Aspects of Asymmetric Nucleophilic Amine Catalysis: Organocatalyst Design and Implementation

MacMillan Group Meeting Ian Storer February 28, 2005

Asymmetric Nucleophilic Catalysis Outline

Introduction to N-centred nucleophilic catalysts

- concepts and definitions
- origins and requirements

Design, development and application of asymmetric nucleophilic catalysts

- Natural alkaloid catalysts Precejus, Wynberg
- Synthetic alkaloid analogues Lectka, Hatekeyama, Deng
- Biomimetic histidine containing peptide catalysts Miller
- Asymmetric DMAP equivalents Fuji / Fu

Useful Reviews:

- Fance, S.; Guerin, D. J.; Miller, S. J.; Leckta, T., Nucleophilic Chiral Amines as Catalysts in Asymmetric Synthesis. *Chem Rev.* **2003**, *103*, 2985-3012.
- Miller, S. J. In Search of Peptide-Based Catalysts for Asymmetric Organic Synthesis. Acc. Chem. Res. **2004**, *37*, 601-610.
- Fu, G. C. Asymmetric Catalysis with "Planar-Chiral" Derivatives of DMAP. *Acc. Chem. Res.* **2004**, *37*, 542-547.

Definitions

Lewis base

A molecular entity (and the corresponding chemical species) able to provide a pair of electrons and thus capable of coordination to a Lewis acid, thereby producing a Lewis adduct.

Nucleophilic Catalysis

Catalysis by a Lewis base, involving formation of a Lewis adduct as a reaction intermediate. For example, the esterification of anhydrides catalysed by DMAP:

IUPAC Compendium of Chemical Terminology



Implication on Choice of catalyst

The catalyst has to be a very effective nucleophile and a good leaving group .

Useful Nucleophilic Catalysts



good Bronsted bases, poor nucleophilic catalysts

Cinchona Alkaloids: Earliest Successful Nucleophilic Asymmetric Catalysts



First used for a resolution of a racemate in 1853 by Pasteur.

Several modes of application

- Asymmetric Bronsted bases
- Asymmetric phase transfer reagents (as salts)
- Asymmetric bifunctional catalysts (acid / base)
- Asymmetric nucleophilic catalysts

Moncure, R. MacMillan Group Meeting, 2003, web. - contains a more fully discussion of cinchona alkaloids

Alkaloids and Derivatives as Nucleophilic Catalysts

Natural alkaloids



Synthetically modified analogues



benzoylquinine Lectka (2000)



Hatakeyama





Blake, A. J.; Friend, C. L.; Outram, R. J.; Simpkins, N. S.; Whitehead, A. J. Tetrahedron Lett. 2001, 42, 2877.



Wynberg's proposed this stereoselectivity model



• In the left TS, the ketene oxygen faces the methylene of the catalyst ring .The chloral approaches the catalyst with the trichloromethyl group facing away from the methyl group of the catalyst to avoid steric strain.

• In the bottom TS, the ring is the dominant form of stereocontrol, and the CCl₃ orients itself away from the ring methylene protons.

Wynberg, H. *J. Am. Chem. Soc.* **1982**, *104*, 166. Wynberg, H. *J. Org. Chem.* **1985**, *50*, 1977.

Development of Functionalised Alkaloid Analogues: Catalytic Asymmetric Cycloadditions

β-lactone synthesis: Romo's expansion of Wynberg's method



- The use of pre-acylated catalyst stops unwanted acylation of the catalyst during the reaction.
- Applies a stoichiometric, non-nucleophilic base to turn over the catalyst.
- *In situ* ketene generation
- Solubility of the Hunigs salt may be crucial to catalyst turnover.

Tennyson, R.; Romo, D. J. Org. Chem. 2000, 65, 7248.

Development of Functionalised Alkaloid Analogues: Catalytic Asymmetric Cycloadditions

β-lactam synthesis: Lectka



- 'Shuttle deprotonation' using excess of a thermodynamic, non-nucleophilic base.
- Concomitant Lewis acid activation of the imine later provided better yields

Taggi, A. E.; Hafez, A. M.; Wack, H.; Young, B.; Drury, W. J.; Lectka, T. J.. *J. Am. Chem. Soc.* **2000**, *122*, 7831. Taggi, A. E.; Hafez, A. M.; Wack, H.; Young, B.; Ferraris, D.; Lectka, T. J.. *J. Am. Chem. Soc.* **2002**, *124*, 6626.



Development of Functionalised Alkaloid Analogues: Catalytic Asymmetric Halogenation



Wack, H.; Taggi, A. E.; Hafez, A. M.; Drury, W. J.; Lectka, T. J. J. Am. Chem. Soc. 2001, 123, 1531.
Hafez, A. M.; Taggi, A. E.; Wack, H.; Esterbrook, J.; Lectka, T. J. Org. Lett. 2001, 3, 2049.
Taggi, A. E.; Wack, H.; Hafez, A. M.; France, S.; Lectka, T. J. Org. Lett. 2002, 4, 627.

Development of Functionalised Alkaloid Analogues: Catalytic Asymmetric Baylis-Hillman Reaction

Natural alkaloid - poor enantioselectivity





Marko, I. E.; Giles, P. R.; Hindley, N. J. Tetrahedron 1997, 53, 1015.

Synthetic analogue - better results



Iwabuchi, Y.; Nakatani, M.; Yokoyama, N.; Hatekeyama, S. J. Am. Chem. Soc. 1999, 121, 10219.



Deng, L. et al. J. Am. Chem. Soc. 2000, 122, 9542.

Designing Asymmetric Variants of Known Nucleophilic Amine Catalysts

Considerations

- Must be nucleophilc, but maintain good leaving group ability
- Must have a valid chiral pocket to transfer asymmetry to product.



Synthetic chiral analogues of common nucleophilic heterocycles



Synthetic chiral imidazole equivalents

Synthetic chiral DMAP equivalents

Biological Role of Nucleophilic Catalysis

Mechanism of some kinases - role of histidine



- Phosphorylation transfer from ATP is ubiquitous in biological chemistry.
- Histidine often serves as *in vivo* acceptors of the γ -phosphate of ATP.
- Proposed mechanism of ATPases and nucleoside diphosphate kinase (NDPK).

In vitro phosphorylation of imidazole



- First *in vitro* example of imidazole participating in phosphoryl transfer from ATP analogues.
- N-nucleophiles found to react 30-100 fold faster than O-nucleophiles at physiological temperatures.
- Carried out in vitro experiments on ATP.
- Supports proposed enzymatic mechanism.

Admiraal, S. J.; Herschlag, D.; J. Am. Chem. Soc., 1999, 121, 5837-5845

Miller N-Methyl Imidazole-Containing Tripeptide

- Synthesis of a nucleophilic tripeptide
- Peptide-substrate interactions are crucial for the fidelity by which an enzyme imparts its selectivity.
- Incorporating an amino acid residue as an analogue of known nucleophilic catalyst N-methyl imidazolium (NMI).
- Work focused on peptides that had propensity to form stable secondary structures (β -turn) in solution.



Modularity of peptide-based systems allows for the rapid synthesis and screening of many analogues.

• Initial designs drew heavily from the peptide design literature in order to generate a relatively rigid β-turn structure

Miller, S. J.; Copeland, G. T.; Papaioannou, N.; Horstmann, T. E.; Ruel, E. M. J. Am. Chem. Soc. 1998, 120, 1629-1630.



- Amide in amino-alcohols necessary for good levels of stereoinduction
- Non-polar solvents which favour H-bonding work best
- Replacement of the amide with an ester provides only poor selectivity

Probing Origins of Selectivity





- Diastereomeric catalysts display enantiodivergence
- Changing a single stereocentre in the catalyst causes a change in secondary structure.
- Changing a single stereocentre in the catalyst causes a complete switch in enantioselectivity.



What is the mechanism of binding and role of the peptide backbone structure?

Copeland, G. T.; Miller, S. J. *J. Am. Chem. Soc.* **1999**, *121*, 4306. Harris, R. F.; Nation, A. J.; Copeland, G. T.; Miller, S. J. *J. Am. Chem. Soc.* **2000**, *122*, 11270.

Miller "Biomimetic" Strategy

Kinetic resolution of amino-alcohols



Mechanistic postulate



Larger rate acceleration than using DMAP - NMI is usually less activating?

Vasbinder, M. M.; Jarvo, E. R.; Miller, S. J. Angew. Chem. Int. Ed. 2001, 40, 2824-2827.

Miller: Development of a Fluorescence Assay

Kinetic resolution of amino-alcohols



Development of a fluorescence probe for rapid reaction conversion analysis



- Fluorescence intensity is a function of acetic acid concentration
- Assisted rapid catalyst discovery.

Copeland, G. T.; Miller, S. J. *J. Am. Chem. Soc.*. **1999**, *121*, 4306. Harris, R. F.; Nation, A. J.; Copeland, G. T.; Miller, S. J. *J. Am. Chem. Soc.*. **2000**, *122*, 11270.

Miller Histidine-Containing Peptide Catalysts

Catalyst for the kinetic resolution of tertiary alcohols





Kinetic resolution acetylation of unfunctionalised alcohols



The conformation / contribution of the peptide backbone is not known

Jarvo, E. R.; Evans, C. A.; Copeland, G. T.; Miller, S. J. J. Org. Chem. 2001, 66, 5522.

Miller: Development of a Desymmetrising Phosphorylation



Asymmetric phosphorylation of meso triols

• Two very different peptides give very highly enantioselective phosphorylation in opposite enantioseries

Sculimbrene, B. R.; Miller, S. J. J. Am. Chem. Soc.. 2001, 123, 10125.

Designing Asymmetric Variants of Known Nucleophilic Amine Catalysts

Considerations

- Must be nucleophilc, but maintain good leaving group ability
- Must have a valid chiral pocket to transfer asymmetry to product.

Alkaloids and analogues



Synthetic chiral analogues of common nucleophilic heterocycles



Synthetic chiral imidazole equivalents

Synthetic chiral DMAP equivalents

Early Attempts to Make Chiral DMAP Equivalents



- This DMAP equivalent is not nucleophilic enough to permit catalytic turnover
- A variety of substrates gave good levels of enantioselectivity

Vedejs, E.; Chen, X. *J. Am. Chem. Soc.* **1996**, *118*, 1809. Vedejs, E.; Chen, X. *J. Am. Chem. Soc.* **1997**, *119*, 2584.



Spivey, A. C et. al.. J. C. S. Perkin 1 2000, 14, 3460.









• Cation-p interaction holds transition state rigid

Kawabata, T.; Nagato, M.; Takasu, K.; Fuji, K. J. Am. Chem. Soc. 1997, 119, 3169-3170.

Fu Planar Chirality: Designing an Alternative Chiral DMAP

Fu built a variety of Sandwich complex derived chiral nucleophilic amines



- In general best results were obtained with DMAP and PPY derivatives
- France, F.; Guerin, D. J.; Miller, S. J.; Lectka, T. *Chem. Rev.* **2003**, *103*, 2985-3012. Fu, Gregory C. *Acc. Chem. Res.* **2004**, *37*, 542-547.

Fu - Planar Chirality: Kinetic Resolution of Alcohols

$\begin{array}{c} OH\\ R^{1} \\ R^{2}\\ racemic \end{array} \qquad \begin{array}{c} O\\ Me \\ \hline \\ 2 \\ Et_{3}N, \ t-amyl \ alcohol \end{array} \qquad \begin{array}{c} OH\\ Me \\ \hline \\ R^{1} \\ R^{2} \\ \end{array} \qquad \begin{array}{c} OH\\ R^{2} \\ \end{array} \qquad \begin{array}{c} OH\\ R^{2} \\ R^{2} \\ \end{array} \qquad \begin{array}{c} OH\\ R^{2} \\ \end{array} \qquad \begin{array}{c} OH\\ R^{2} \\ R^{2} \\ \end{array} \qquad \begin{array}{c} OH\\ R^{2} \\ \end{array} \qquad$



Very high enantioselectivities for Aryl and Cinnamyl alcohols

Kinetic Resolution of Alcohols



Ruble, J. C.; Latham, H. A.; Fu, G. C. *J. Am. Chem. Soc.* **1997**, *119*, 1492-1493. Ruble, J. C.;Tweddell, J.; Fu, G. C. *J. Org. Chem.* **1998**, *63*, 2794-2795. Fu Chiral DMAP: Origins of Enantioselectivity

Geometry of acyl rotamers



- Acetyl rotamer consistent with minimization of sterics between the methyl R group and ferrocene.
- Selectivity increases as the size of the alkyl group R increases.
- Nucleophile approaches from the top face of the DMAP catalyst.

Kinetic Resolution of Amines



First non-enzymatic acylation catalyst for kinetic resolution of amines

Very high enantioselectivities for Aryl and Cinnamyl alcohols



Arai, S.; Laponnaz-Bellemin, S.; Fu, G. C. Angew. Chem. Int. Ed., 2001, 40, 234-236.

Rearrangement of O-Acylated Azalactones



Ruble, J. C.; Fu, G. C. J. Am. Chem. Soc., 1998, 120, 11532-11533.



Hills, I. D.; Fu, G. C. Angew. Chem. Int. Ed., 2003, 42, 3921-3924.



• Currently limited to α -aryl substrates

Mermerian, A. H.; Fu, G. C. J. Am. Chem. Soc., 2003, 125, 4050-4051.

Chiral Planar DMAP Analogues: Reactions with Ketenes



• Reaction successful for both symmetrical and unsymmetrical ketenes

Hodous, B. L.; Fu, G. C. J. Am. Chem. Soc. 2002, 124, 1578-1579.

Summary and Conclusions

- Chiral amines have become functional nucleophilic catalysts, having been largly overlooked until recently.
- Natural and synthetic alkaloids continue to have success across a variety of reaction types, although definitive stereochemical rationale is not always possible
- Miller's combinatorial approach has successfully obtained catalysts that show high selectivity
- Fu has developed the first truly general synthetic DMAP analogue catalyst system. Led the way in showing that nucleophilic catalysis is a general and useful concept.