

# *Synthesis and Applications of Fluorinated $\alpha$ -Amino Acids*

*Overview and Recent Advances*



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MacMillan Group Meeting  
Princeton University  
11/15/2017*

# *Outlines*

## ■ Introduction

- Interest of artificial amino acids
- Unique properties of the C-F bond
- Natural fluorinated amino acid

## ■ Applications of fluorinated $\alpha$ -amino acids

- Protein engineering
- Medicinal chemistry
- $^{19}\text{F}$ -NMR
- $^{18}\text{F}$ -PET

## ■ Incorporation of fluorinated $\alpha$ -amino acids

## ■ Synthesis of fluorinated $\alpha$ -amino acids

- Strategy I
- Strategy II
- Strategy III
- Strategy IV
- Strategy V
- Strategy VI

## *Introduction*

### *Interest of Artificial $\alpha$ -Amino Acids*

- Peptides and proteins occupy innumerable biological functions

- Metabolic

- Structural

- Serve as inspiration in many research fields

- New therapeutic agents

- Biomaterials

- Appealing features

- Low toxicity

- Structural simplicity

- Low immune response

- Intrinsic limitations

- Enzymatic degradation (protease)

- Thermic stability

- Denaturation in organic solvents

- The use of non-canonical amino acids may overcome some of those limitations

- D- $\alpha$ -amino acids

- $\beta$ -amino acids

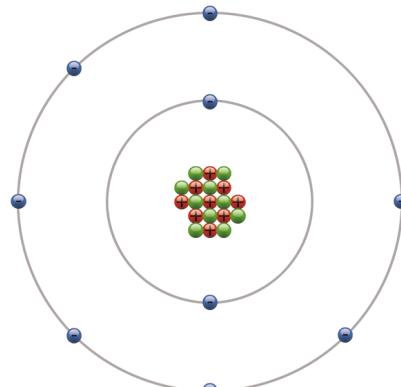
- Artificial amino acids**

- Artificial amino acids expand the functional diversity and open the way to tailored protein design

## $C-H \rightarrow C-F$ Bond Substitution

### ■ Minor steric impact

F: smallest Van der Waals radius after H and Ne



### ■ Major electronic impact

F: most electronegative element

Inverted polarization vs C-H

### ■ Unique features of the C-F bond

Often described as the strongest organic bond (BDE up to 544 KJ/mol)

Most polar covalent bond

Can influence polarization on adjacent bonds

Low polarizability of C-F can give rise to “Fluorous effect”



### ■ Fluorous effect

Auto-segregation of fluorine-rich organic species

Intermolecular: fluorous phase

Intermolecular: fluorophile domain on protein

Can those unique features be explored in peptide/protein engineering ?

Fluorinated  $\alpha$ -amino acids ( $\alpha$ -AAF)

Buer, B. C.; Marsh, E. N. G. *Protein Sci.* **2012**, *21*, 453.

Cametti, M.; Crousse, B.; Metrangolo, P.; Milani, R.; Resnati, G. *Chem. Soc. Rev.* **2012**, *41*, 31.

## Natural Fluorinated $\alpha$ -Amino Acid

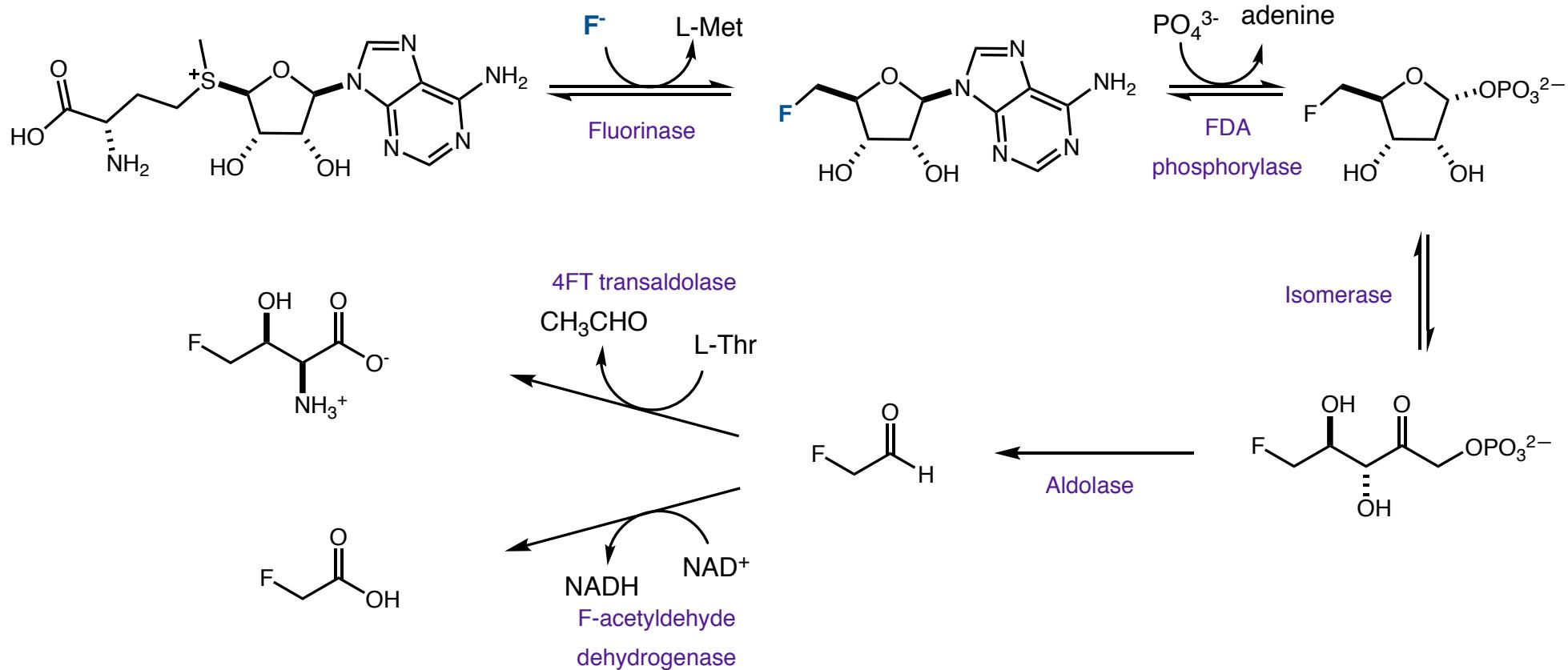
### ■ 4-Fluoro-L-threonine (4F-Thr): only natural $\alpha$ -AA<sub>F</sub> known

Produced by *Streptomyces cattleya* (Gram-positive)

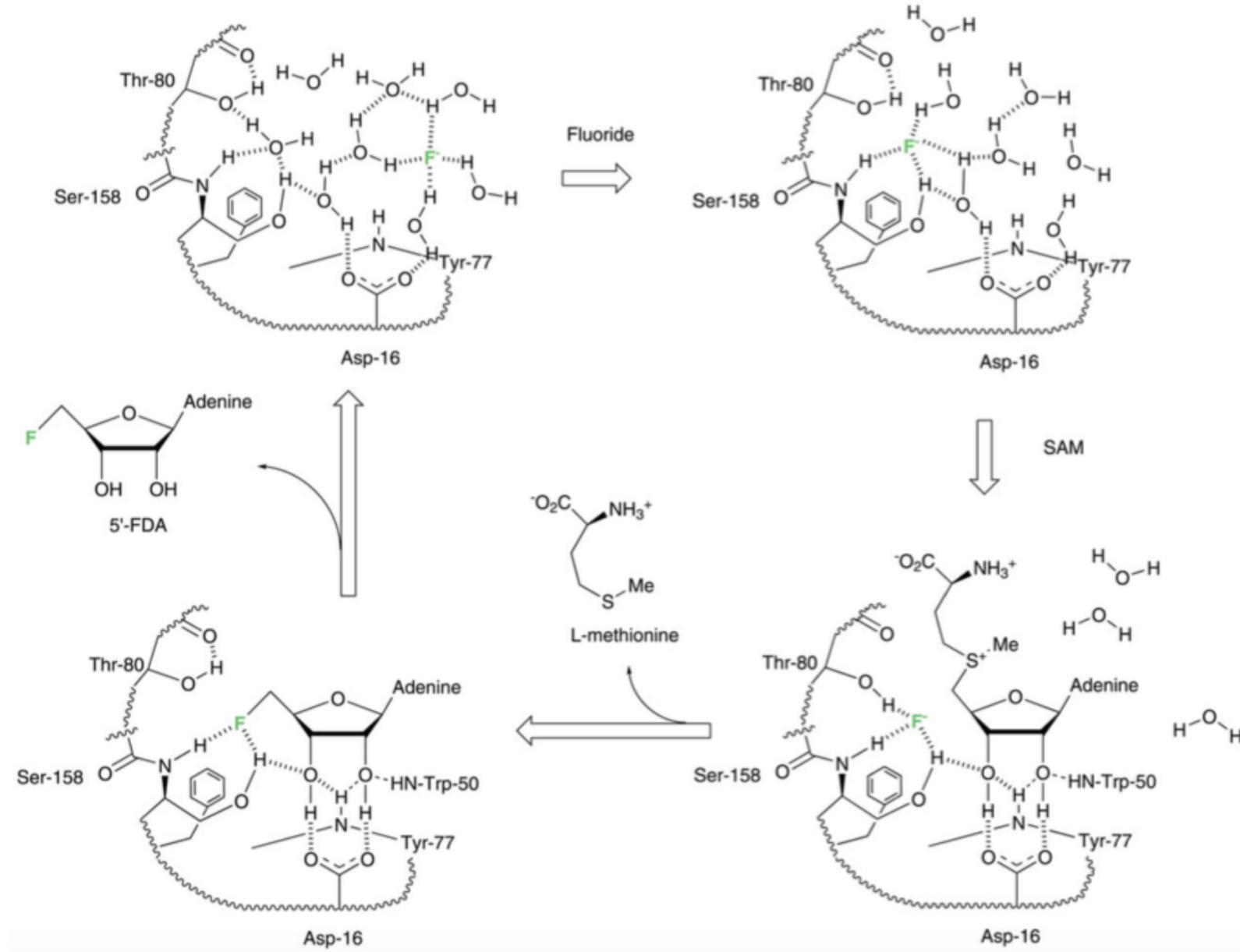
Antibiotic agent

Not identified in any peptide or protein so far

### ■ Biosynthesis involves incorporation of fluoride by a fluorinase enzyme



## Incorporation of Inorganic Fluoride by Fluorinase Enzyme



# *Application of $\alpha$ -AA<sub>F</sub>*

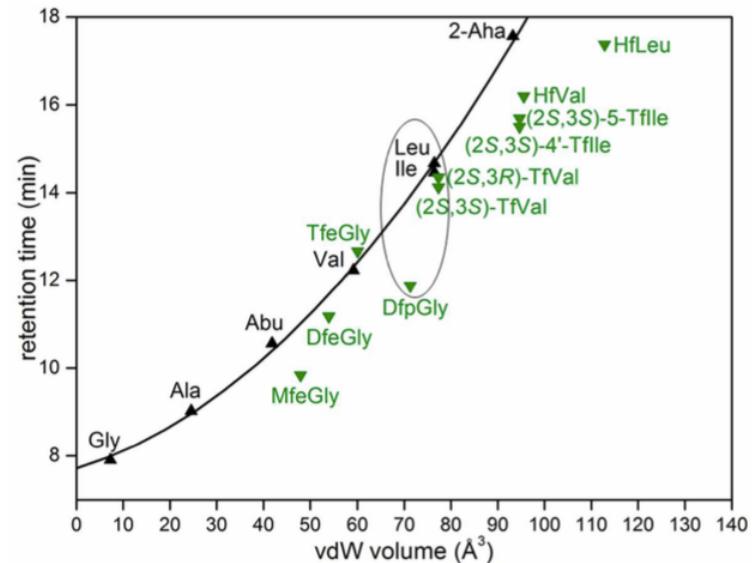
## *Protein Engineering: General Considerations*

- Many properties can be influenced by the substitution of canonical  $\alpha$ -amino acids by fluorinated analogues

Protein folding  
Protein-protein interactions  
Ribosomal translation  
Lipophilicity  
Acidity/basicity character  
Optimal pH  
Stability: temperature, protease, organic solvents

- Example: lipophilicity vs. VdW volume of side chains

$\alpha$ -AA<sub>F</sub> can fill a desired space but with a different lipophilicity from  $\alpha$ -AA



- Often claimed to generally improve stability and properties in out-of-field literature

- But, not that simple !

*« Fluorine has been shown to impart often favourable but seldom predictable properties to peptides and proteins »*

-Beate Koksch (2017)-

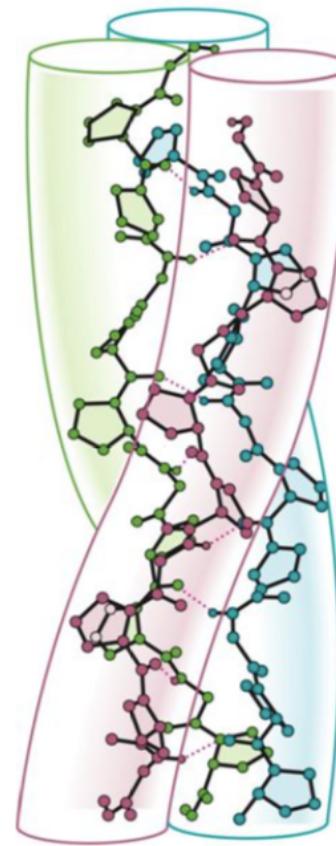
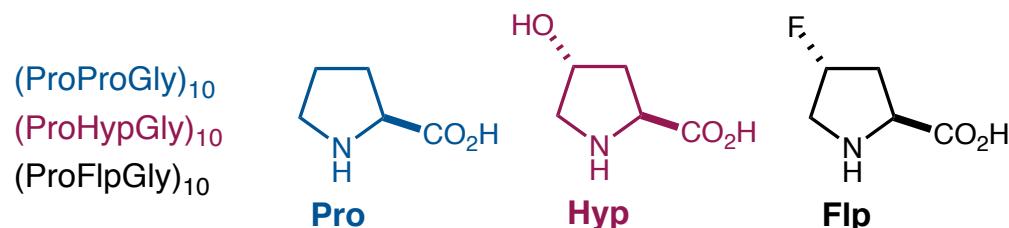
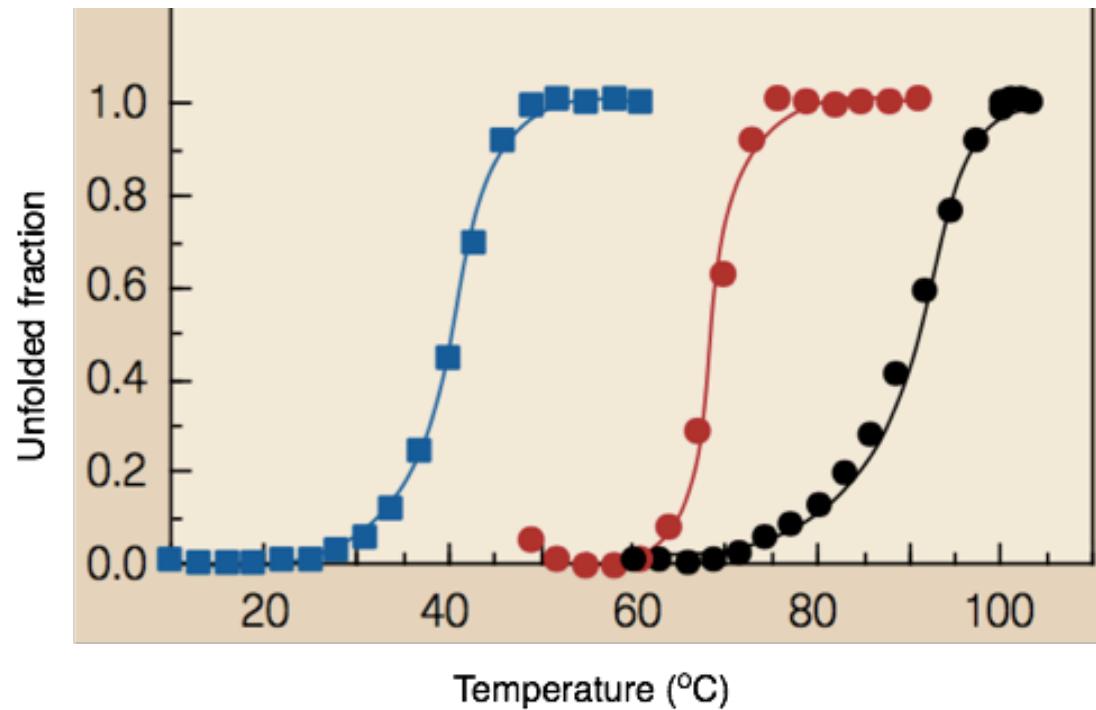
Merkel, L.; Schauer, M.; Antranikian, G.; Budisa, N. *ChemBioChem* **2010**, *11*, 1505.

Berger, A. A.; Völler, J.-S.; Budisa, N.; Koksch, B. *Acc. Chem. Res.* **2017**, *50*, 2093.

## *Applications of $\alpha$ -AAF*

### *Protein Engineering*

- Classic example: stabilization of the folded triple helix structure of collagen at high temperature



# *Applications of $\alpha$ -AA<sub>F</sub>*

## *Medicinal Chemistry of Peptides*

### ■ Interest for peptides in medicinal chemistry

- Low toxicity
- Simple structure / synthesis
- Low immune response

### ■ Drawbacks of peptides in medicinal chemistry

- Biodisponibility
- Low membrane permeability
- Enzymatic cleavage

Can  $\alpha$ -AA<sub>F</sub> help to overpass those limitations ?

### ■ Impact of fluorine in medicinal chemistry

- 150 fluorinated small molecules reaching the market since 1950
- In 2010, 20% of all administered drugs
- 30% by 2017**

Can this impact be translated to peptides ?

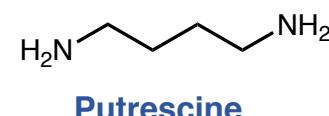
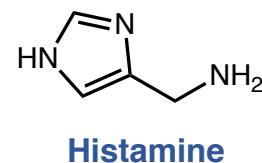
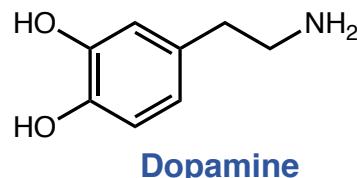
## *Applications of $\alpha$ -AAF*

### *Inhibitors*

- Because of their resemblance with canonical  $\alpha$ -amino acids, some  $\alpha$ -AAF can act as inhibitors

Known action on amino acid decarboxylase enzymes

Inhibition of the production of biologically relevant amines, including:



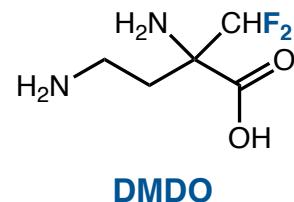
- $\alpha$ -(difluoromethyl)ornithine (DMDO)

Commercial name: Eflornithine

Treatment of hirsutism and African trypanosomiasis (sleeping sickness)

Inhibitor of ornithine decarboxylase

Acts on putrescine biosynthesis



Kollonitsch, J.; et al. *Nature* **1978**, 274, 906.

Seki, M.; Suzuki, M.; Matsumoto, K. *Biosci. Biotechnol. Biochem.* **1993**, 57, 1024.

# *Applications of $\alpha$ -AAF*

*$^{19}\text{F-NMR}$  Spectroscopy*

■ Widely used non-native NMR-active nucleus for interpretation of biological structure and functions

■ Many advantages

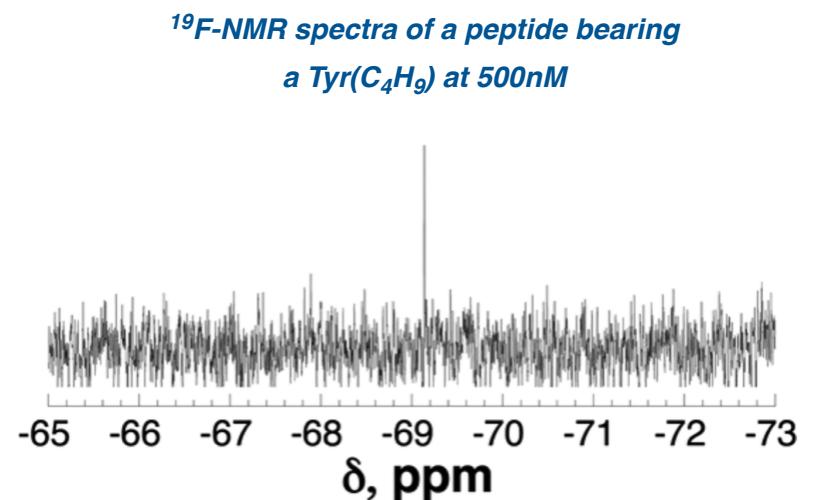
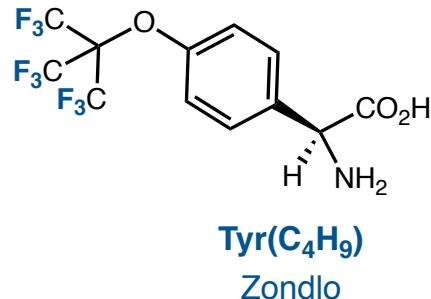
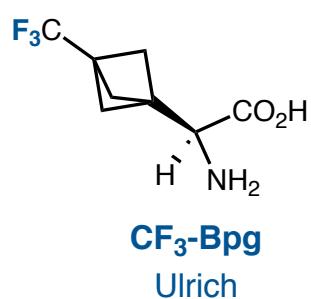
No marking necessary

No background in biological systems

Sensible

Can be used at biorelevant concentrations

■ Some  $\alpha$ -AAF specifically designed as NMR probes

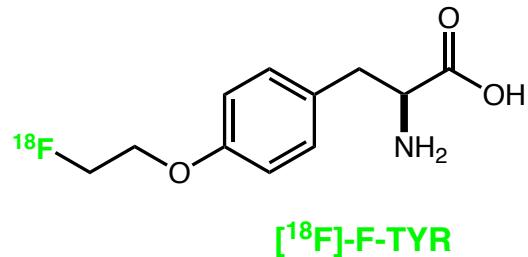
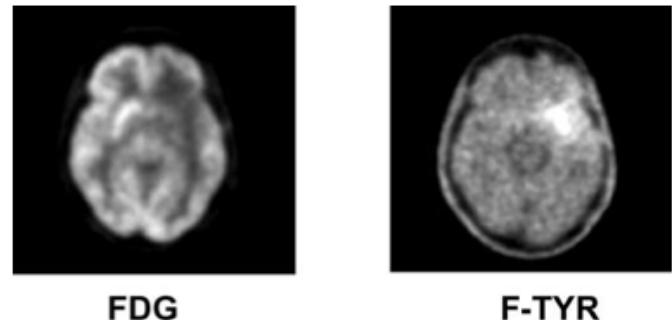


Salwiczek, M.; Mikhailiuk, P. K.; Afonin, S.; Komarov, I. V.; Ulrich, A. S.; Koksch, B. *Amino Acids* **2010**, *39*, 1589.  
Tressler, C. M.; Zondlo, N. J. *Org. Lett.* **2016**, *18*, 6240.

## *Applications of $\alpha$ -AAF*

$^{18}\text{F}$ -Positron Emission Tomography ( $^{18}\text{F}$ -PET)

- $^{18}\text{F}$  :  $t_{1/2} = 109.77$  min
- $^{18}\text{F}$ -PET mainly used in oncology
- Main marker used:  $^{18}\text{F}$ -FDG  
 $(^{18}\text{F})$ -fluorodeoxyglucose  
Discrimination of cancer cells due to their higher metabolism
- Consumption of  $\alpha$ -amino acids also higher in cancer cells  
 $\alpha$ -AAF explored as new markers  
Better resolution in some cases
- $\alpha$ -AAF could also be used for more refine analysis  
Marking based on active transport events  
Transport protein LAT1 upregulated in cancer cells

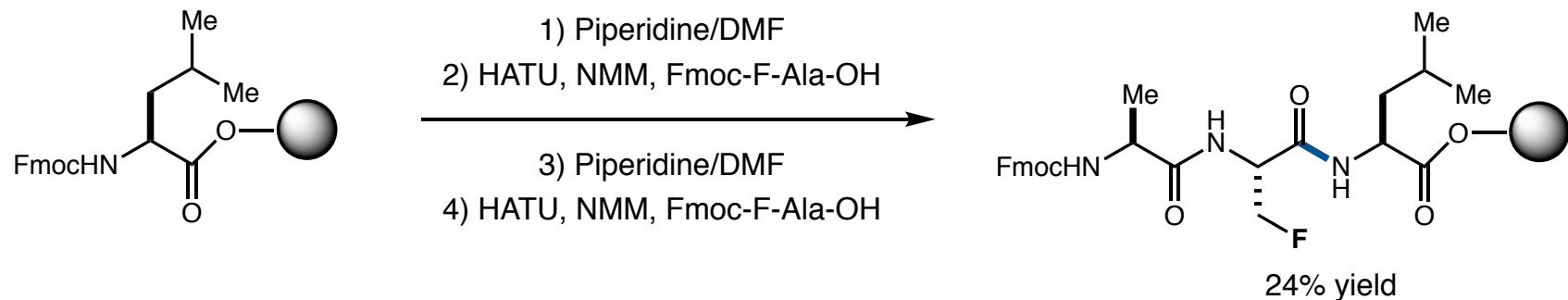


Kotzerke, J.; et al. *Eur. J. Nucl. Med. Mol. Imag.* **2003**, *30*, 1004.  
Nodwell, M. B.; Britton, R.; et al. *J. Am. Chem. Soc.* **2017**, *139*, 3595.

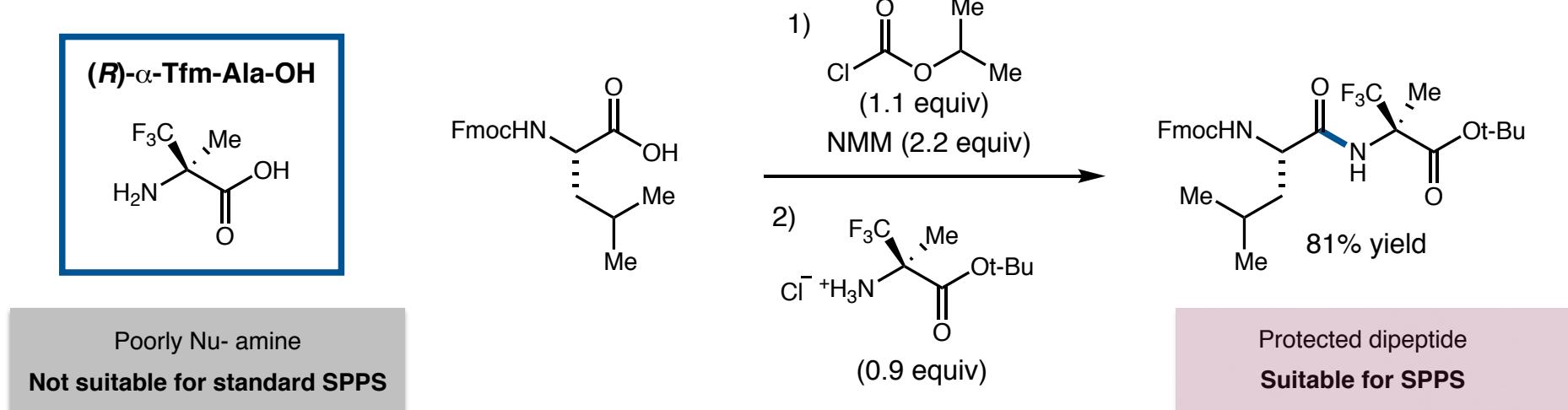
## Incorporation of Fluorinated $\alpha$ -Amino Acids in Peptides and Proteins

Synthetic Incorporation with Solid-Phase Peptide Synthesis (SPSS)

### ■ A lot of $\alpha$ -AA<sub>F</sub> can be used in the context of SPSS



### ■ Coupling in solution can overpass compatibility issues with problematic $\alpha$ -AA<sub>F</sub>



Carpentier, C.; Godbout, R.; Otis, F.; Voyer, N. *Tetrahedron Lett.* **2015**, *56*, 1244.

Devillers, E.; Pytkowicz, J.; Chelain, E.; Brigaud, T. *Amino Acids* **2016**, *48*, 1457.

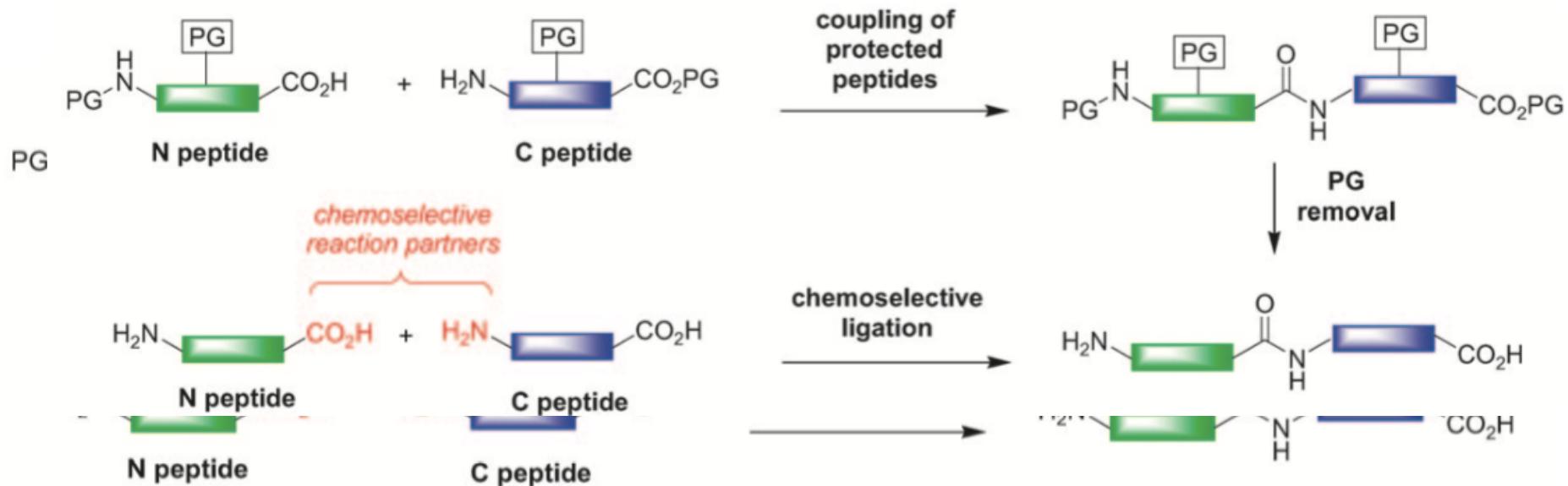
# Incorporation of Fluorinated $\alpha$ -Amino Acids in Peptides and Proteins

## Synthetic Incorporation Into Larger Peptide/Proteins

### ■ Limitations of SPSS

Compatibility issues with some  $\alpha$ -AAF  
Specific protecting groups required  
Not suitable for more than 50-100  $\alpha$ -AA

### ■ Coupling of fragments in solution may be used for longer sequences



# Incorporation of Fluorinated $\alpha$ -Amino Acids Into Peptides and Proteins

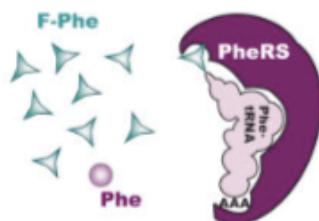
*In Vivo* Ribosomal Translation of Non-Canonical  $\alpha$ -Amino Acids

## ■ Residue specific

Relies on native translational machinery of the cell  
Incorporates the modified  $\alpha$ -AA at every position occupied by the parental  $\alpha$ -AA

Needs strains of bacteria that are auxotrophic for the parental  $\alpha$ -AA  
 $\alpha$ -AAF has to be sufficiently similar to its natural analog  
Particularly suitable for monoofluorinated  $\alpha$ -AAF

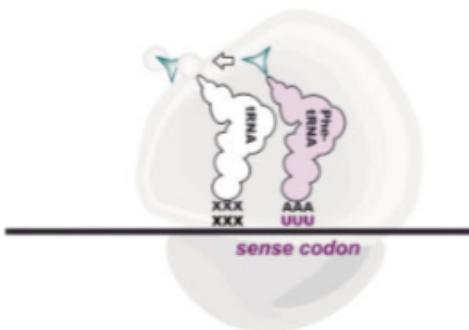
aminoacylation of tRNA



endogenous PheRS charges Phe-tRNA with F-Phe in the virtual absence of Phe

in a Phe auxotrophic strain

translation



F-Phe is incorporated in place of Phe

protein



all Phe residues replaced by F-Phe

# Incorporation of Fluorinated $\alpha$ -Amino Acids Into Peptides and Proteins

In Vivo Ribosomal Translation of Non-Canonical  $\alpha$ -Amino Acids

## ■ Site specific

*Nonsense suppression:* associate one of the three stop codons with an artificial  $\alpha$ -AA

Design of an appropriately engineered pair: **aminoacyl tRNA synthetase (aaRS) / Suppressor tRNA**

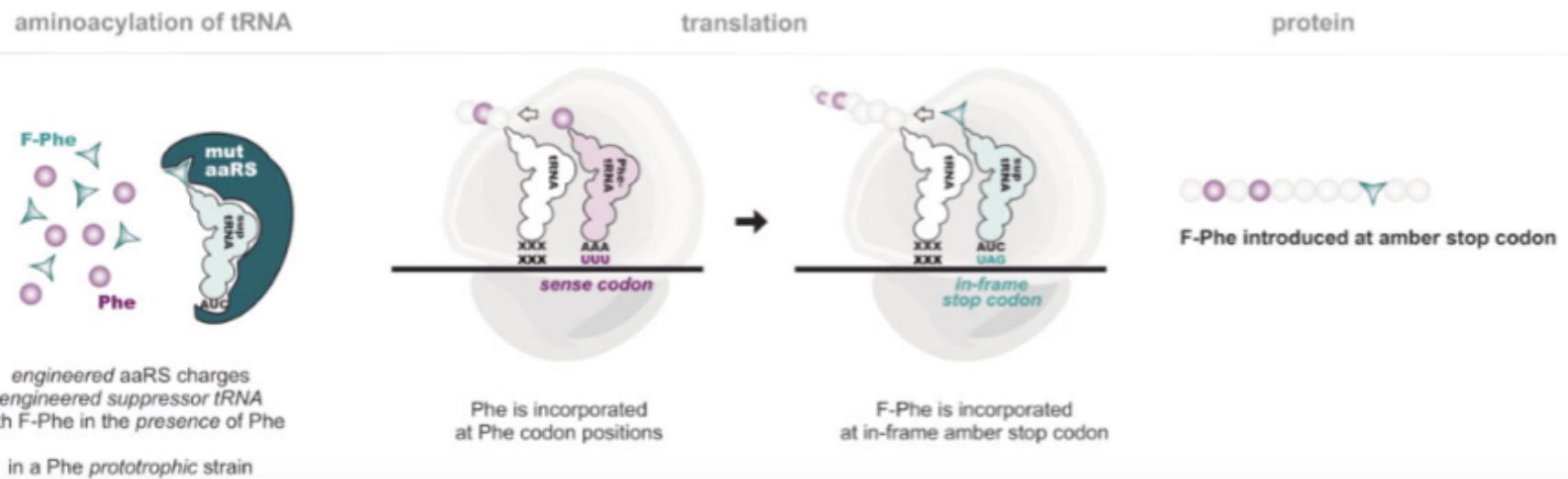
Corresponding sequence needs to be implemented in the genetic code in order to express the desired protein

### Suppressor tRNA

Recognizes stop codon (via anti-codon)  
Incorporates artificial  $\alpha$ -AA in the protein

### Mutant aaRS

Brings together artificial  $\alpha$ -AA and suppressor tRNA



Odar, C.; Winkler, M.; Wiltschi, B. *Biotechnol. J.* **2015**, *10*, 427.

Biava, H.; Budisa, N. *Engineering in Life Sciences* **2014**, *14*, 340.

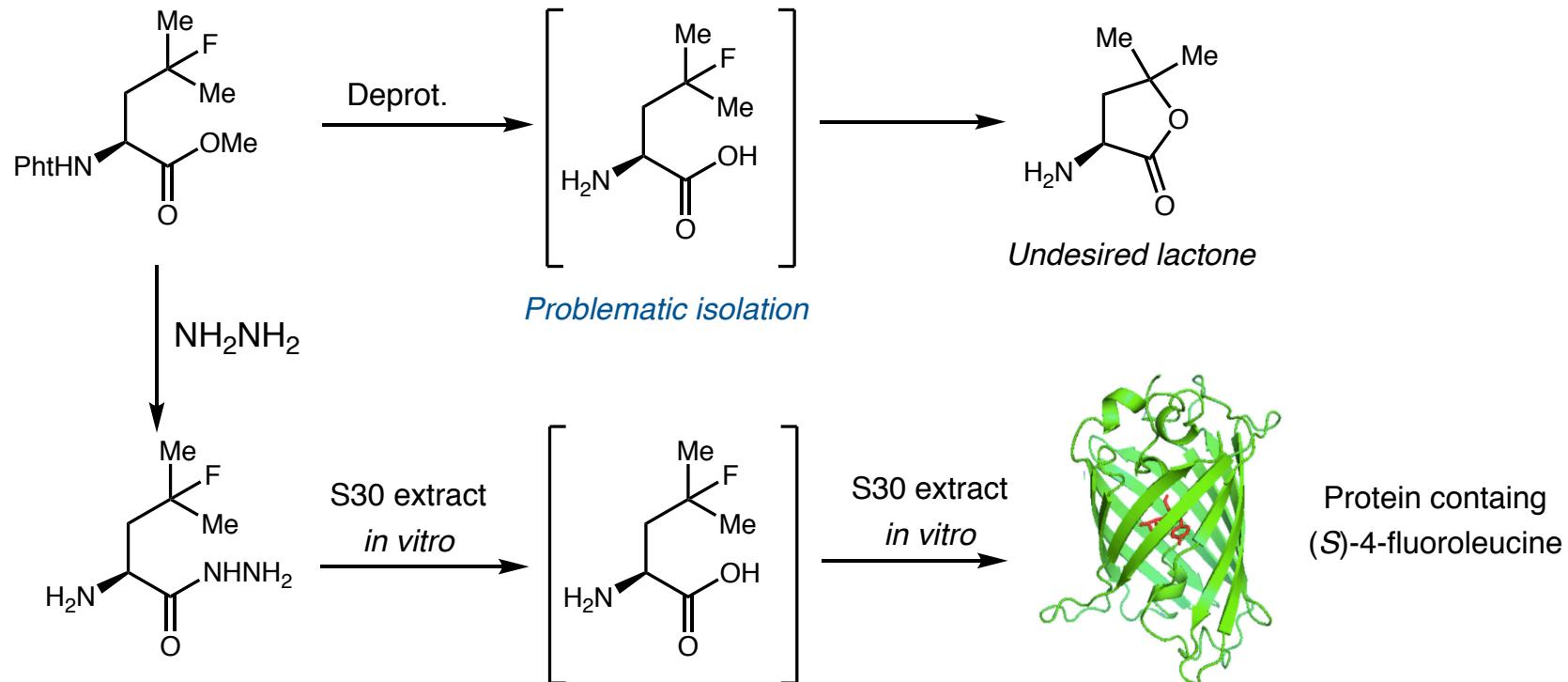
# Incorporation of Fluorinated $\alpha$ -Amino Acids Into Peptides and Proteins

Cell-Free Ribosomal Translation of Non-Canonical  $\alpha$ -Amino Acids

- Recently developed as *in vitro* alternatives to the *in vivo* approaches

Rely on the same translational machinery  
Need prior extraction and partial purification of cellular material  
More tunable systems

- Other enzymes can be added to promote other transformations



# Synthesis of $\alpha$ -AAF: Proposed Classification

## Approach A

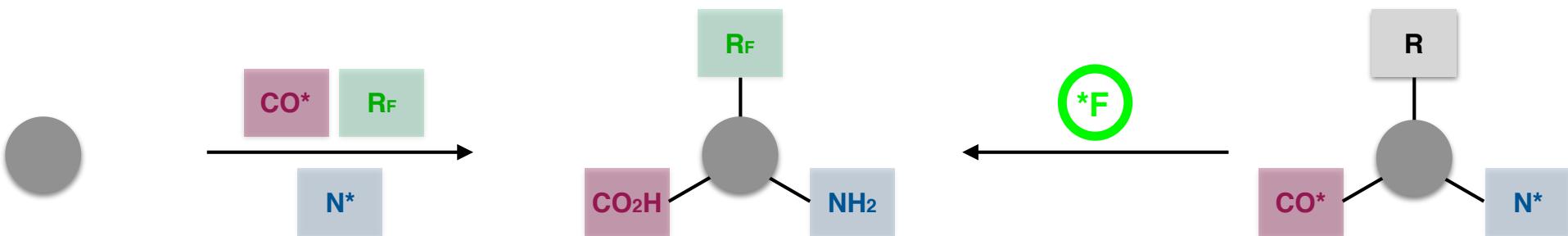
Installation of functionalities

(Strategies 1-5)

## Approach B

Fluorination of side chain

(Strategy 6)



N\*

Amine or *nitrogen function*

CO\*

Carbox. acid or oxygenated function

RF

Fluorinated side chain

R

Non fluorinated side chain

\*F

Fluorinating agent

●

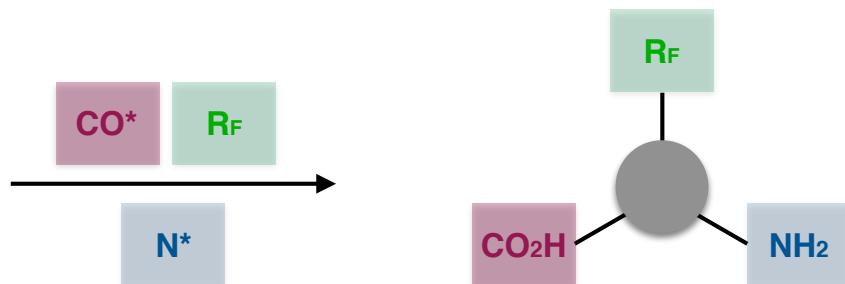
Amino acid C $\alpha$

# *Synthesis of $\alpha$ -AAF: Proposed Classification*

## Approach A

Installation of functionalities

(Strategies 1-5)



$\text{N}^*$

Amine or *nitrogen function*

$\text{CO}^*$

Carbox. acid or oxygenated function

$\text{R}_F$

Fluorinated side chain

$\text{R}$

Non fluorinated side chain

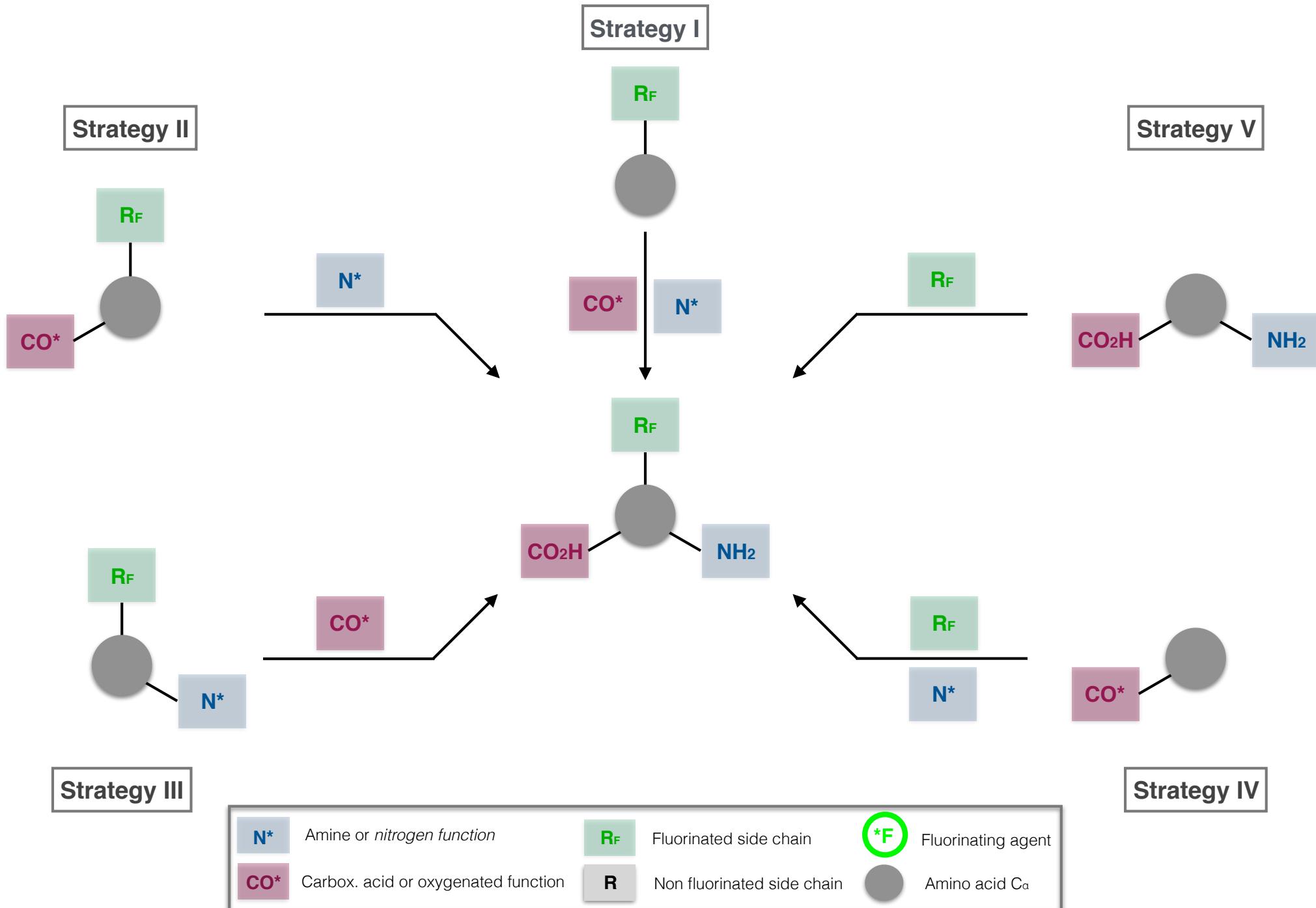
$^*\text{F}$

Fluorinating agent



Amino acid  $\text{C}_\alpha$

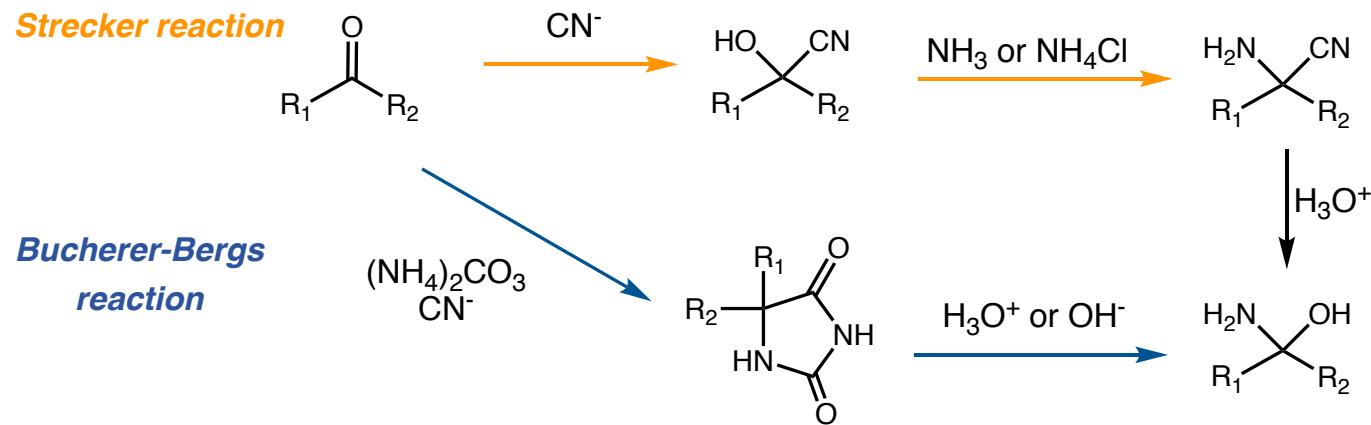
# Synthesis of $\alpha$ -AAF: Proposed Classification [Approach A]



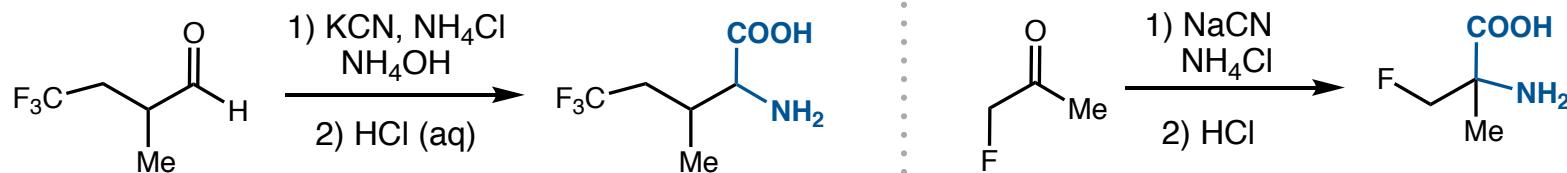
## Strategy I

Methodologies from Classic  $\alpha$ -Amino Acid Chemistry

### ■ Installation of carboxylic acid and amine functionalities on a carbonyl precursor



### ■ Examples in the synthesis of $\alpha$ -AAF: Strecker reaction



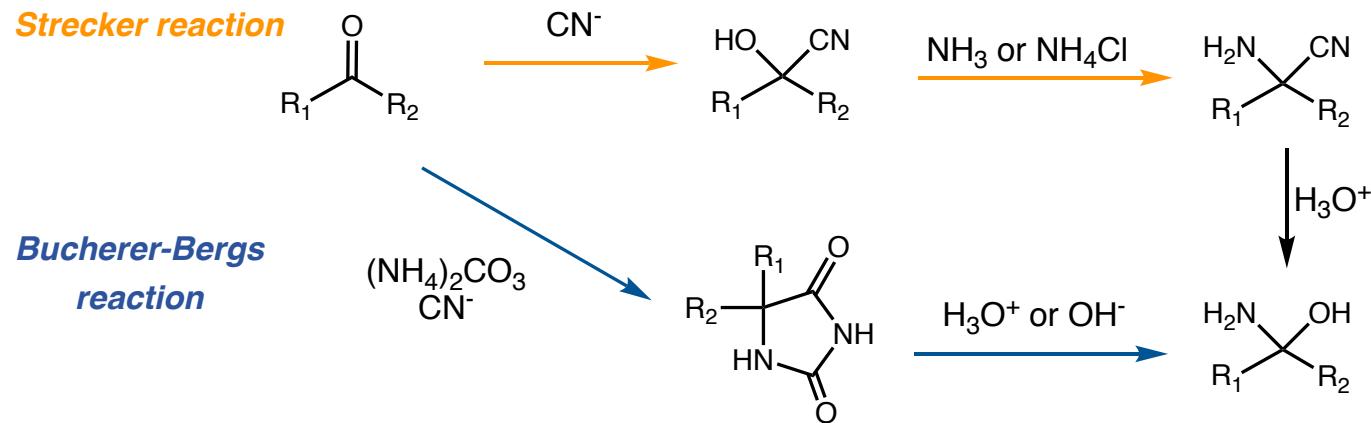
Bergmann, E. D.; Shani, A. *J. Chem. Soc.* **1963**, 3462.

Muller, N. *J. Fluorine Chem.* **1987**, *36*, 163.

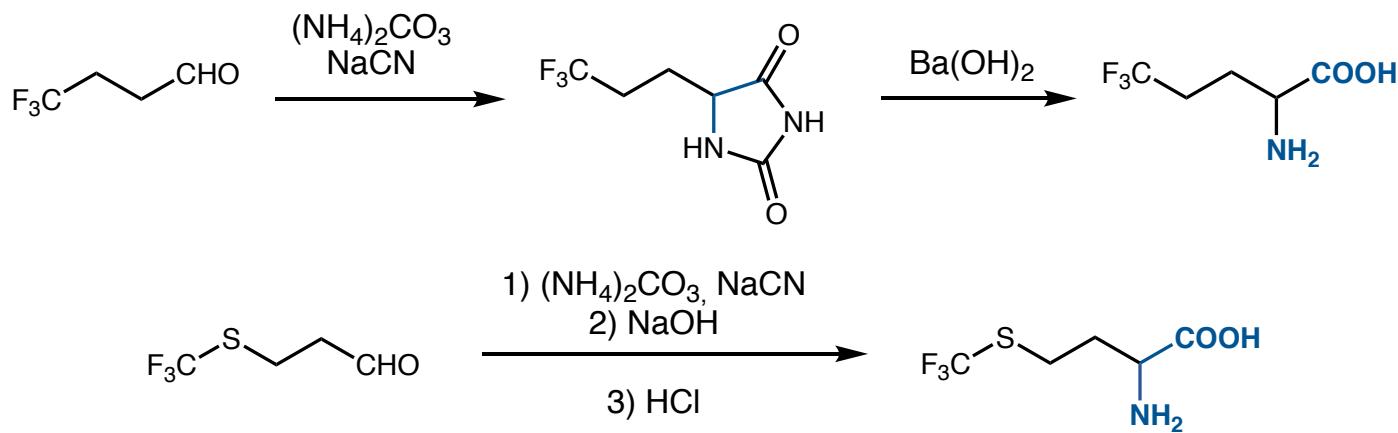
# Strategy I

*Methodologies from Classic  $\alpha$ -Amino Acid Chemistry*

- Installation of carboxylic acid and amine functionalities on a carbonyl precursor



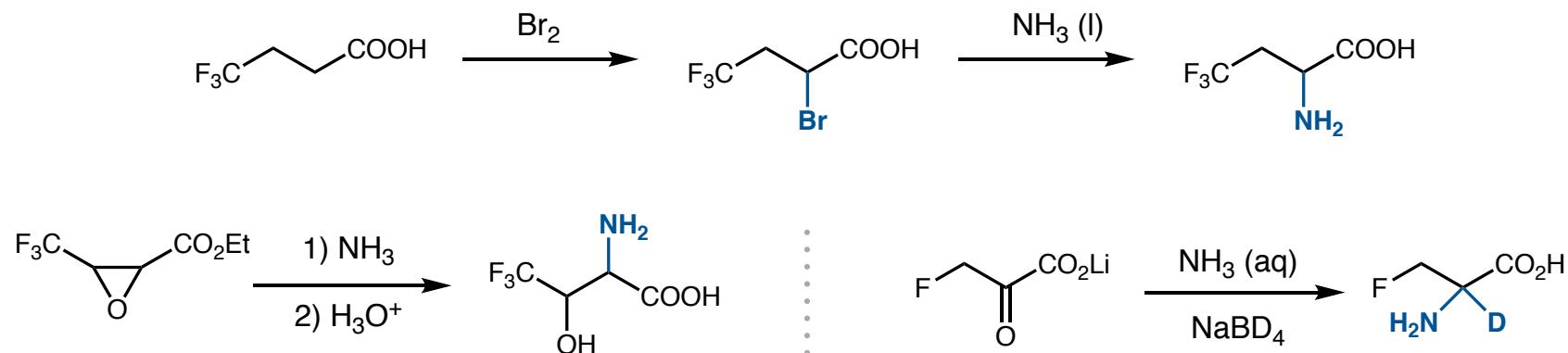
- Examples in the synthesis of  $\alpha$ -AA<sub>F</sub>: Bucherer-Bergs reaction



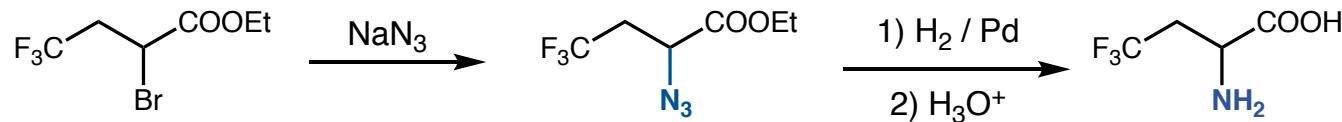
## Strategy II

### Classic Amination Reactions

#### ■ Direct amination with ammonia



#### ■ Azide as nucleophilic aminating agent

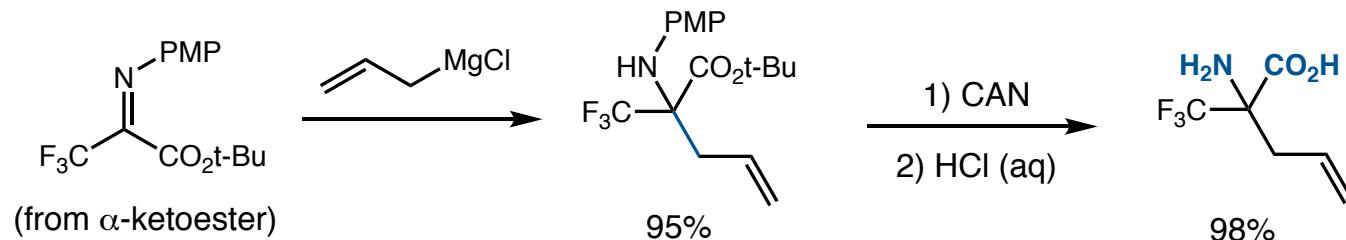


Tolman, V. *Amino Acids* **1996**, *11*, 15.

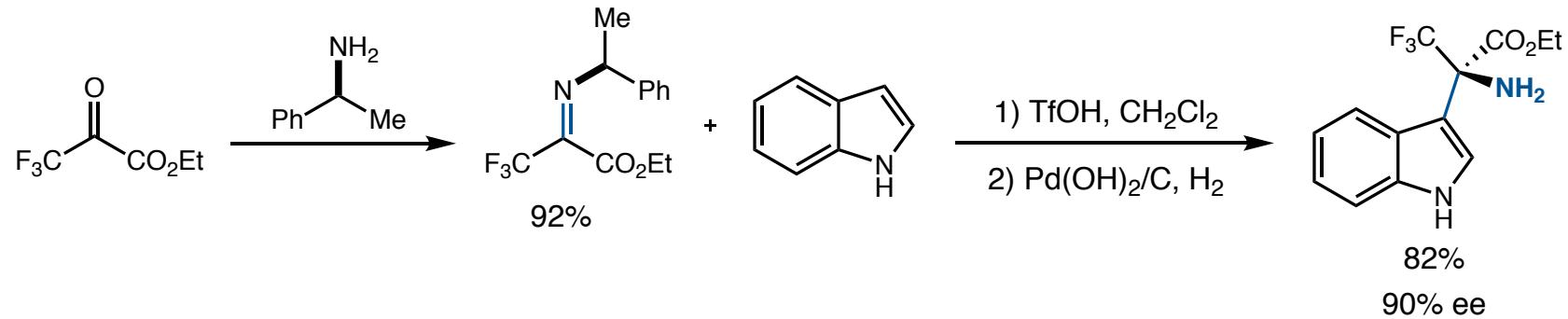
Walborsky, H. M.; Baum, M. E. *J. Org. Chem.* **1956**, *21*, 538.

*Strategy II*  
*Via the Formation of an Imine*

- Imines can be reduced by organometallic reagents



- Chiral imines can influence stereoselectivity on following reaction at  $\text{C}\alpha$

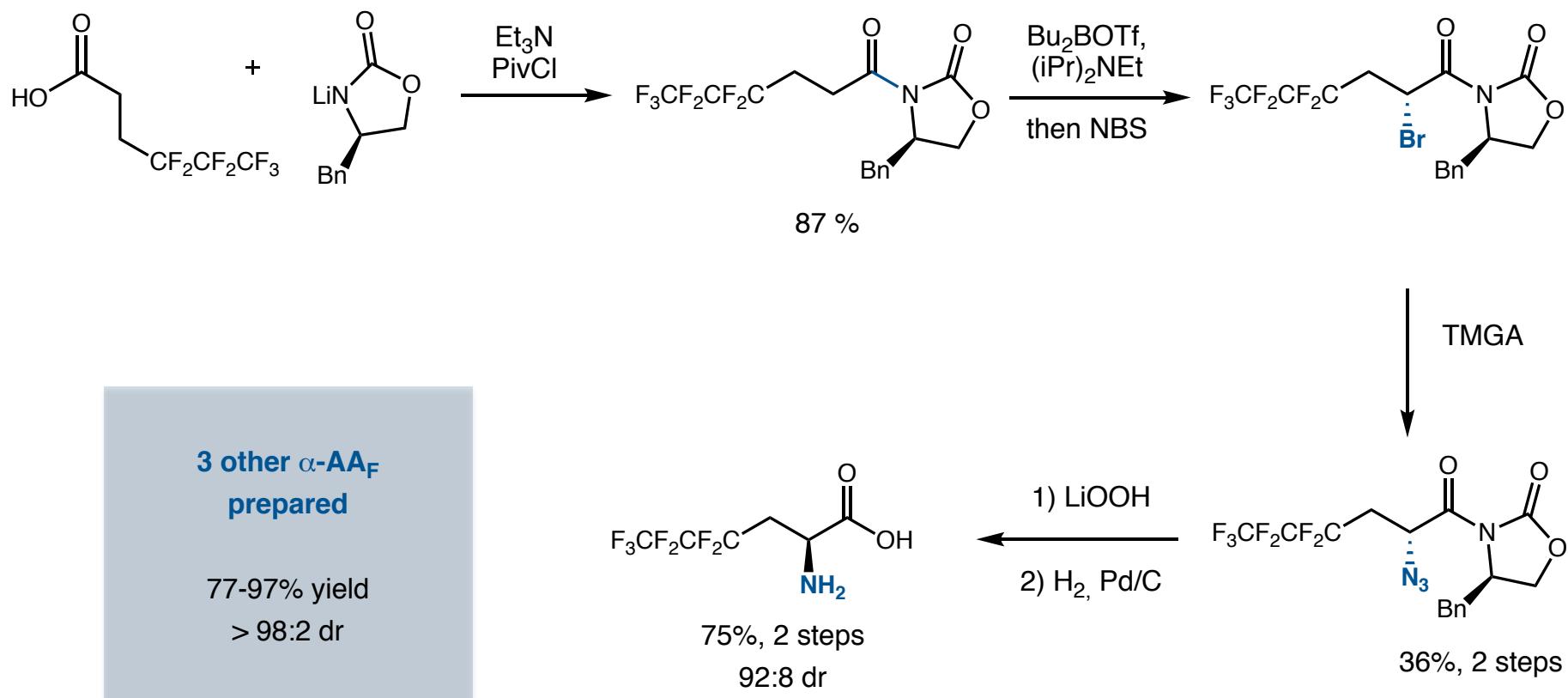


Uneyama, K.; Katagiri, T.; Amii, H. *J. Synth. Org. Chem. Jpn* **2002**, *60*, 1069.

Abid, M.; Teixeira, L.; Torok, B. *Org. Lett.* **2008**, *10*, 933.

*Strategy II*  
*Stereoselective Amination*

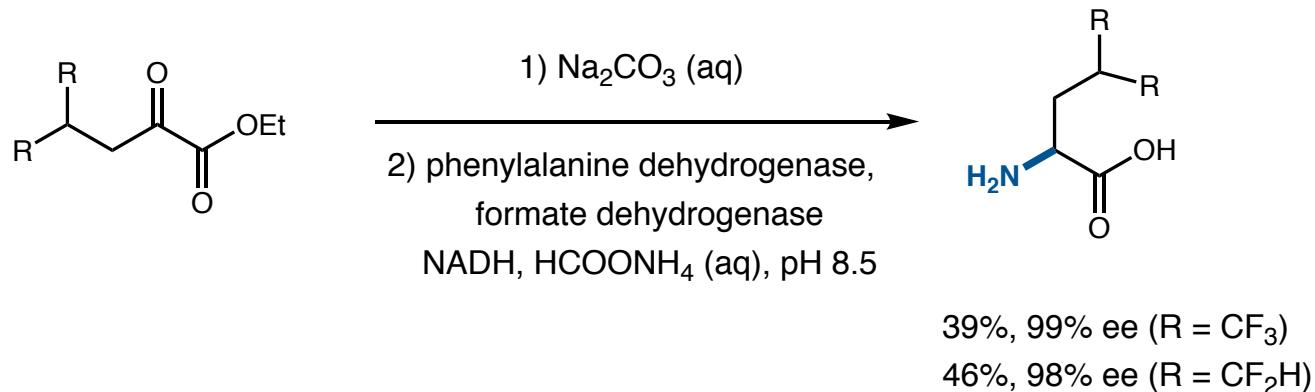
■ Amination directed by Evans auxilairy



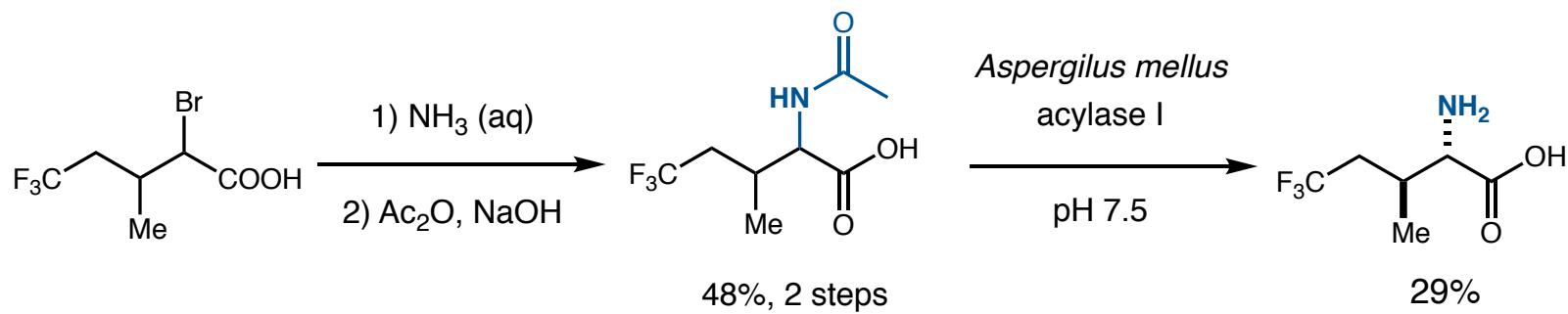
## Strategy II

*Obtention of Single Stereoisomer with Biocatalysts*

### ■ Enzymatic transamination



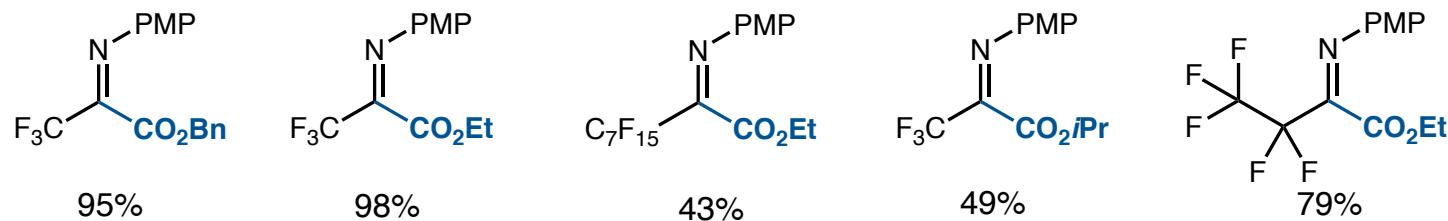
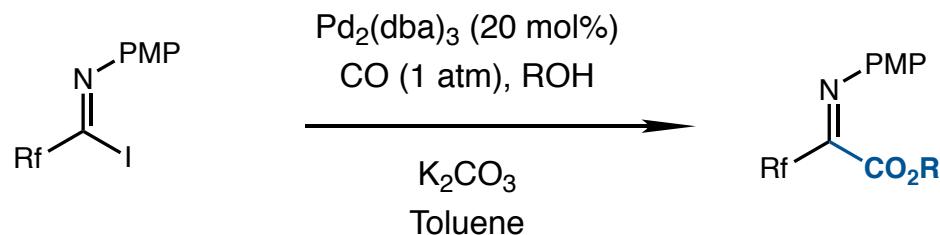
### ■ Kinetic resolution



Chiu, H. P.; Cheng, R. P. *Org. Lett.* **2007**, *9*, 5517.  
 Biava, H.; Budisa, N. J. *Fluorine Chem.* **2013**, *156*, 372.

*Strategy III*  
*Palladium-Catalyzed Alkoxy carbonylation*

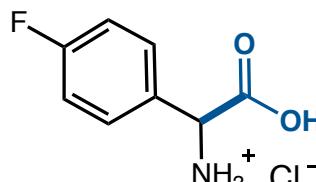
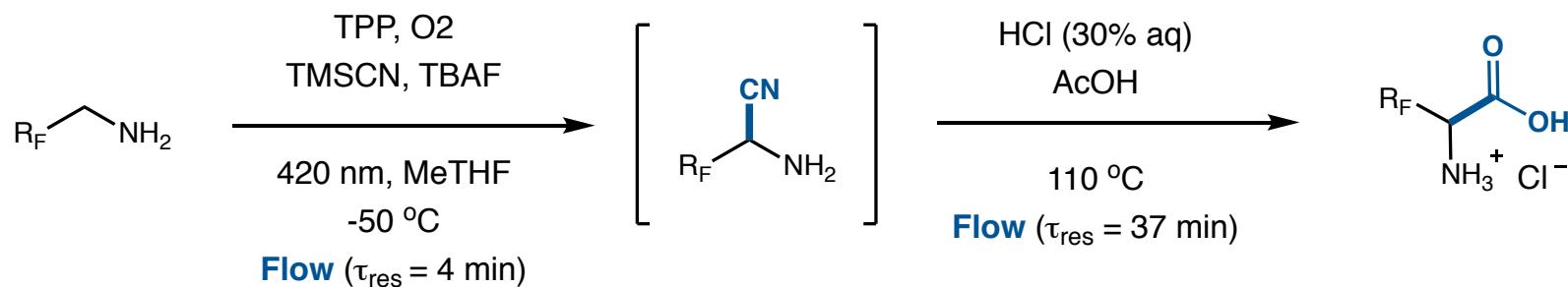
■ Installation of an ester on an imidoyl halide



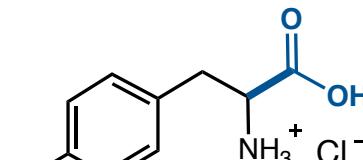
Biava, H.; Budisa, N. Watanabe, H.; Hashizume, Y.; Uneyama, K. *Tetrahedron Lett.* **1992**, *33*, 4333.  
Amii, H.; Kishikawa, Y.; Kageyama, K.; Uneyama, K. *J. Org. Chem.* **2000**, *65*, 3404.

***Strategy III***  
*Photooxidative Cyanation*

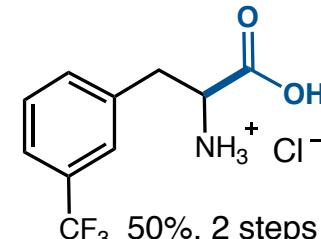
■ Formation and hydrolysis of an  $\alpha$ -aminonitrile in a semi-continuous process



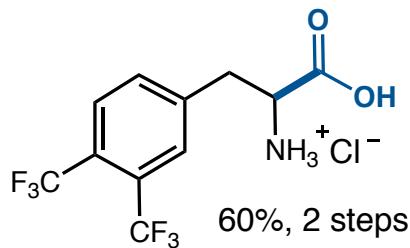
64%, 2 steps



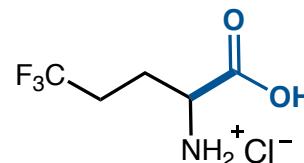
67%, 2 steps



50%, 2 steps



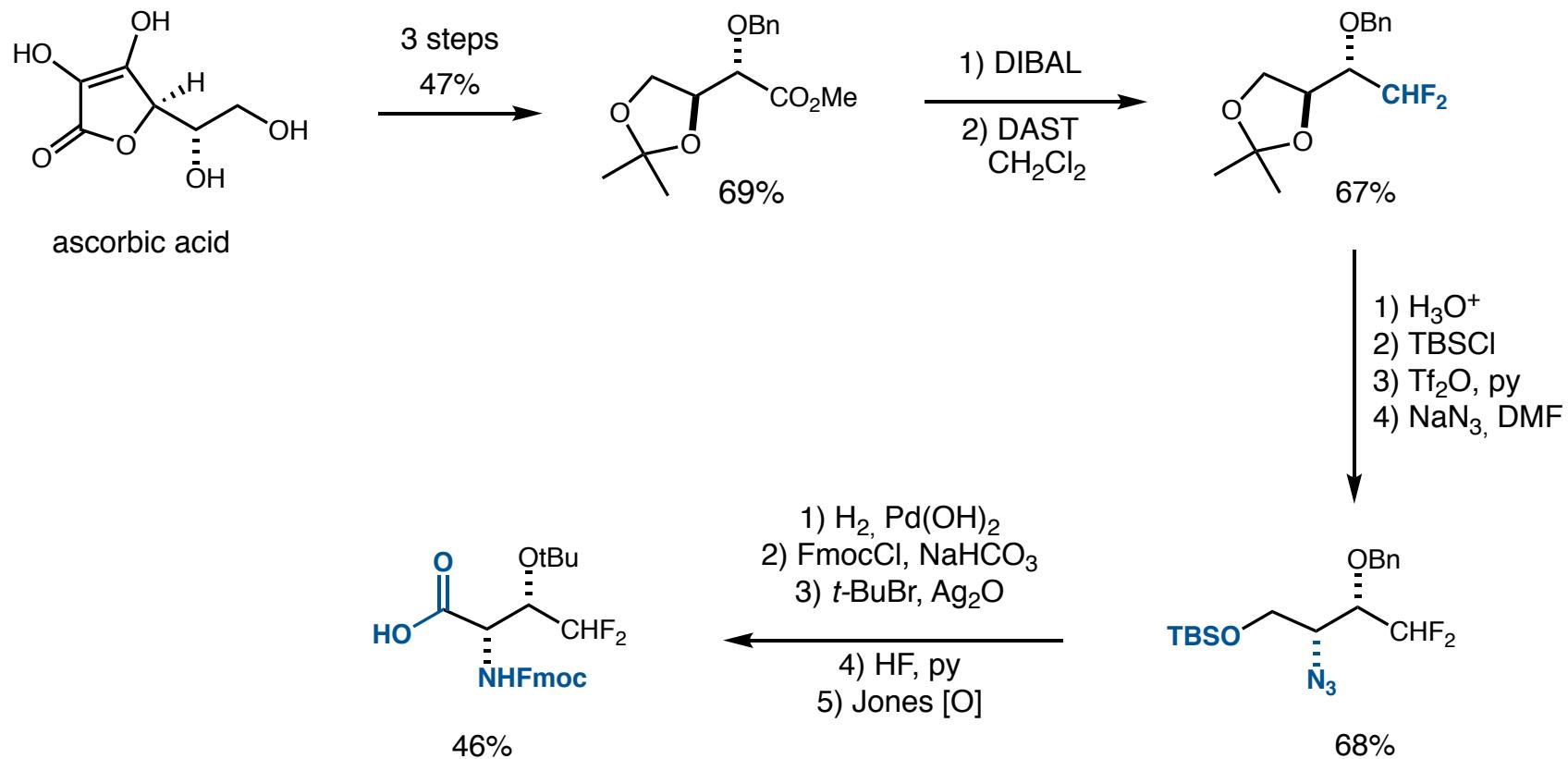
60%, 2 steps



63%, 2 steps

*Strategy IV*  
*Deoxofluorination Followed by Amination*

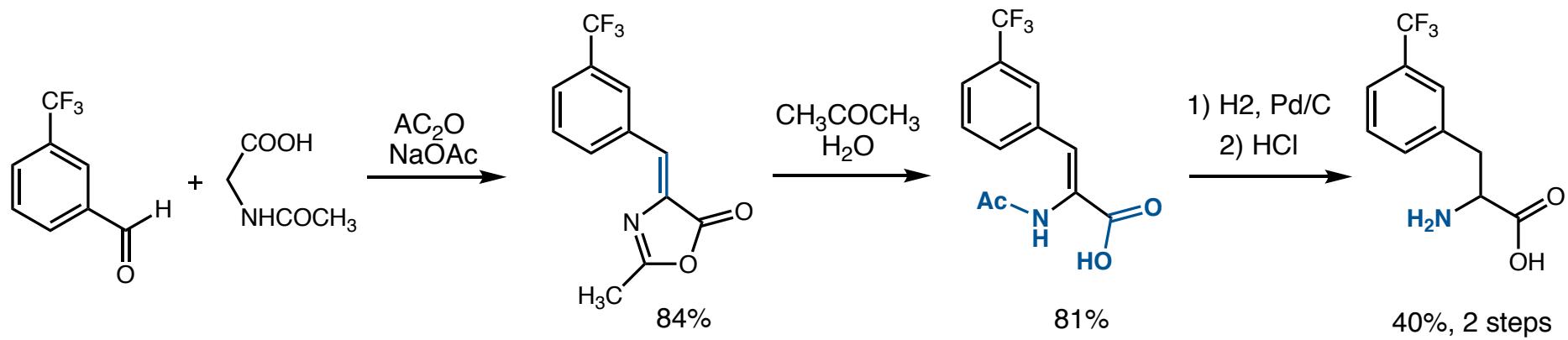
■ Preparation of protected  $\gamma$ -difluorothreonine from ascorbic acid



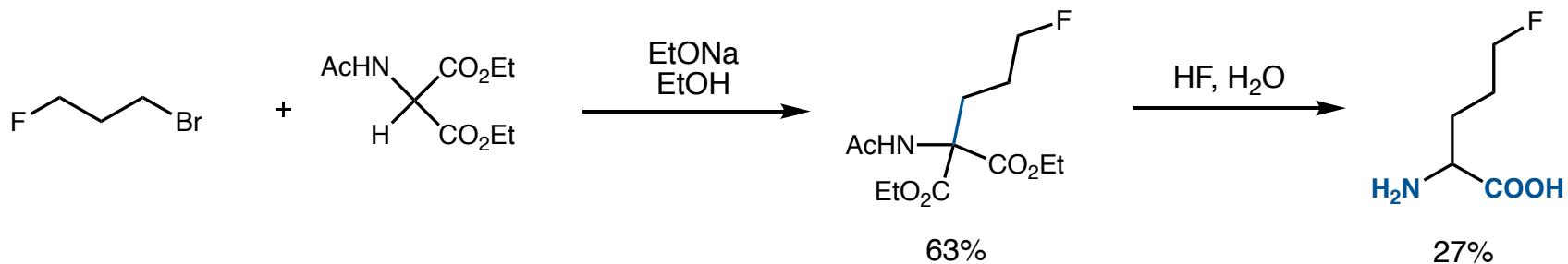
## Strategy V

*Exploiting Basicity of C $\alpha$  on the Glycine Moiety*

### ■ Erlenmeyer synthesis (azlactone) - First $\alpha$ -AAF synthesis (1932)



### ■ S<sub>N</sub>2 reaction with an alkyl halide

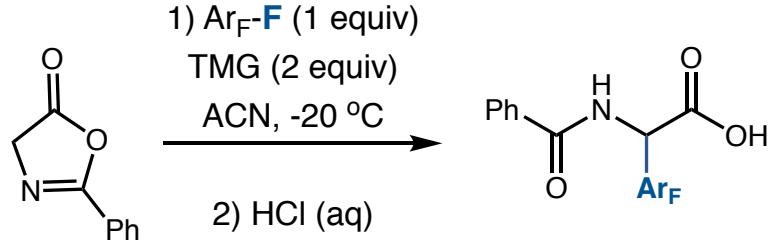


Raasch, M. S. *J. Org. Chem.* **1958**, 23, 1567.  
Tolman, V. *Amino Acids* **1996**, 11, 15.

## Strategy V

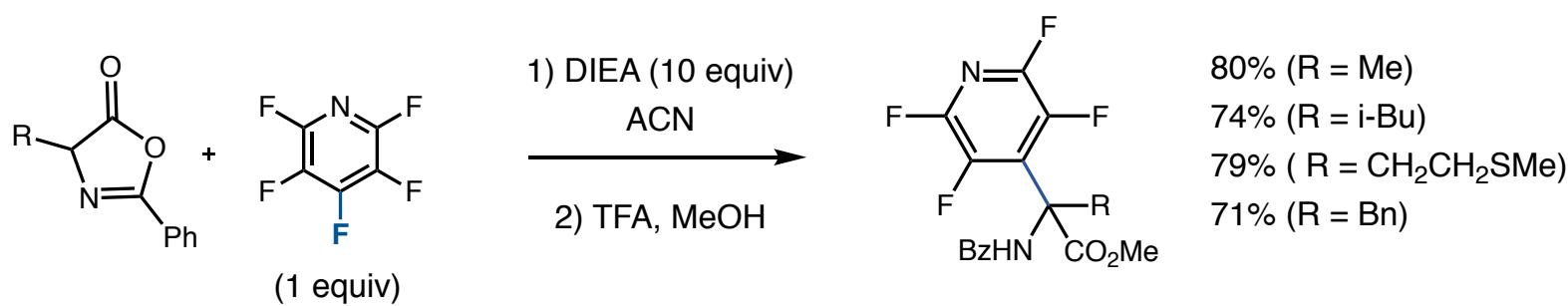
*Use of Oxazolone as  $\alpha$ -Amino Acid Template*

- Oxazolone enolates can be used to perform nucleophilic aromatic substitution



$\text{Ar}_\text{F}-\text{F}$				
$\alpha\text{-AA}_\text{F}$ (%)	87	86	82	78

- Fully substituted  $\alpha$ -amino esters also accessible



# *Synthesis of $\alpha$ -AAF: Proposed Classification*

## Approach A

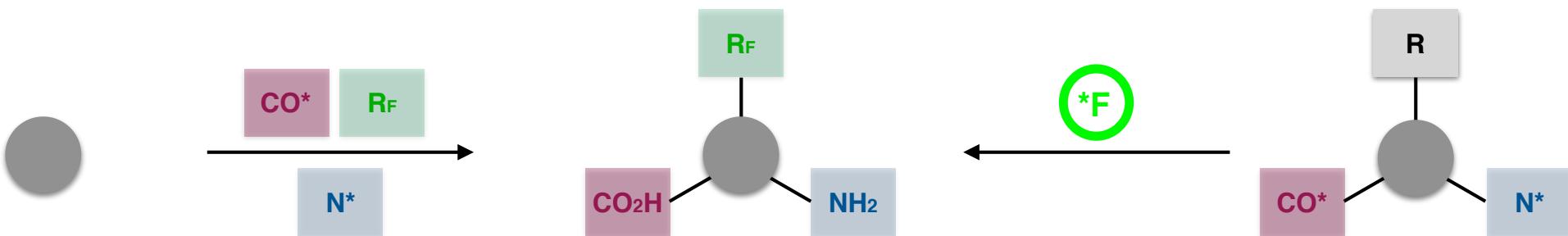
Installation of functionalities

(Strategies 1-5)

## Approach B

Fluorination of side chain

(Strategy 6)



N\*

Amine or *nitrogen function*

CO\*

Carbox. acid or oxygenated function

R<sub>F</sub>

Fluorinated side chain

R

Non fluorinated side chain

\*F

Fluorinating agent

●

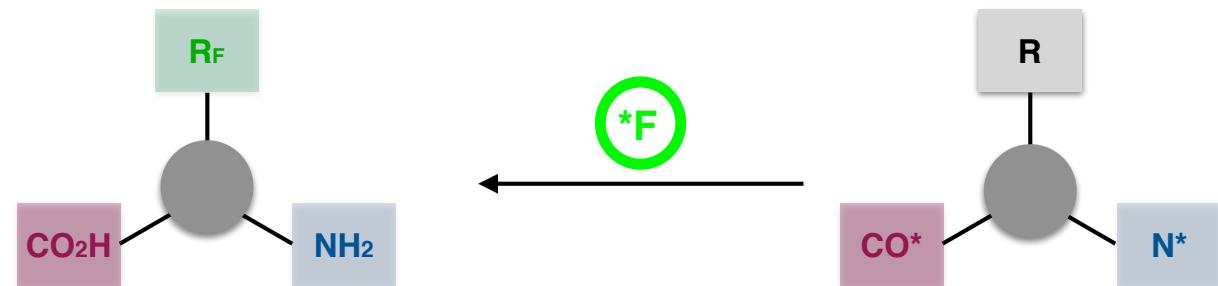
Amino acid C $\alpha$

## *Synthesis of $\alpha$ -AAF: Proposed Classification*

### Approach B

Fluorination of side chain

#### (Strategy 6)



N\*

Amine or *nitrogen function*

R<sub>F</sub>

Fluorinated side chain

CO\*

Carbox. acid or oxygenated function

R

Non fluorinated side chain

\*F

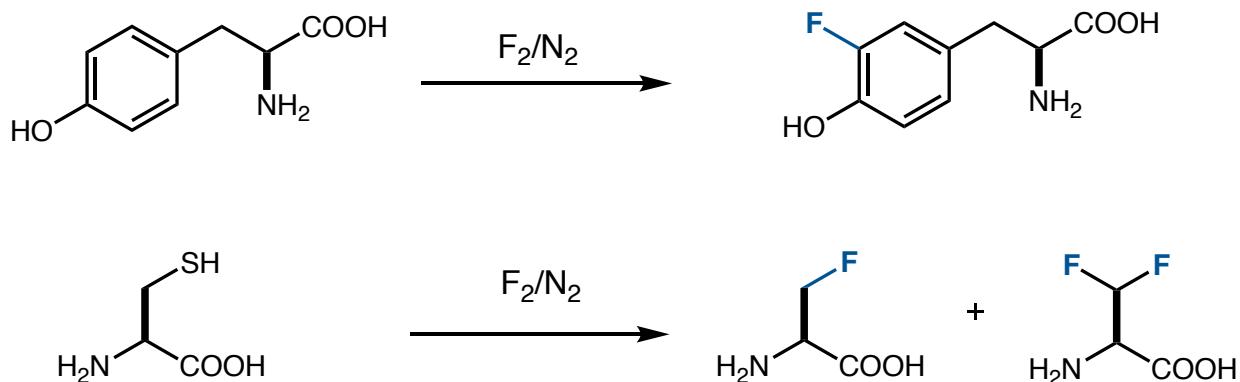
Fluorinating agent



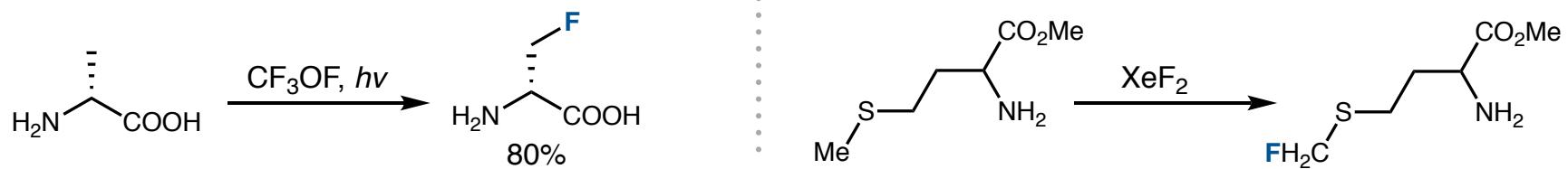
Amino acid C<sub>α</sub>

*Strategy VI*  
*Pioneering Works*

■ Fluorination of canonical  $\alpha$ -AA with  $F_2$



■  $XeF_2$  and  $CF_3OF$  also explored



Smith, R. W.; et al *J. Fluorine Chem.* **1987**, *37*, 267.

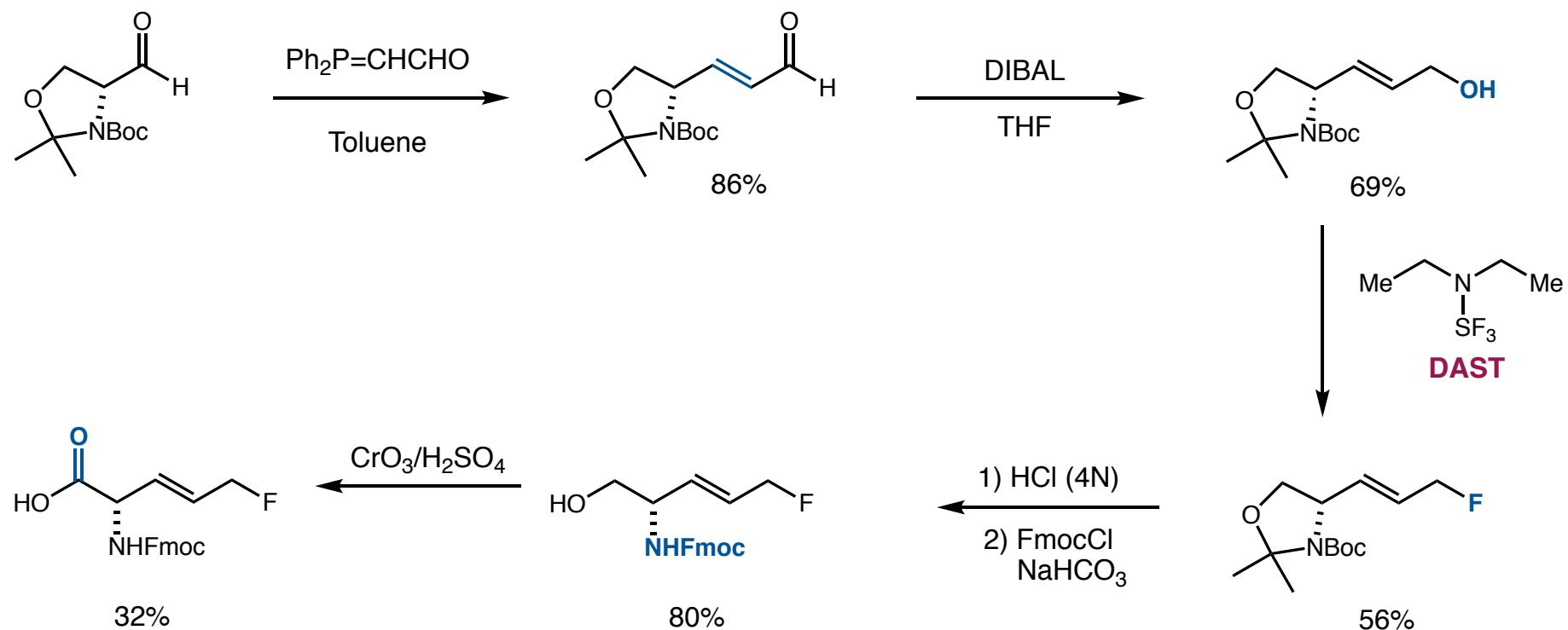
Kollonitsch, J.; Marburg, S.; Perkins, L. M. *J. Org. Chem.* **1976**, *41*, 3107.

Kollonitsch, J.; Barash, L. *J. Am. Chem. Soc.* **1976**, *98*, 5591.

Janzen, A. F.; Wang, P. M. C.; Lemire, A. E. *J. Fluorine Chem.* **1983**, *22*, 557.

*Strategy VI*  
*Deoxofluorination on Side Chain*

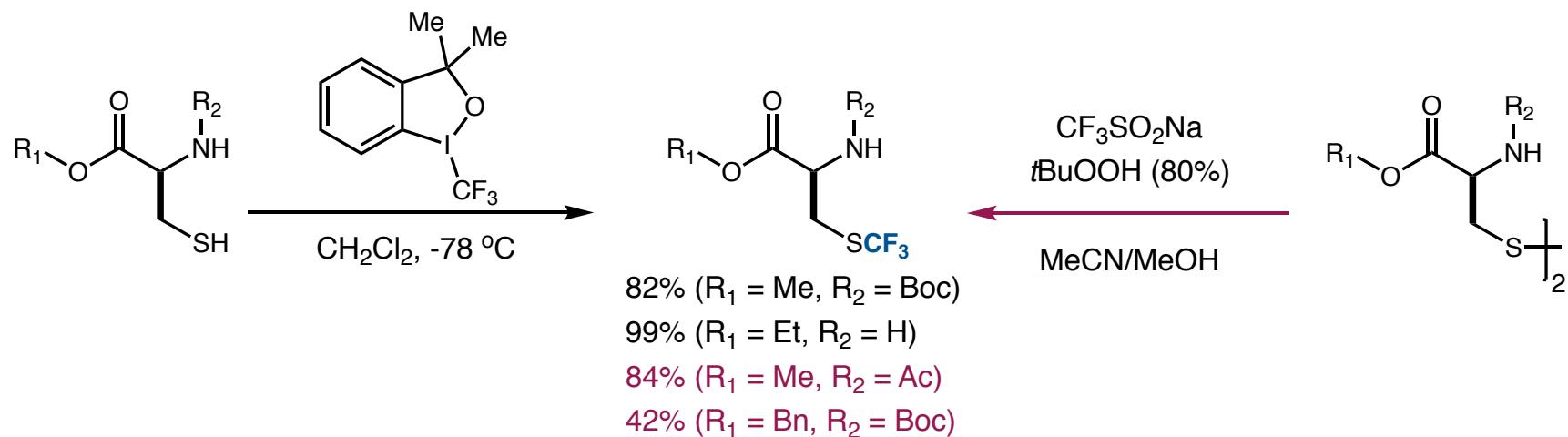
■ Preparation of  $\alpha$ -AAF from Garner's aldehyde



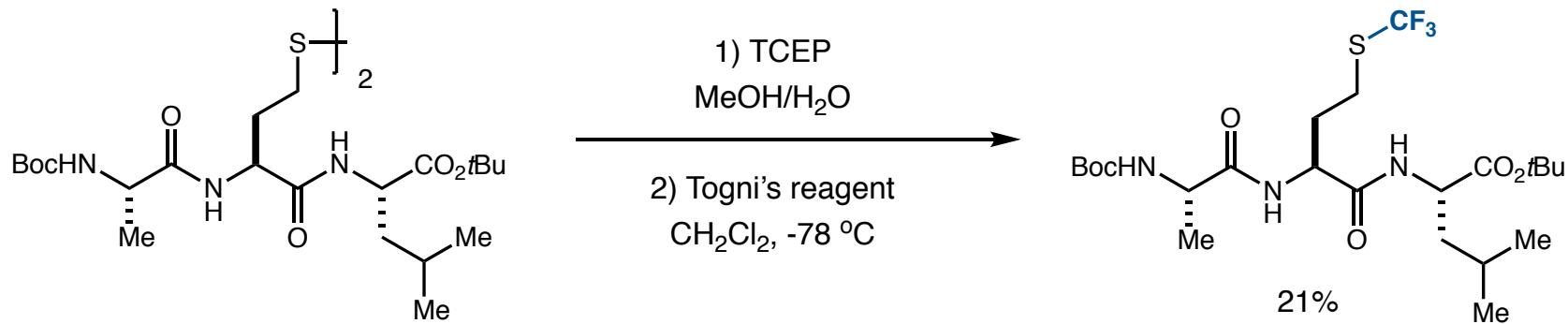
## Strategy VI

*Addition of Trifluoromethyl Radical*

### ■ Trifluoromethylation of L-cysteine



### ■ Togni's reagent has shown potential for late-stage functionalisation on tripeptides

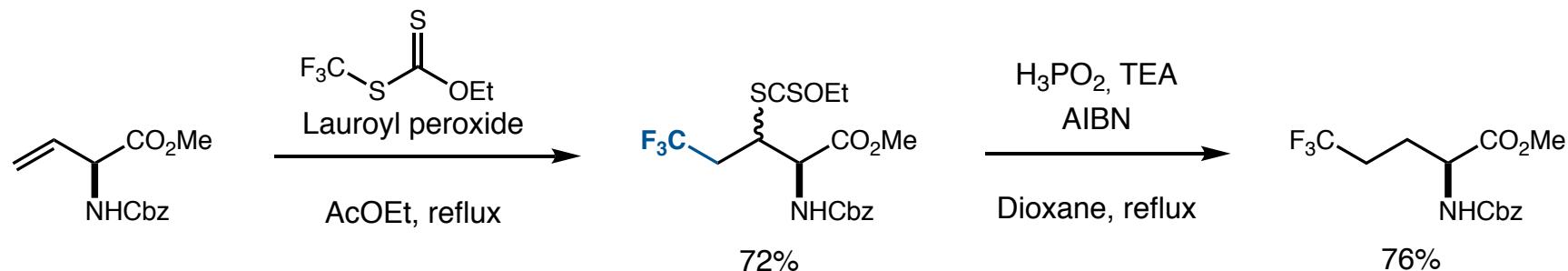


Kielsch, I.; Eisenberger, P.; Togni, A. *Angew. Chem., Int. Ed.* **2007**, *46*, 754.

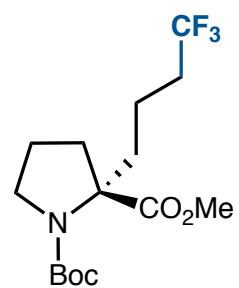
Gadais, C.; Saraiva-Rosa, N.; Chelain, E.; Pytkowicz, J.; Brigaud, T. *Eur. J. Org. Chem.* **2017**, *2017*, 246.

*Strategy VI*  
*Via the Addition of a Xanthate*

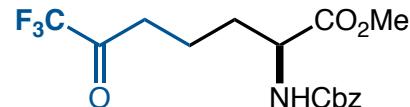
- Trifluoromethylation of protected vinyl glycine by a  $\text{CF}_3$  radical generated from a xanthate



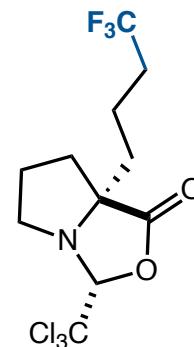
- Other type of fluorinated side chains accessed from different olefins



60%, 2 steps



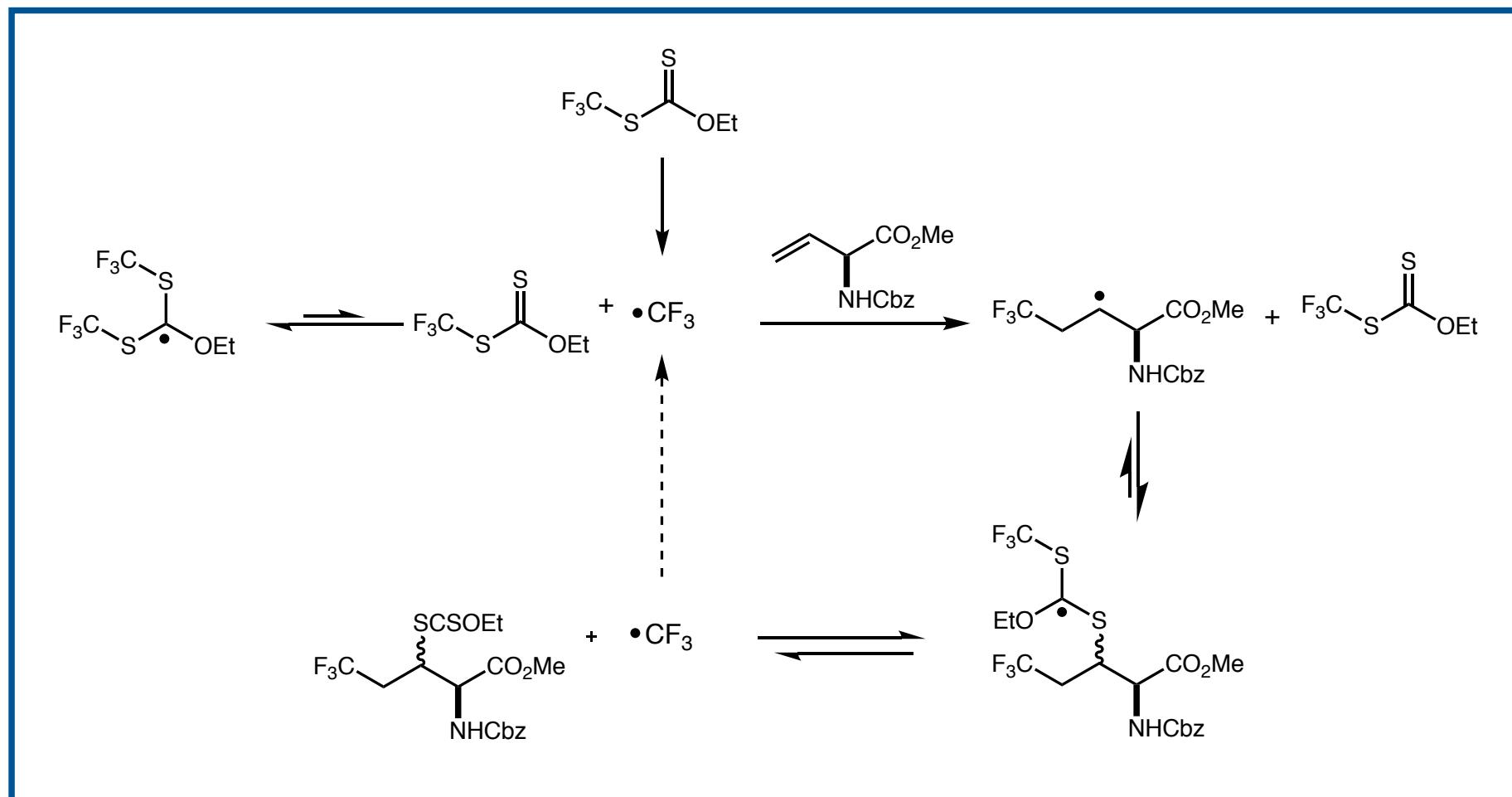
53%, 2 steps



52%, 2 steps

*Strategy VI*  
*Via the Addition of a Xanthate*

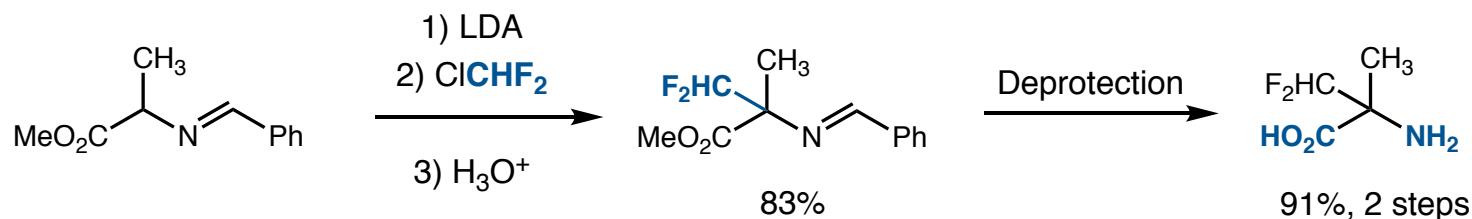
■ Xanthate acts as a  $\text{CF}_3$  “resevoir”



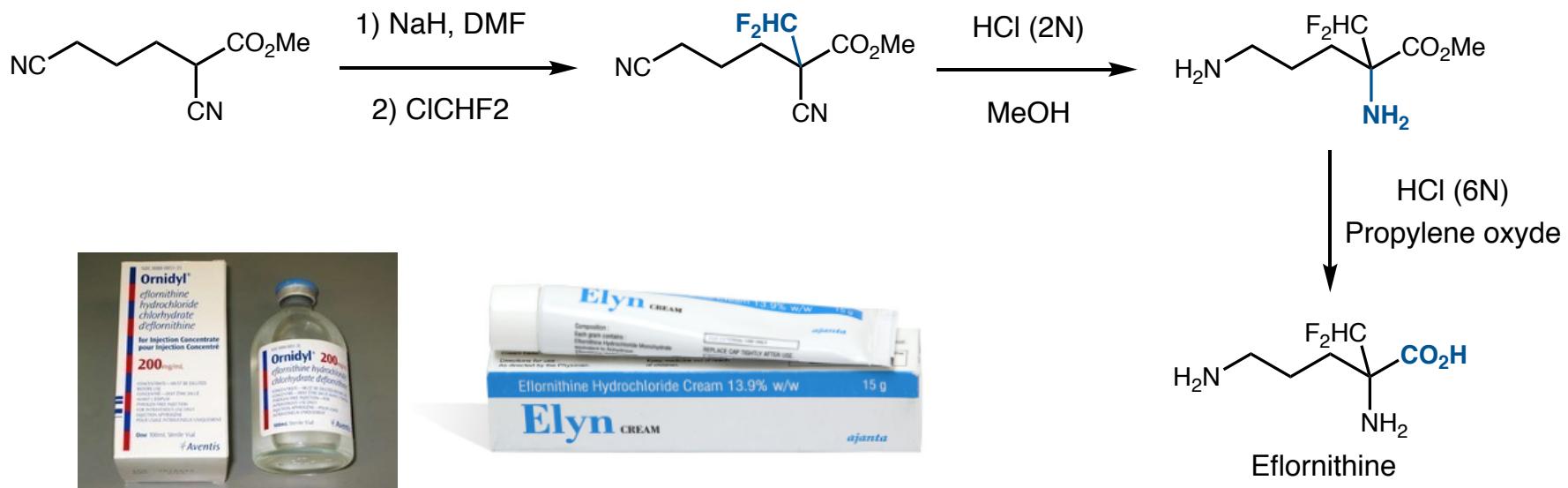
## Strategy VI

*Difluoromethylation with Difluorocarbene Source*

### ■ Trapping of a difluorocarbene by a carbanion generated at C $\alpha$



### ■ Strategy employed in the synthesis of eflornithine



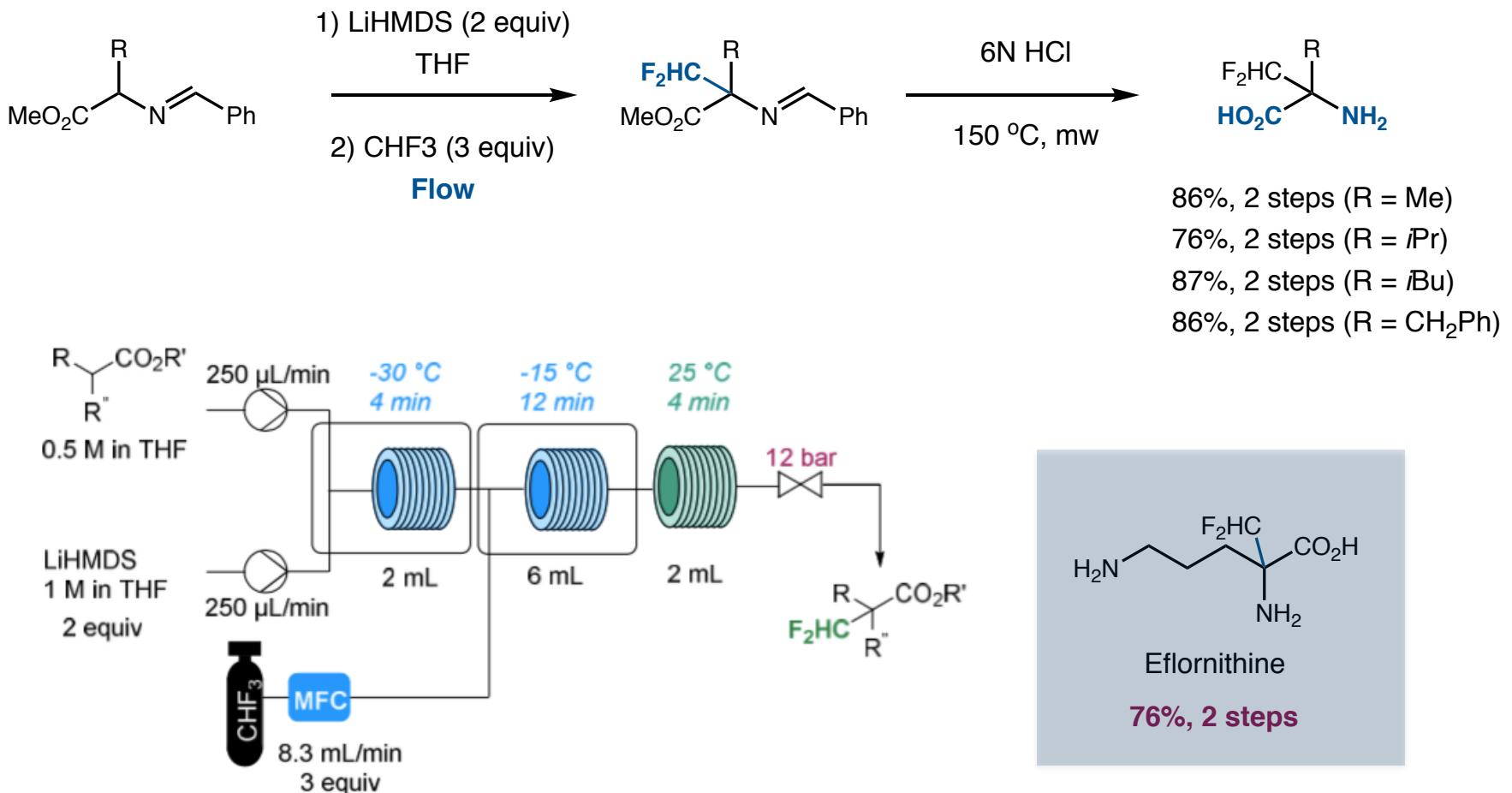
Bey, P.; Vevert, J. P.; Van Dorsselaer, V.; Kolb, M. *J. Org. Chem.* **1979**, *44*, 2732.

Seki, M.; Suzuki, M.; Matsumoto, K. *Biosci. Biotechnol. Biochem.* **1993**, *57*, 1024.

## Strategy VI

*Difluoromethylation with Difluorocarbene Source*

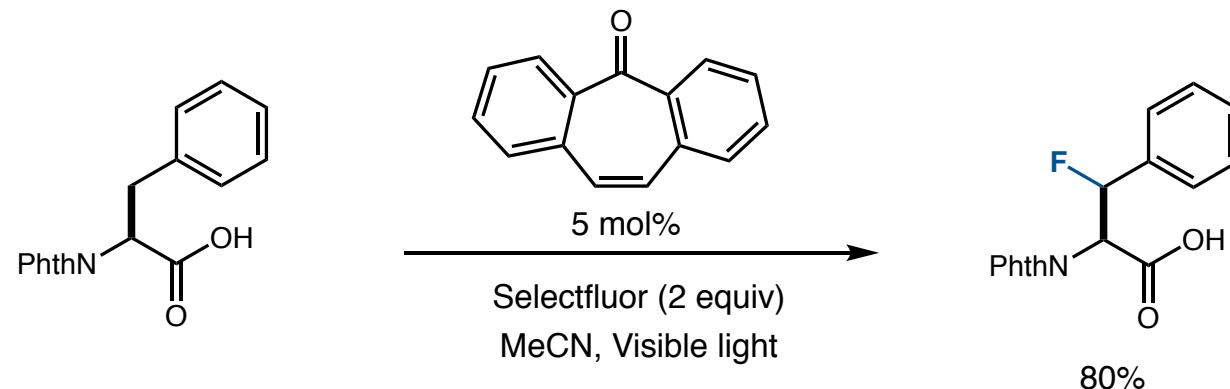
### ■ Use of fluoroform as a $\text{CHF}_2$ source



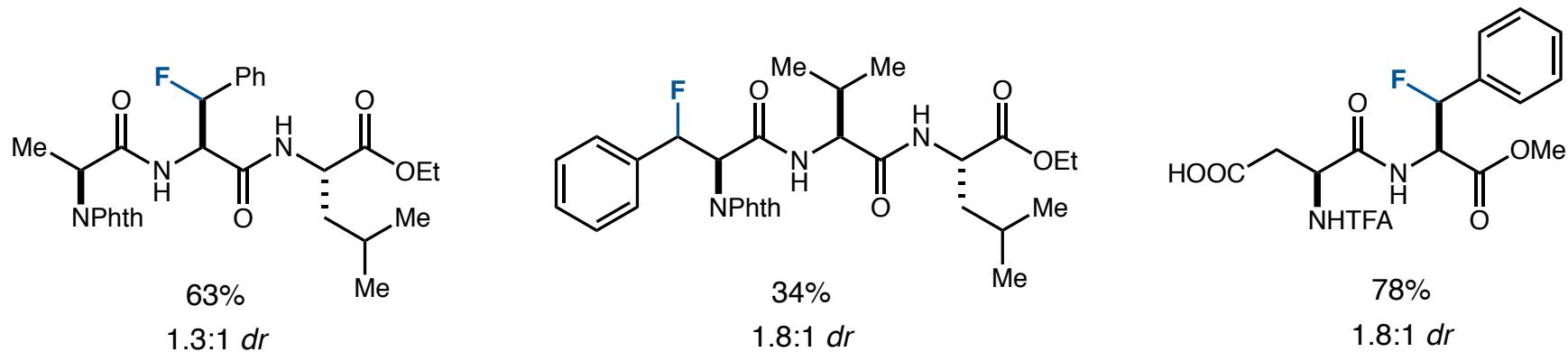
## Strategy VI

### Benzylic C-H Fluorination

#### ■ Visible-light sensitized benzlic C-H fluorinated with Selectfluor



#### ■ Selective late-stage fluorination of benzylic position in peptides



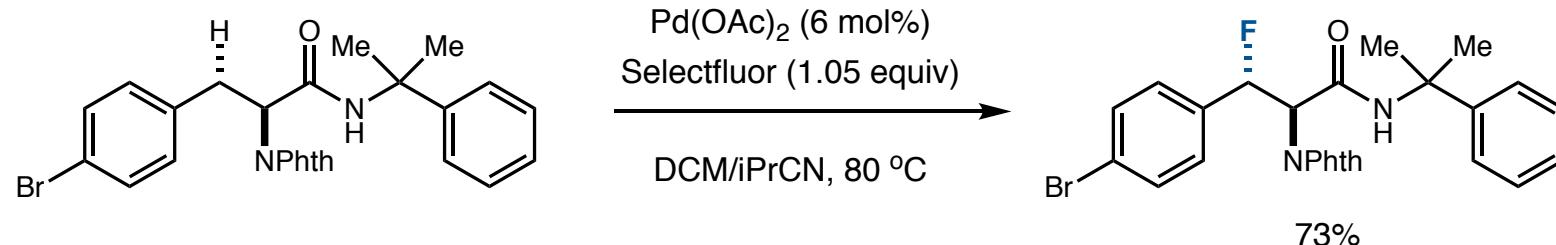
Bume, D. D.; Pitts, C. R.; Jokhai, R. T.; Lectka, T. *Tetrahedron* **2016**, *72*, 6031.

Bloom, S.; McCann, M.; Lectka, T. *Org. Lett.* **2014**, *16*, 6338.

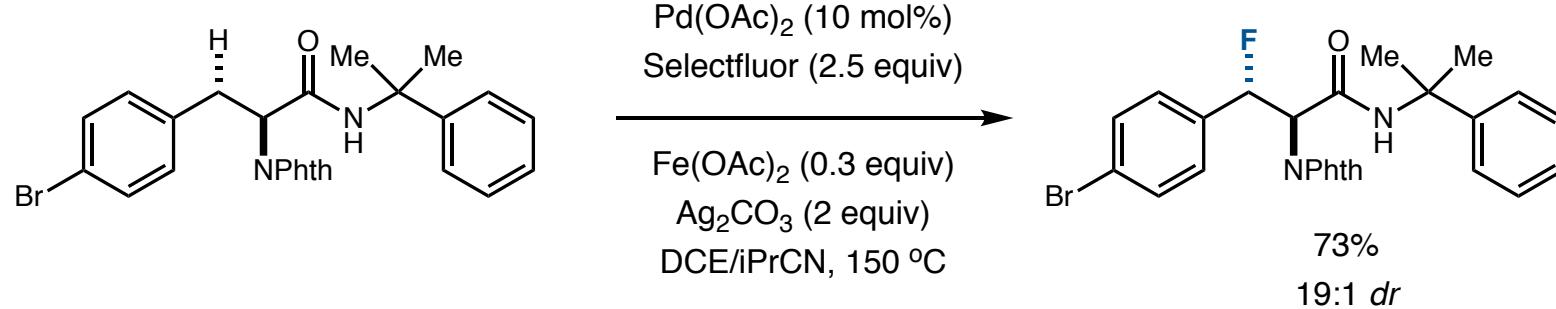
*Strategy VI*  
*Benzylic C-H Fluorination*

■ Pd-catalyzed fluorination using 2-(pyridin-2-yl)isopropyl (PIP) as protecting and directing group

■ Shi (JACS, 2015)



■ Ge (Org Lett, 2015)



Zhang, Q.; Yin, X.-S.; Chen, K.; Zhang, S.-Q.; Shi, B.-F. *J. Am. Chem. Soc.* **2015**, *137*, 8219.

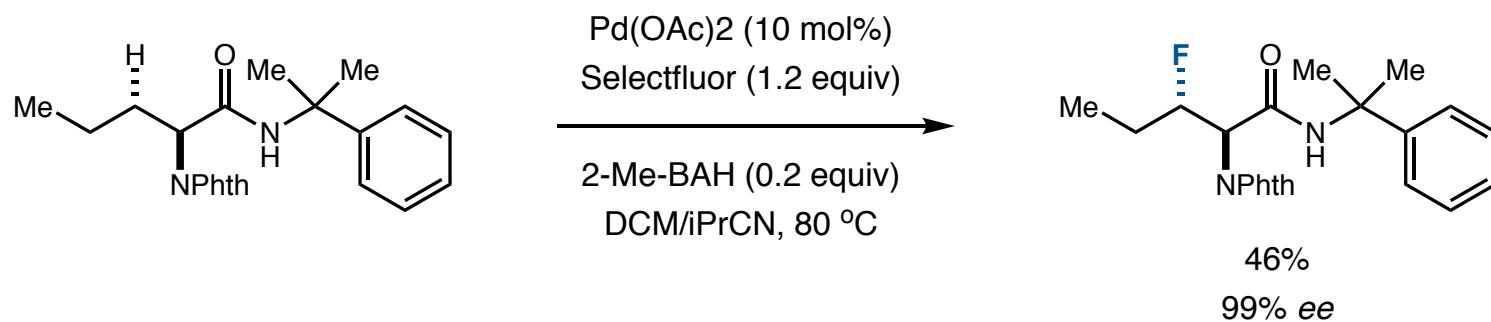
Miao, J.; Yang, K.; Kurek, M.; Ge, H. *Org. Lett.* **2015**, *17*, 3738.

## Strategy VI

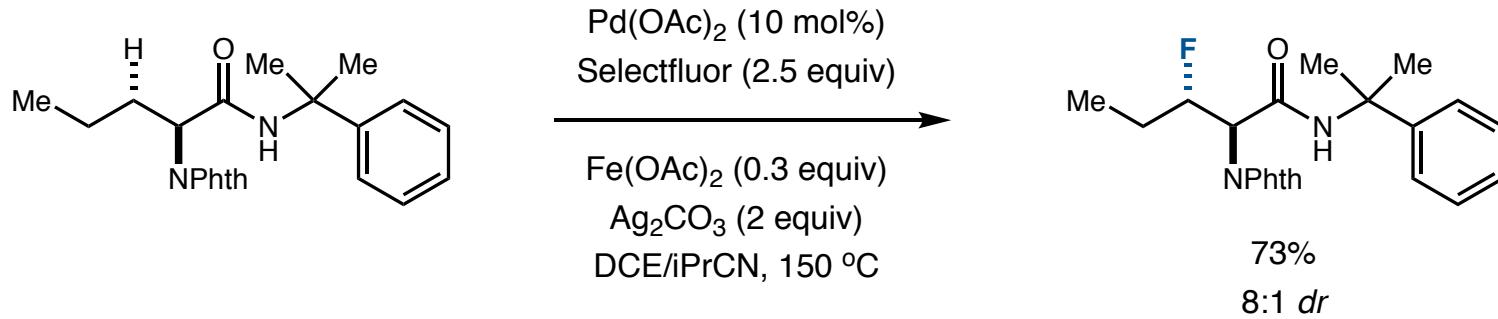
*Fluorination of Unactivated C-H Bonds*

■ Pd-catalyzed fluorination using 2-(pyridin-2-yl)isopropyl (PIP) as protecting and directing group

■ Shi (JACS, 2015)



■ Ge (Org Lett, 2015)



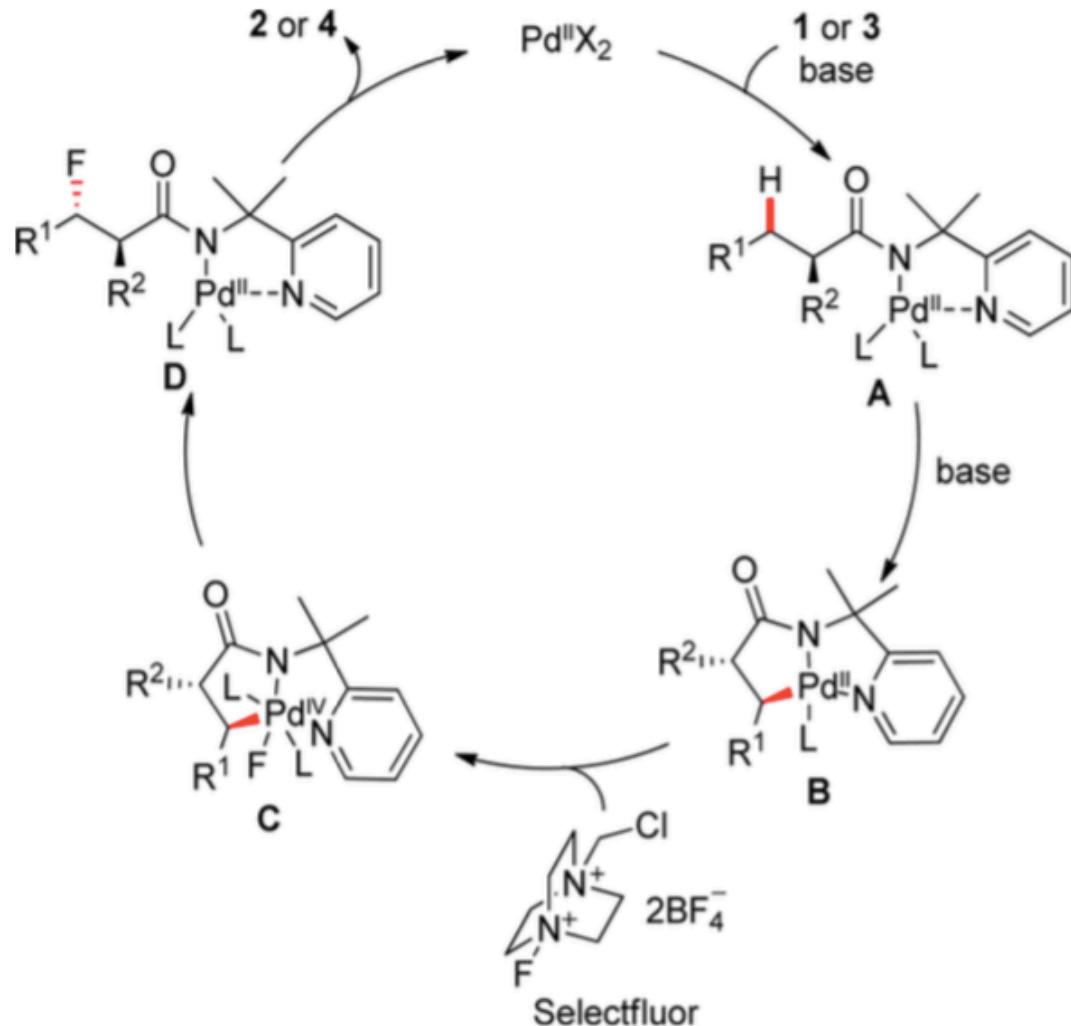
Zhang, Q.; Yin, X.-S.; Chen, K.; Zhang, S.-Q.; Shi, B.-F. *J. Am. Chem. Soc.* **2015**, *137*, 8219.

Miao, J.; Yang, K.; Kurek, M.; Ge, H. *Org. Lett.* **2015**, *17*, 3738.

## Strategy VI

Fluorination of Unactivated C-H bonds

- Ge's proposed mechanism invokes fluorination of a Pd(II) palladacyclic center

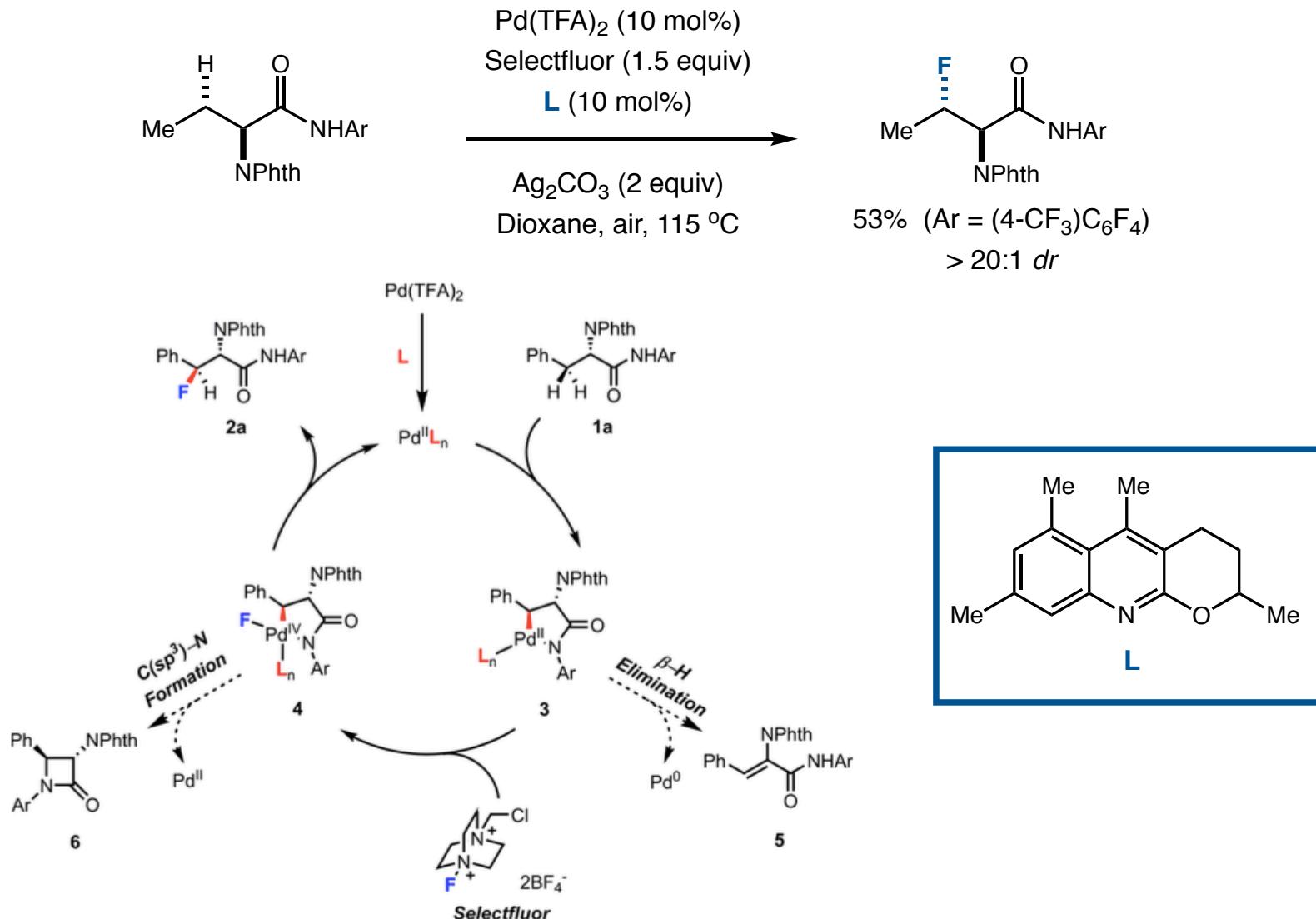


Miao, J.; Yang, K.; Kurek, M.; Ge, H. *Org. Lett.* **2015**, *17*, 3738.

## Strategy VI

*Fluorination of Unactivated C-H Bonds*

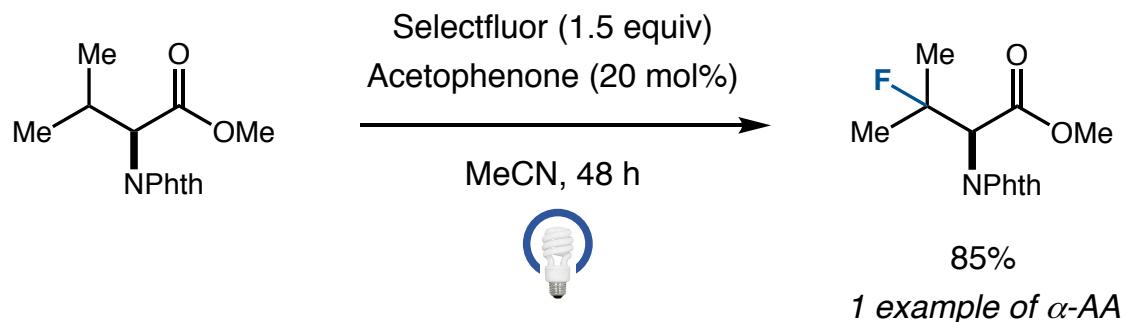
### ■ Similar system driven by a Quinoline-based ligand



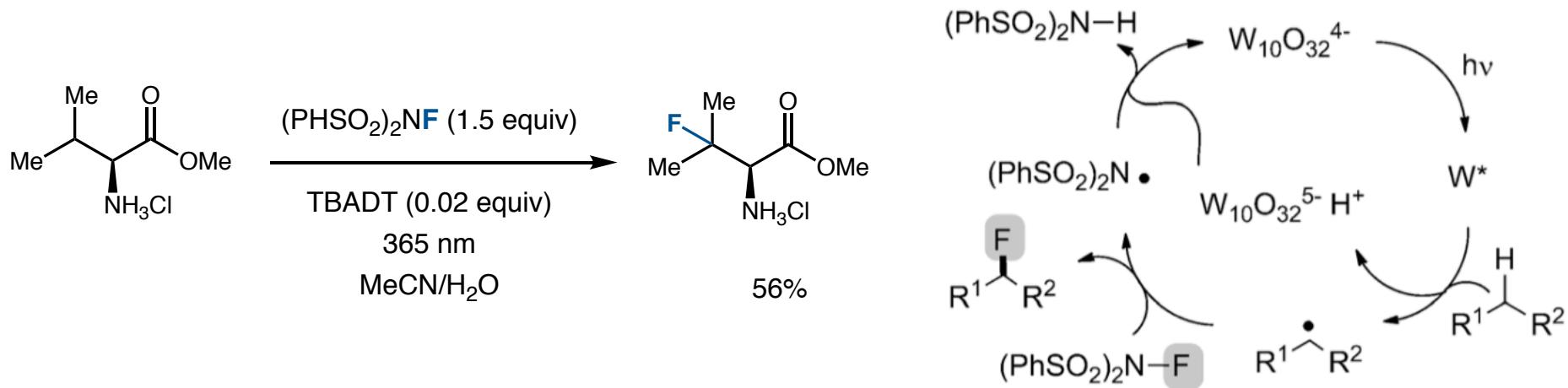
## Strategy VI

*Fluorination of Unactivated C(sp<sup>3</sup>)-H Bonds*

### ■ Visible-light promoted fluorination with benzophenone



### ■ Decatungstate-catalyzed C-H fluorination of branched aliphatic with NFSI



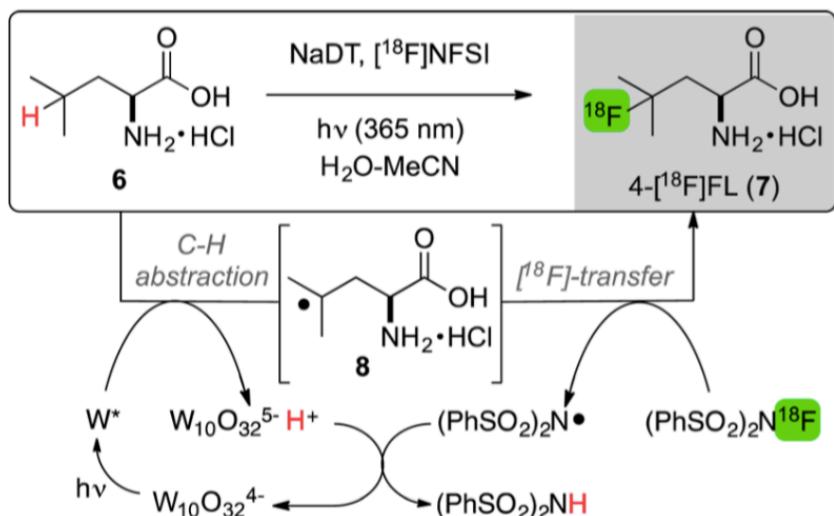
Xia, J.-B.; Zhu, C.; Chen, C. *Chem. Commun.* **2014**, *50*, 11701.

Halperin, S. D.; Fan, H.; Chang, S.; Martin, R. E.; Britton, R. *Angew. Chem., Int. Ed.* **2014**, *53*, 4690.

## Strategy VI

*Fluorination of Unactivated C(sp<sup>3</sup>)-H Bonds*

■ Decatungstate-catalyzed C-H fluorination: suitable system for preparation of <sup>18</sup>F-labeled  $\alpha$ -AAF



- Preparation/purification of marked NFSI: 10 min
- Reaction time: 40 min
- Purification time: 10 min
- Total process from <sup>18</sup>F generation: 60 min
- Product isolated directly as formulation for IV injection

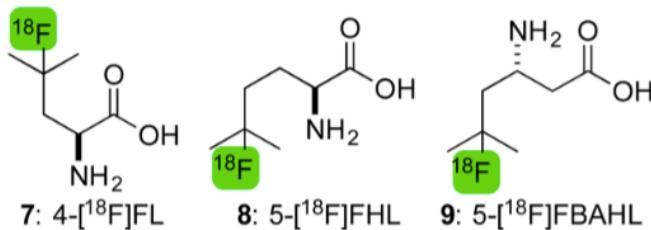
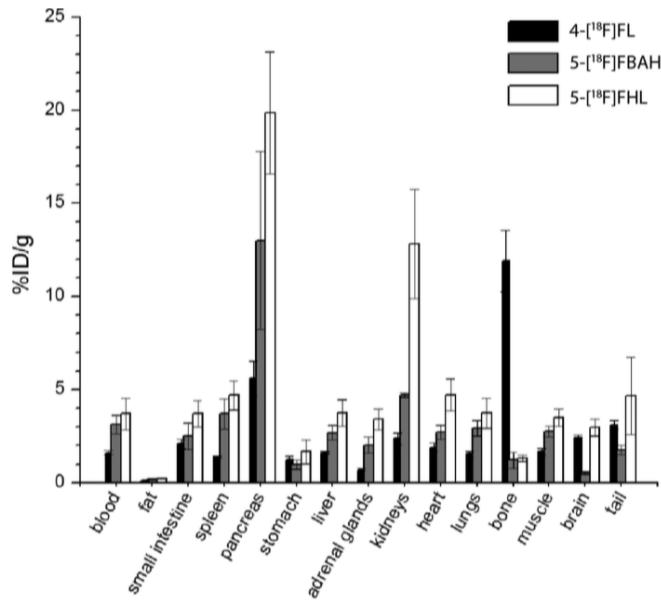
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<sup>18</sup> F-labelled amino acid	7: 4-[ <sup>18</sup> F]FL	8: 5-[ <sup>18</sup> F]FHL	9: 5-[ <sup>18</sup> F]FBAHL	10: 3-[ <sup>18</sup> F]FV	11: 3-[ <sup>18</sup> F]FI
radiochemical yield (%) <sup>b,c</sup>	23.3 ±3.3%	27.9 ±3.3%	29.8 ±0.7%	6.4 ±0.4%	<5% <sup>d</sup>

## *In Vivo Biological Investigation*

*<sup>18</sup>F Marked  $\alpha$ -AAF Prepared by Fluorination of Unactivated C-H Bonds*

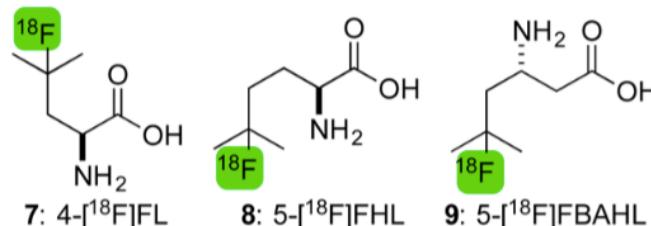
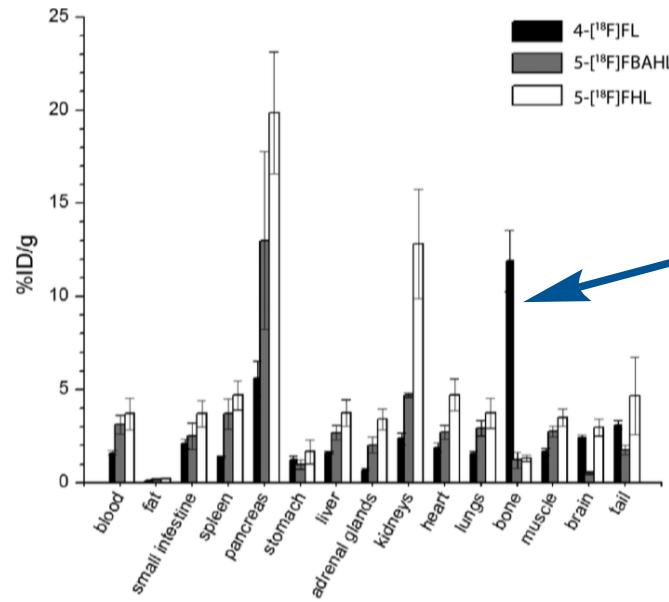
- 5-[<sup>18</sup>F]FHL showed a normal distribution profile in healthy mice after 60 minutes



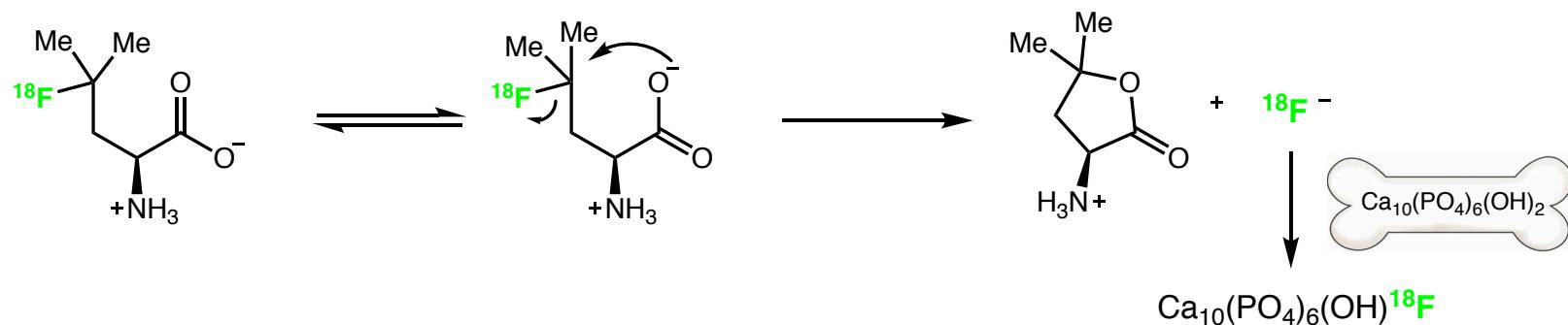
## In Vivo Biological Investigation

$^{18}\text{F}$  Marked  $\alpha$ -AAF Prepared by Fluorination of Unactivated C-H Bonds

- 5-[ $^{18}\text{F}$ ]FHL showed a normal distribution profile in healthy mice after 60 minutes



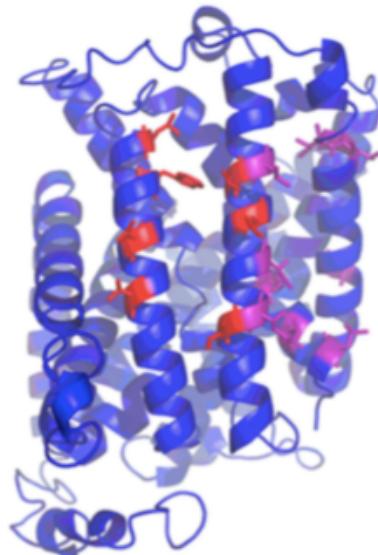
- 4-[ $^{18}\text{F}$ ]FL accumulation in bones reveals *in vivo* production of fluoride by lactonisation



## *In Vivo Biological Investigation*

$^{18}\text{F}$  Marked  $\alpha$ -AAF Prepared by Fluorination of Unactivated C-H Bonds

- LAT1 surexpression in tumor can be revealed by 5-[ $^{18}\text{F}$ ]FHL in PET imaging



LAT1

$\alpha$ -AA transport protein

Surexpressed in many cancer cell lines

