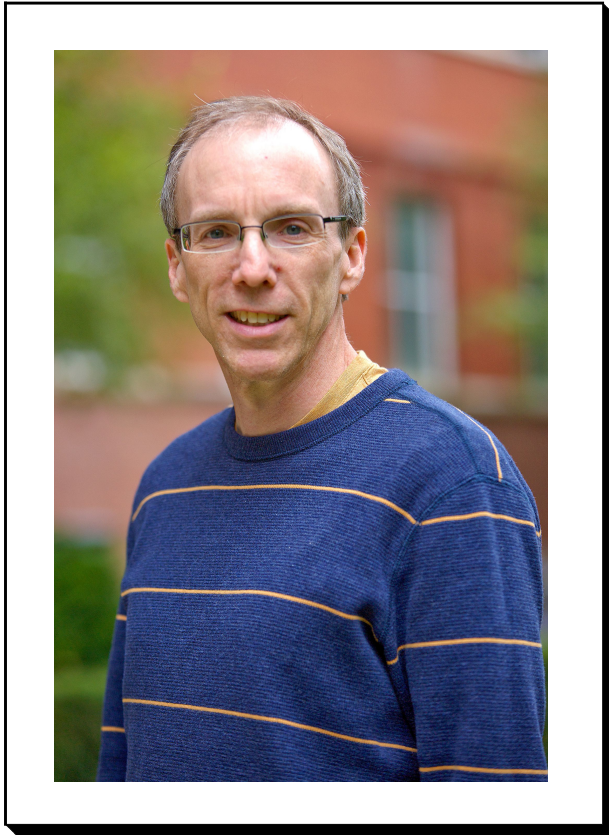


Highlights from the Career of Jeffrey S. Moore



**Group Meeting
January 31th, 2013**

Christopher R. Jamison



Jeffrey Moore:
Professor of Chemistry and Materials Science

PhD – UIUC, 1989. Advisor: Samuel Stupp

Post Doc. – Caltech. Advisor: Bob Grubbs

Assistant Prof. – University of Michigan, 1990

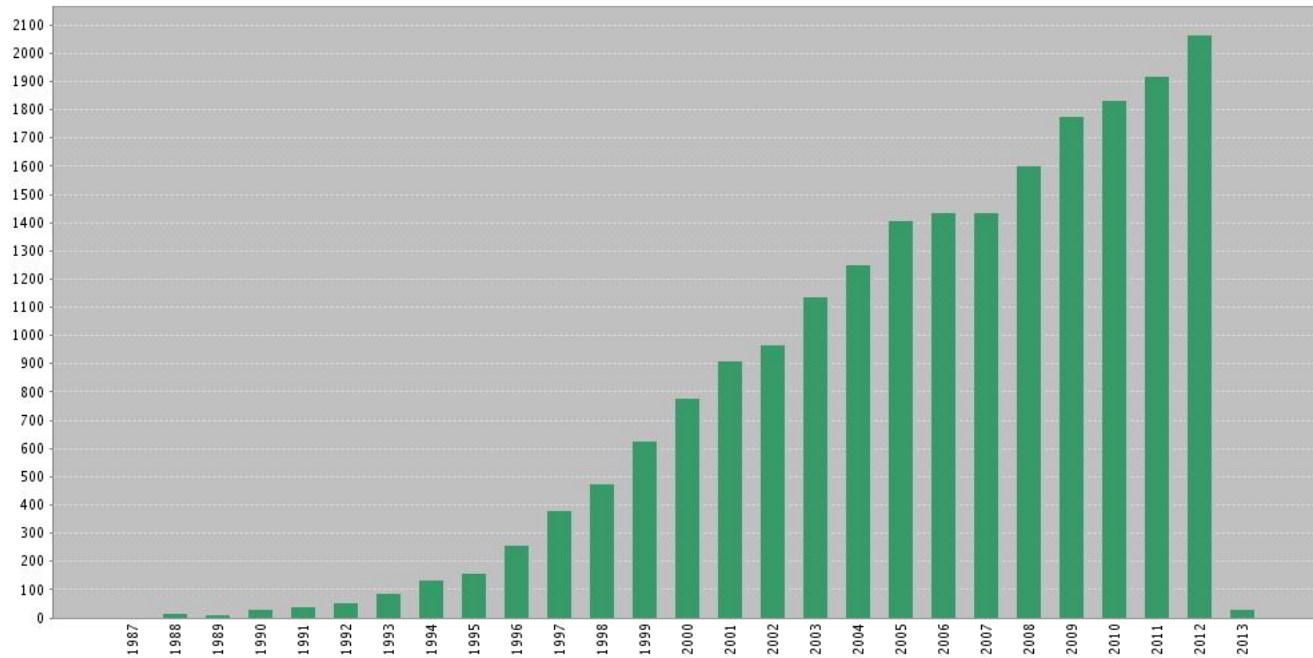
Professor – UIUC 1993 – present

Fellow AAAS – 2008

H-Index: 77



Citations Per Year



Total publications: 318
Total citations: 20,828
Citations per paper: 65.5
H-index: 77

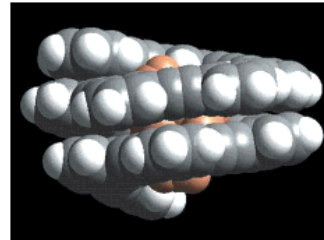
The Moore Group

"In general, our group uses the tools of synthetic and physical organic chemistry to address problems at the interface of chemistry and materials science."

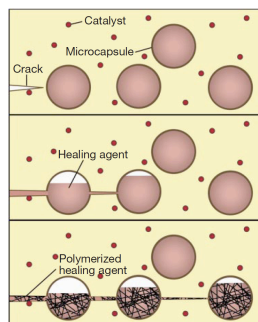


Areas of Research

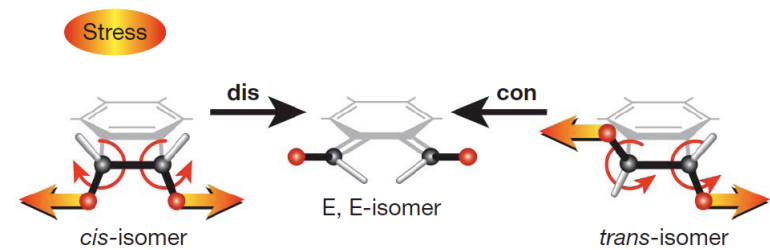
New Macromolecular Architectures



Self-Healing Materials

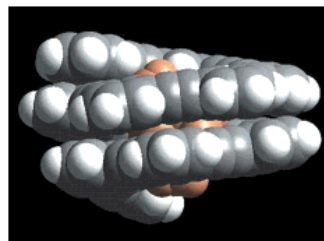


Mechanochemistry



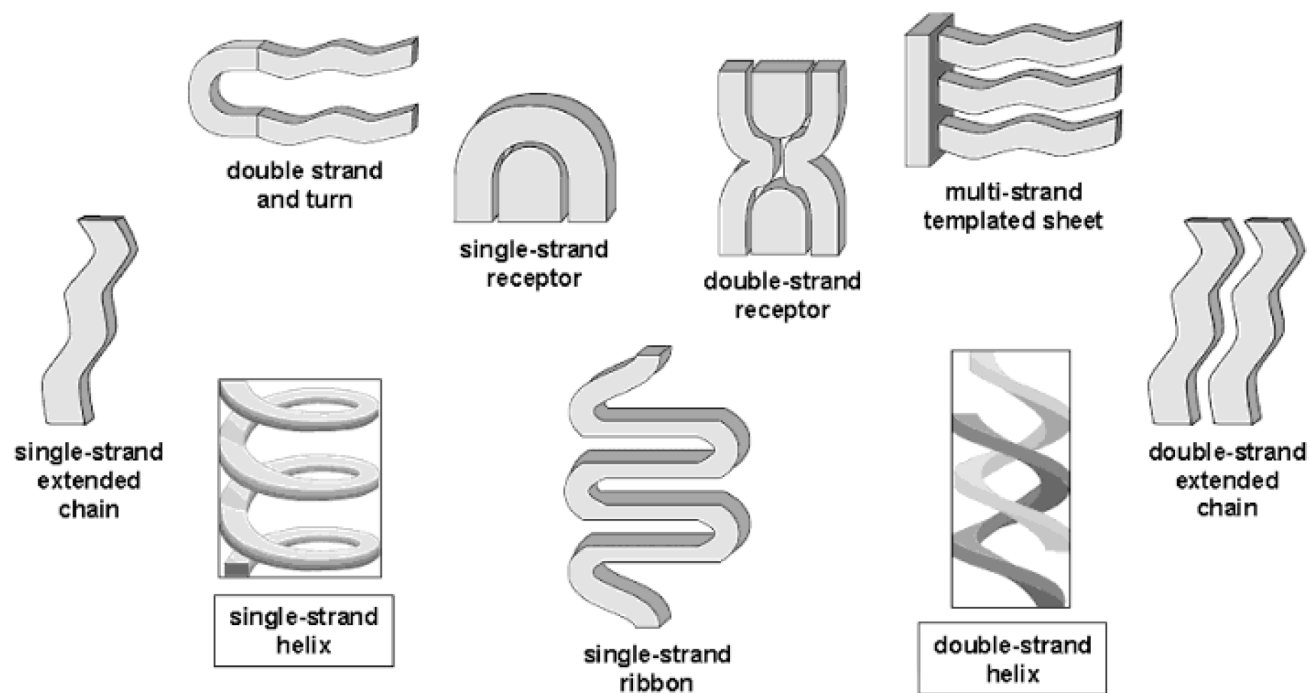
Areas of Research

New Macromolecular Architectures



New Macromolecular Architectures

Non-Biological Foldamers

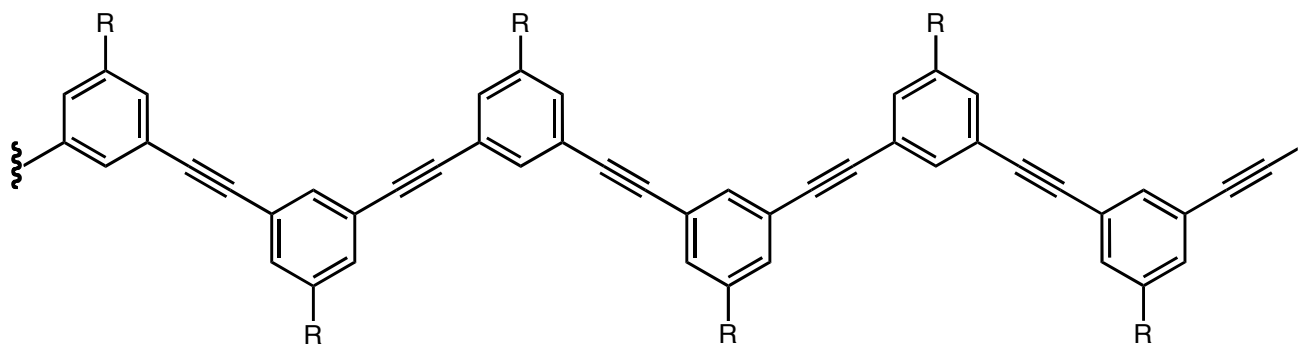


Foldamer

Any oligomer that folds into a conformationally ordered state in solution, the structures of which are stabilized by a collection of noncovalent interactions between nonadjacent monomer units.

New Macromolecular Architectures

Arylene-Ethynylene Polymers



organic semiconductors

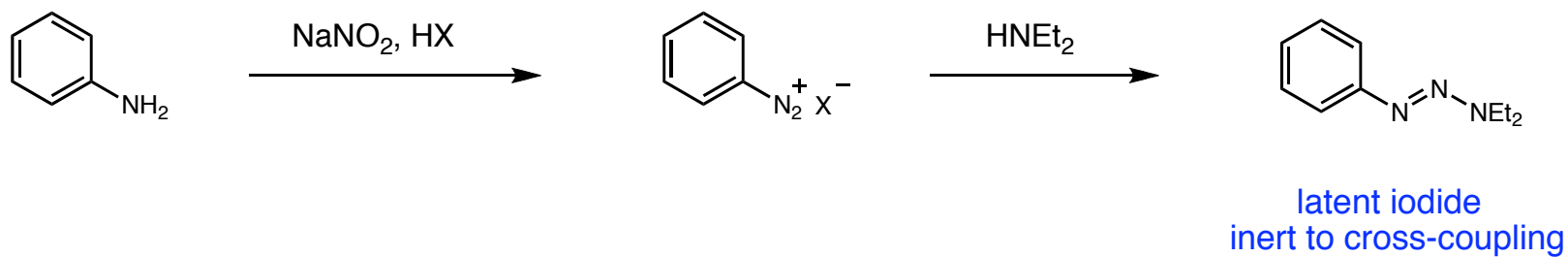
useful optical properties

propensity to self assemble by π - π interactions

restricted conformational freedom aids in predictable self-assembly

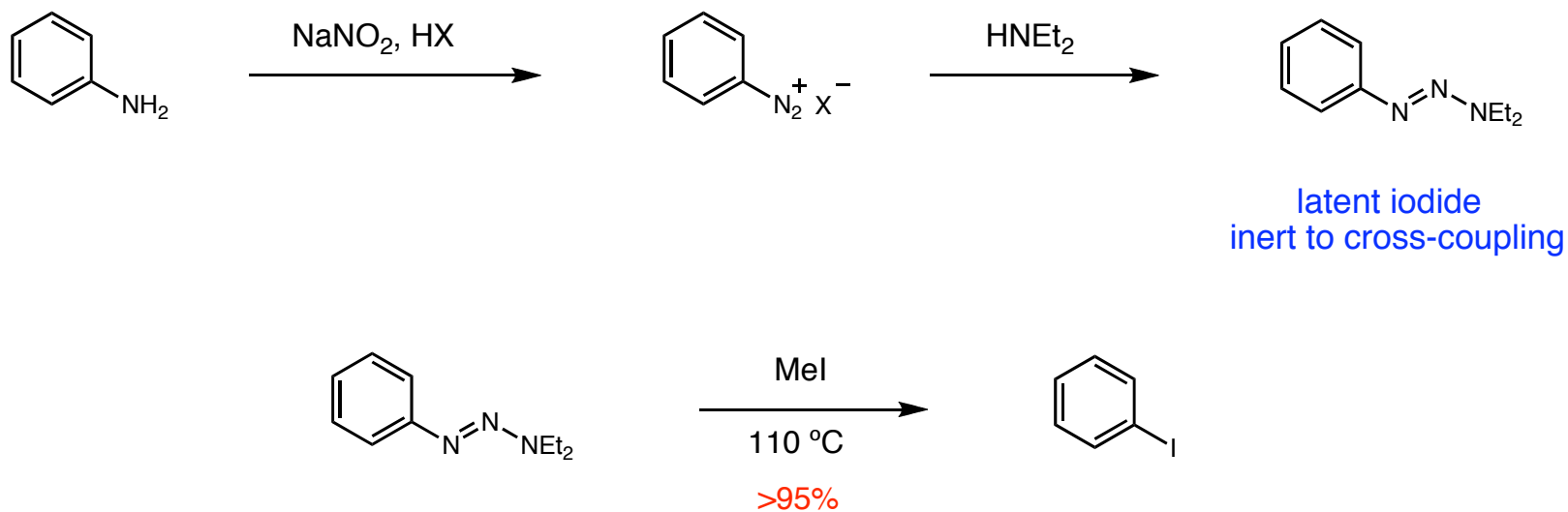
New Macromolecular Architectures

An Aryl Iodide Masking Group



New Macromolecular Architectures

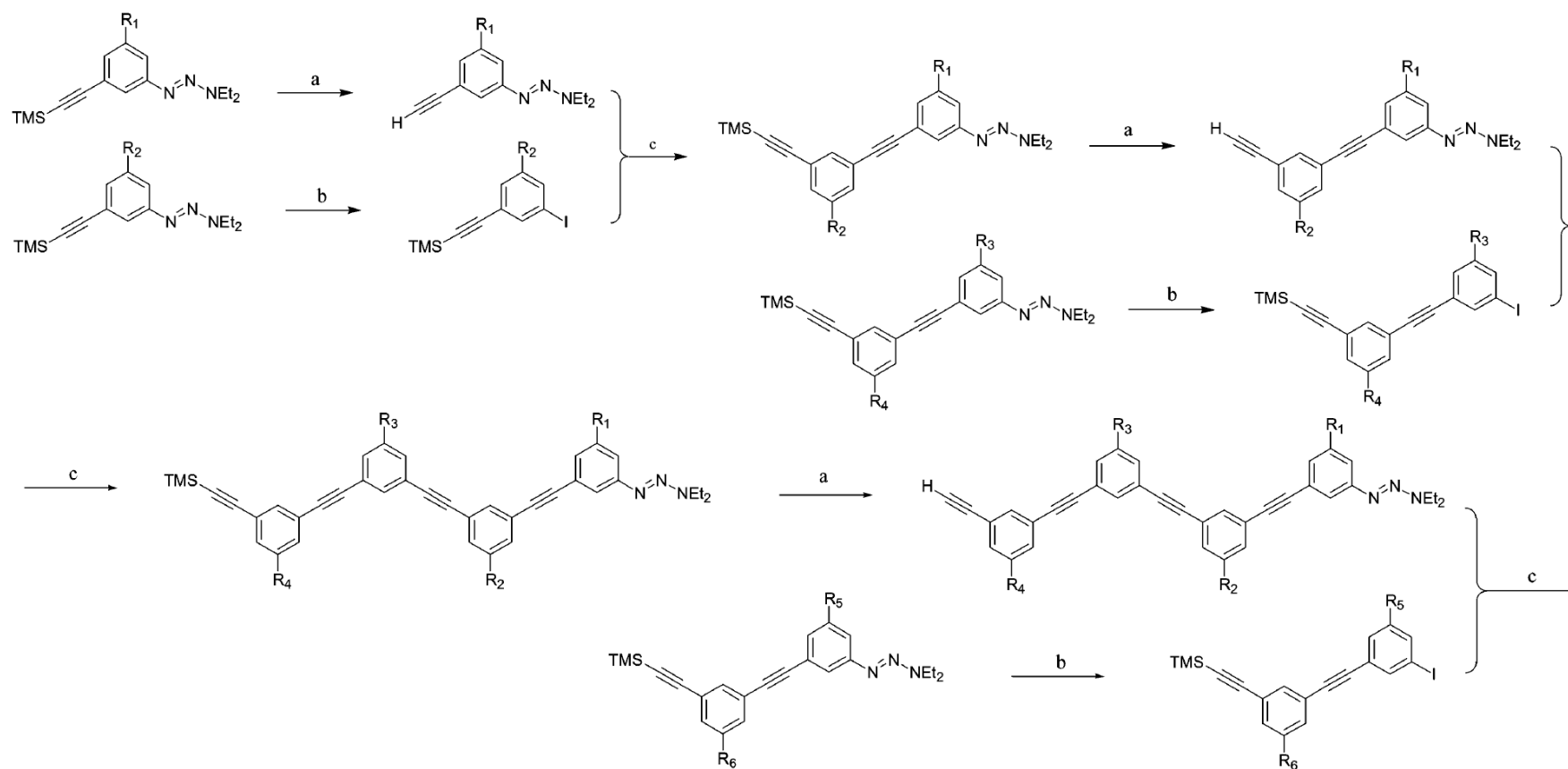
An Aryl Iodide Masking Group



Allows for controlled, step-wise synthesis
of oligomers with precise structure

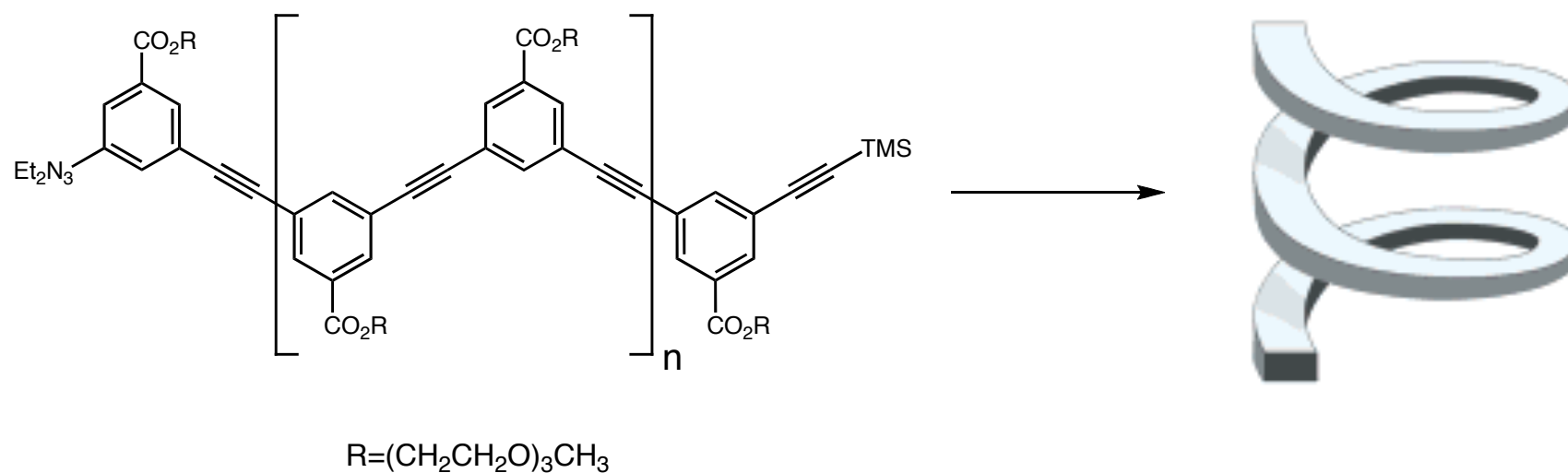
New Macromolecular Architectures

Synthesis of Arylene-Ethynylene Oligomers



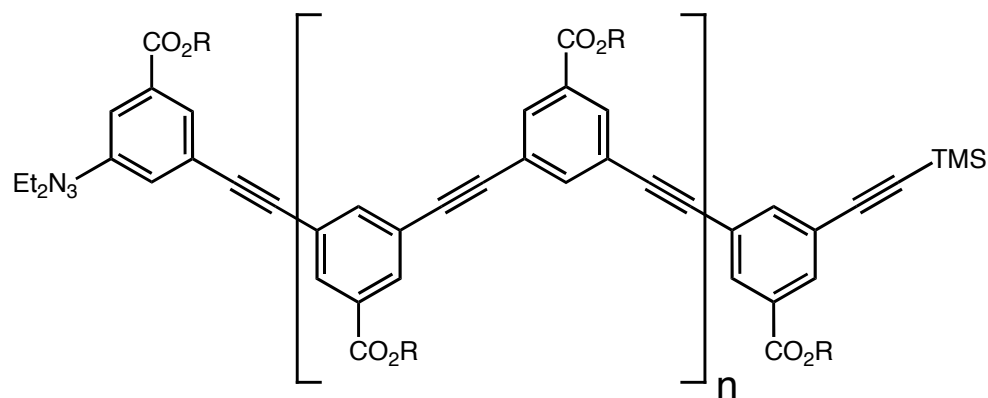
New Macromolecular Architectures

Folding of Arylene-Ethynylene Oligomers into Helices



New Macromolecular Architectures

Folding of Arylene-Ethynylene Oligomers into Helices

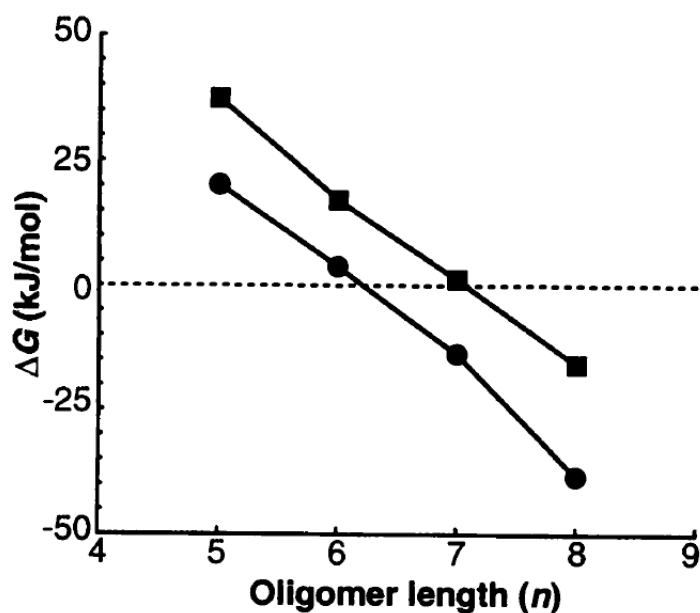


Interested in generating secondary structure without hydrogen bonding to gauge solvophobic contributions

A step toward synthetic polymers with folded architectures on par with biological macromolecules

New Macromolecular Architectures

Folding of Arylene-Ethynylene Oligomers into Helices



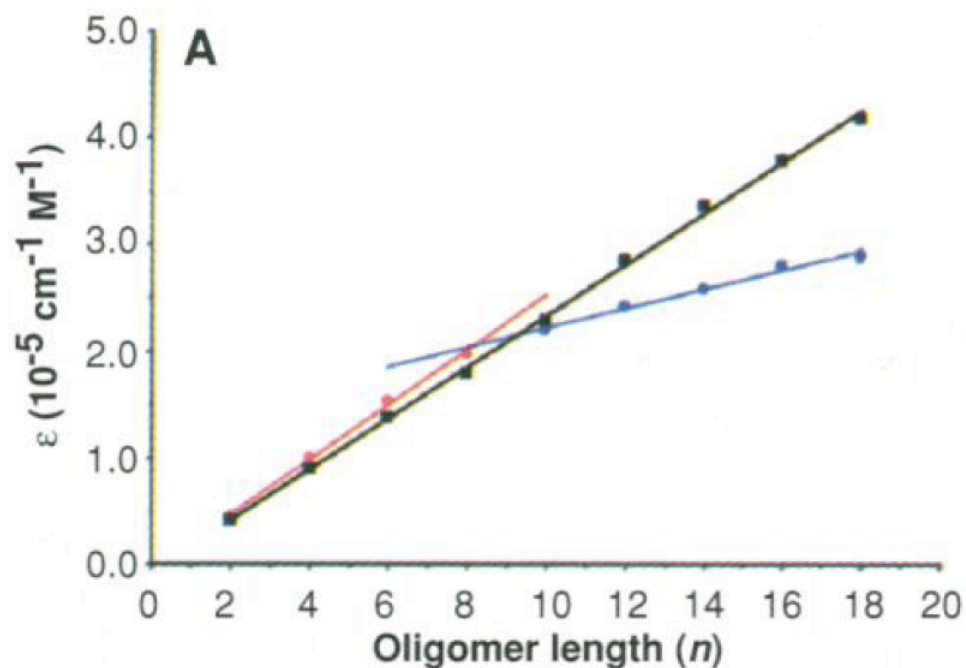
Computational studies predict stable helices are possible for oligomers of at least 8 units

Fig. 2. Estimated free energy of helix formation ΔG versus chain length

New Macromolecular Architectures

Folding of Arylene-Ethynylene Oligomers into Helices

Fig. 3. (A) The molar extinction coefficient ϵ (303 nm) chloroform (black) and acetonitrile (red and blue)



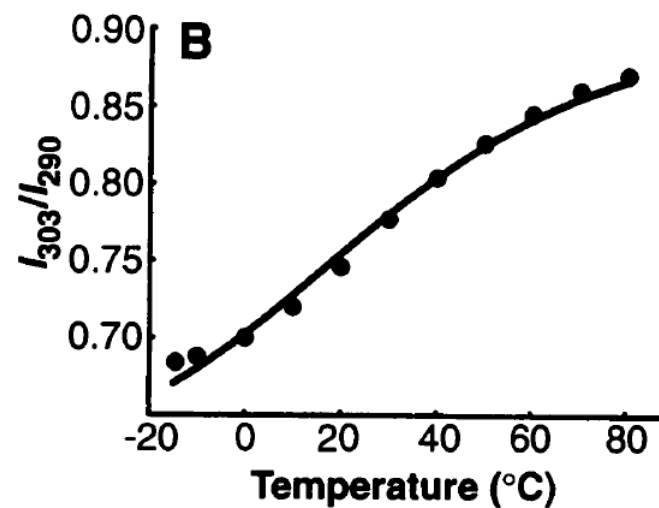
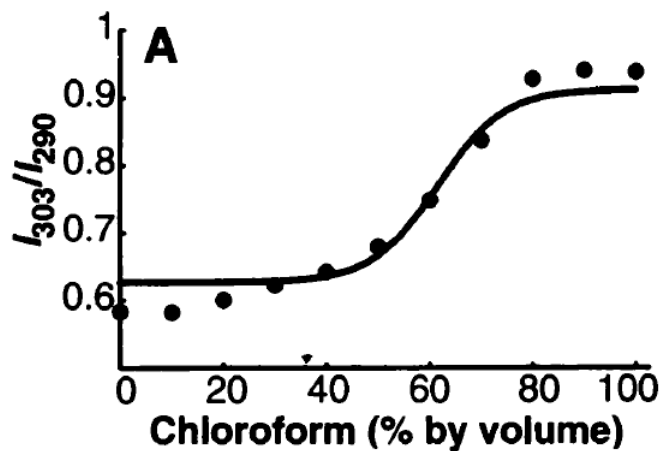
Each oligomer's ϵ was measured over a concentration range to rule out intermolecular associations

Beer-Lambert Law

$$A = \epsilon cl$$

New Macromolecular Architectures

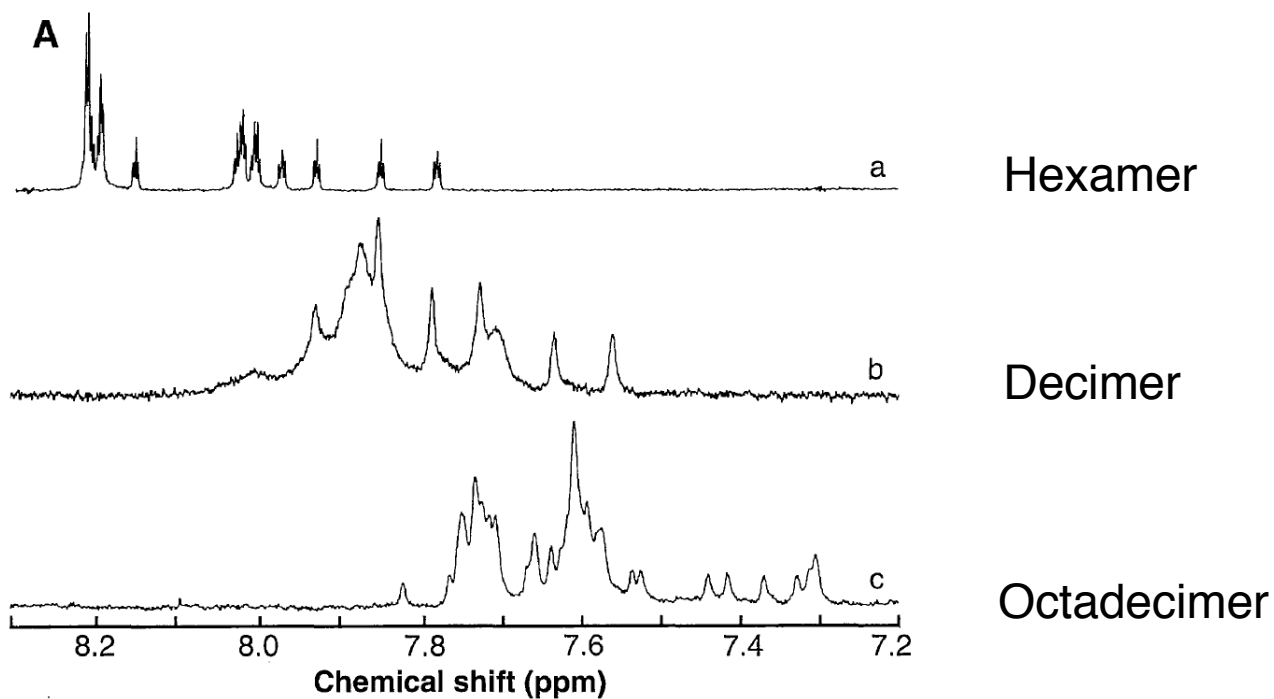
Folding of Arylene-Ethynylene Oligomers into Helices



Both heat and chloroform denature the octadecimer helix

New Macromolecular Architectures

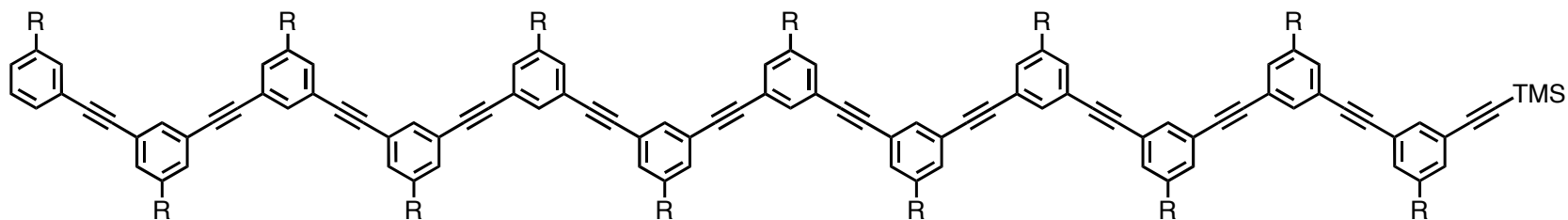
Folding of Arylene-Ethynylene Oligomers into Helices



^1H NMR in CD_3CN

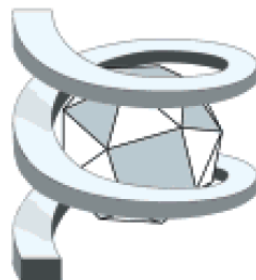
New Macromolecular Architectures

Foldamer-Based Molecular Recognition



1, R = $\text{CO}_2(\text{CH}_2\text{CH}_2\text{O})_3\text{CH}_3$

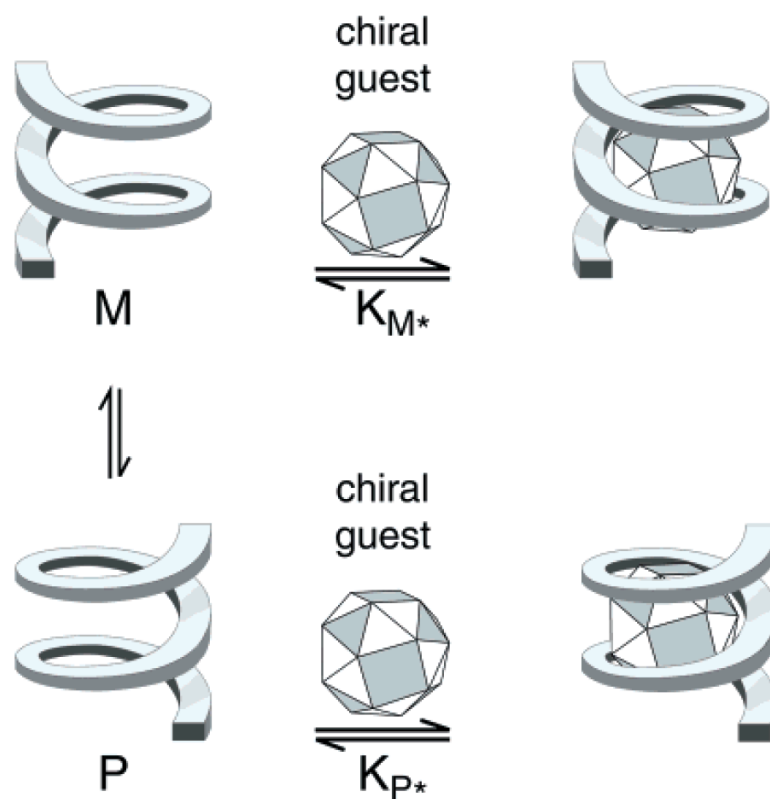
(-)- α -pinene



Prince, R.; Barnes, S.; Moore, J. S. *J. Am. Chem. Soc.* **2000**, 122, 2758

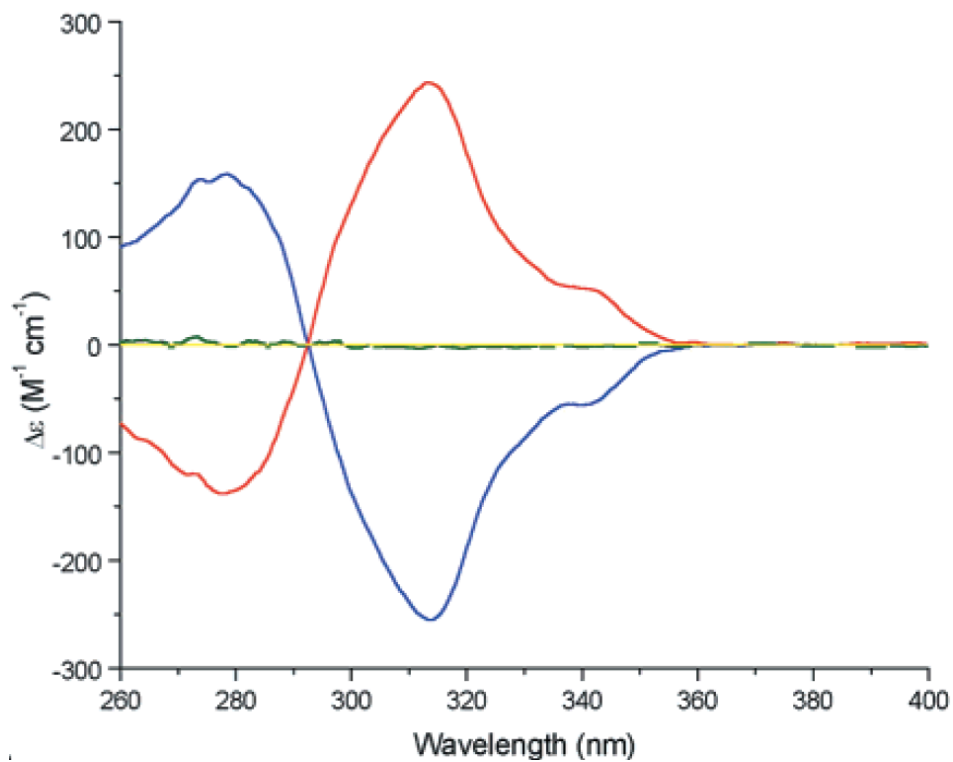
New Macromolecular Architectures

Foldamer-Based Molecular Recognition



New Macromolecular Architectures

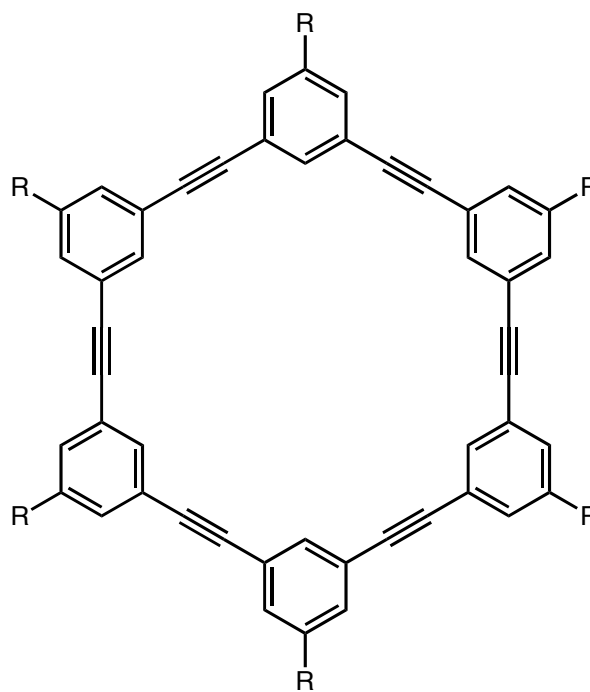
Foldamer-Based Molecular Recognition



CD spectra. (–)- α -Pinene (yellow line, 420 μM), oligomer 1 (green line) in the presence of 100 equiv of (–)- α -pinene (blue line) and 100 equiv of (+)- α -pinene (red line).

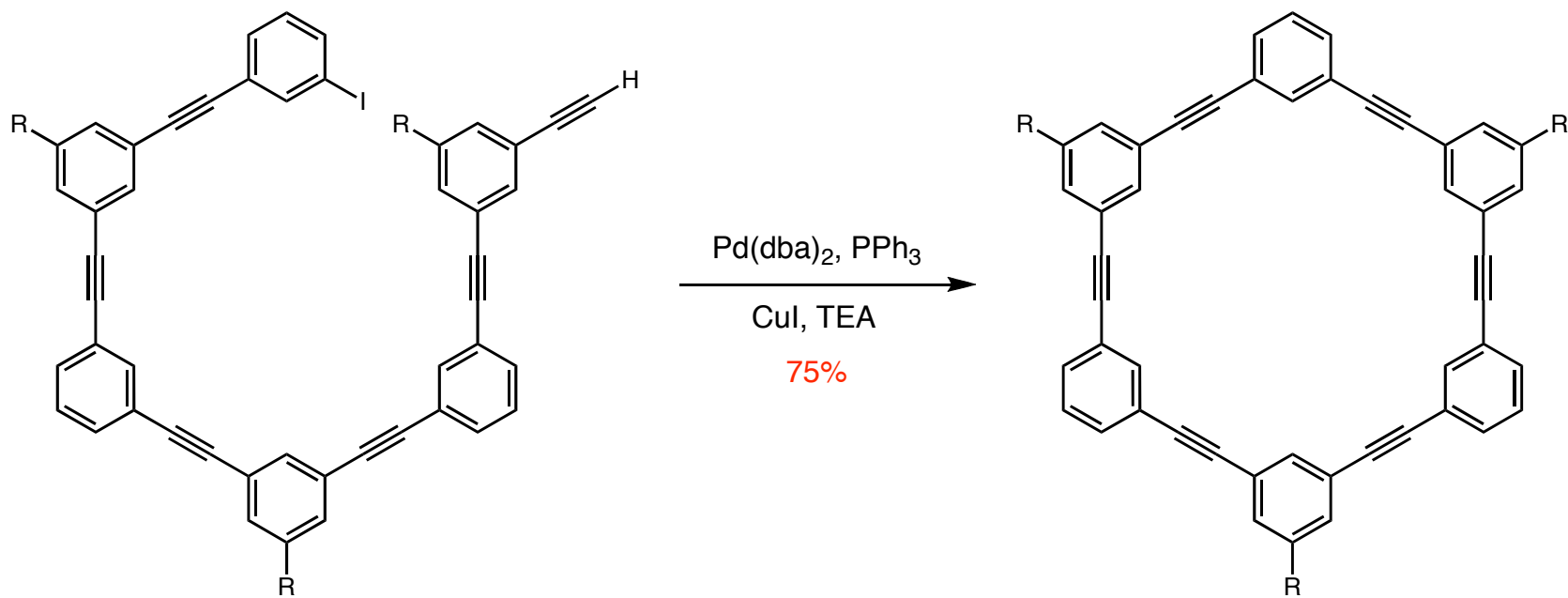
New Macromolecular Architectures

Shape-Persistent Macrocycles



New Macromolecular Architectures

Synthesis of Shape-Persistent Macrocycles

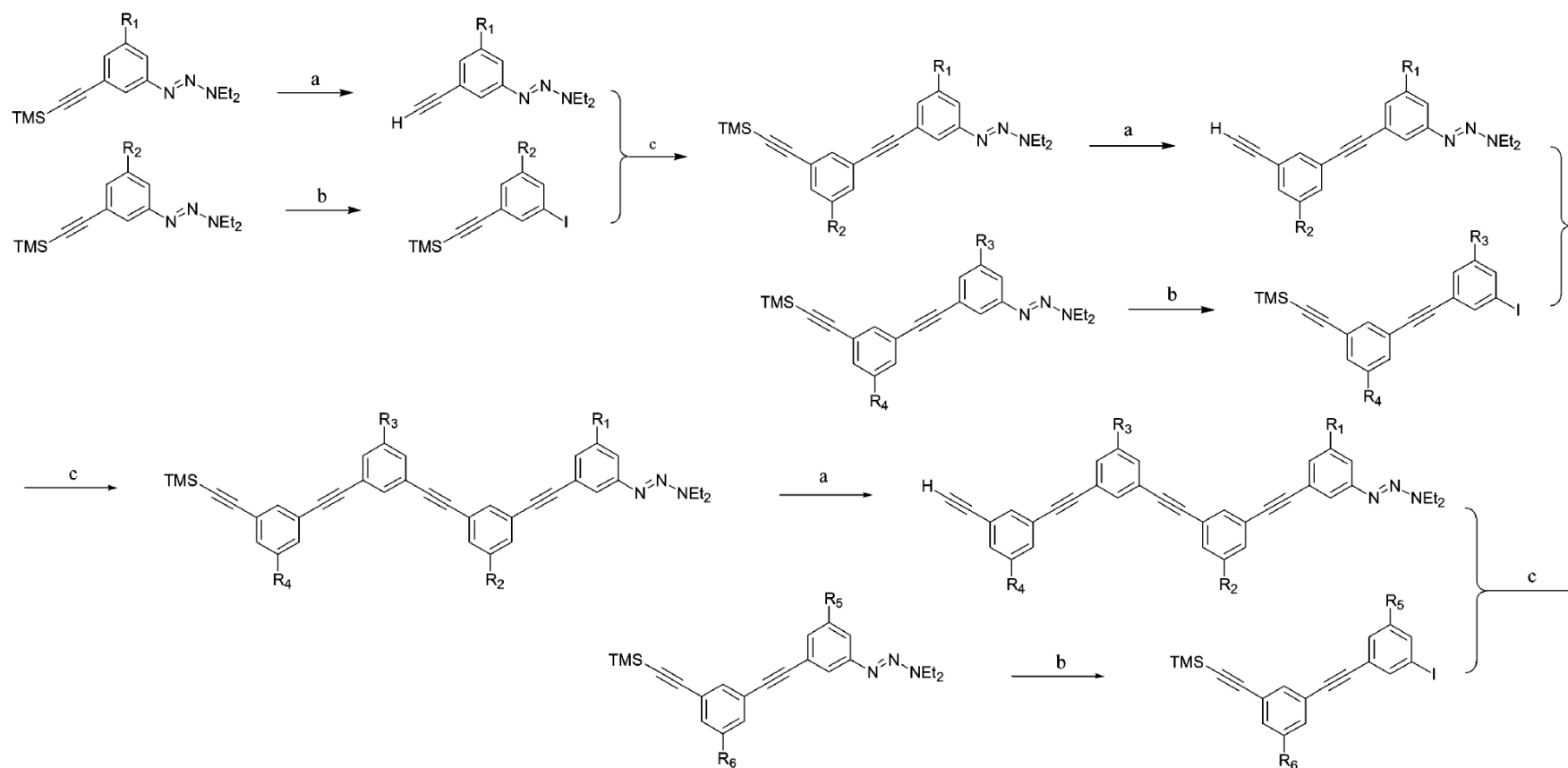


syringe-pump addition

1. Zhang, J.; Moore, J. S. *ACIEE.*, **1992**, 31, 922-923
2. Zhao, D.; Moore, J. S. *Chem. Comm.*, **2003**, 807-818

New Macromolecular Architectures

Synthesis of Shape-Persistent Macrocycles

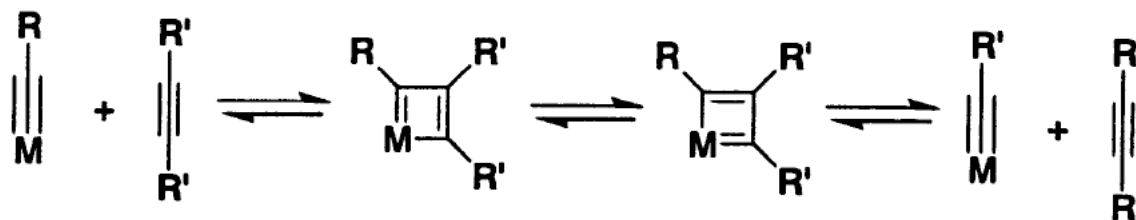


1. Zhang, W.; Moore, J. S. *ACIEE.*, **2006**, 45, 4416-4439
2. Zhao, D.; Moore, J. S. *Chem. Comm.*, **2003**, 807-818

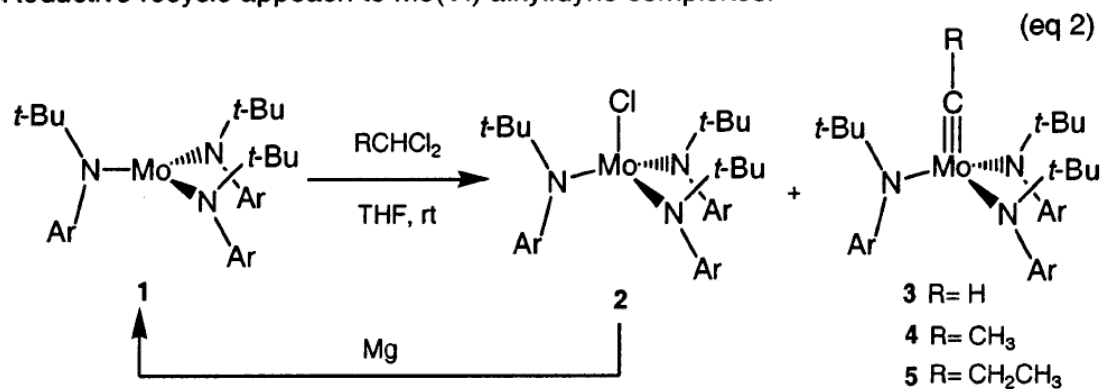
New Macromolecular Architectures

Synthesis of Shape-Persistent Macrocycles

Alkylidyne Mechanism of Alkyne Metathesis

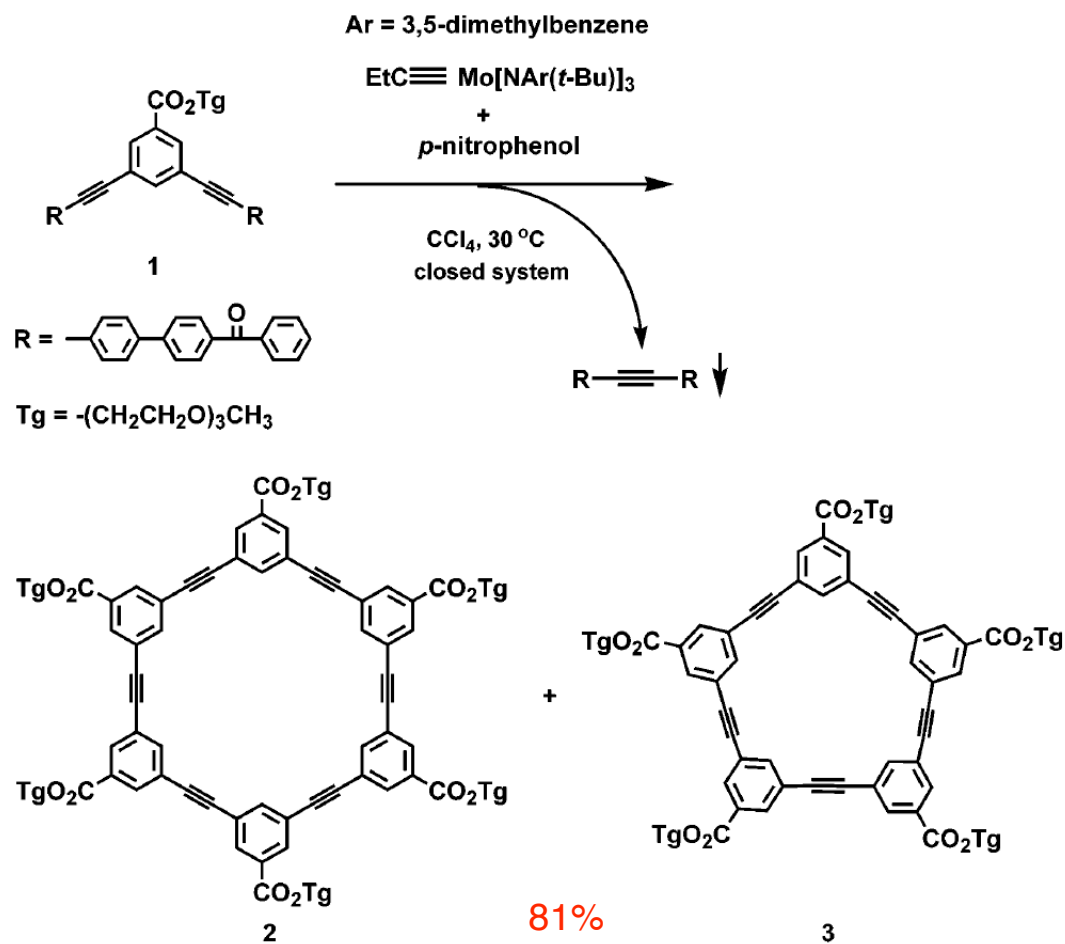


Reductive recycle approach to Mo(VI) alkylidyne complexes.



New Macromolecular Architectures

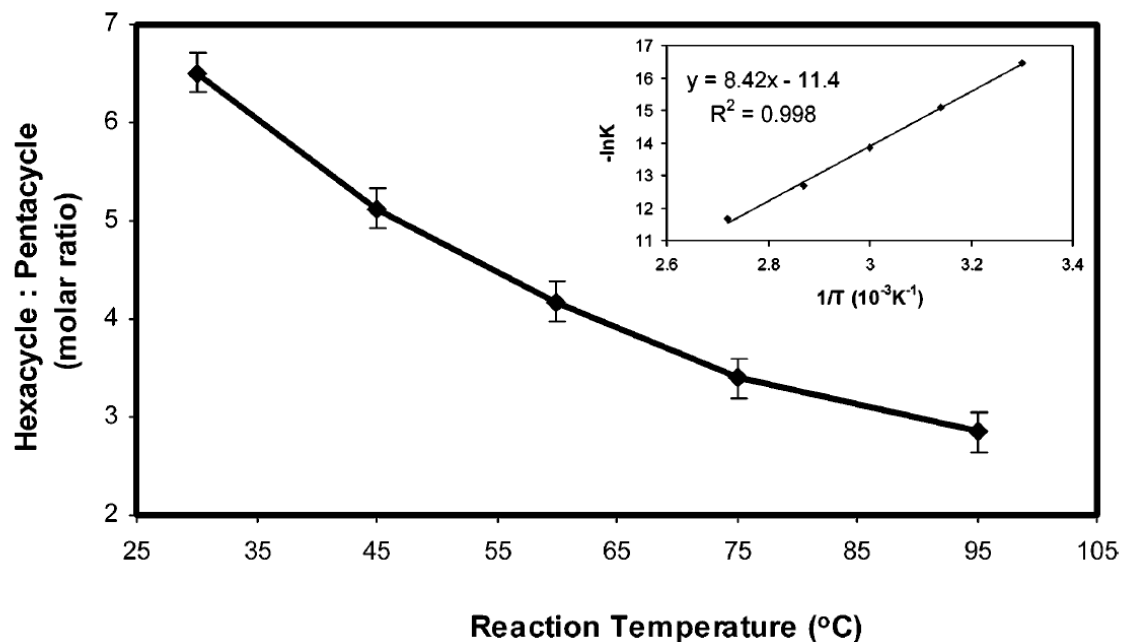
Synthesis of Shape-Persistent Macrocycles



Zhang, W.; Moore, J. S. *J. Am. Chem. Soc.* **2005**, 127, 11863-11867

New Macromolecular Architectures

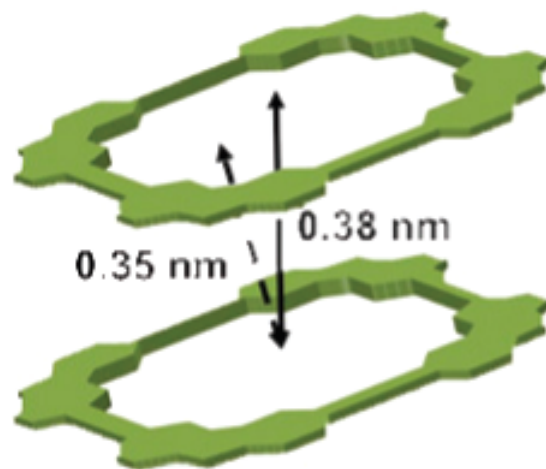
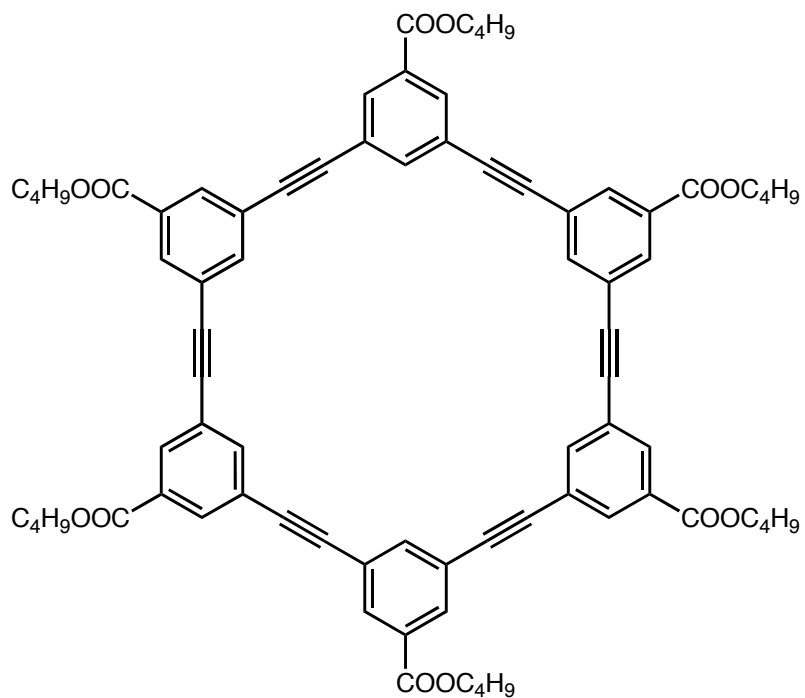
Synthesis of Shape-Persistent Macrocycles



The pentamer is entropically favored despite its higher ring strain and is a larger component of the equilibrium at higher temperatures. ie $\Delta G = \Delta H - T\Delta S$

New Macromolecular Architectures

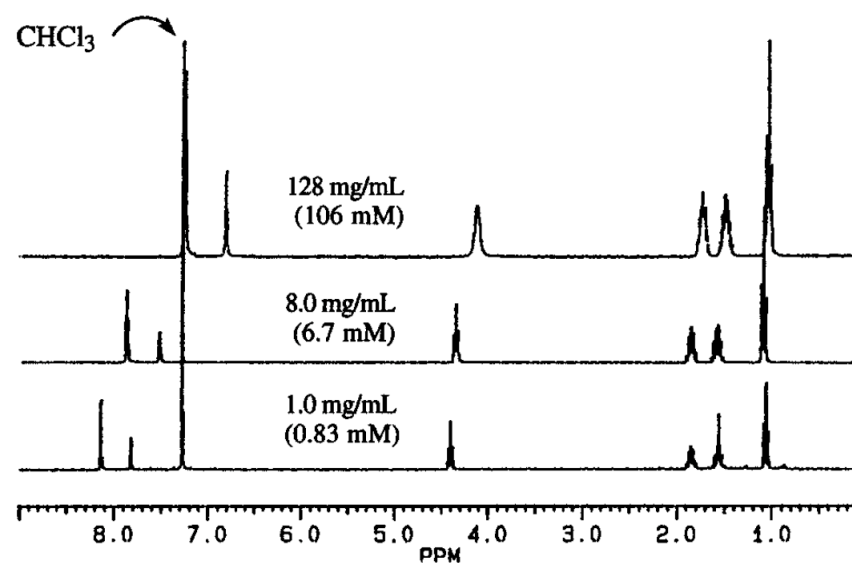
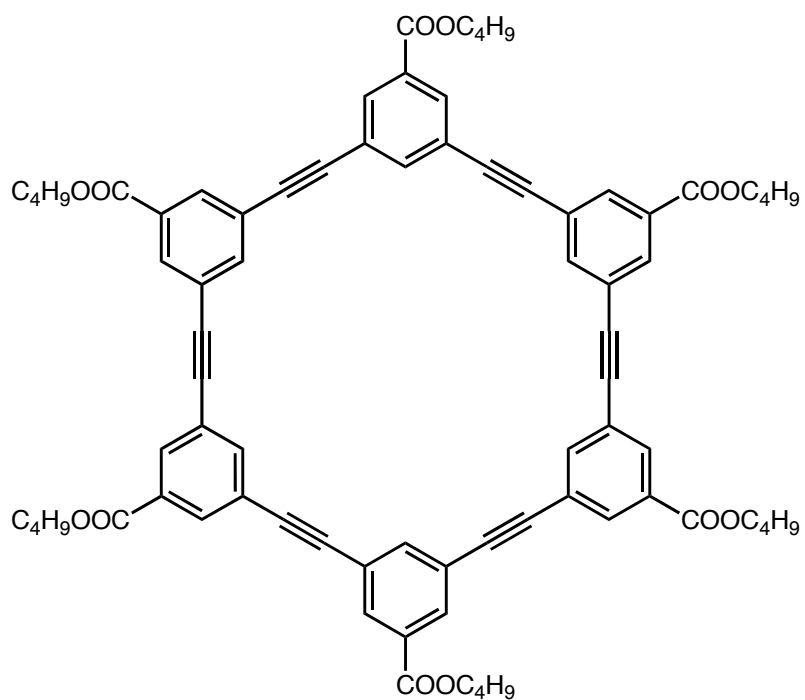
Supramolecular Organization of Arylene-Ethynylene Macrocycles



1. Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1992**, 114, 9701-9702
2. Shetty, S.; Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1996**, 118, 1019

New Macromolecular Architectures

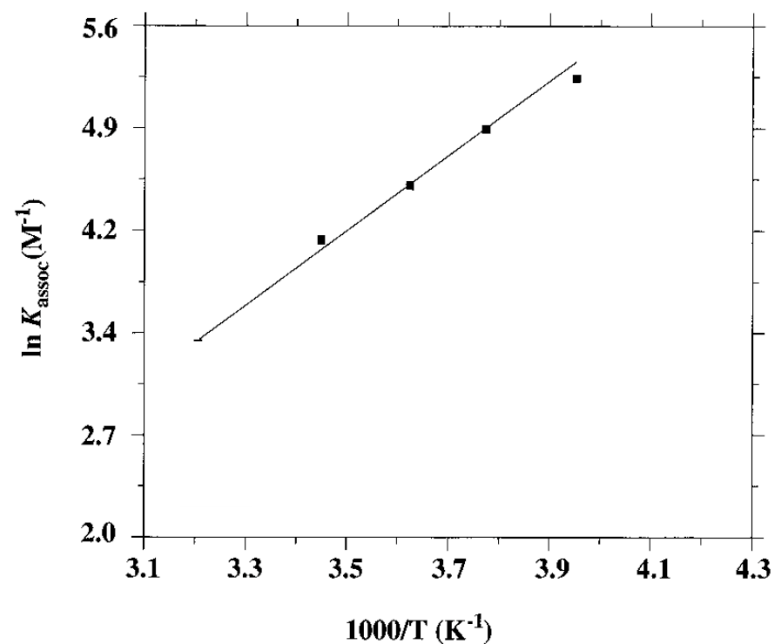
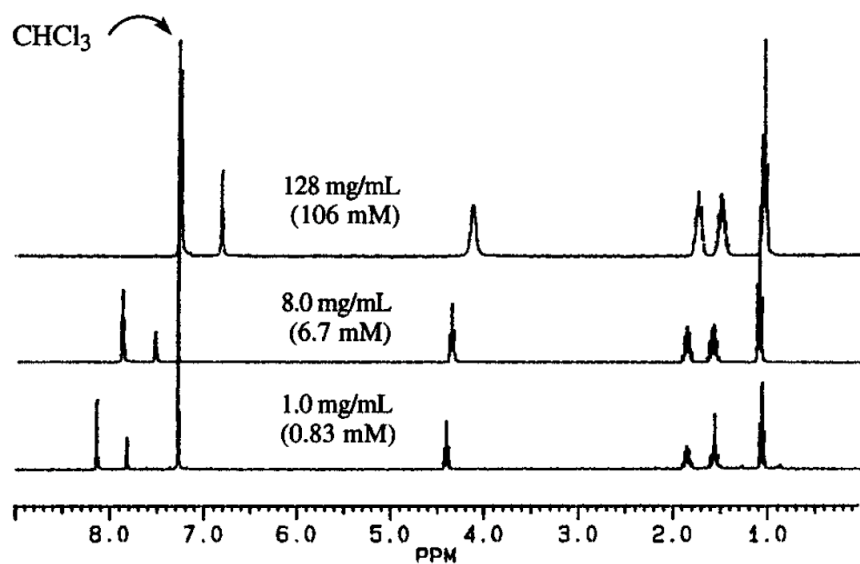
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New Macromolecular Architectures

Supramolecular Organization of Arylene-Ethynylene Macrocycles



van't Hoff plots of self-association constants (K_{assoc})

$$\ln K = -\Delta H/T + \Delta S/R$$

1. Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1992**, 114, 9701-9702
2. Shetty, S.; Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1996**, 118, 1019

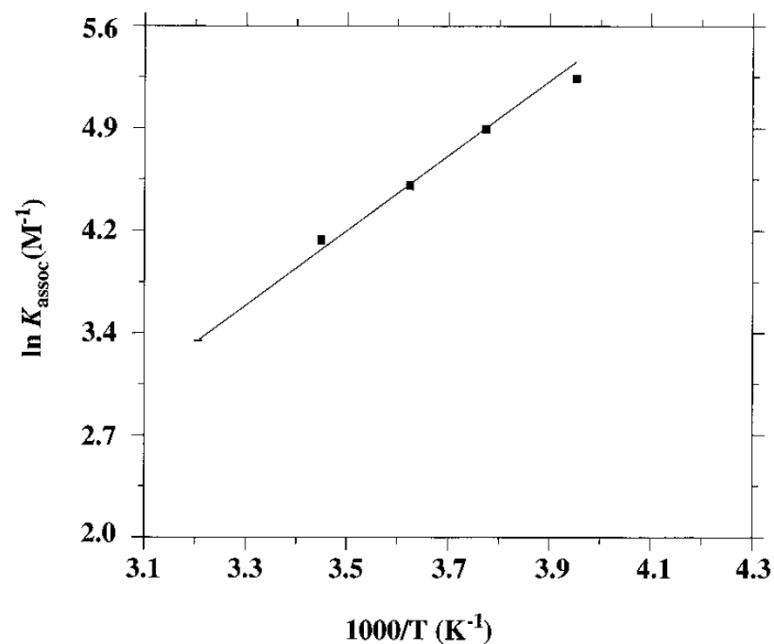
New Macromolecular Architectures

Supramolecular Organization of Arylene-Ethynylene Macrocycles

Thermodynamic Data for Self-Association

K_{assoc} (M^{-1}) ^a	ΔG (kcal/mol) ^a	ΔH (kcal/mol)	ΔS (cal/mol·K)
60	-2.4	-5.0 ± 0.2	-9.2 ± 0.8

Assuming that dimerization is the only significant aggregation state (confirmed by VPO).



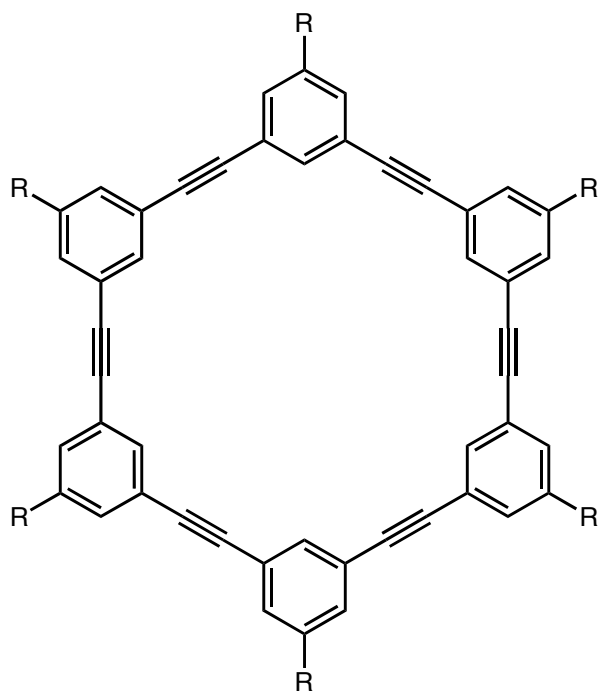
van't Hoff plots of self-association constants (K_{assoc})

$$\ln K = -\Delta H/T + \Delta S/R$$

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2. Shetty, S.; Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1996**, 118, 1019

New Macromolecular Architectures

π -Stacking Studies



A model system for studying π - π interactions

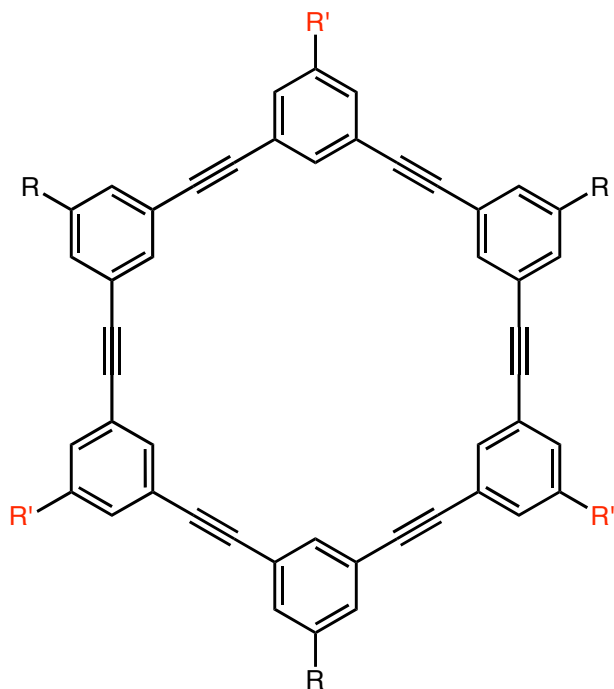
Stacking results in minimal entropic penalty and large enthalpic gain

π - π interactions should be amplified

1. Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1992**, 114, 9701-9702
2. Shetty, S.; Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1996**, 118, 1019

New Macromolecular Architectures

π -Stacking Studies



1 R=R'=CO₂C₄H₉

2 R=R'=OC₄H₉

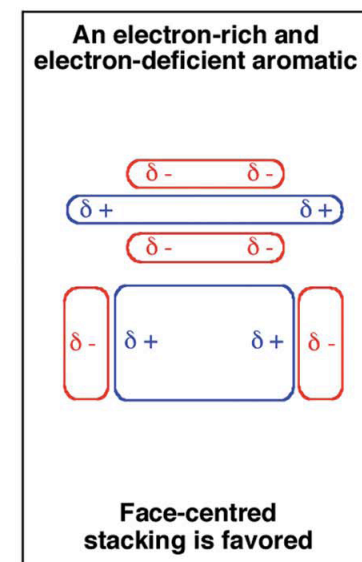
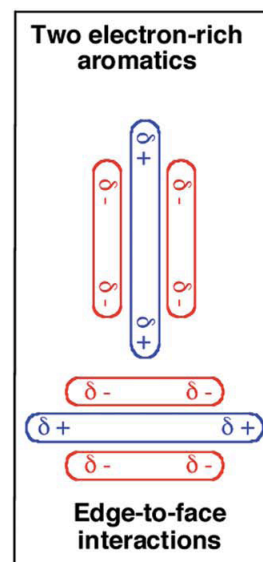
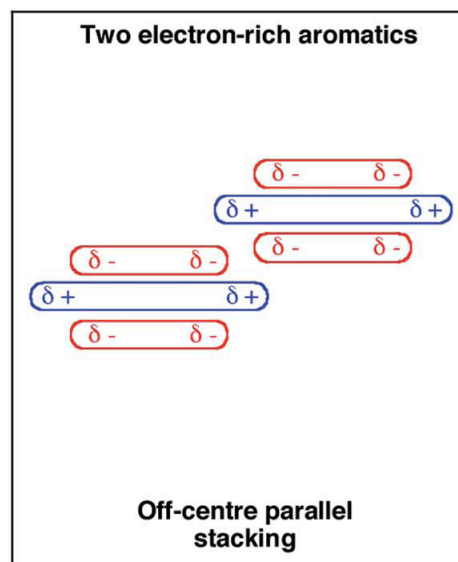
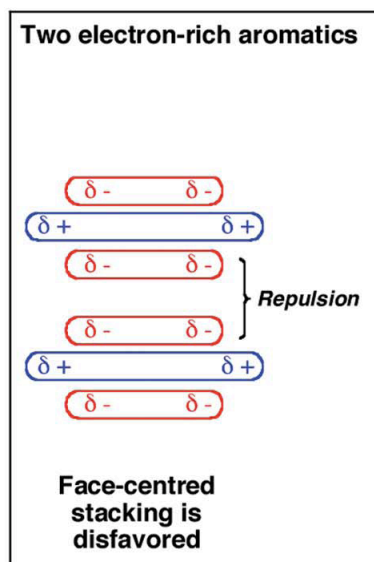
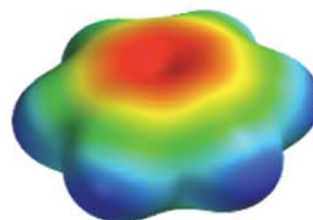
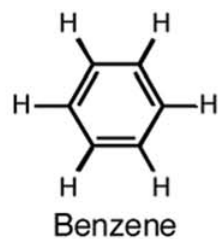
3 R=OC₄H₉, R'=CO₂C₄H₉

cmpd	K _{assoc} (M ⁻¹)	Δ G (kcal/mol)	Δ H (kcal/mol)	Δ S (kcal/mol)
1	60	-2.4	-5.0	-9.2
2	0	---	---	---
3	18	-1.7	-5.1	-13.6

1. Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1992**, 114, 9701-9702
2. Shetty, S.; Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1996**, 118, 1019

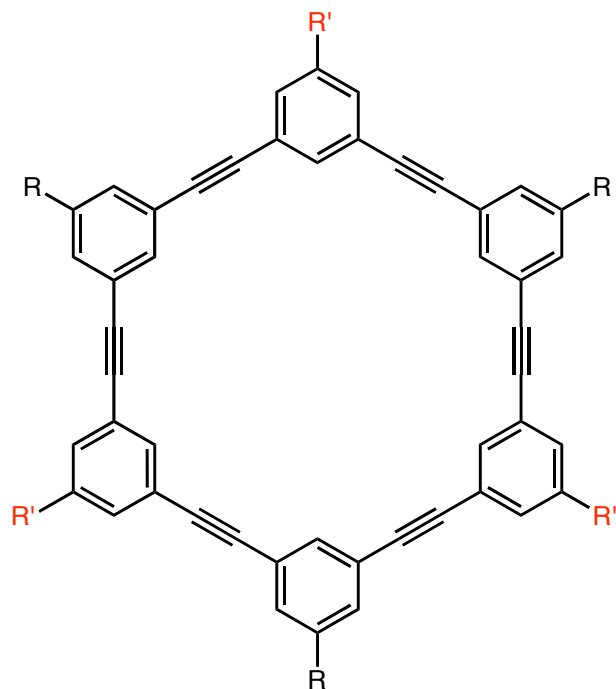
New Macromolecular Architectures

π -Stacking Studies



New Macromolecular Architectures

π -Stacking Studies



- 1 R=R'=CO₂C₄H₉
- 2 R=R'=OC₄H₉
- 3 R=OC₄H₉, R'=CO₂C₄H₉

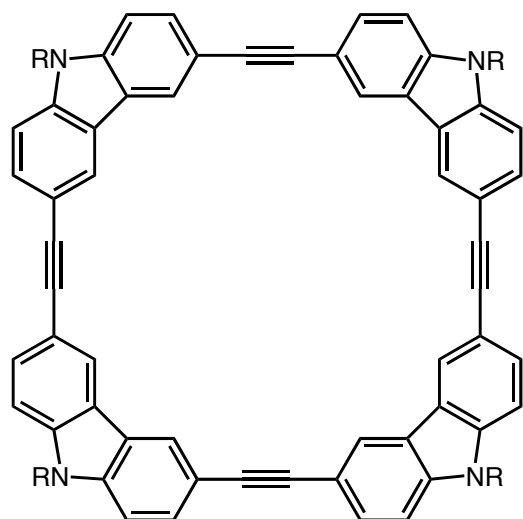
cmpd	K _{assoc} (M ⁻¹)	Δ G (kcal/mol)	Δ H (kcal/mol)	Δ S (kcal/mol)
1	60	-2.4	-5.0	-9.2
2	0	---	---	---
3	18	-1.7	-5.1	-13.6

Face-centered stacking is positively affected by van der Waals interactions and negatively affected by quadrupole moment.

1. Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1992**, 114, 9701-9702
2. Shetty, S.; Zhang, J.; Moore, J. S. *J. Am. Chem. Soc.* **1996**, 118, 1019

New Macromolecular Architectures

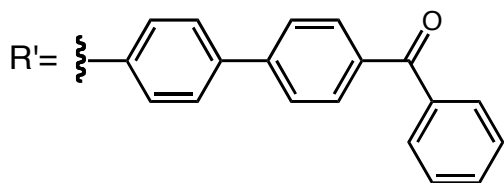
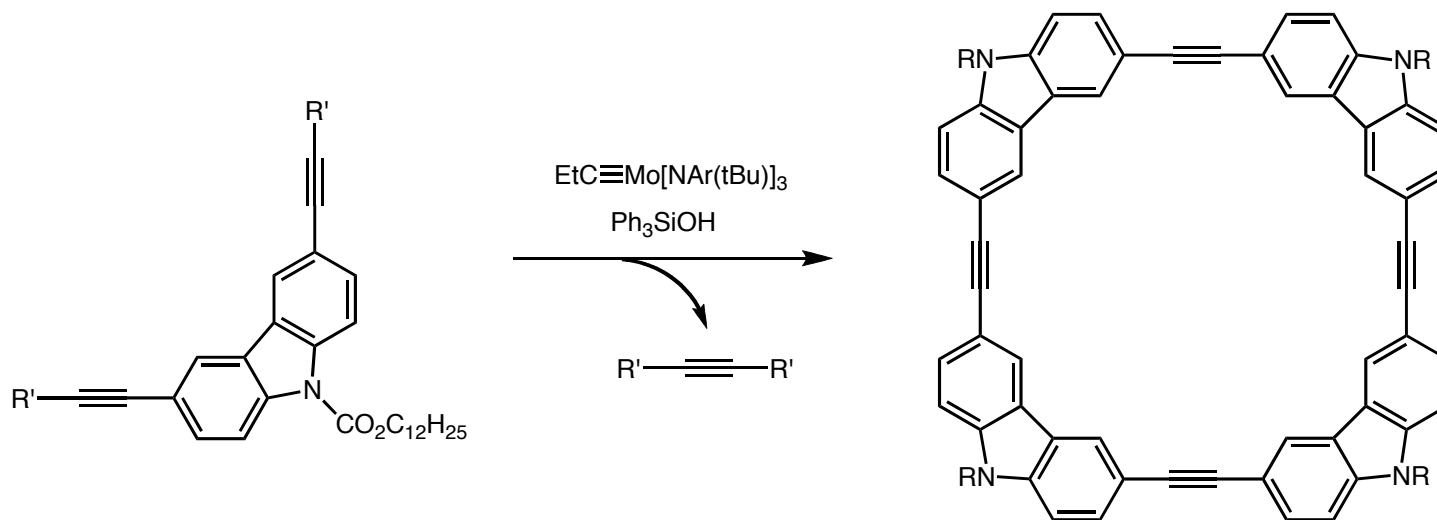
Detection of Explosives with a Nanofibril Film



Moore's group has utilized the supramolecular organization of related carbazole-macrocycles to design materials capable of detecting explosives

New Macromolecular Architectures

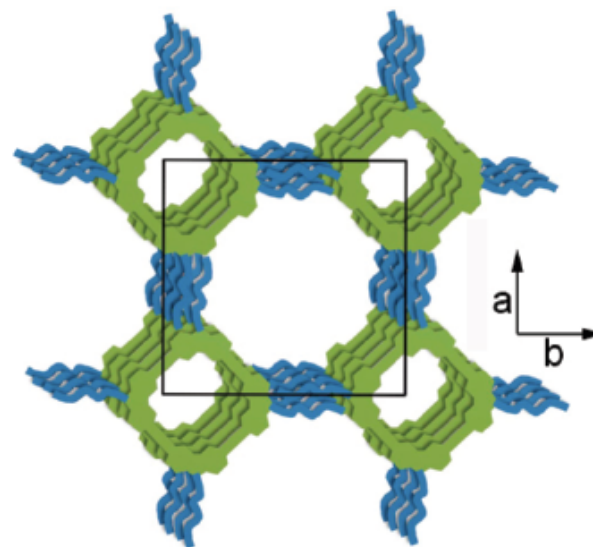
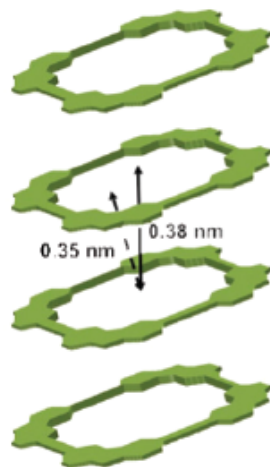
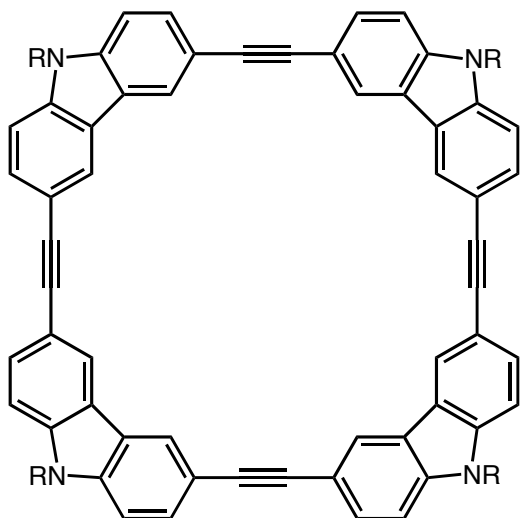
Detection of Explosives with a Nanofibril Film



84% yield as single product

New Macromolecular Architectures

Detection of Explosives with a Nanofibril Film

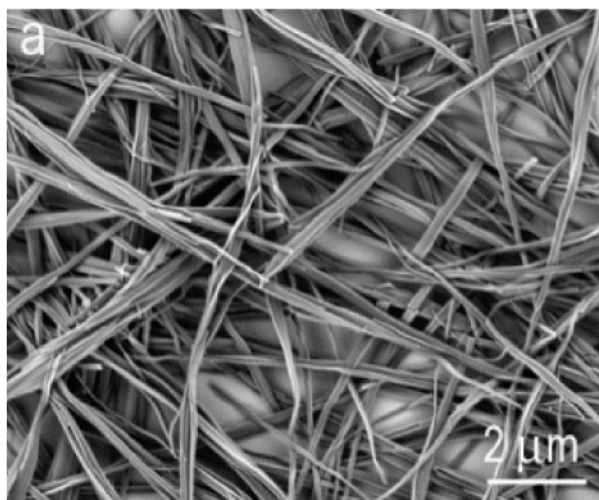


π - π stacking causes the carbazole-based tetracycles to assemble into 1D nanofibrils

1. Naddo, T. *et al. J. Am. Chem. Soc.*, **2007**, 129, 6978-6979.
2. Che, Y. *et al. J. Am. Chem. Soc.*, **2012**, 134-4978.

New Macromolecular Architectures

Detection of Explosives with a Nanofibril Film

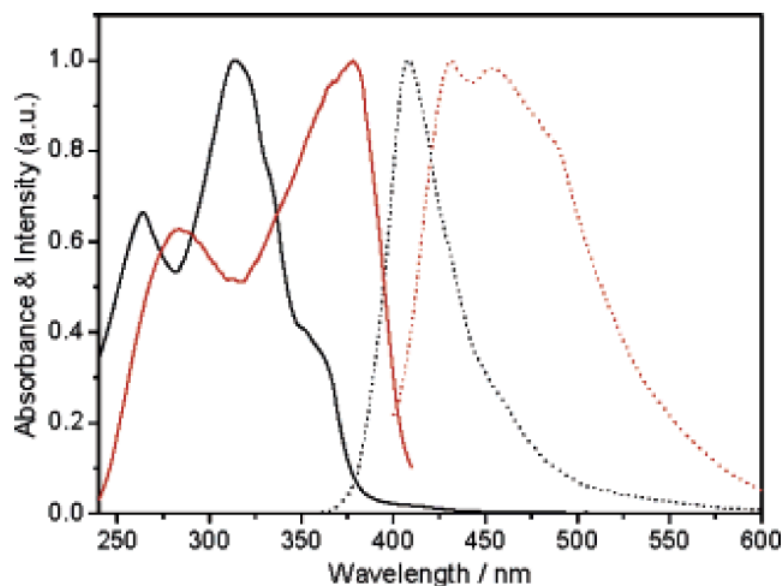


SEM image of porous nanofibers generated from piling the fibers by spin-casting

1. Naddo, T. *et al. J. Am. Chem. Soc.*, **2007**, 129, 6978-6979.
2. Che, Y. *et al. J. Am. Chem. Soc.*, **2012**, 134-4978.

New Macromolecular Architectures

Detection of Explosives with a Nanofibril Film



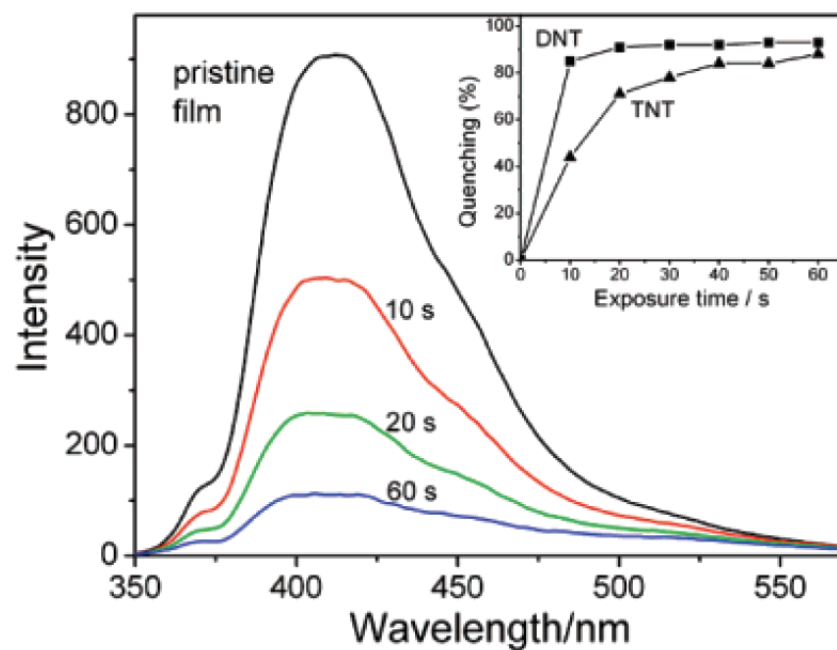
Absorption (solid) and fluorescence (dotted) spectra
solution (black) and dried gel (red)

π - π stacking results in intermolecular electronic coupling as evinced by perturbation of the absorption and fluorescence spectra

The supramolecular assembly behaves somewhat like one large conjugated molecule

New Macromolecular Architectures

Detection of Explosives with a Nanofibril Film

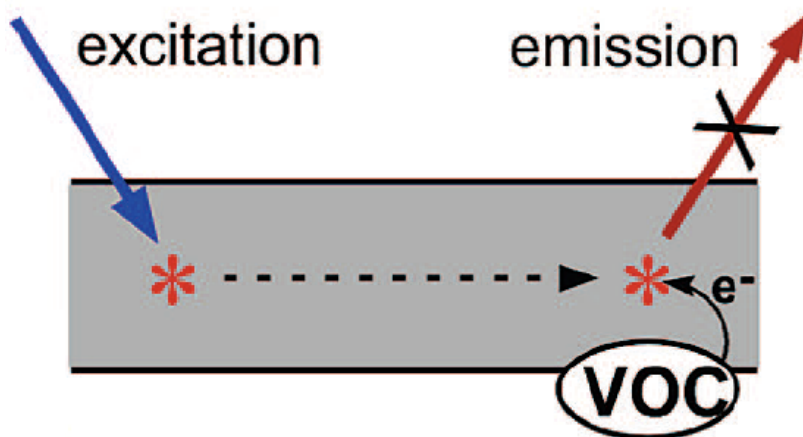


Fluorescence spectra of a 90 nm thick ACTC film upon exposure vapor of TNT (5 ppb) at different times.

1. Naddo, T. *et al. J. Am. Chem. Soc.*, **2007**, 129, 6978-6979.
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New Macromolecular Architectures

Detection of Explosives with a Nanofibril Film



π - π stacking is favorable for long range exciton migration

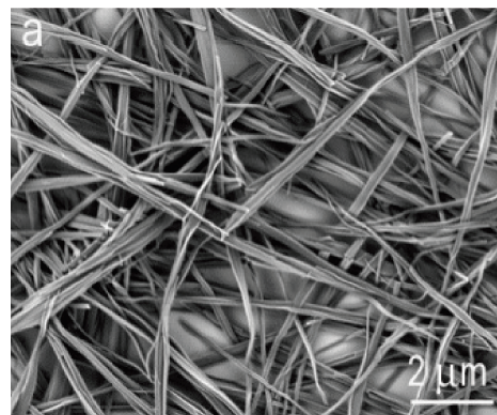
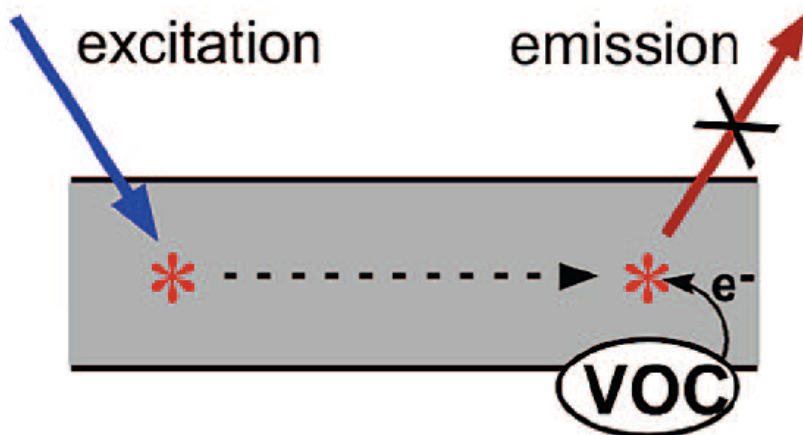
Fluorescent emission λ is significantly higher than TNT's absorption λ , which precludes quenching by energy transfer

Fluorescence quenching is due to photoinduced electron transfer from the fibril to TNT

1. Naddo, T. *et al. J. Am. Chem. Soc.*, **2007**, 129, 6978-6979.
2. Zhang, L. *et al. Accounts Chem. Res.*, **2008**, 41, 1596.

New Macromolecular Architectures

Detection of Explosives with a Nanofibril Film

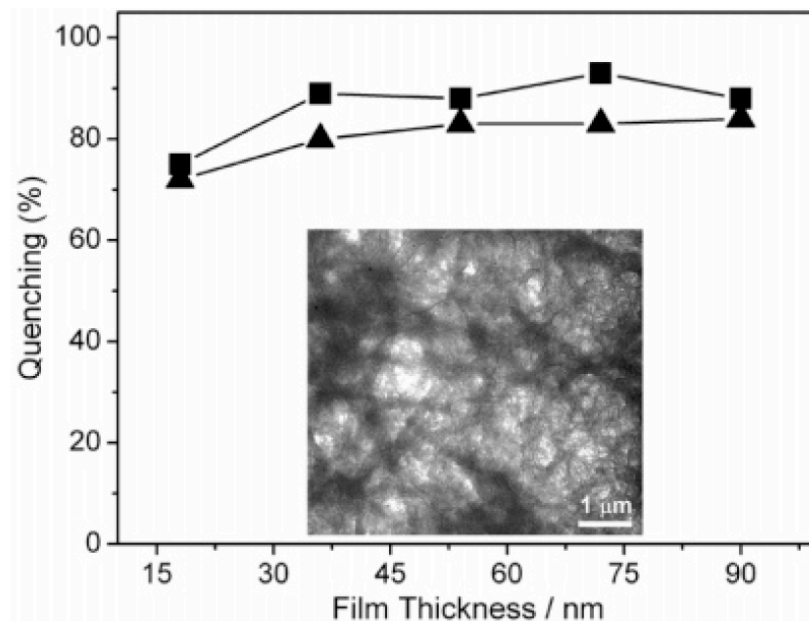


Nanofibers assembled via π - π interactions have high quenching efficiency by both enabling long range exciton migration and diffusion of gaseous TNT

1. Naddo, T. *et al. J. Am. Chem. Soc.*, **2007**, 129, 6978-6979.
2. Zhang, L. *et al. Accounts Chem. Res.*, **2008**, 41, 1596.

New Macromolecular Architectures

Detection of Explosives with a Nanofibril Film

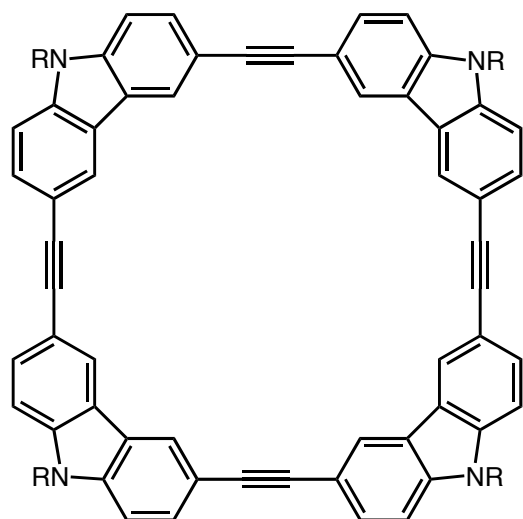


The efficiency of the fluorescence quenching is largely independent of the thickness of the film, a useful property not seen in most thin films

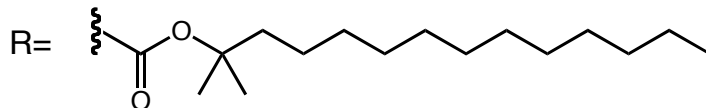
1. Naddo, T. *et al. J. Am. Chem. Soc.*, **2007**, 129, 6978-6979.
2. Zhang, L. *et al. Accounts Chem. Res.*, **2008**, 41, 1596.

New Macromolecular Architectures

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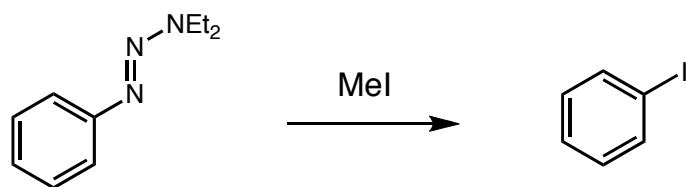


Can selectively detect TNT over other nitroaromatics at a vapor pressure of 20 parts per trillion

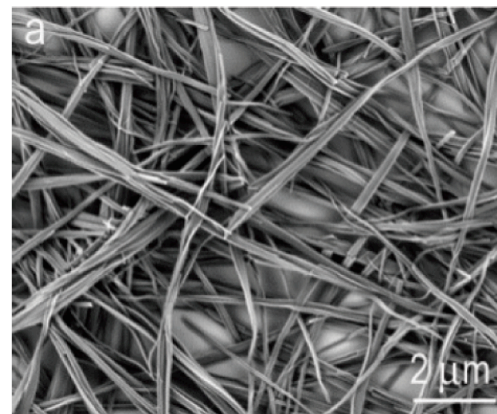


Conclusion

A Twenty-One Year Progression



1991



2012