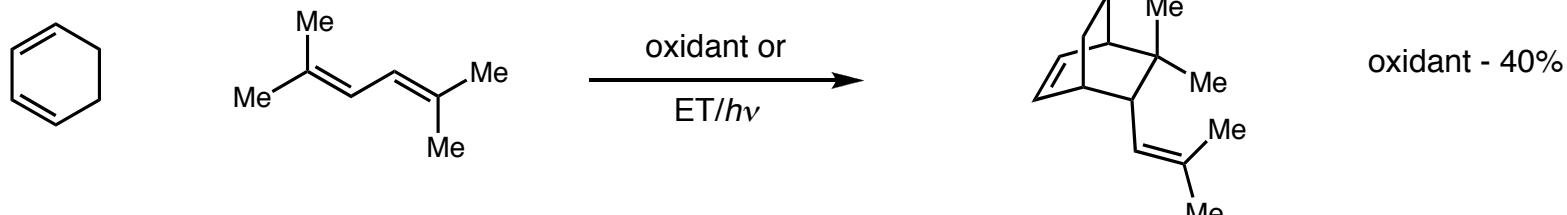
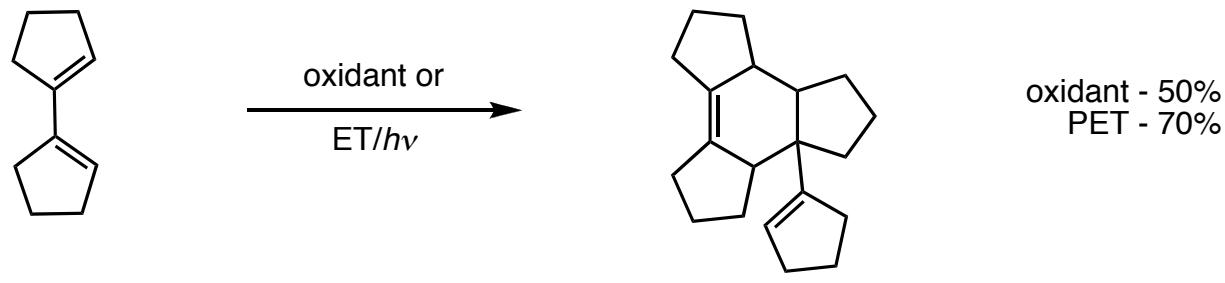
- A useful improvement to diene dimerization



ox. = $\text{Ar}_3\text{N}^+\text{SbCl}_6^-$
an "aminium" salt

Radical Cation Pericyclic Reactions

- Sterically encumbered dienes react well



"(2,5-dimethyl-2,4-hexadiene) had not previously been induced to participate in a Diels-Alder addition as either the dienic or dienophilic component"

ox. = $\text{Ar}_3\text{N}^+\text{SbCl}_6^-$
an "aminium" salt

Radical Cation Pericyclic Reactions

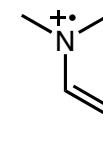
■ Some terminology

1987

catical = cation radical

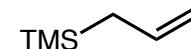
2007

SOMO



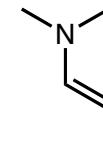
caticophile = neutral component which is
the reaction partner of the catical

SOMO-phile



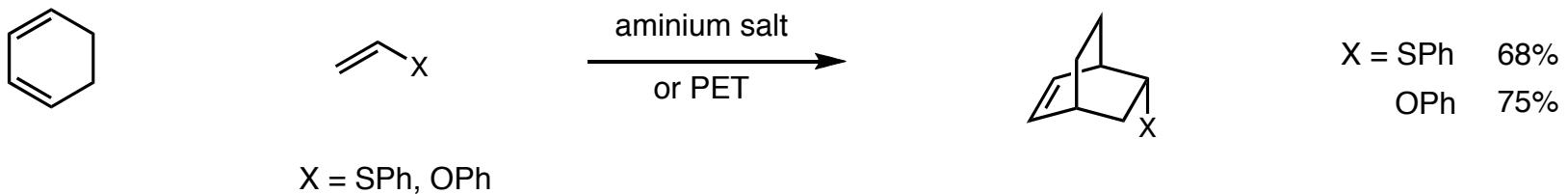
caticogen = neutral component which generates the catical

enamine

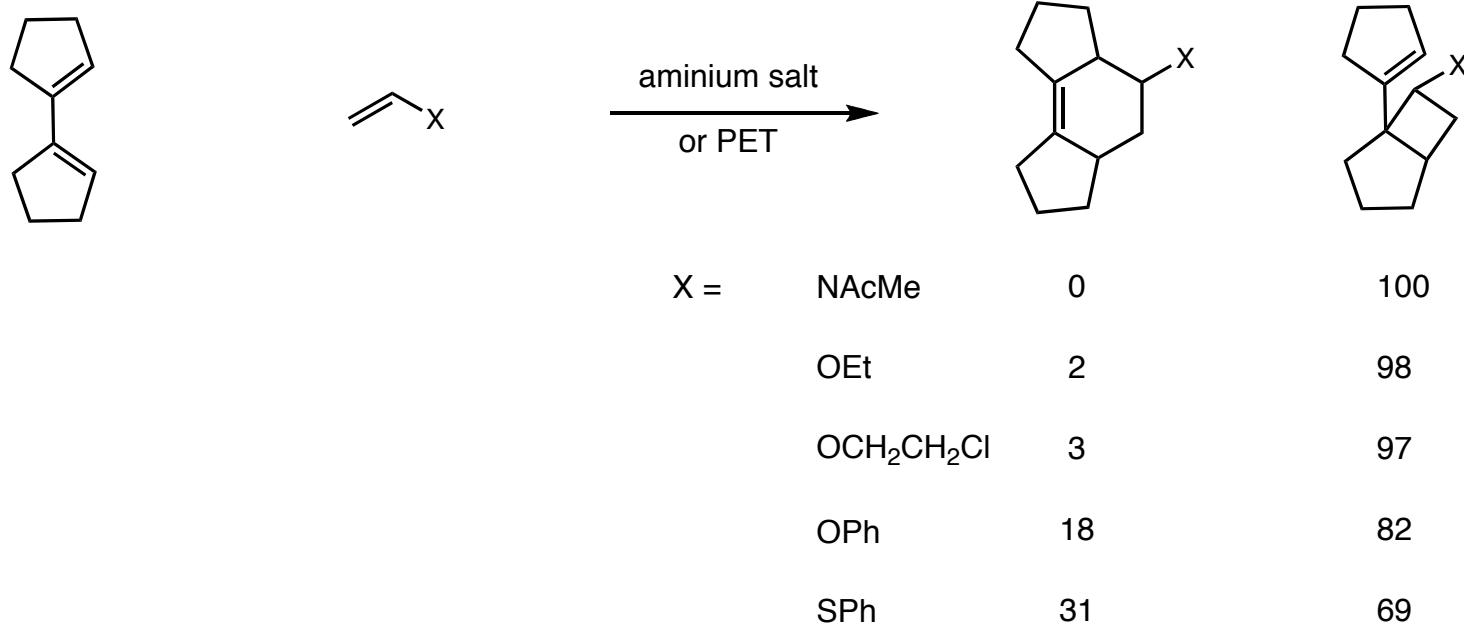


Radical Cation Pericyclic Reactions

■ Diels-Alders beyond diene homo- and heterodimerization

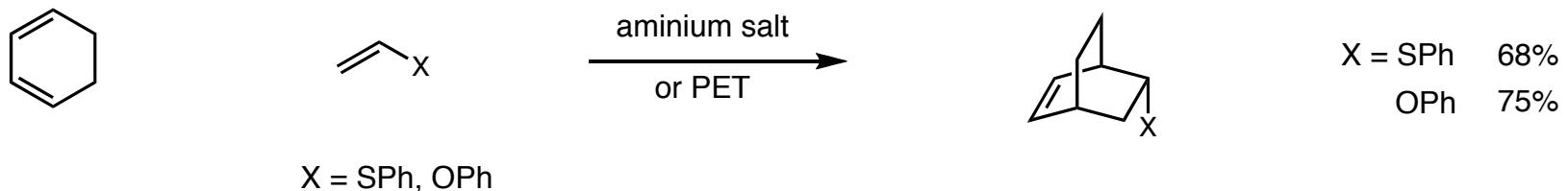


■ Acyclic dienes preferentially yield cyclobutanation products

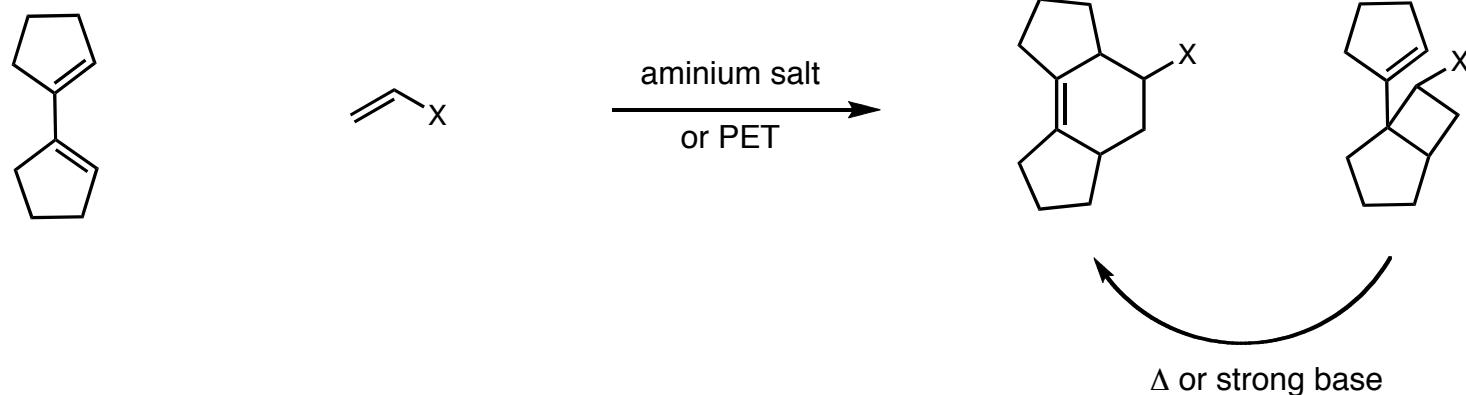


Radical Cation Pericyclic Reactions

■ Diels-Alders beyond diene homo- and heterodimerization

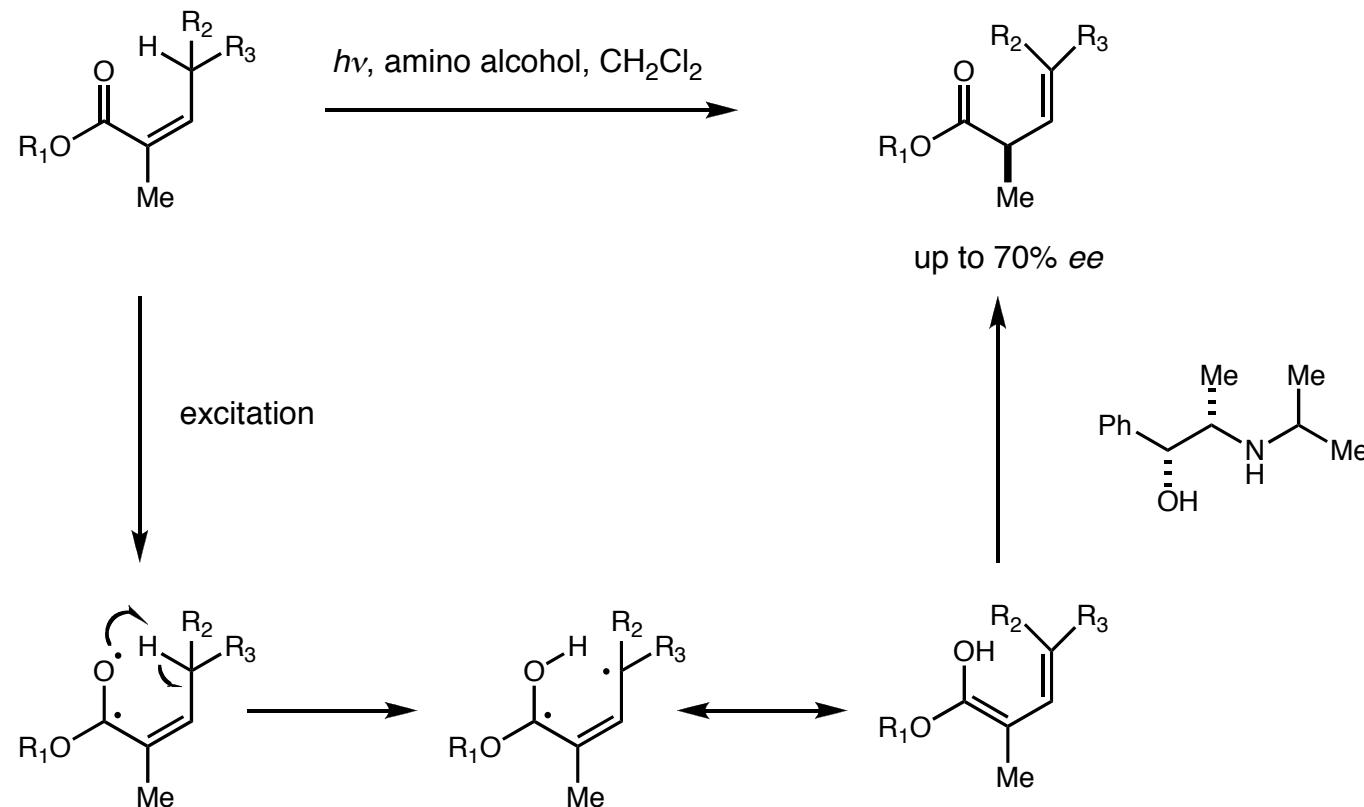


■ Acyclic dienes preferentially yield cyclobutanation products



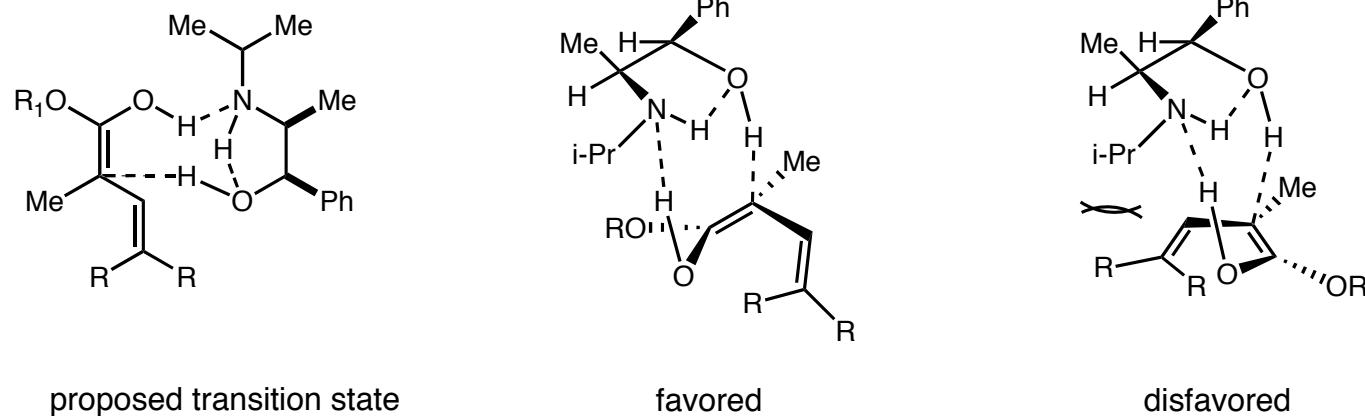
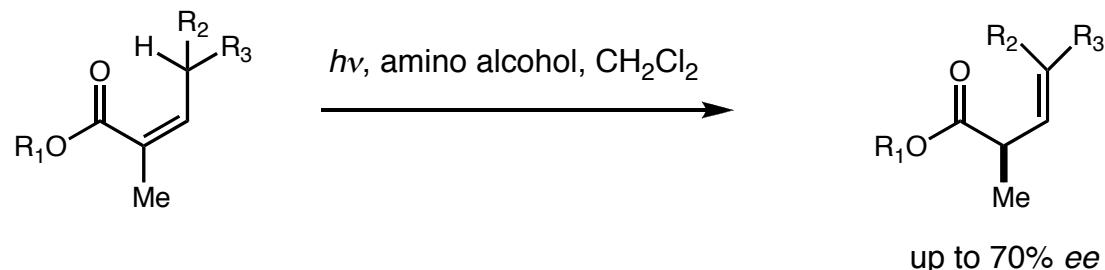
Enantioselective Photodeconjugation of Enoates

■ Norrish type II rearrangement/protonation leads to enantioenriched β,γ unsaturated esters

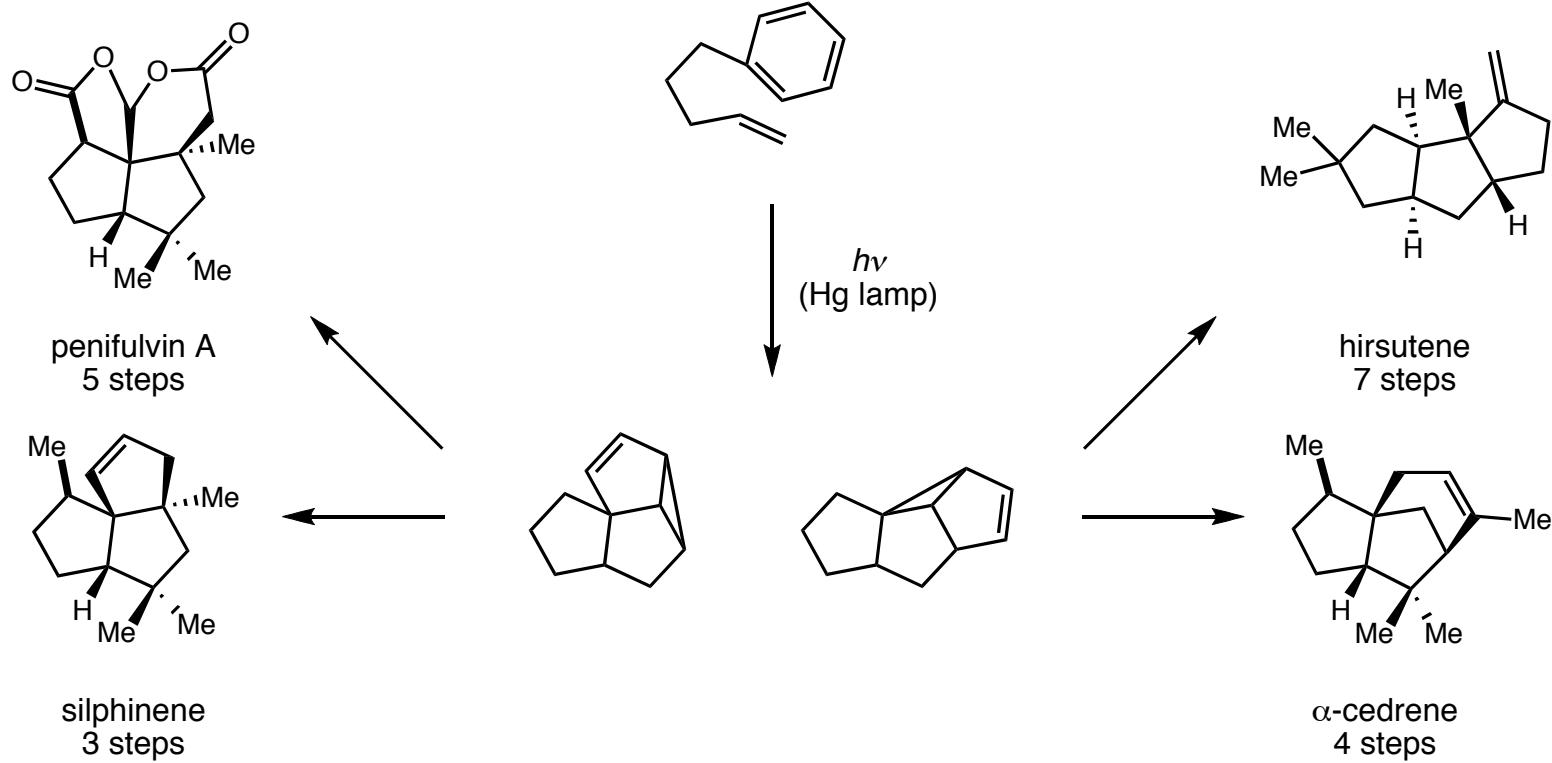


Enantioselective Photodeconjugation of Enoates

■ Norrish type II rearrangement/protonation leads to enantioenriched β,γ unsaturated esters



Intramolecular Arene-Olefin Photocycloaddition



(Intermolecular) Discovery:

Intramolecular variant:

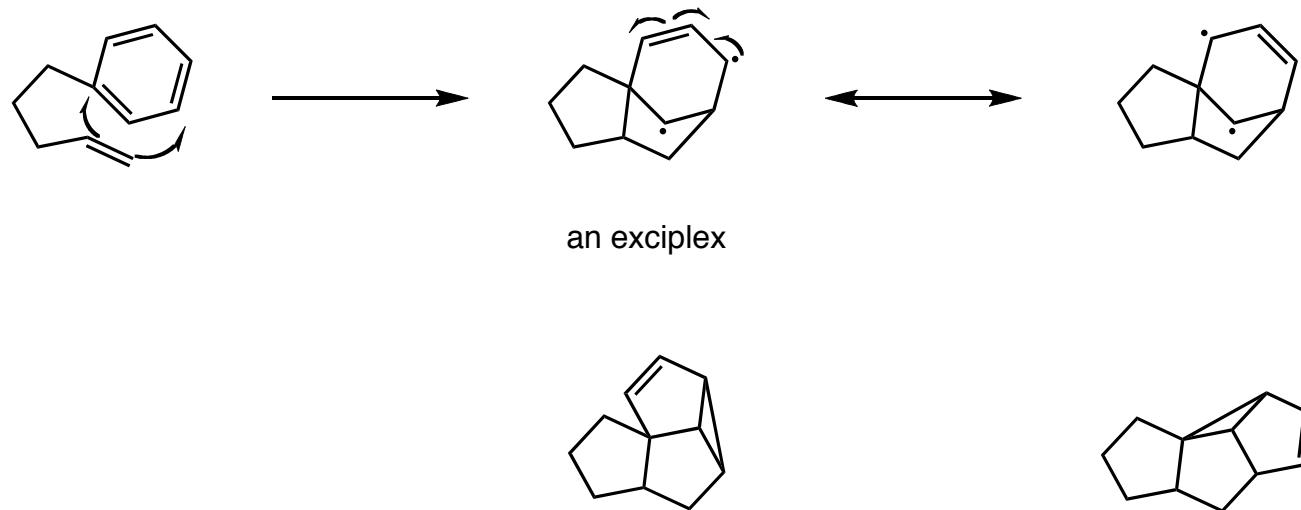
Histurene:
Cedrene:
Silphinene:
Penifulvins:

Wilzbach, K. E.; Kaplan, L. *J. Am. Chem. Soc.* **1966**, *88*, 2066-2067.
 Bryce-Smith, D.; Gilbert, A.; Orger, B. H. *Chem. Commun.* **1966**, 512.
 Morrison, H.; Ferree, W. *J. Chem Soc., Chem. Commun.* **1969**, 268.
 Ferree, W.; Grutzner, J.; Morrison, H. *J. Am. Chem. Soc.* **1971**, *93*, 5502-5512.
 Wender, P. A.; Howbert, J. J. *Tetrahedron Lett.* **1982**, *23*, 3983-3986.
 Wedner, P. A.; Howbert, J. J. *J. Am. Chem. Soc.* **1981**, *103*, 688-690.
 Wender, P. A.; Ternansky, R. J. *Tetrahedron Lett.* **1985**, *26*, 2625-2628.
 Gaich, T.; Mulzer, J. *J. Am. Chem. Soc.* **2009**, *131*, 452-453.
 Gaich, T.; Mulzer, J. *Org. Lett. ASAP*.
 Russell, A. T.; Chappell, D. *Org. Biomol. Chem.* **2006**, *4*, 4409-4430.

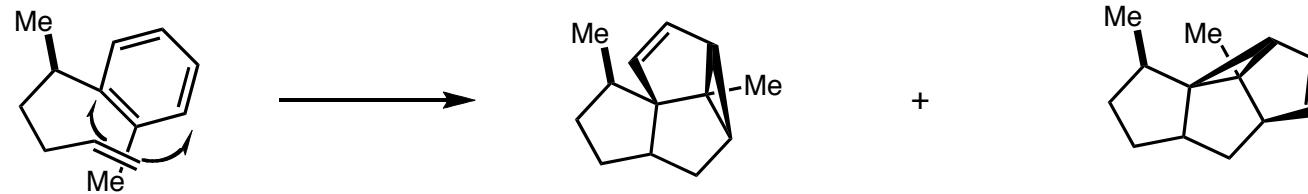
Review of arene-olefin photocycloaddition in total synthesis:

Intramolecular Arene-Olefin Photocycloaddition

- Photoexcitation of a bichromophoric arene-olefin leads to exciplex formation/cyclization



- The presence of a benzylic stereocenter serves to impart diastereoselectivity



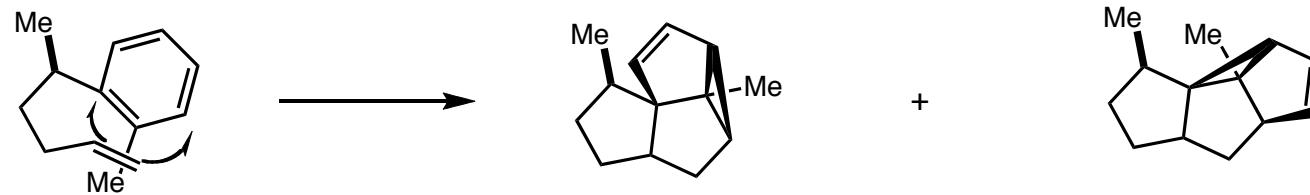
Intramolecular Arene-Olefin Photocycloaddition

■ Model for diastereoselectivity



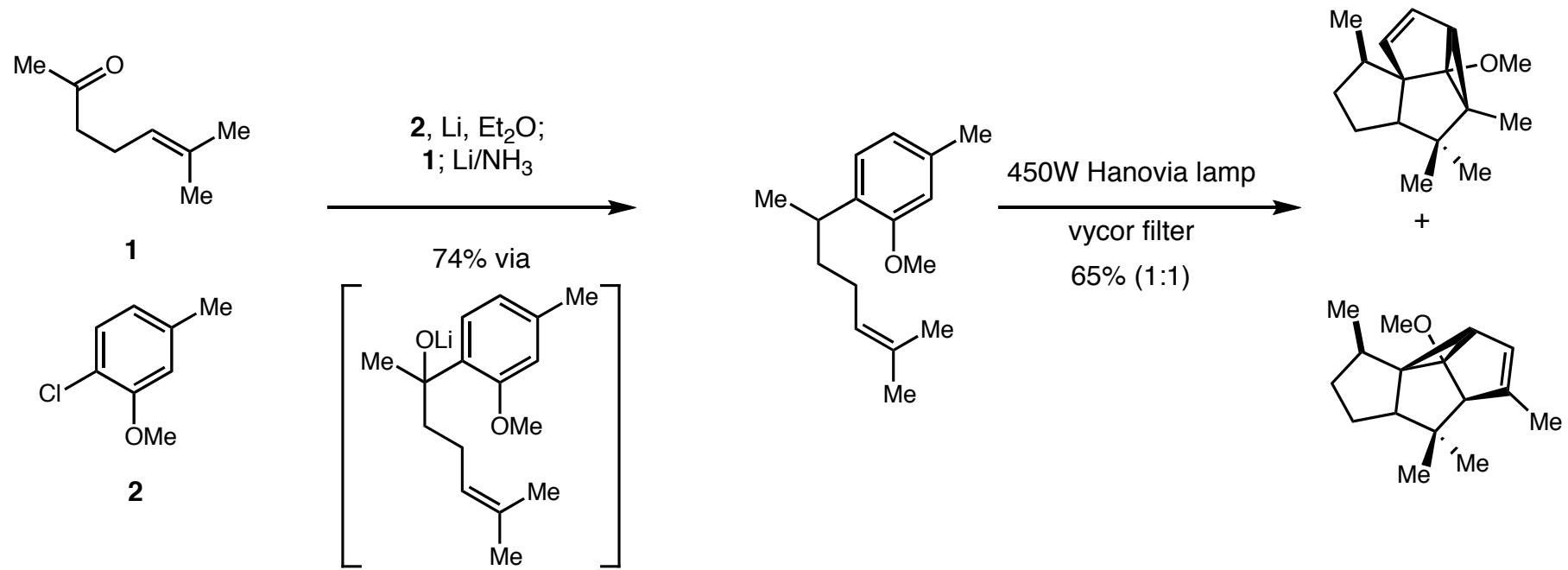
single diastereomer observed for each regioisomer

■ The presence of a benzylic stereocenter serves to impart diastereoselectivity



Intramolecular Arene-Olefin Photocycloaddition

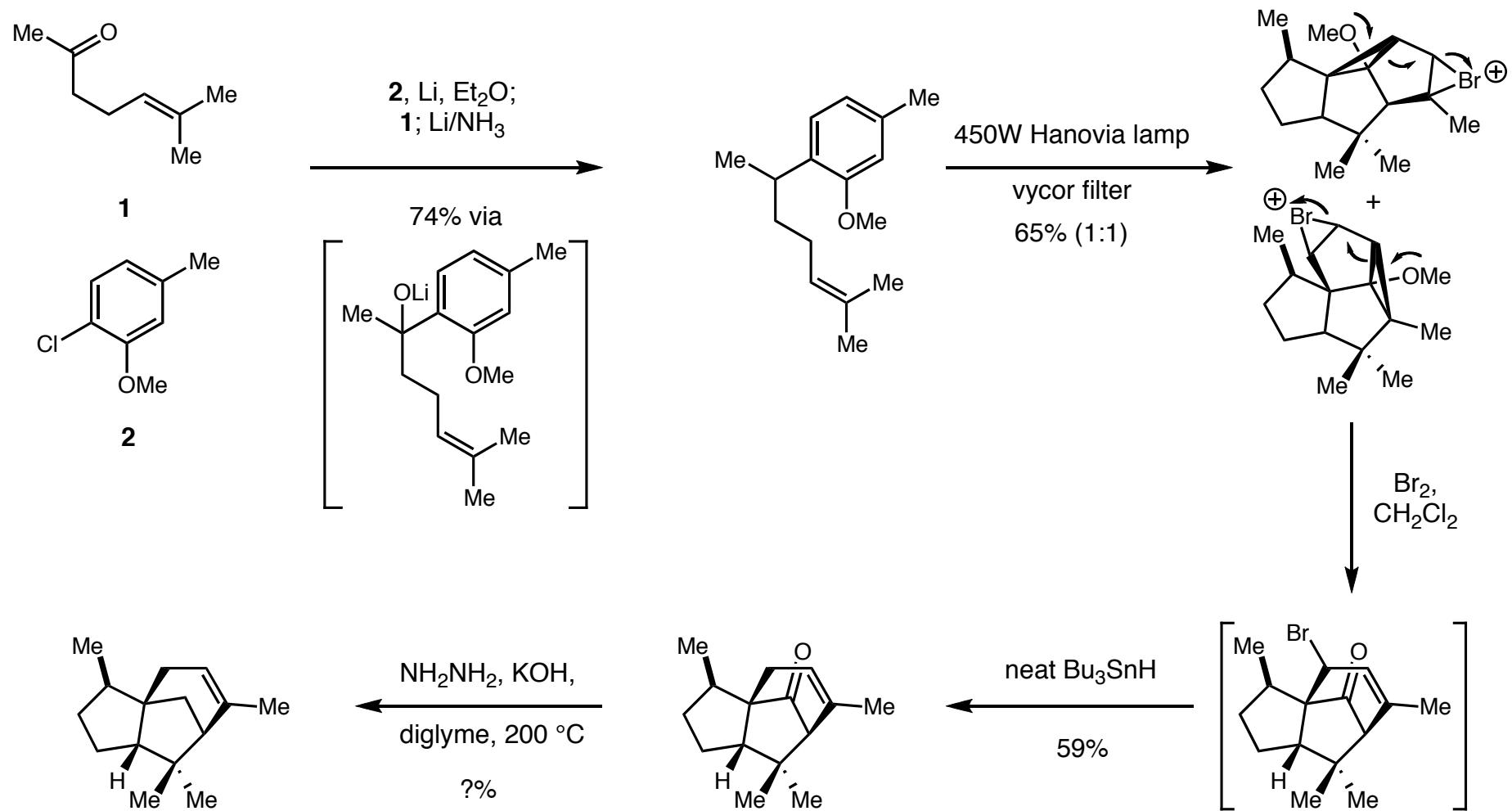
■ Cyclized products can be advanced to a variety of terpenoid natural products



Hall alkylation: Hall, S. S.; McEnroe, F. J. *J. Org. Chem.* **1975**, *40*, 271-275.

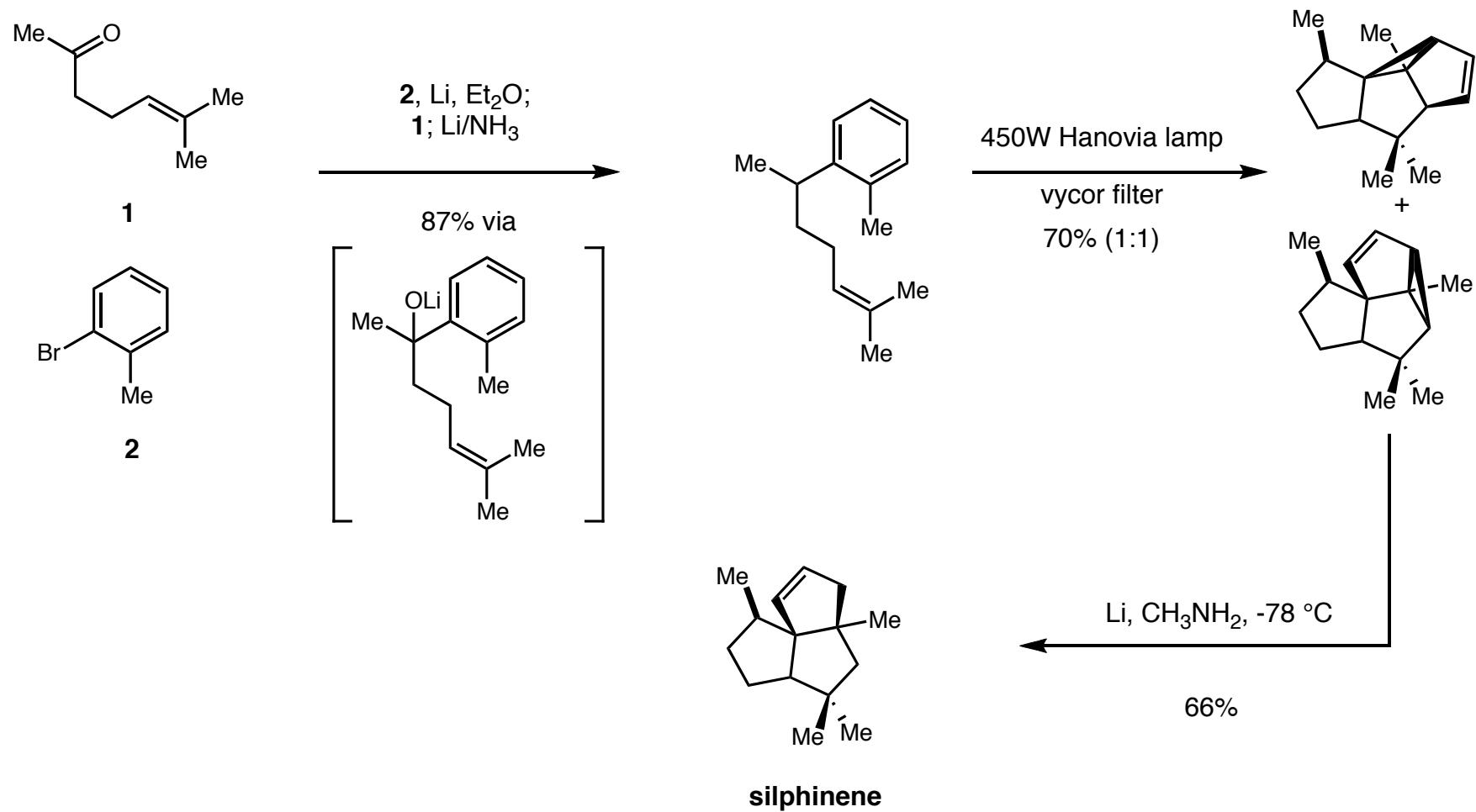
Intramolecular Arene-Olefin Photocycloaddition

■ Cyclized products can be advanced to a variety of terpenoid natural products



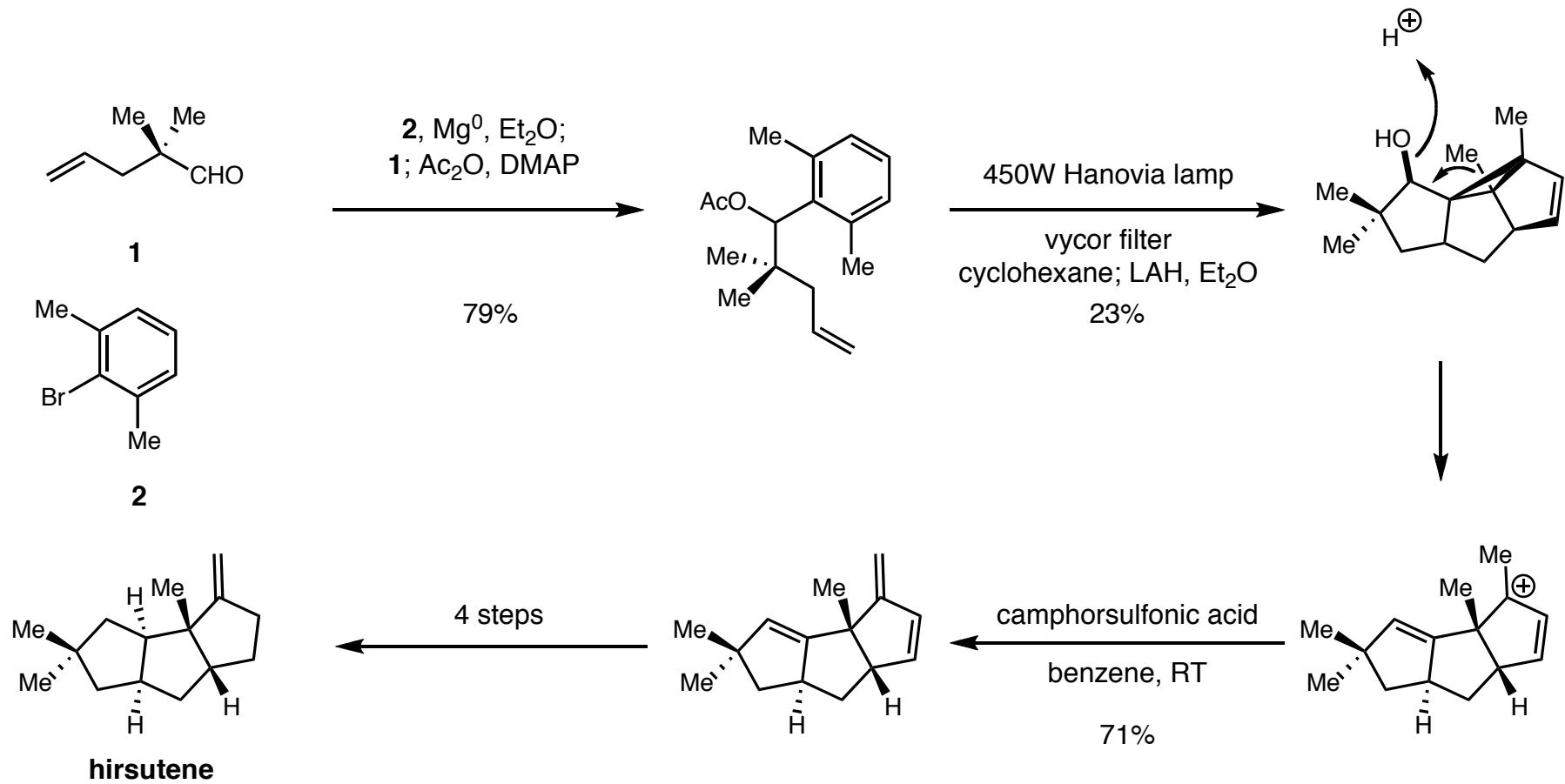
Intramolecular Arene-Olefin Photocycloaddition

■ Reductive cleavage leads to a different series of products



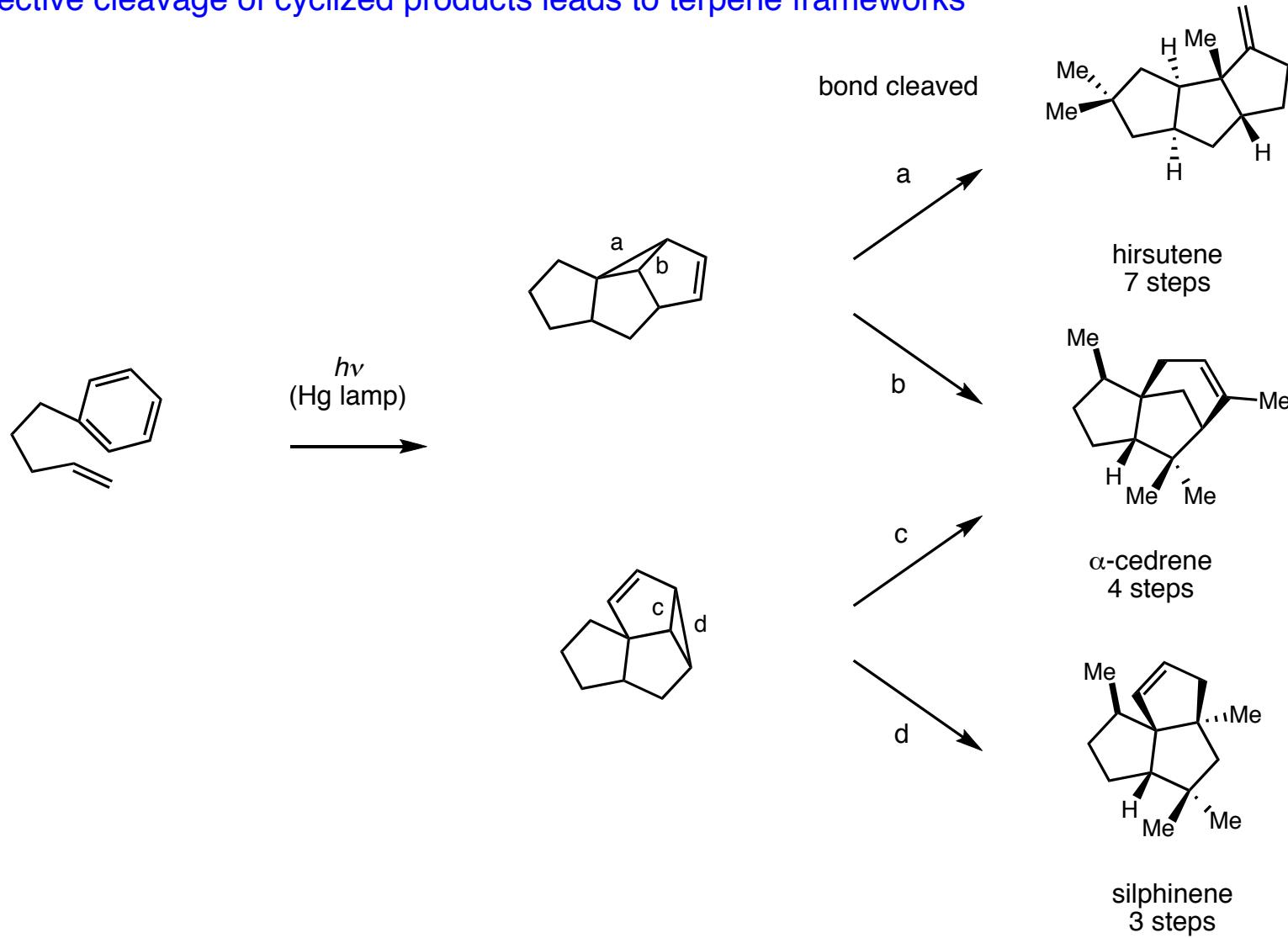
Intramolecular Arene-Olefin Photocycloaddition

■ A third mode of cleavage leads to triquinanes



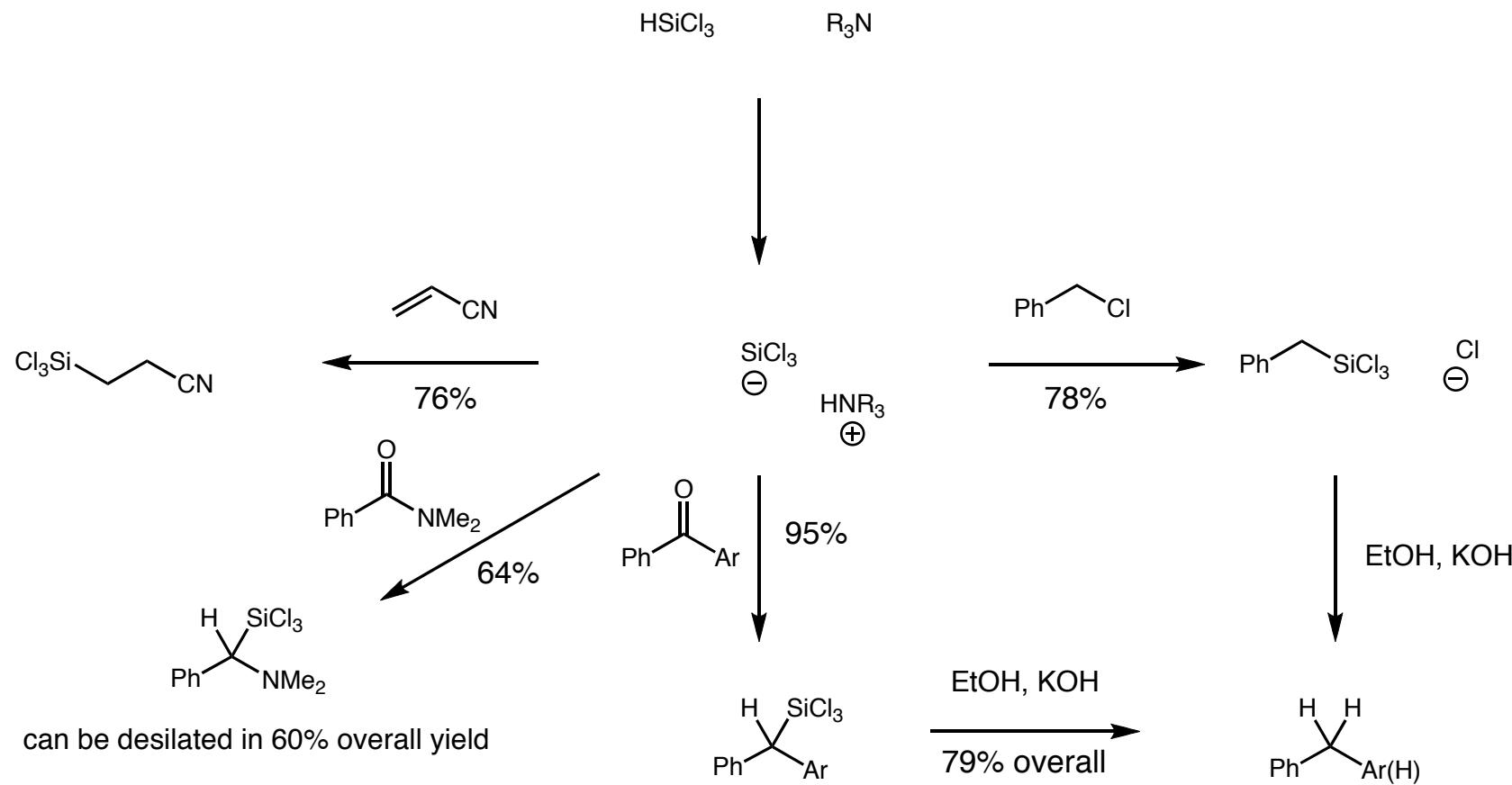
Intramolecular Arene-Olefin Photocycloaddition

- Selective cleavage of cyclized products leads to terpene frameworks



Trichlorosilane-Tertiary Amine Combinations - The Benkeser Reaction

- Trichlorosilane is inexpensive and abundant due to its use in transistor-grade silicon

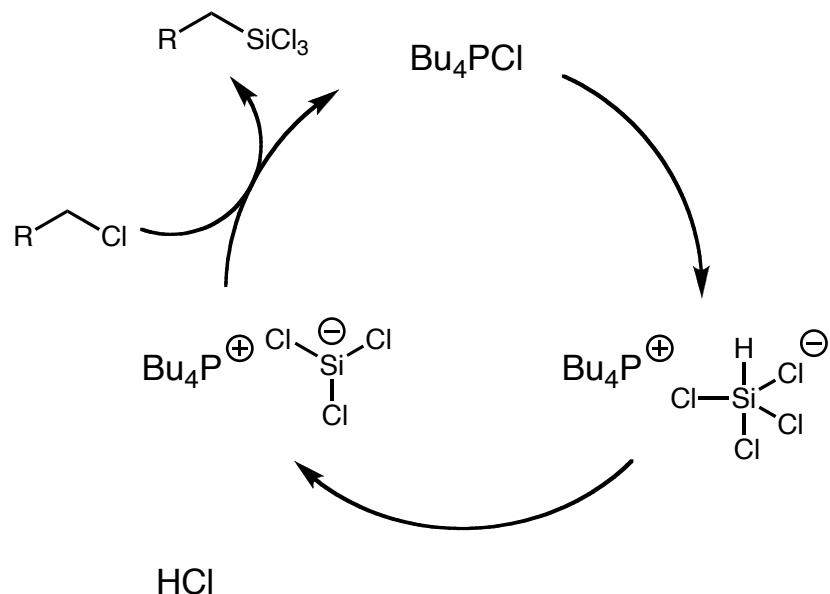
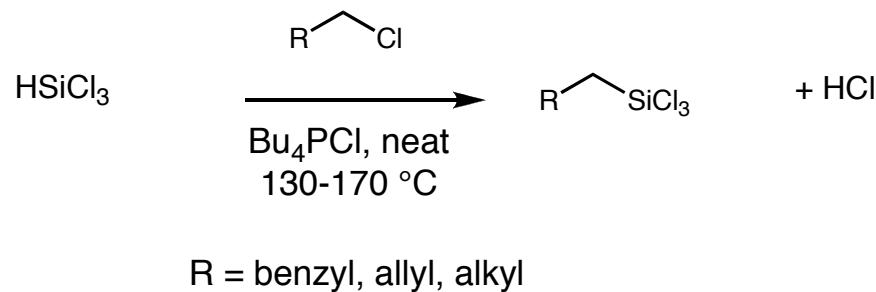


- Unfortunately, these reactions only work with HSiCl_3 , not even HMeSiCl_2 works

Benkeser, R. B. *Acc. Chem. Res.* **1971**, 4, 94-100.

Trichlorosilane-Tertiary Amine Combinations - The Benkeser Reaction

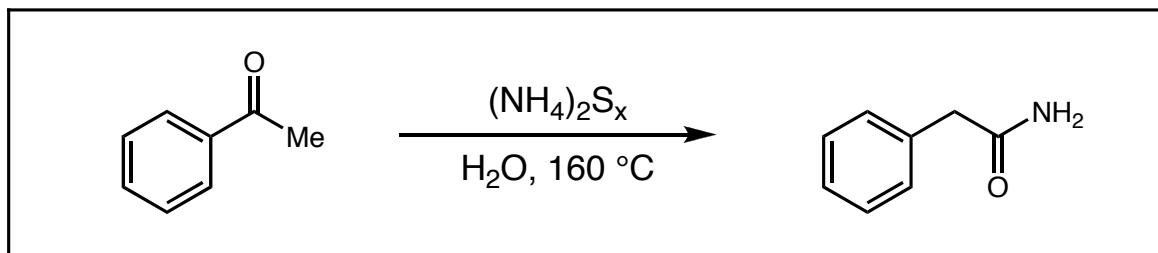
- A recent (2001) JACS paper updated this methodology, expanding the nature of the electrophile



Jung, I. N. et al. *J. Am. Chem. Soc.* **2001**, 123, 5584-5585.

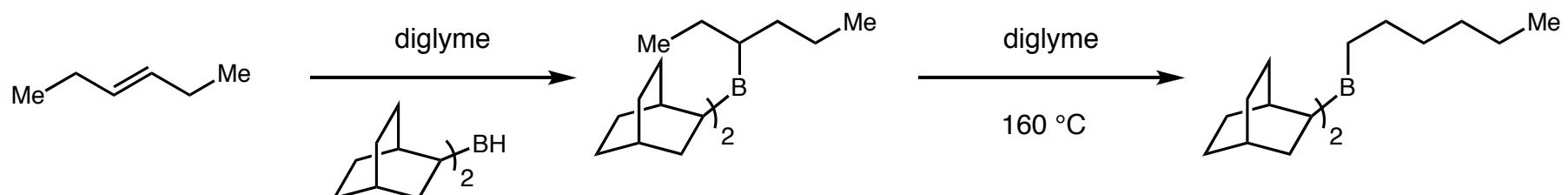
The Willgerodt-Kindler Reaction

- Reaction of aryl alkyl ketones with an amine and sulfur at high temperatures

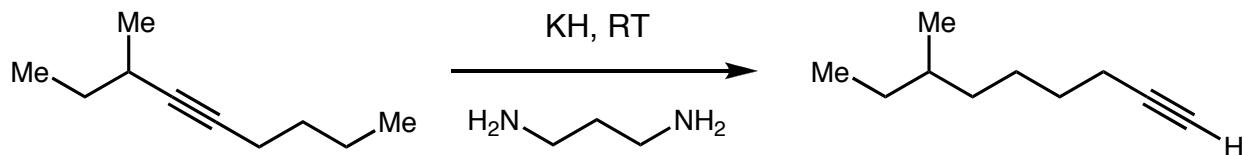


Willgerodt, C. *Ber.* **1887**, *20*, 2467-2470.
Kindler, K. *Ann. Chem.* **1923**, *431*, 187.

- Conceptually related to the (H. C.) Brown borane/olefin isomerization and the (C. A.) Brown acetylene zipper reactions



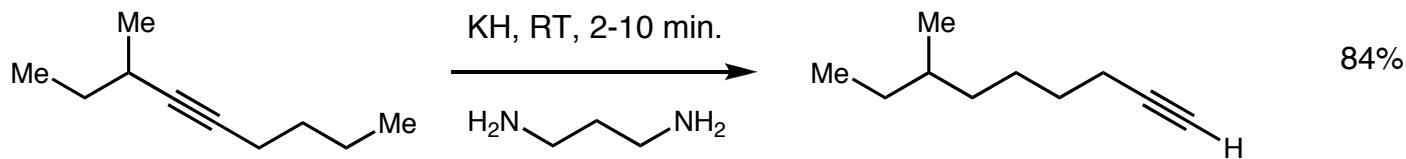
Brown, H. C.; Racherla, U. S. *J. Organometal. Chem.* **1983**, *241*, C37-C40.



Brown, C. A.; Yamashita, A. *J. Am. Chem. Soc.* **1974**, *97*, 891-891.

The Willgerodt-Kindler Reaction - Related Contrathermodynamic Isomerizations

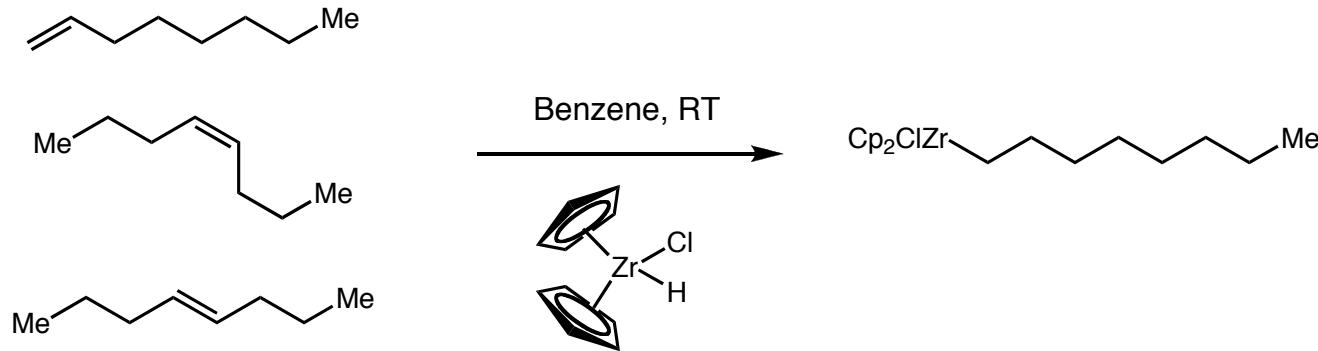
- C. A. Brown's KAPA contribution makes the alkyne zipper synthetically useful



Previous conditions required > 100 °C, longer than 2 hours, and only worked on methyl alkynes

Brown, C. A.; Yamashita, A. *J. Am. Chem. Soc.* **1974**, 97, 891-891.

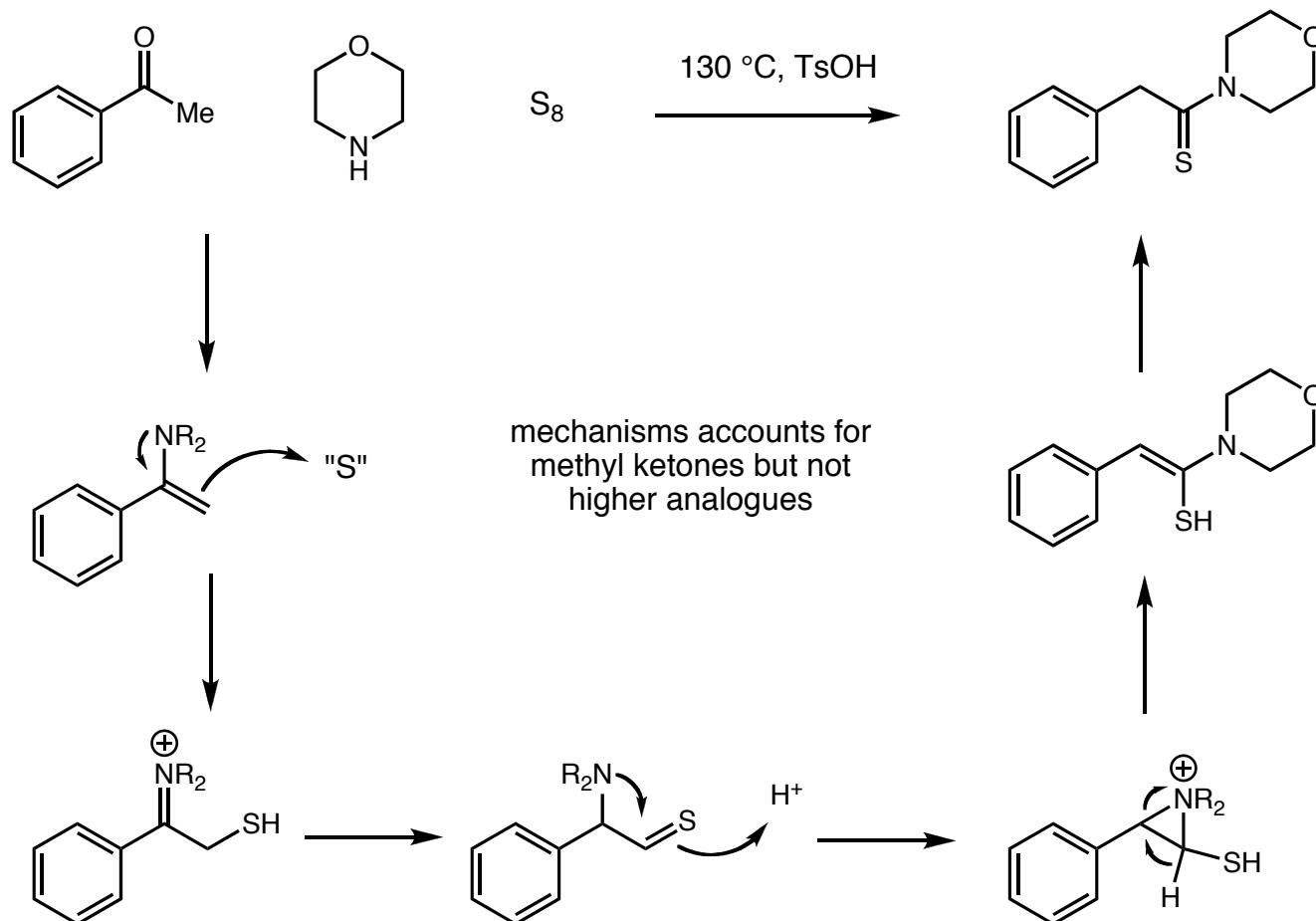
- J. Schwartz's hydrozirconation isomerization trumps H. C. Brown's hydroboration isomerization



Hart, D. W.; Schwartz, J. *J. Am. Chem. Soc.* **1974**, 96, 8115-8116.

The Willgerodt-Kindler Reaction - Mechanism

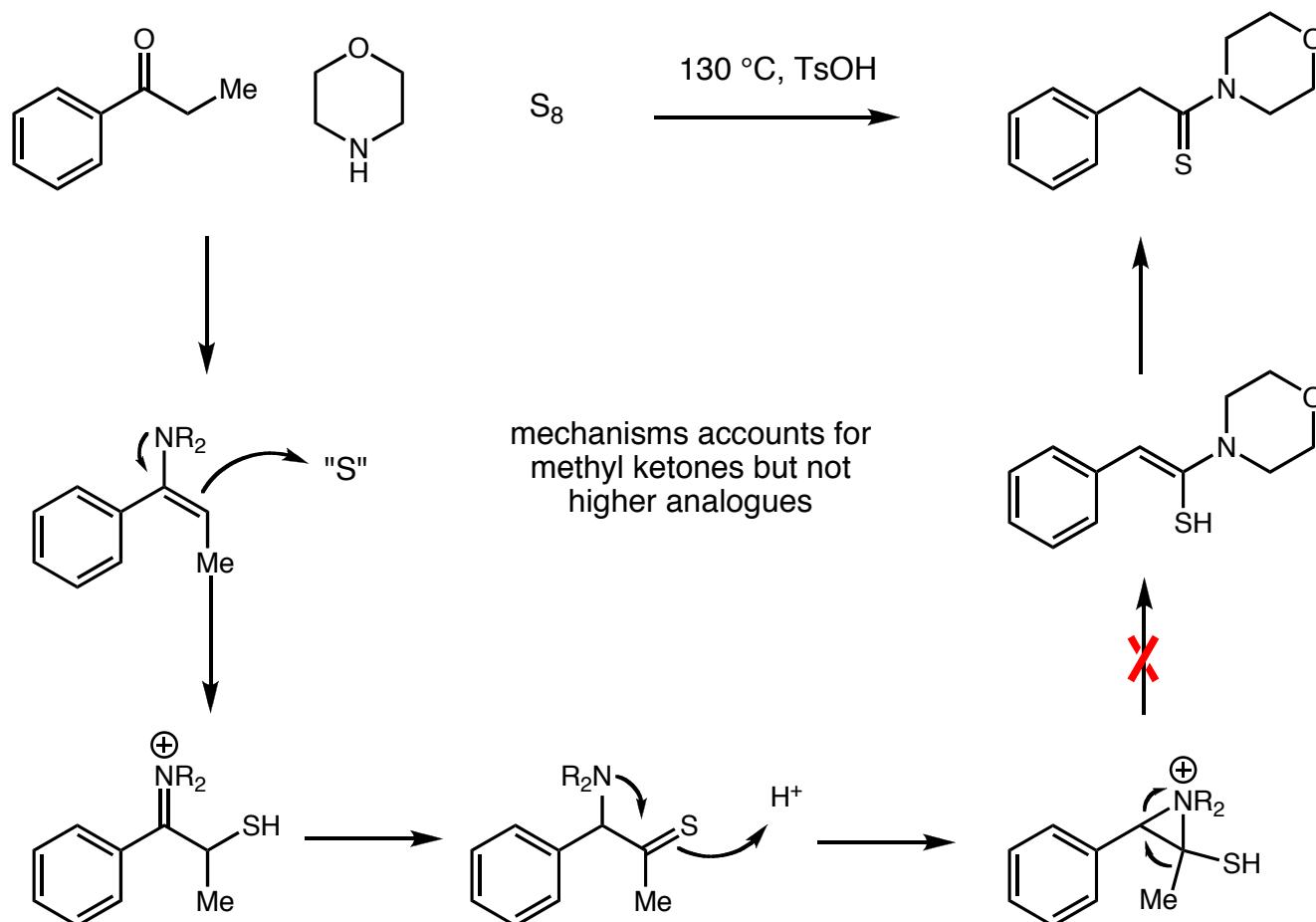
- At high temperatures, many mechanistic oddities are possible



Carmack, M., *J. Heterocyclic Chem.* **1989**, 26, 1319-1323.

The Willgerodt-Kindler Reaction - Mechanism

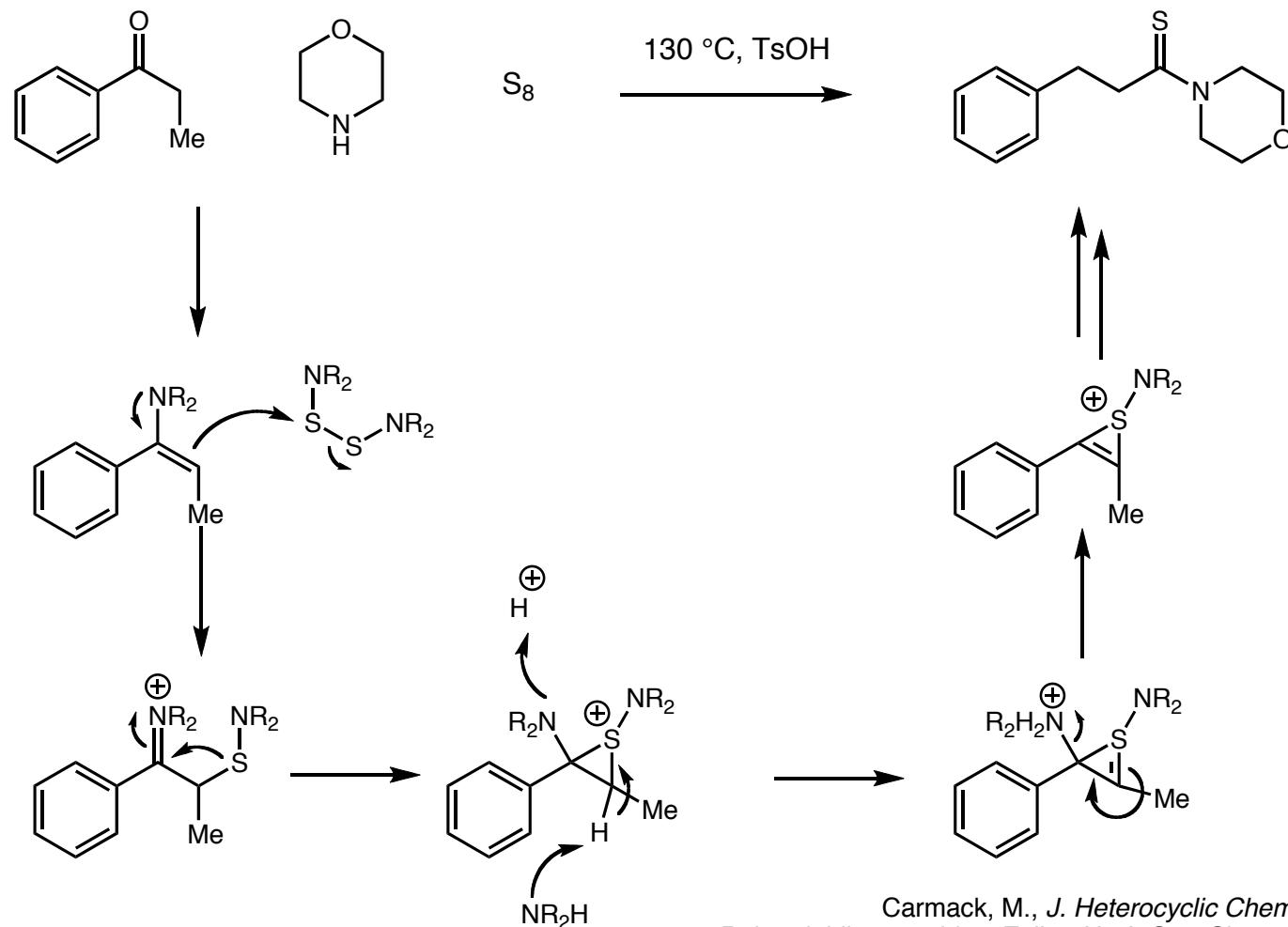
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Carmack, M., *J. Heterocyclic Chem.* **1989**, 26, 1319-1323.

The Willgerodt-Kindler Reaction - Mechanism

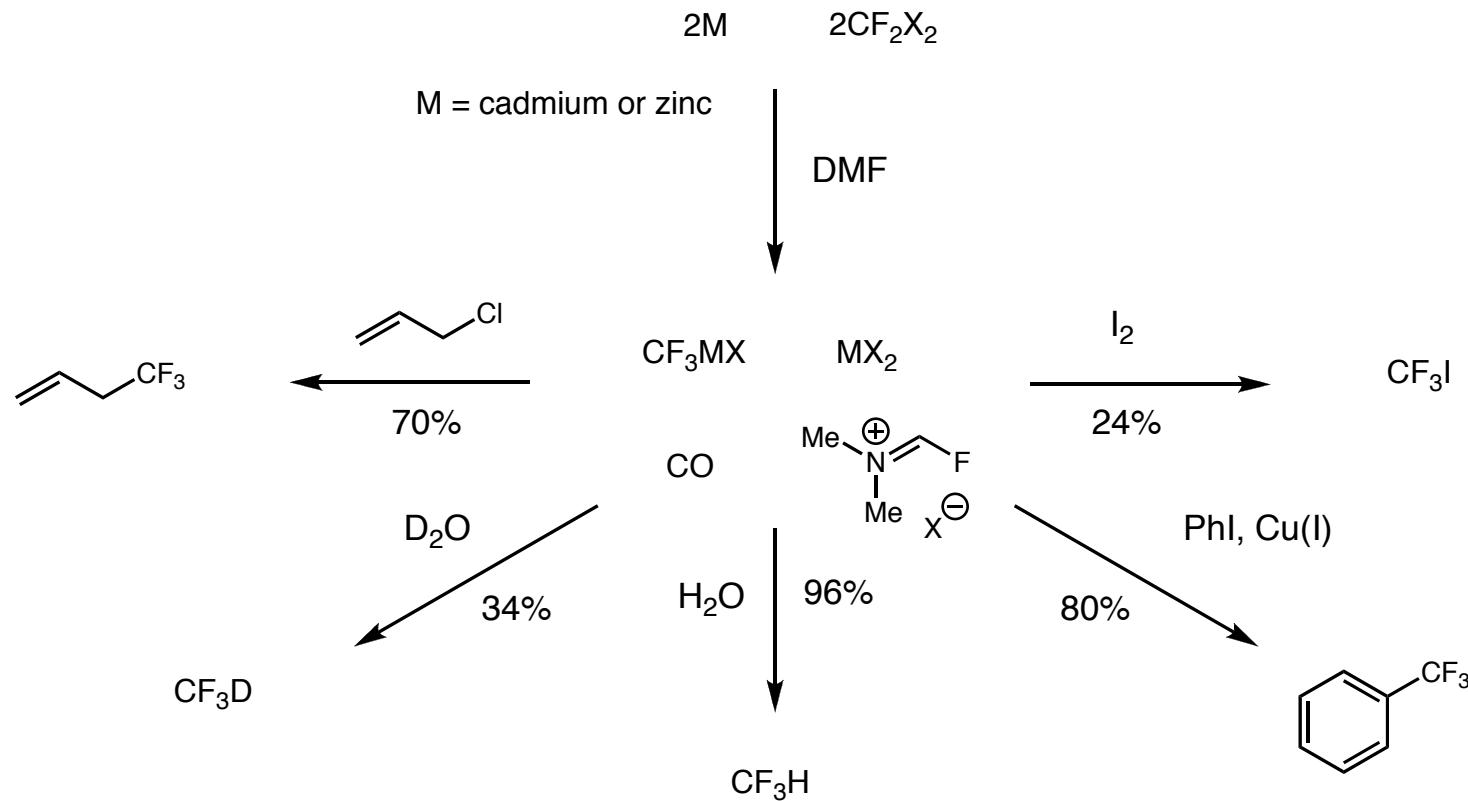
- In order to walk the carbonyl along the chain, a new mechanistic proposal was needed



Carmack, M., *J. Heterocyclic Chem.* **1989**, *26*, 1319-1323.
Related thiirene oxides: Zoller, U. *J. Org. Chem.* **1985**, *50*, 1107-1110.

Burton Trifluoromethyl Organometallics

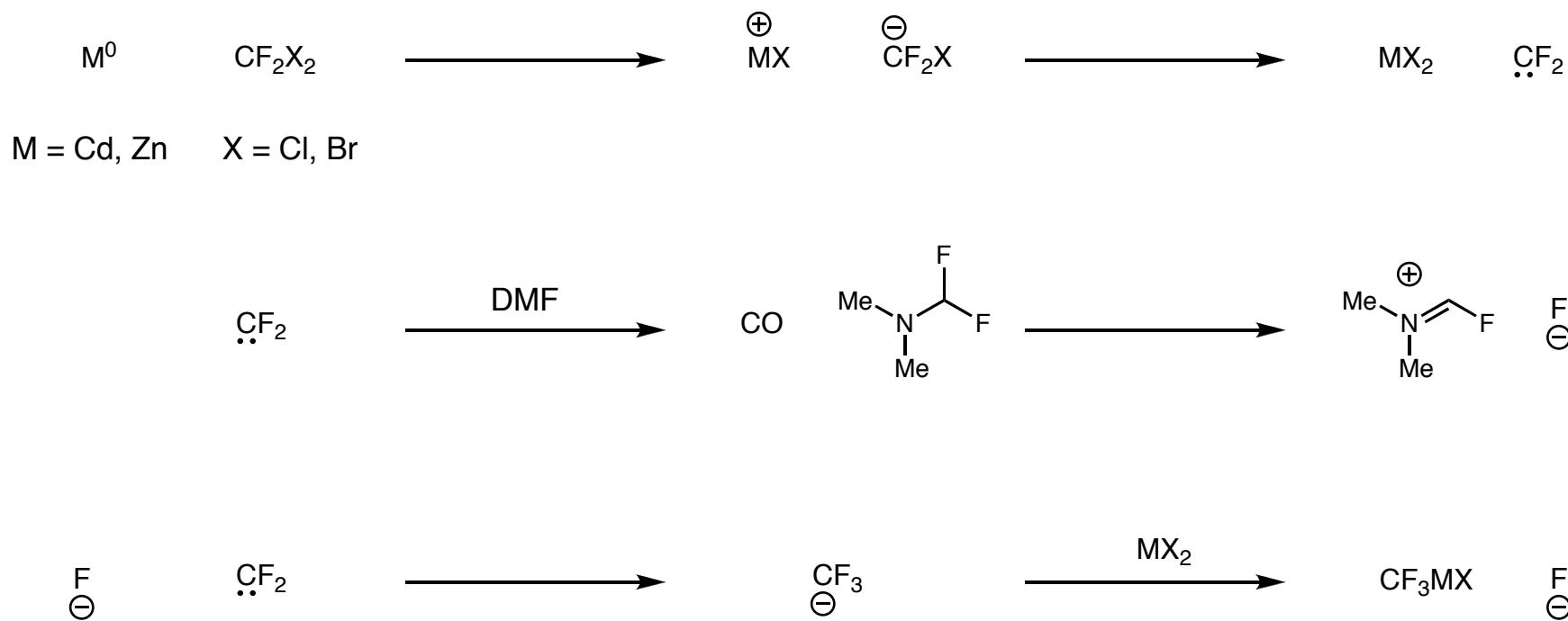
■ Preparation of trifluoromethyl organometallics via reduction of CF_2X_2



Burton, D. J.; Wiemers, D. M. *J. Am. Chem. Soc.* **1985**, *107*, 5014-5015.
Burton, D. J.; Wiemers, D. M. *J. Am. Chem. Soc.* **1986**, *108*, 832-834.

Burton Trifluoromethyl Organometallics

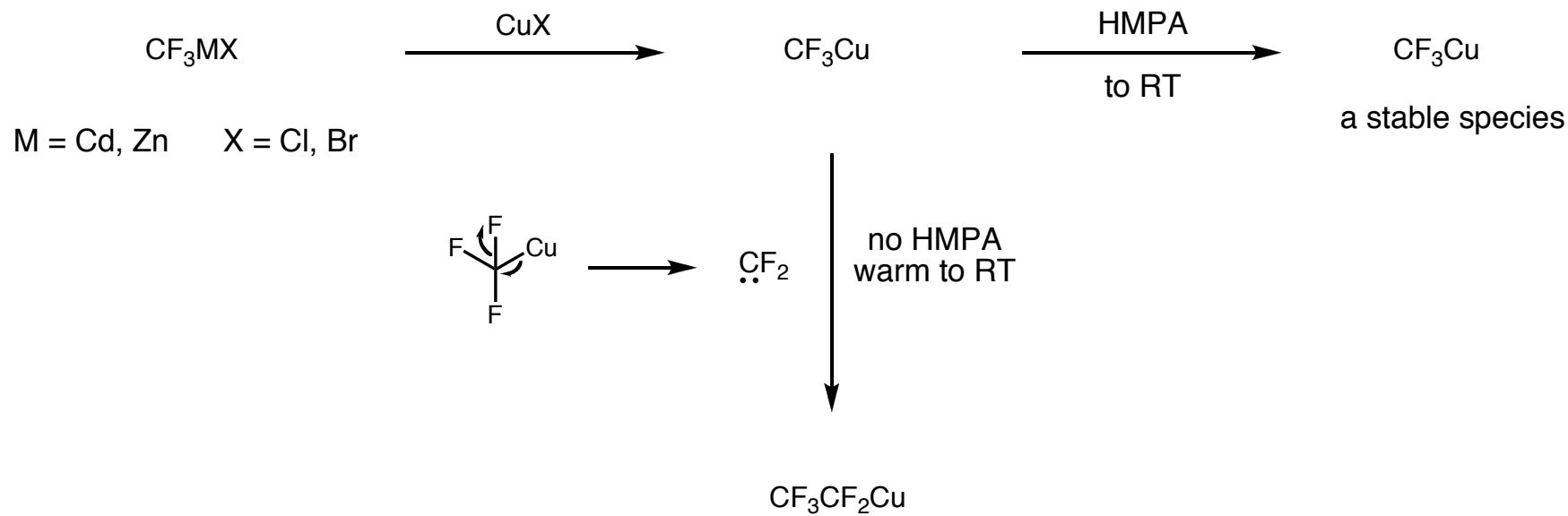
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Burton Trifluoromethyl Organometallics

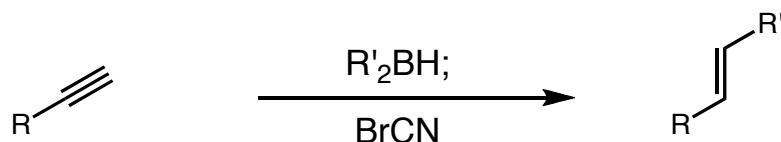
■ Preparation of trifluoromethyl organometallics via reduction of CF_2X_2



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Burton, D. J.; Wiemers, D. M. *J. Am. Chem. Soc.* **1986**, *108*, 832-834.

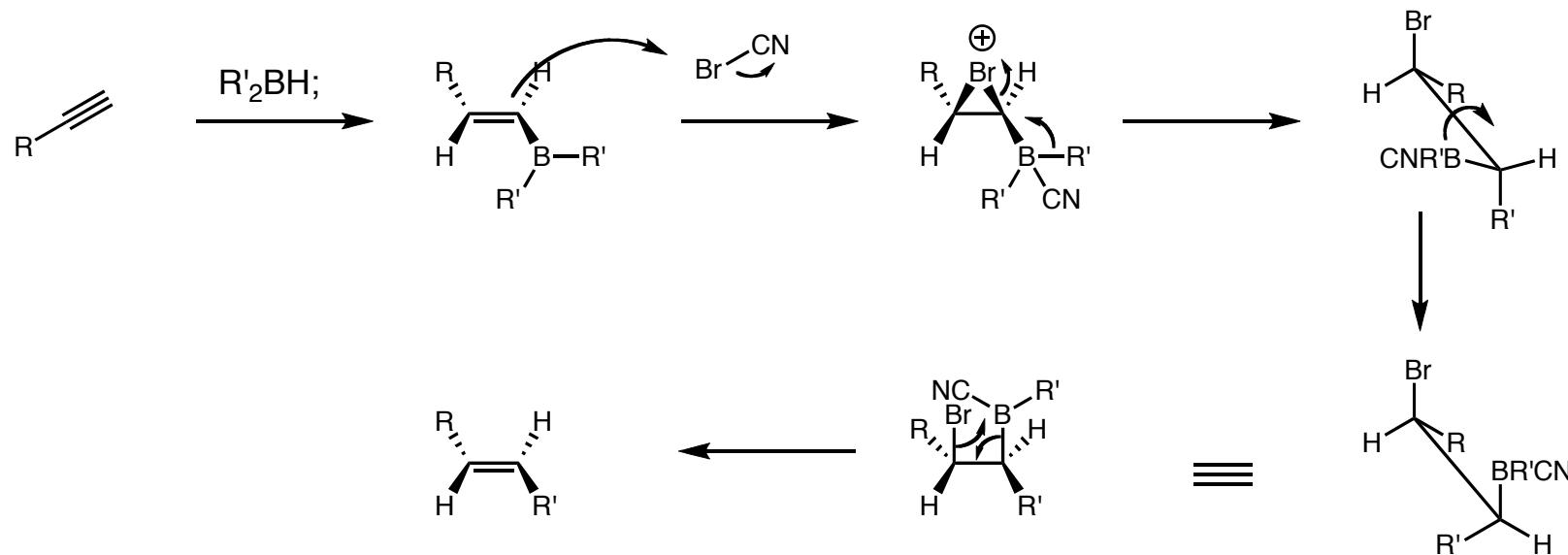
The Zweifel Olefin Synthesis

- Hydroboration of an alkyne leads to a vinyl borane, which can undergo net reductive elimination



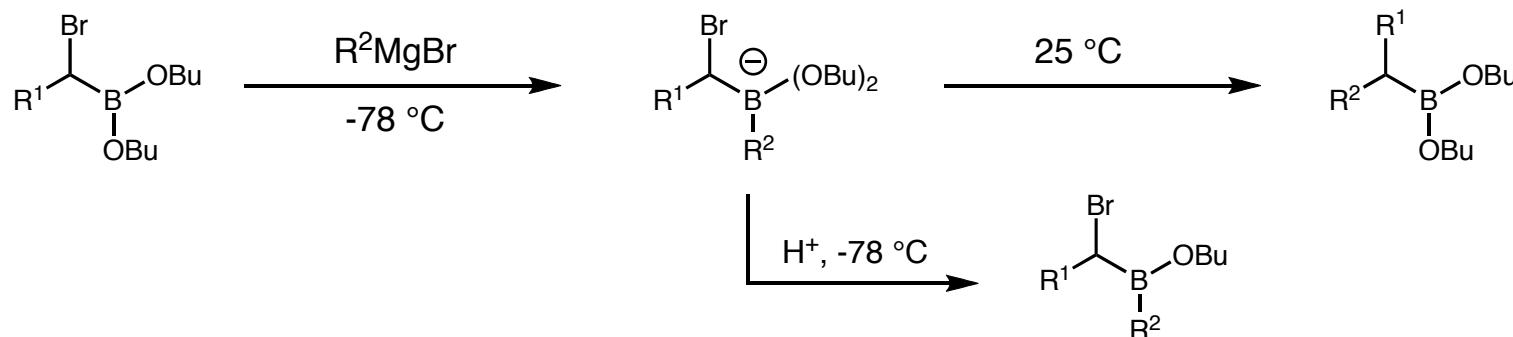
Zweifel, G.; Fisher, R. P.; Snow, J. T.; Whitney, C. C. *J. Am. Chem. Soc.* **1972**, *94*, 6560-6571.

- Hydroboration, alkyl migration and B-X elimination are all stereospecific

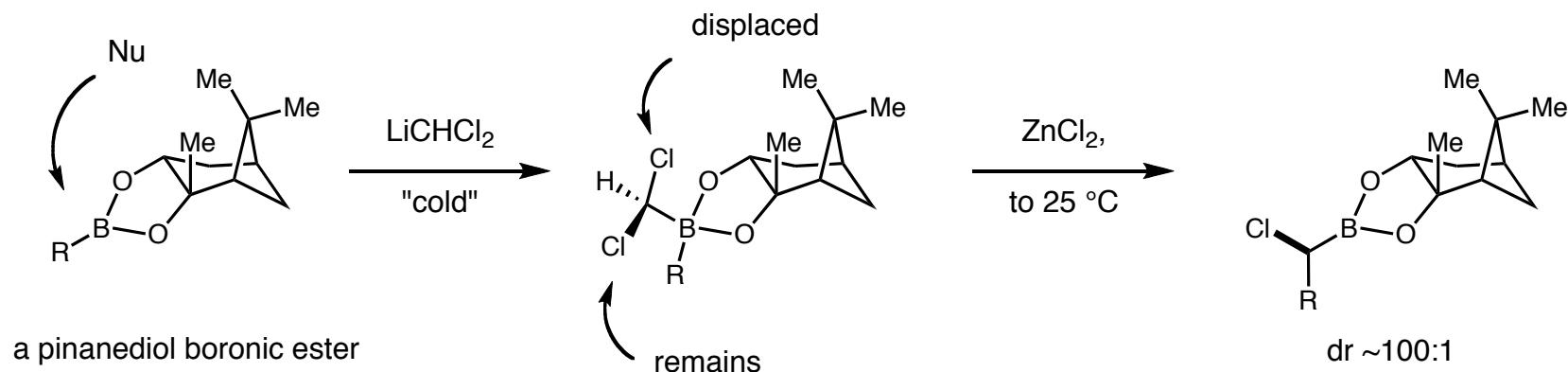


The Matteson Alkylation

- Addition of alkyl lithium and grignard reagents to α -halo boronic esters leads to net displacement



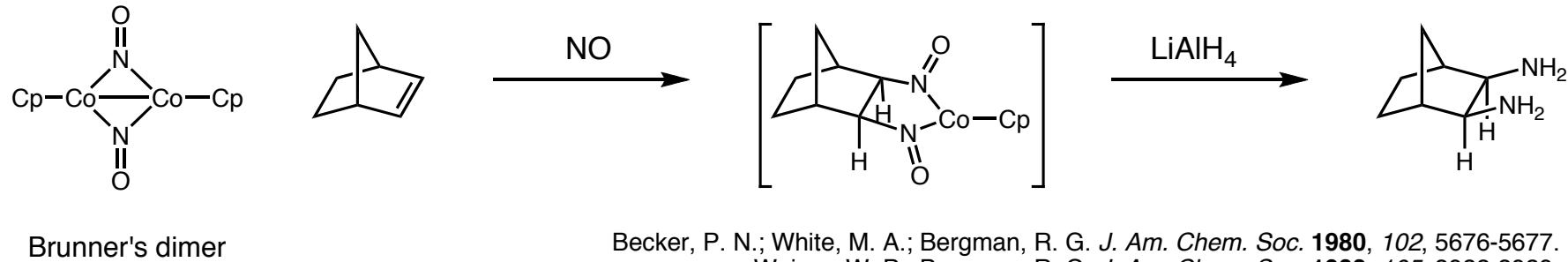
- Very high levels ($>1000:1$) of diastereoselection can be achieved with chiral boronic esters



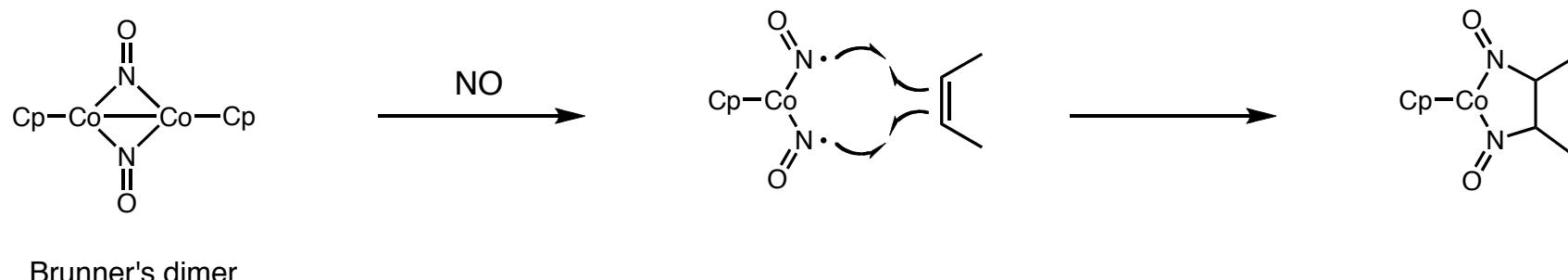
Matteson, D. S. *Tetrahedron* **1998**, 54, 10555-10607.

Bergman Chemistry Involving $(\text{CoCpNO})_2$

- So much CO chemistry (hydroformylation) is known, why is there so little NO chemistry?



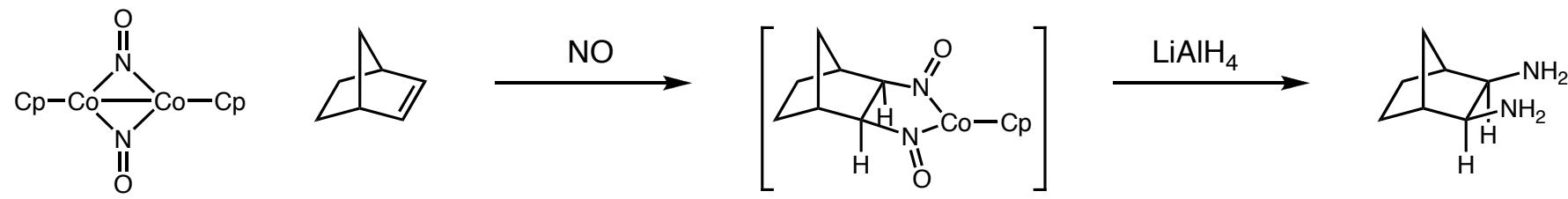
- The reaction is proposed to proceed via $\text{CpCo}(\text{NO})_2$



- An interesting case of a metal's ligands interacting with an olefin, rather than the metal itself (another example is osmium tetroxide)

Bergman Chemistry Involving $(\text{CoCpNO})_2$

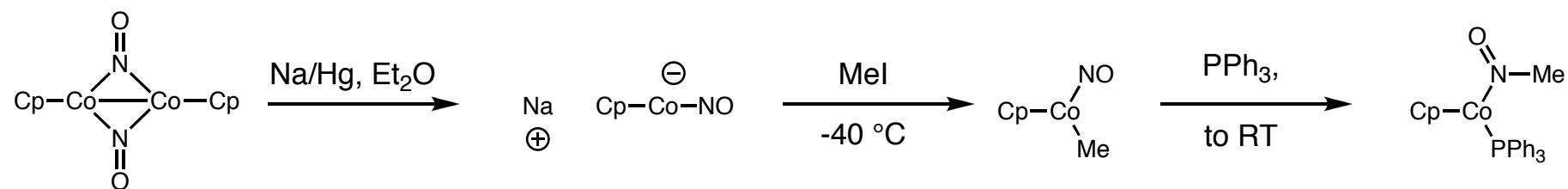
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Brunner's dimer

Becker, P. N.; White, M. A.; Bergman, R. G. *J. Am. Chem. Soc.* **1980**, *102*, 5676-5677.
 Weiner, W. P.; Bergman, R. G. *J. Am. Chem. Soc.* **1983**, *105*, 3922-3929.
 Becker, P. N.; Bergman, R. G. *J. Am. Chem. Soc.* **1983**, *105*, 2985-2995.
 Becker, P. N.; Bergman, R. G. *Organometallics* **1983**, *2*, 787-796.
 Schomaker, J. M.; Boyd, W. C.; Stewart, I. C.; Toste, F. D.; Bergman, R. G. *J. Am. Chem. Soc.* **2008**, *130*, 3777-3779.

- NO can undergo migratory insertion analogous to CO



Bergman Chemistry Involving $(\text{CoCpNO})_2$

■ Recently, the groups of Bergman and Toste looked at some new chemistry involving nitrosoalkanes

