

Nickel-Catalyzed Cycloadditions

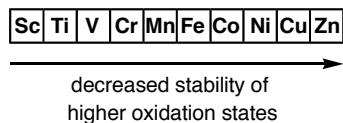
Tristan Lambert
MacMillan Group Meeting
January 31, 2001

- Properties of nickel
- Introduction to metal-catalyzed cycloadditions
- Nickel-catalyzed cycloadditions

Review of metal-catalyzed cycloadditions: Lautens, *Chem. Rev.*, **1996**, *96*, 49.

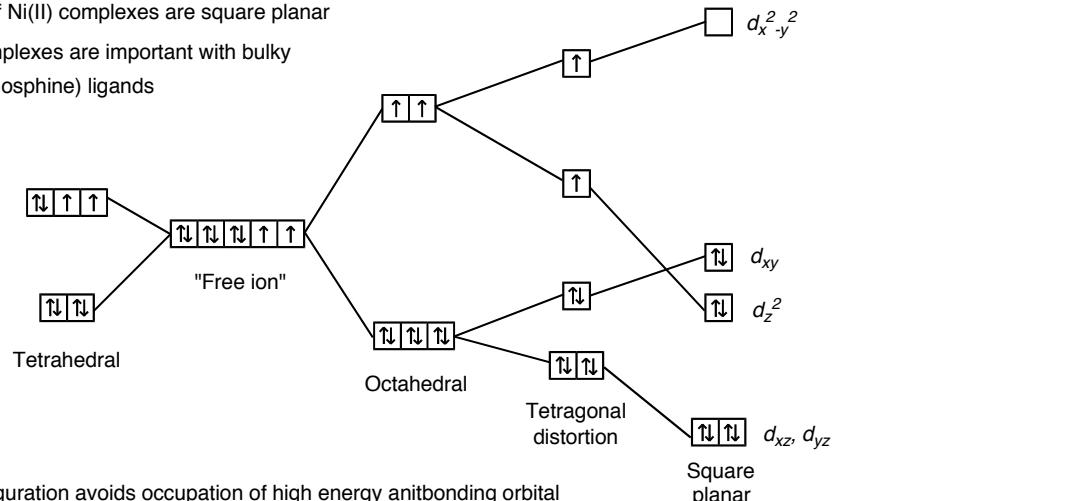
Physical Properties of Nickel

- Across the first row of transition metals, there is a trend towards decreased stability of higher oxidation states such that only Ni(II) occurs in the normal chemistry of the element



Cotton, *Advanced Inorganic Chemistry*, 6th ed.; p. 835

- The vast majority of Ni(II) complexes are square planar
- Tetrahedral Ni complexes are important with bulky (i.e. phosphine) ligands



- Square planar configuration avoids occupation of high energy antibonding orbital

Greenwood, *Chemistry of the elements*; p. 1347

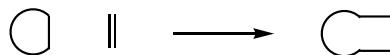
Introduction to Metal-Catalyzed Cycloadditions

■ Definition of Cycloaddition:

-A reaction of two separate π systems in which a ring is formed with two more σ bonds and two fewer π bonds than the reactants



■ Note: As an exception to the above definition, in certain cases one of the π systems can be a reactive σ bond.



- Cycloadditions have been promoted by heat, light, Lewis acids, high pressure, and sonication

- Many of these conditions require the presence of polarized functional groups in the substrate to facilitate the transformation. In general reaction of unactivated olefins, acetylenes, or dienes is notoriously poor and extreme conditions are necessary to achieve good yields of cycloadducts
 - Metal catalysts provide new opportunities for highly selective cycloaddition reactions since complexation of a metal to a reactant significantly modifies its reactivity, thus allowing for improved reactivity and novel chemistry
 - It should be emphasized that while the products of these reactions are clearly cycloadducts, most if not all reactions proceed in a stepwise fashion and probably involve a cyclization as a key event

Lautens, *Chem. Rev.*, 1996, 96, 49.

[2 + 1] Cycloadditions: Ni(II) Catalyzed Cyclopropanations of Electron Deficient Olefins

Metal-Stabilized Alkyl Carbenes

		Yield (with respect to CH_2Br_2)	
	10 mol\% NiBr_2 $0.6 \text{ eq NaI}, 0.8 \text{ eq Zn}$ <hr/> $0.5 \text{ eq CH}_2\text{Br}_2, 0^\circ\text{C}, 42\text{h}$		92%
	10 mol\% NiBr_2 $1.0 \text{ eq NaI}, 0.8 \text{ eq Zn}$ <hr/> $0.5 \text{ eq CH}_2\text{Br}_2, 0^\circ\text{C}, 24\text{h}$		90%
	10 mol\% NiBr_2 $1.0 \text{ eq NaI}, 0.8 \text{ eq Zn}$ <hr/> $0.5 \text{ eq CH}_2\text{Br}_2, 0^\circ\text{C}, 96\text{h}$		97%

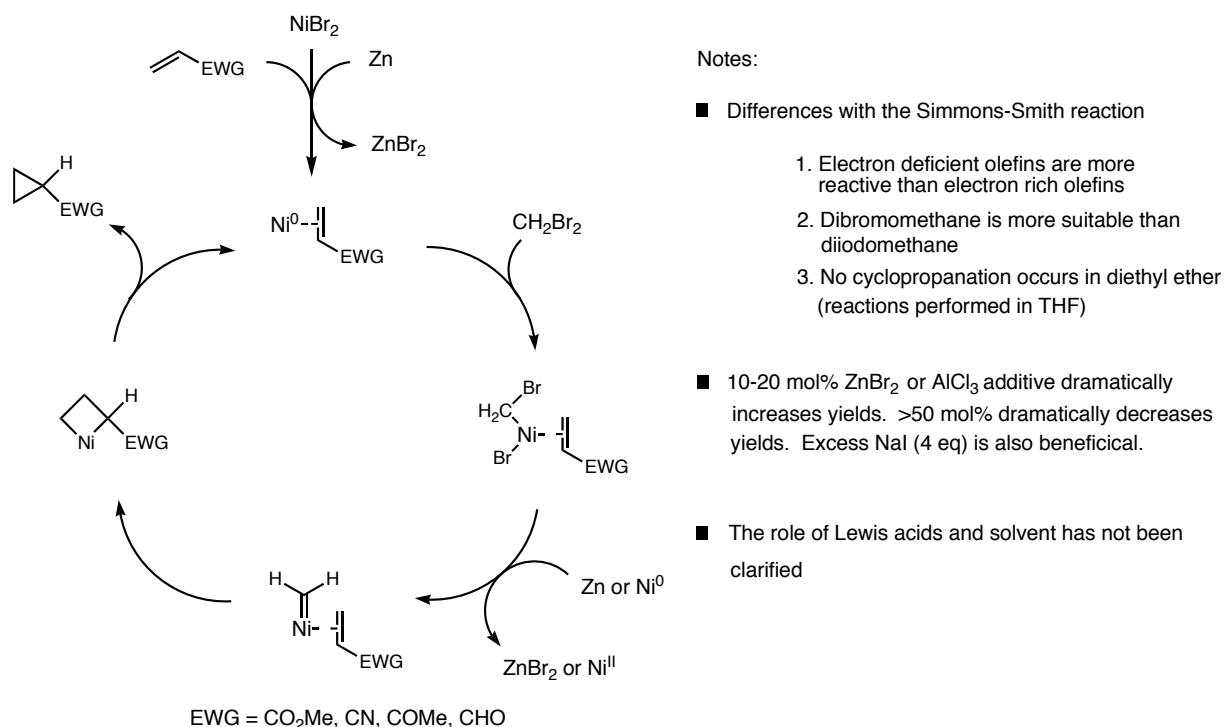
Kanai, *Chem. Lett.*, **1979**, 1979.
Kanai, *Bull. Chem. Soc. Jpn.*, **1983**, 53, 1025.

General Reactivity Trends with Various Catalysts

catalyst	reactive olefins
$\text{Rh}_2(\text{OAc})_4$	styrene, enol ethers
$\text{Cu}(\text{acac})_2$, $\text{CuClP}(\text{OMe})_3$	enamines, alkyl substituted olefins
$\text{Cu}(\text{OTf})_2$, $\text{Cu}(\text{OTf})$, $\text{Cu}(\text{BF}_4)_2$	terminal olefins
$\text{Pd}(\text{OAc})_2$, PdCl_2 , $\text{Pd}(\text{PPh}_3)_4$	styrene, strained, conjugated terminal olefins; α,β -unsaturated carbonyl compounds
$\text{Ni}(\text{COD})_2$, $\text{Ni}(\text{PPh}_3)_4$, NaI/Zn and NiBr_2	α,β -unsaturated carbonyl compounds, acrylonitrile

Catalytic Cycle of Ni(II) Catalyzed Cyclopropanation

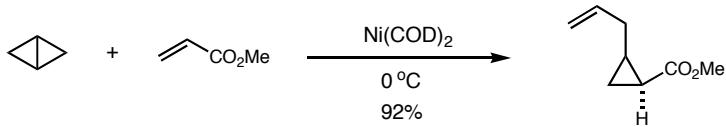
■ Catalytic Cycle



Kanai, *Bull. Chem. Soc. Jpn.*, **1983**, 53, 1025.

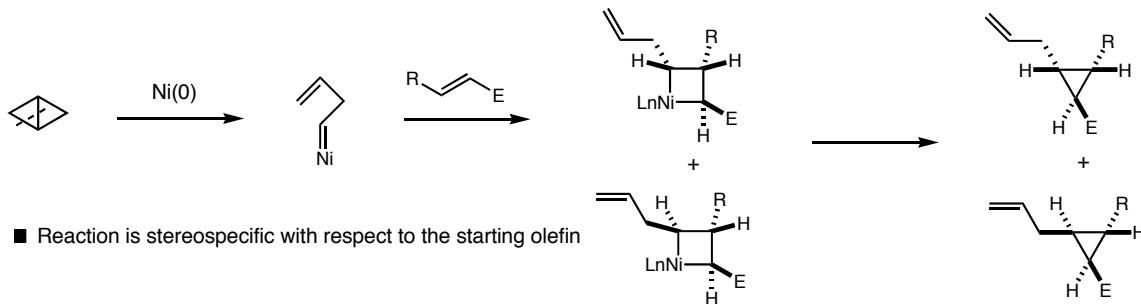
[2 + 1] Cycloadditions: Nickel Carbenes From Highly Strained Hydrocarbons

■ Bicyclo[1.1.0]butane: strain energy 66 kcal/mol

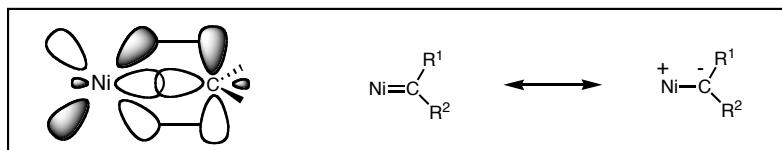


Noyori, *Tetrahedron Lett.*, **1974**, 1749.

■ Mechanism:



■ Nature of bonding in Ni-carbene

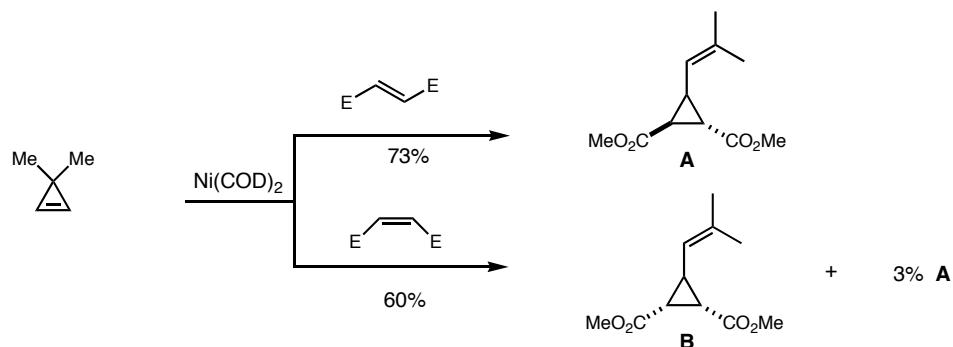


■ Carbenoids involving metals with high backbonding capacity have considerable ylide character

■ Nickel has low electronegativity → carbenenoid carbon nucleophilic

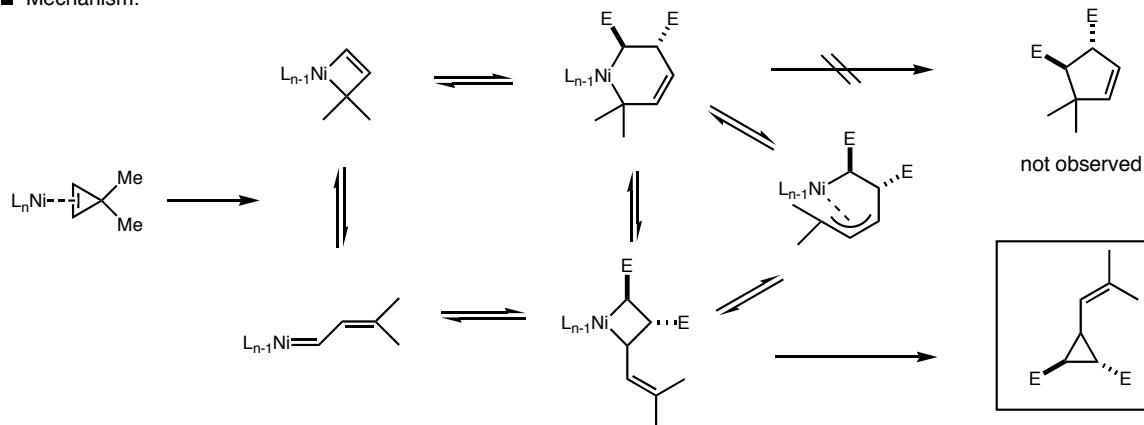
Noyori, *Tetrahedron Lett.*, **1973**, 1691.

[2 + 1] Cycloadditions: Vinyl Carbenes



■ Cis maleate undergoes Ni-catalyzed isomerization leading to small amounts of trans product

■ Mechanism:

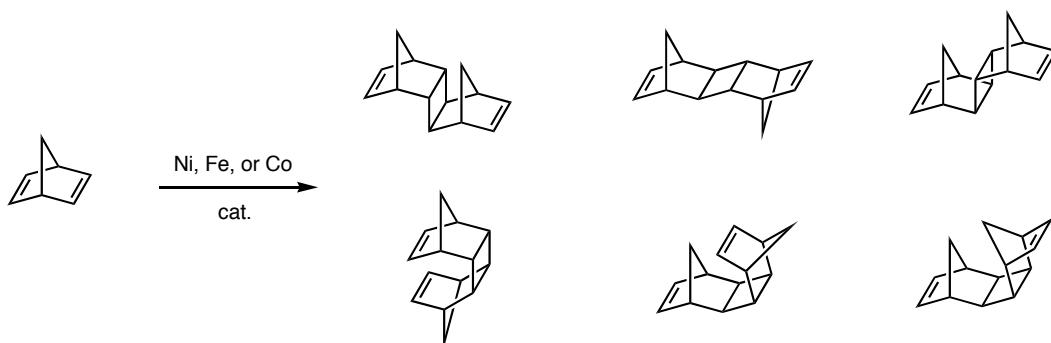


Binger, *Chem. Ber.*, 117, 1984, 1551.

[2 + 2] Cycloadditions: Catalysis of Thermally Forbidden Processes

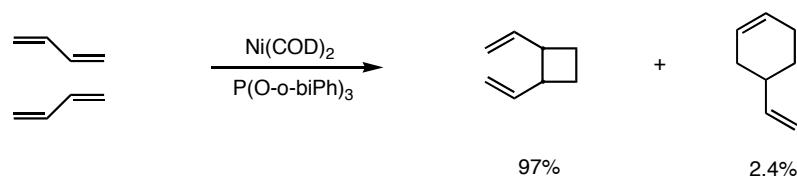
Early Examples

■ Dimerization of Norbornadiene



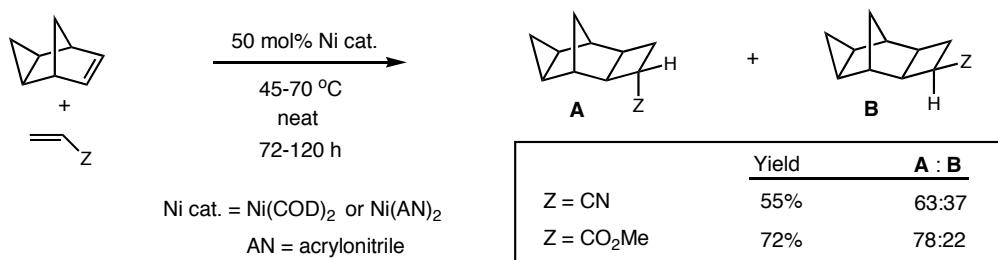
Vallerino, *J. Chem. Soc.*, 1957, 2287.

■ Dimerization of 1,3-butadiene



Heimbach, *ACIEE*, 1967, 6, 800.

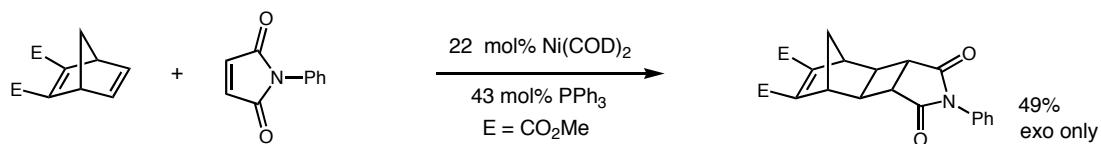
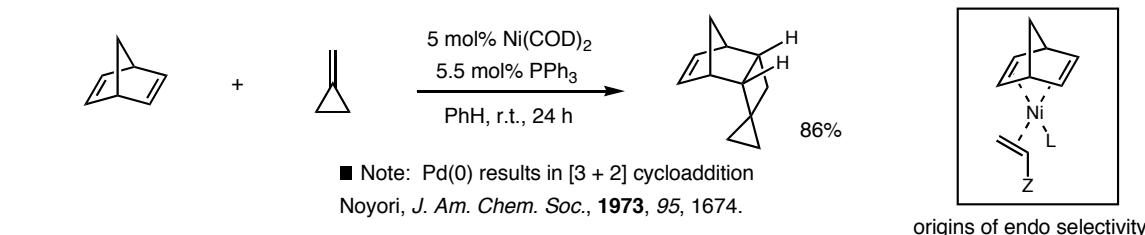
[2 + 2] Cycloadditions: Strained Alkenes and Electron Deficient Olefins



Noyori, Bull. Chem. Soc. Jpn., 55, 1982, 852.

■ Ni(0) catalyzed reaction of electron deficient olefins with norbornadienes usually gives [2 + 2 + 2] homo-Diels-Alder adducts

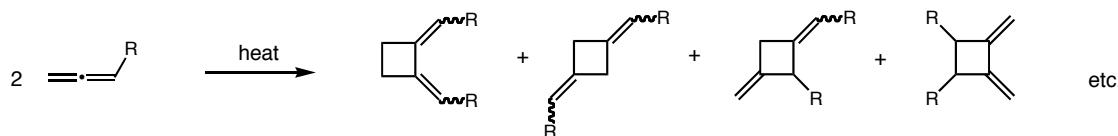
■ Certain strained or highly reactive enophiles undergo exclusive [2 + 2] pathway



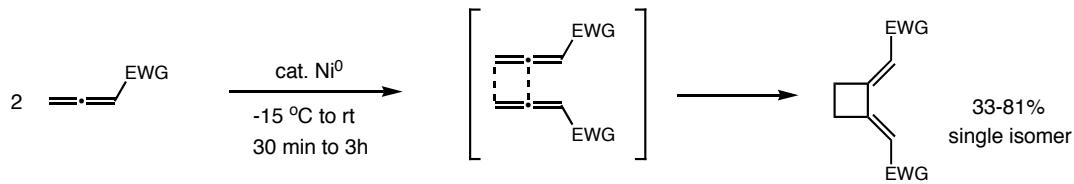
Lautens, J. Am. Chem. Soc., 117, 1995, 10276.

[2 + 2] Cycloadditions: Electron-Deficient Allenes

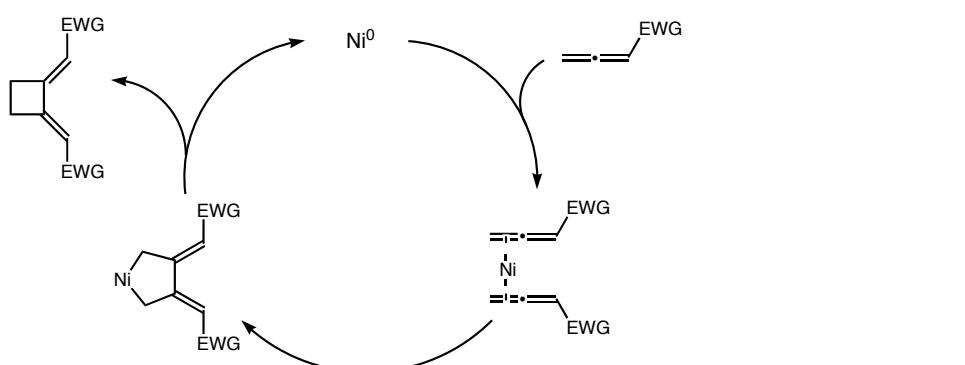
■ Thermal [2 + 2] with allenes gives mixture of regioisomers



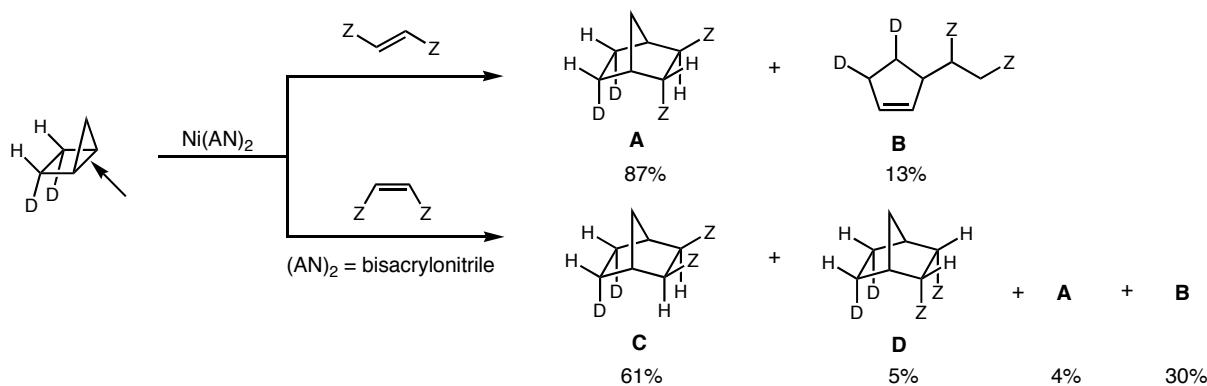
■ Nickel-catalyzed reaction is highly selective



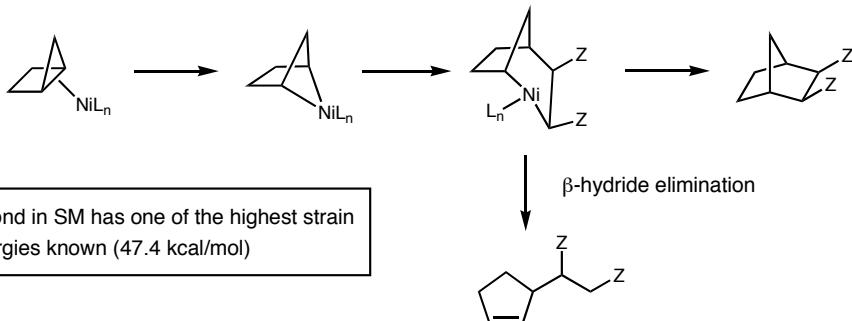
■ Catalytic cycle



[3_{2σ} + 2_{2π}] Cycloadditions: Metallocyclobutanes

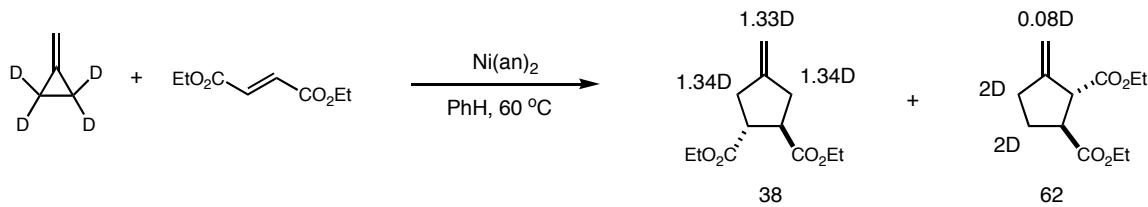


■ Mechanism:



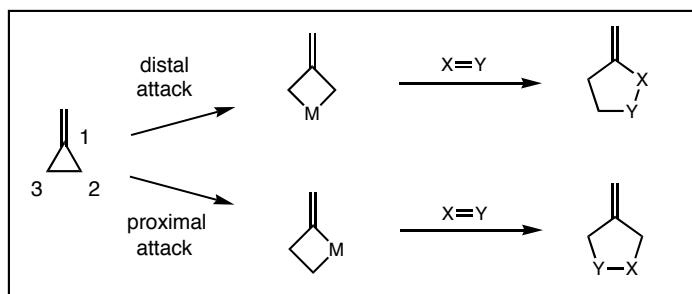
Noyori, *J. Am. Chem. Soc.*, **96**, 1974, 634.

[3_{2σ} + 2_{2π}] Cycloadditions: Methylenecyclopropane



■ Two possible pathways

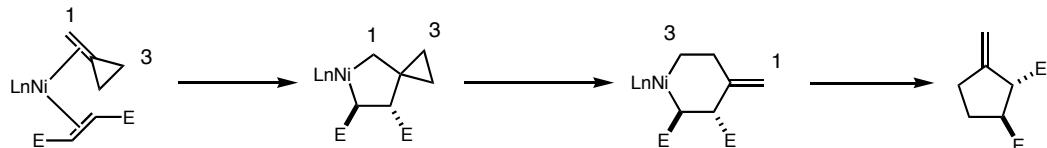
Noyori, *Tetrahedron Lett.*, **1978**, 4823.



- "Naked" nickel catalysts such as $Ni(COD)_2$ favor proximal ring opening
- Phosphine ligands result in a preference for distal ring opening
- Palladium reacts exclusively at the distal position

Binger, *Top. Curr. Chem.*, **1987**, 135, 77.

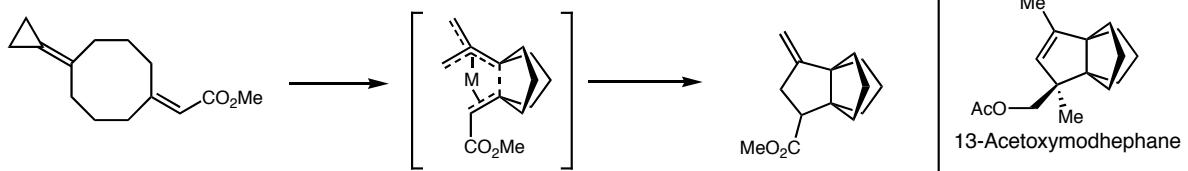
■ Mechanism (proximal attack):



Binger, *Top. Curr. Chem.*, **1983**, 116, 1.

Intramolecular [3_{2σ} + 2_{2π}] Cycloadditions With Methylenecyclopropanes

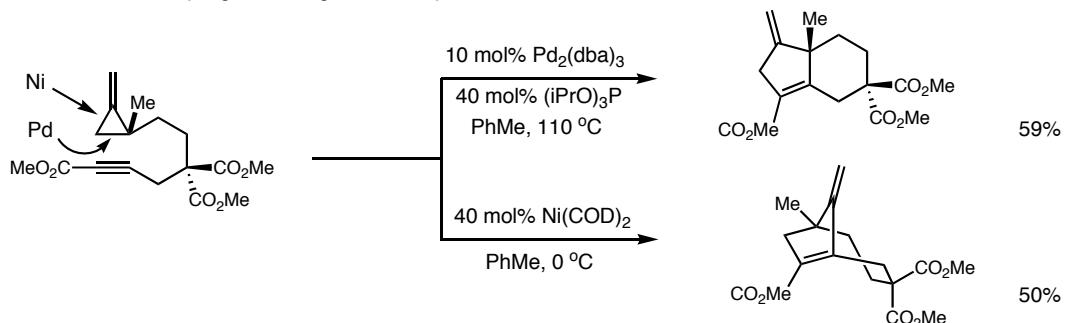
■ Synthesis of 13-Acetoxymodhephane



catalyst (20 mol%)	temp.	yield
Ni(COD) ₂ , PPh ₃	110 °C	74%
(PPh ₃) ₂ PdCl ₂ , DIBAL	130 °C	98%

Nakamura, *J. Chem. Soc., Chem. Comm.*, 1988, 1112.

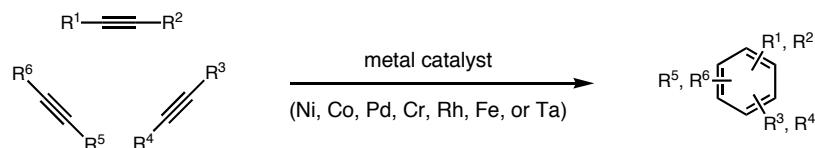
■ Selection of metal catalyst governs regiochemical product



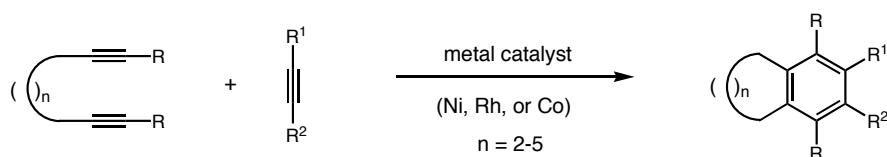
Motherwell, *Tetrahedron Lett.*, 1989, 7107.

[2_{2π} + 2_{2π} + 2_{2π}] Cycloadditions: Alkyne Cyclotrimerizations

- Although in principle thermal [2 + 2 + 2] cycloadditions are symmetry allowed, the entropic barriers associated with bringing three reaction partners together and enthalpic activation energy contributions mitigate against such processes
- The use of a transition metal catalyst enables entropic constraints to be circumvented by coordination of the reaction partners to the metal complex in a stepwise process



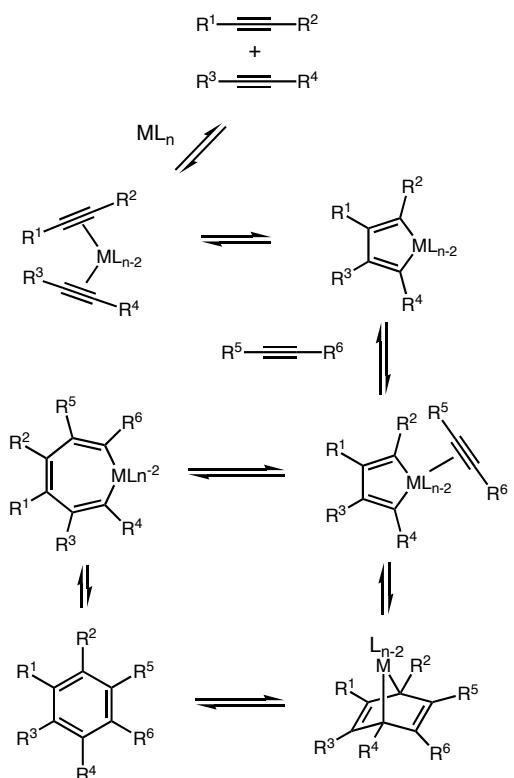
- Intermolecular reaction suffers from chemo- and regioselectivity problems → complex product mixture



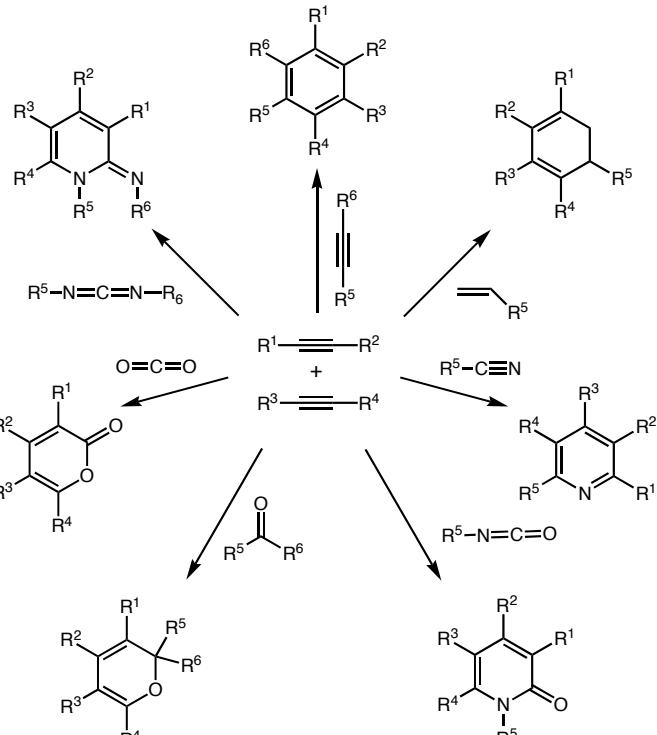
- Partially intramolecular cyclotrimerization has become a very useful synthetic procedure

[2_{2π} + 2_{2π} + 2_{2π}] Cyclotrimerizations: Mechanism and Scope

■ General Mechanism:

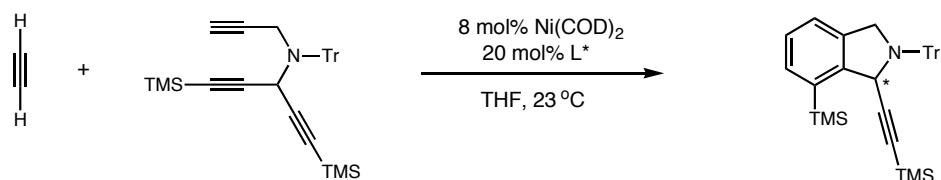


■ Modifications of Cyclotrimerization

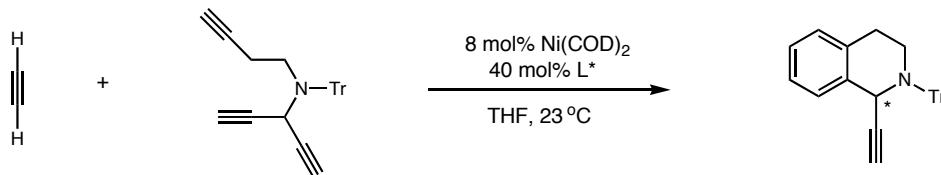
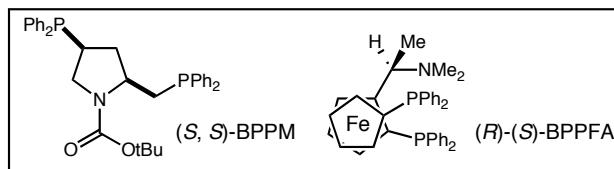


Lautens, *Chem. Rev.*, 96, 1996, 49.

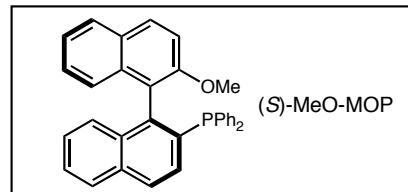
[2_{2π} + 2_{2π} + 2_{2π}] Cycloadditions: Semi-Intramolecular Alkyne Cyclotrimerizations Asymmetric Synthesis of Isoindoline and Isoquinoline Derivatives



L*	Yield	ee
(S, S)-BPPM	92%	60%
(R)-(S)-BPPFA	52%	73%



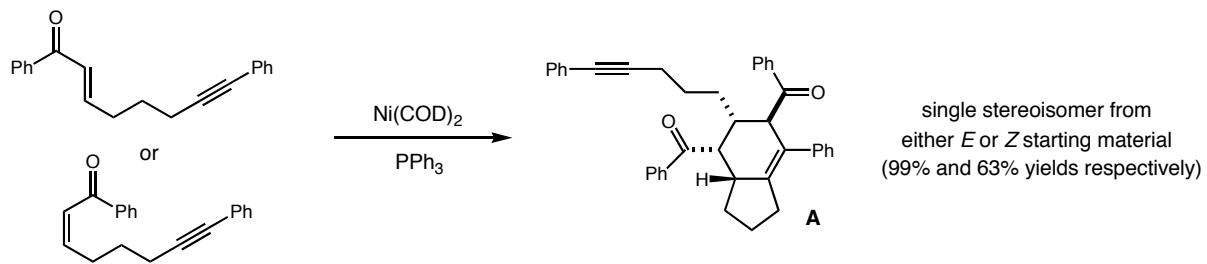
L*	Yield	ee
(S, S)-BPPM	42%	6%
(S)-MeO-MOP	62%	54%



Mori, *J. Org. Chem.*, 59, 1994, 6133.

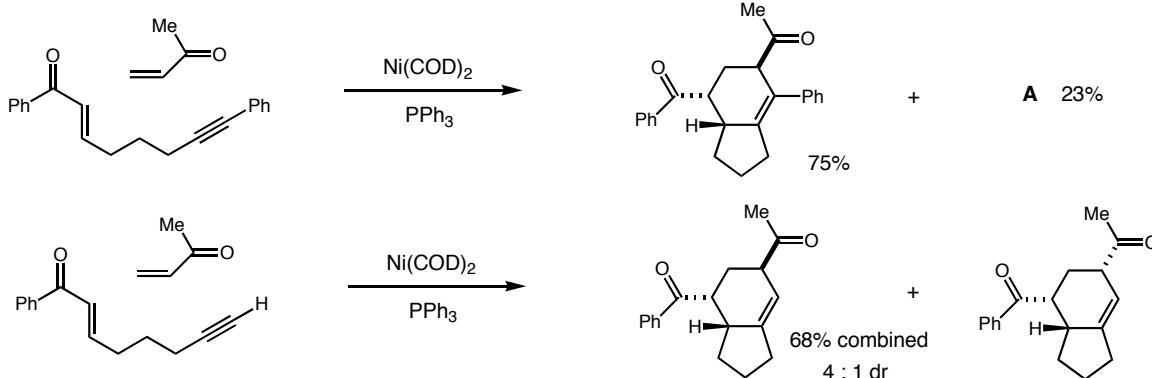
[2 + 2 + 2] Cycloadditions: Alkynyl Enone / Alkene Cyclotrimerizations

■ Montgomery (w/ Jeongbeob Seo)



- Reaction sets two rings and four contiguous stereocenters with complete stereoselectivity
- Unusual chemoselectivity for alkene over alkyne in [2 + 2 + 2] cyclotrimerization

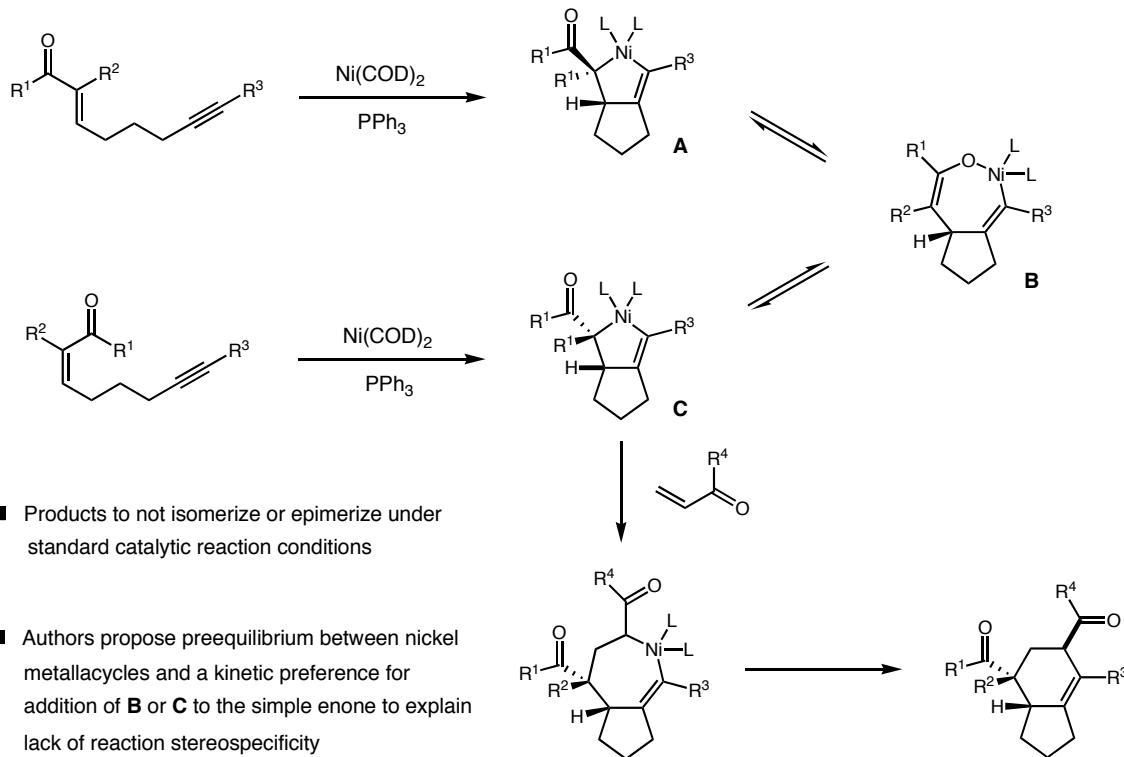
■ Simple enones



Montgomery, *J. Am. Chem. Soc.*, **1999**, *121*, 477.

[2 + 2 + 2] Cycloadditions: Alkynyl Enone / Alkene Cyclotrimerizations

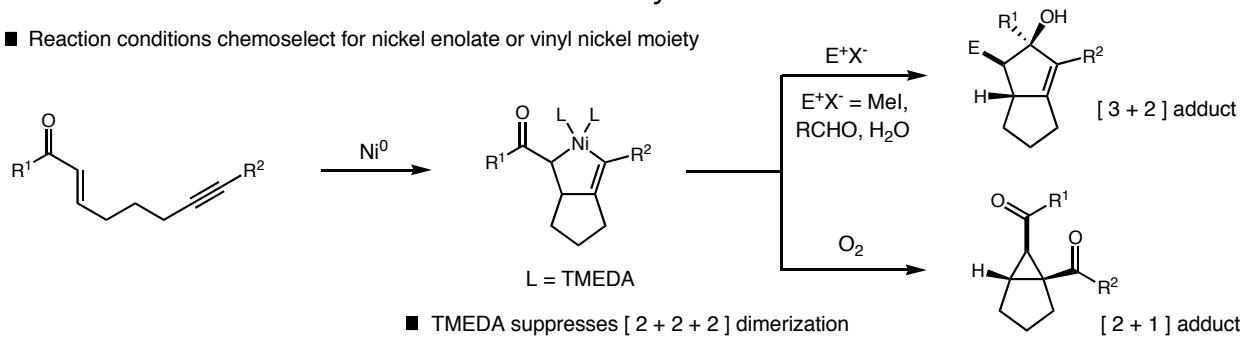
Mechanism



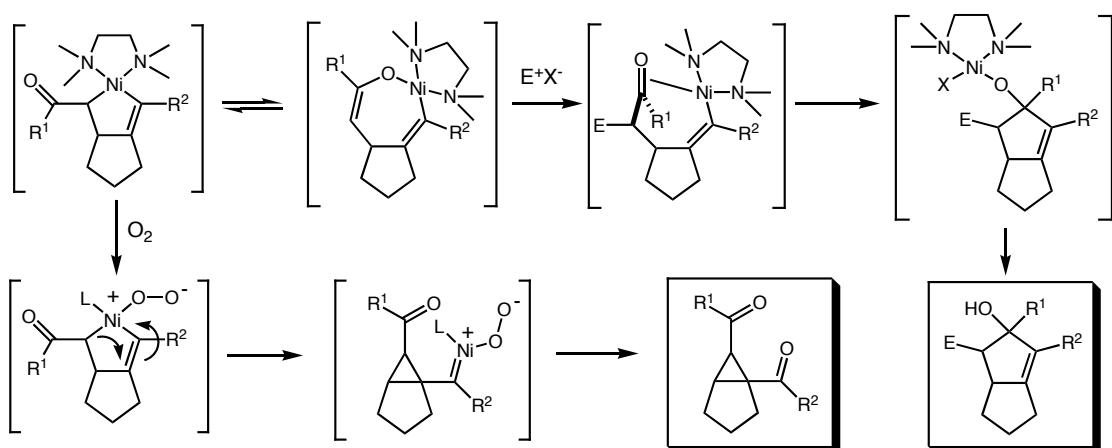
Montgomery, *J. Am. Chem. Soc.*, **1999**, *121*, 477.

[3 + 2] and [2 + 1] Cycloadditions: Diverse Reaction Manifolds From a Common Nickel Metallocycle

■ Reaction conditions chemoselect for nickel enolate or vinyl nickel moiety

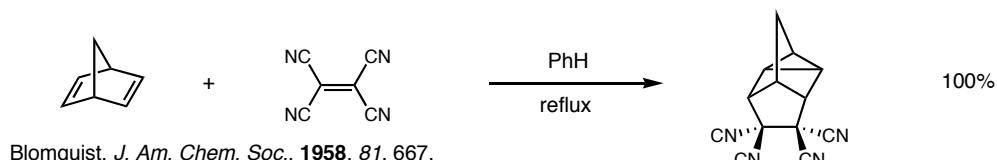
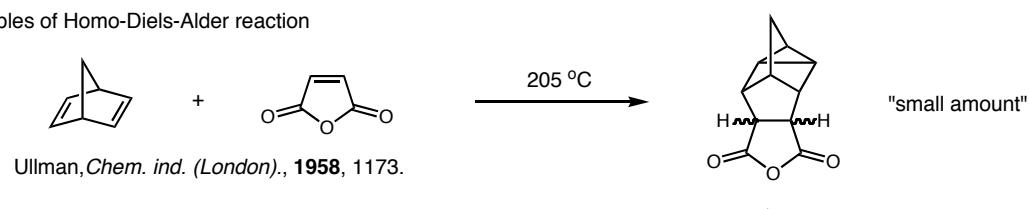


■ Mechanisms



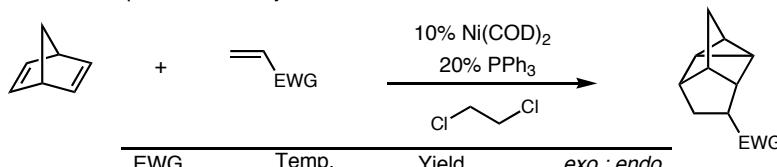
[2_{2π} + 2_{2π} + 2_{2π}] Homo-Diels-Alder Cycloadditions

■ First examples of Homo-Diels-Alder reaction



■ Less activated olefins result in greatly diminished yields—acrylonitrile and norbornadiene (200 °C, 12h) gives only 12% yield

■ Nickel catalysts greatly broaden scope and efficiency of HDA

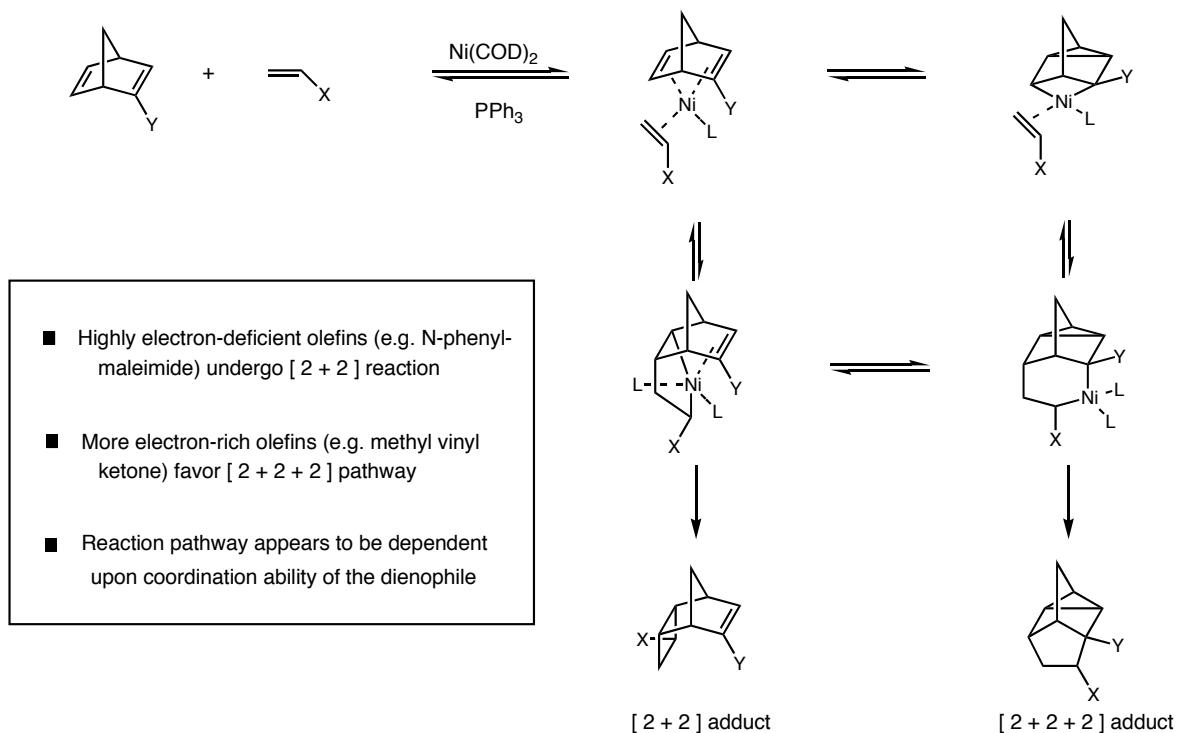


EWG	Temp.	Yield	exo : endo
COMe	80 °C	99%	>20 : 1
CHO	r.t.	58%	3 : 1
CN	80 °C	82%	4 : 1
SO ₂ Ph	r.t.	75%	1 : 1
SOPh	r.t.	73%	>19 : 1 ^a

^a: P(OPh)₃ used instead of PPh₃

Lautens, *J. Am. Chem. Soc.*, **1995**, 117, 10276.

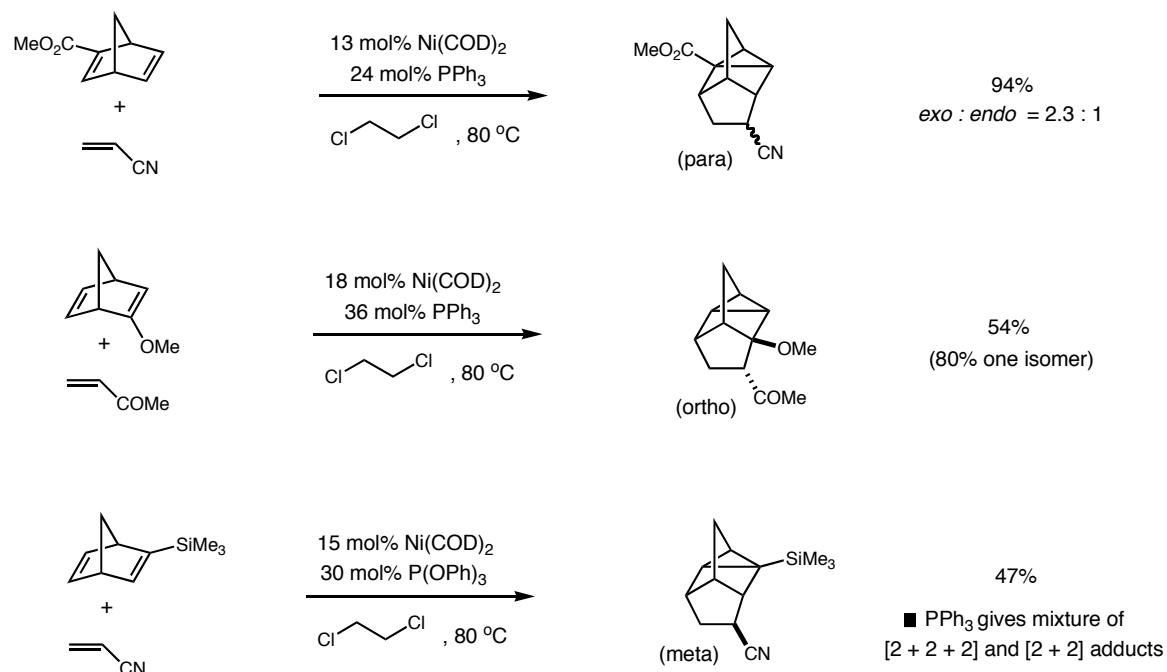
Norbornadienes: [2 + 2 + 2] Vs. [2 + 2]



Lautens, *J. Am. Chem. Soc.*, 1995, 117, 10276.

[2_{2π} + 2_{2π} + 2_{2π}] Homo-Diels-Alder Cycloadditions : Regio- and Stereoselectivity

2-Substituted Norbornadienes

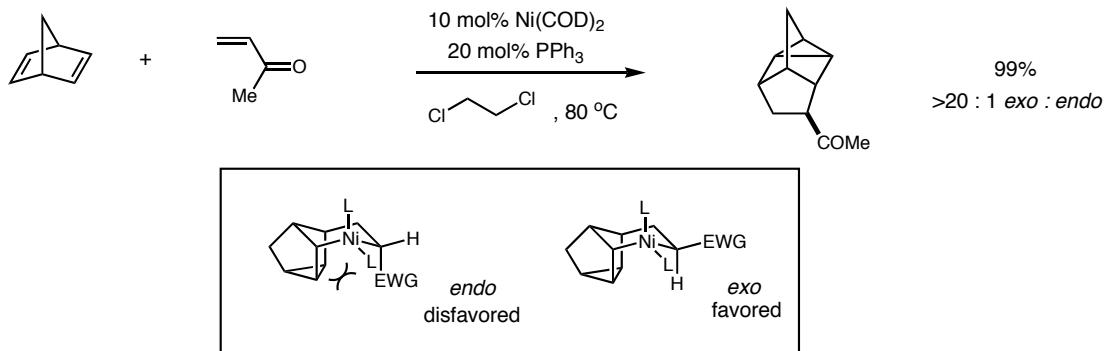


■ Diene and dienophile substituents as well as ligands shown to have dramatic effect on selectivities

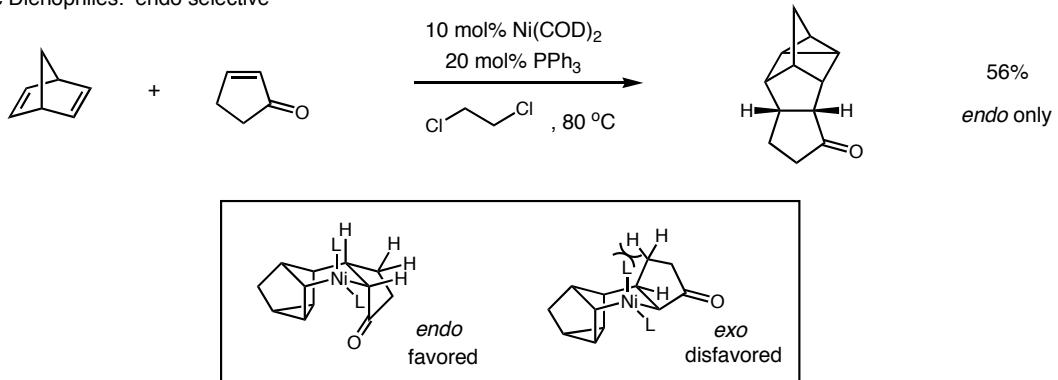
Lautens, *J. Am. Chem. Soc.*, 1995, 117, 10276.

$[2_{2\pi} + 2_{2\pi} + 2_{2\pi}]$ Homo-Diels-Alder: Cyclic Vs. Acyclic Dienophiles

■ Acyclic Dienophiles: exo-selective

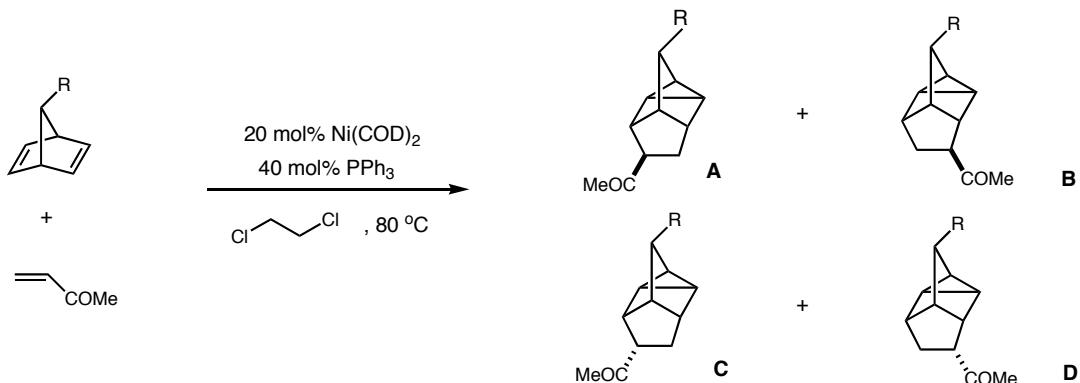


■ Cyclic Dienophiles: endo selective



Lautens, J. Am. Chem. Soc., 1995, 117, 10276.

$[2_{2\pi} + 2_{2\pi} + 2_{2\pi}]$ Homo-Diels-Alder: 7-Substituted Norbornadienes



R	Yield	A : B : C : D	anti : syn	exo : endo
n-hexyl	83%	40 : 58 : 1.6 : 0.4	42 : 58	98 : 2
Ph	84%	54 : 45 : 0.8 : 0.2	55 : 45	99 : 1
Cl	60%	71 : 28 : 0.8 : 0.2	72 : 28	99 : 1
OCOPh	97%	80 : 20 : 0 : 0	80 : 20	100 : 0
OTIPS	90%	90 : 9 : 1 : 0	91 : 9	99 : 1
OMEM	89%	88 : 9 : 3 : 0	91 : 9	97 : 3
OtBu	95%	95 : 5 : 0 : 0	95 : 5	100 : 0

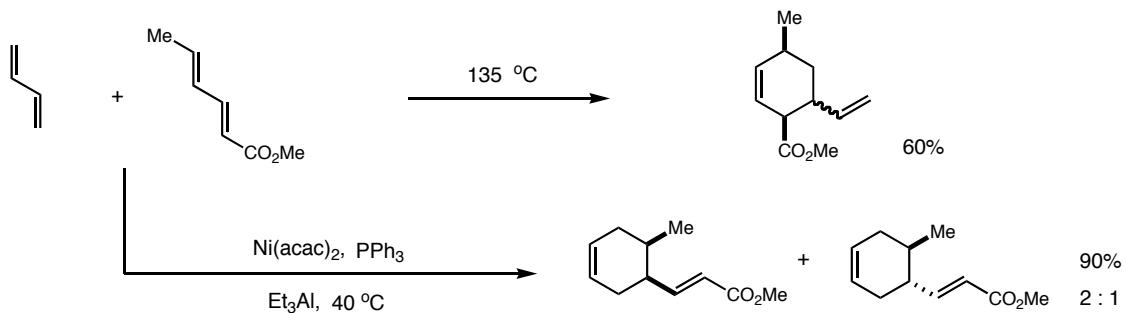
■ Anti : syn selectivity increases as group electronegativity of 7-substituent increases

■ Ab initio calculations indicate a shift of electron density from the anti- π olefin to the syn- π olefin as electronegativity increases

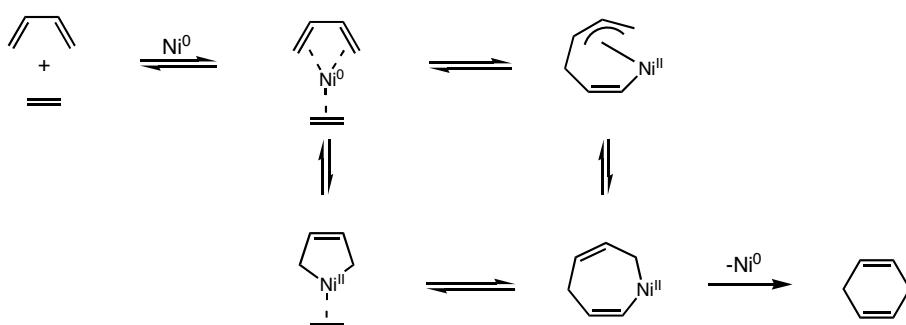
Lautens, J. Am. Chem. Soc., 1995, 117, 6863.

[4_{4π} + 2_{2π}] Cycloadditions: Nickel-Catalyzed Diels-Alder Reactions

■ Reversal of thermal reactivity



■ Mechanism:

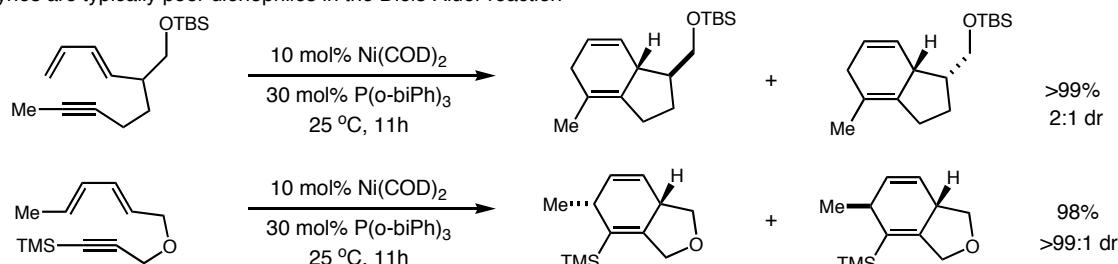


Wender, *J. Am. Chem. Soc.*, 1989, 111, 6432.

[4_{4π} + 2_{2π}] Cycloadditions: Nickel-Catalyzed Intramolecular Diels-Alder Reactions

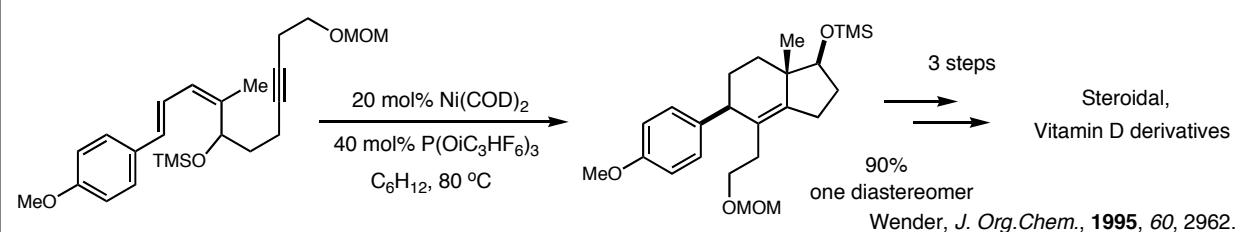
Alkyne and Allene Dienophiles

■ Alkynes are typically poor dienophiles in the Diels-Alder reaction



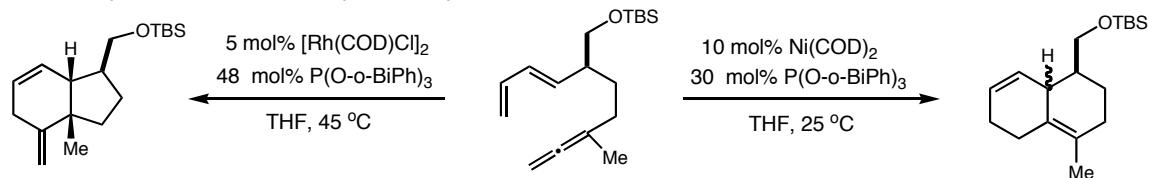
Wender, *J. Am. Chem. Soc.*, 1989, 111, 6432.

■ Transition metal catalysts make these processes facile and synthetically useful



Wender, *J. Org. Chem.*, 1995, 60, 2962.

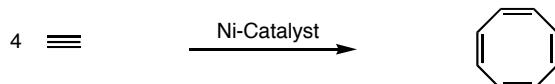
■ Allenes: Catalyst-controlled chemocomplementarity



Wender, *J. Am. Chem. Soc.*, 1995, 117, 1843.

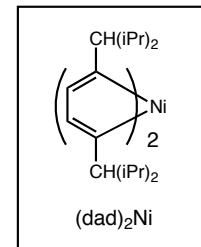
$[2_{2\pi} + 2_{2\pi} + 2_{2\pi} + 2_{2\pi}]$ Cycloadditions: Cyclooctatetraene Synthesis

■ First example by Reppe, has been used on an industrial scale

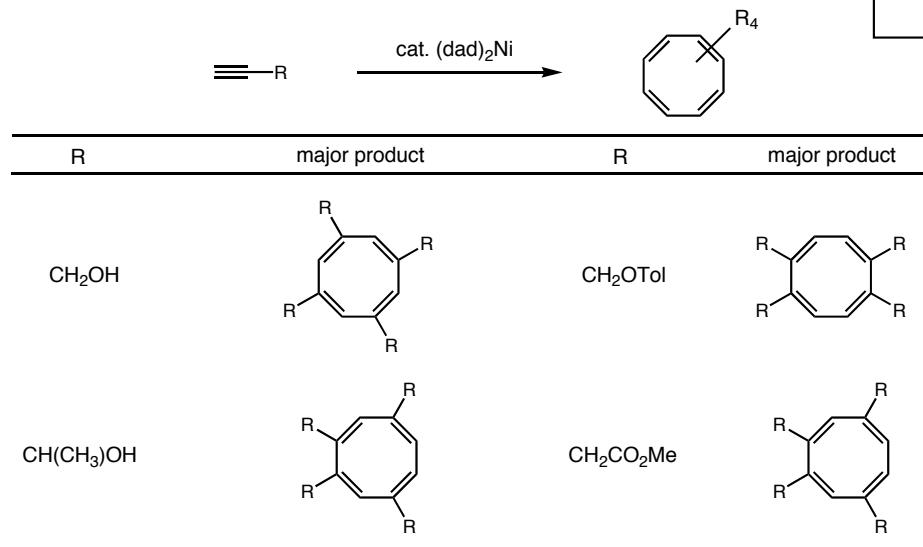


Ni-Catalyst = $\text{NiBr}_2 / \text{CaC}_2, \text{Ni(acac)}_2, \text{Ni(COT)}_2$

Reppe, *Leibigs Ann. Chem.*, **1948**, 560, 1.



■ tom Dieck developed first method for regioselective cycloadditions with monosubstituted alkynes

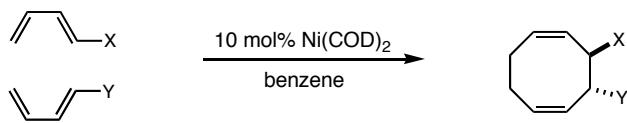


tom Dieck, *Chem. Ber.*, **1985**, 118, 428.

$[4_{4\pi} + 4_{4\pi}]$ Cycloadditions: Cyclooctadiene Synthesis

Intermolecular

■ Substituted butadiene dimerization

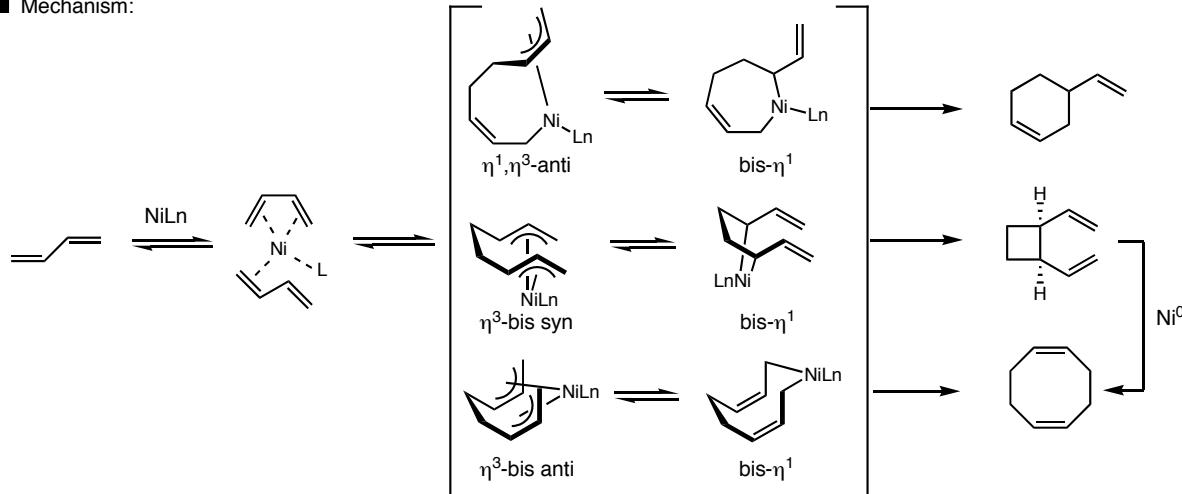


X, Y = OSiMe ₃	(18d)	90%
X, Y = CO ₂ Me	(24h)	70%
X = H, Y = CO ₂ Me	(72h)	33%

■ Use of butadiene results in significant [4 + 2] adduct

Waegell, *Tetrahedron Lett.* **1983**, 24, 385.

■ Mechanism:

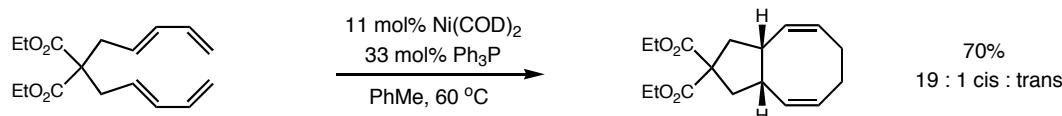


■ Conditions can be optimized to minimize vinylcyclohexene and divinylcyclobutane formation

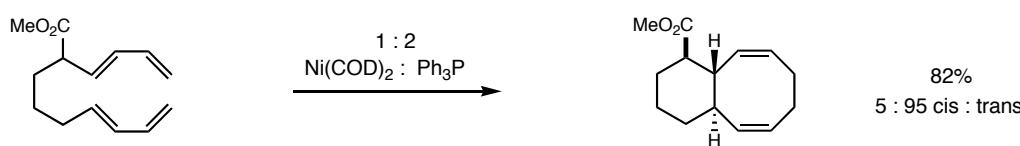
Wilke, *ACIEE*, **1988**, 27, 185.

[4_{4π} + 4_{4π}] Cycloadditions: Intramolecular Cyclooctadiene Synthesis

■ 3-atom tethers → cis-fused



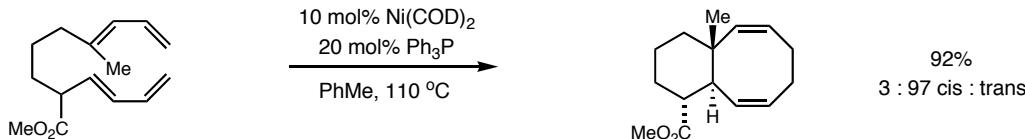
■ 4-atom tethers → trans-fused



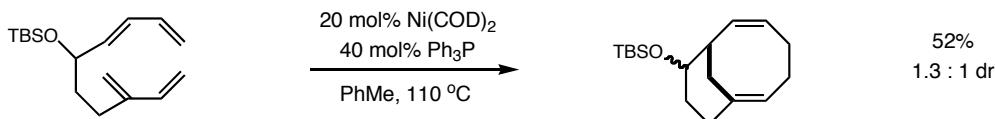
■ Note: other substrates give much lower selectivities

Wender, *J. Am. Chem. Soc.*, **1986**, *108*, 4678.

■ Type I



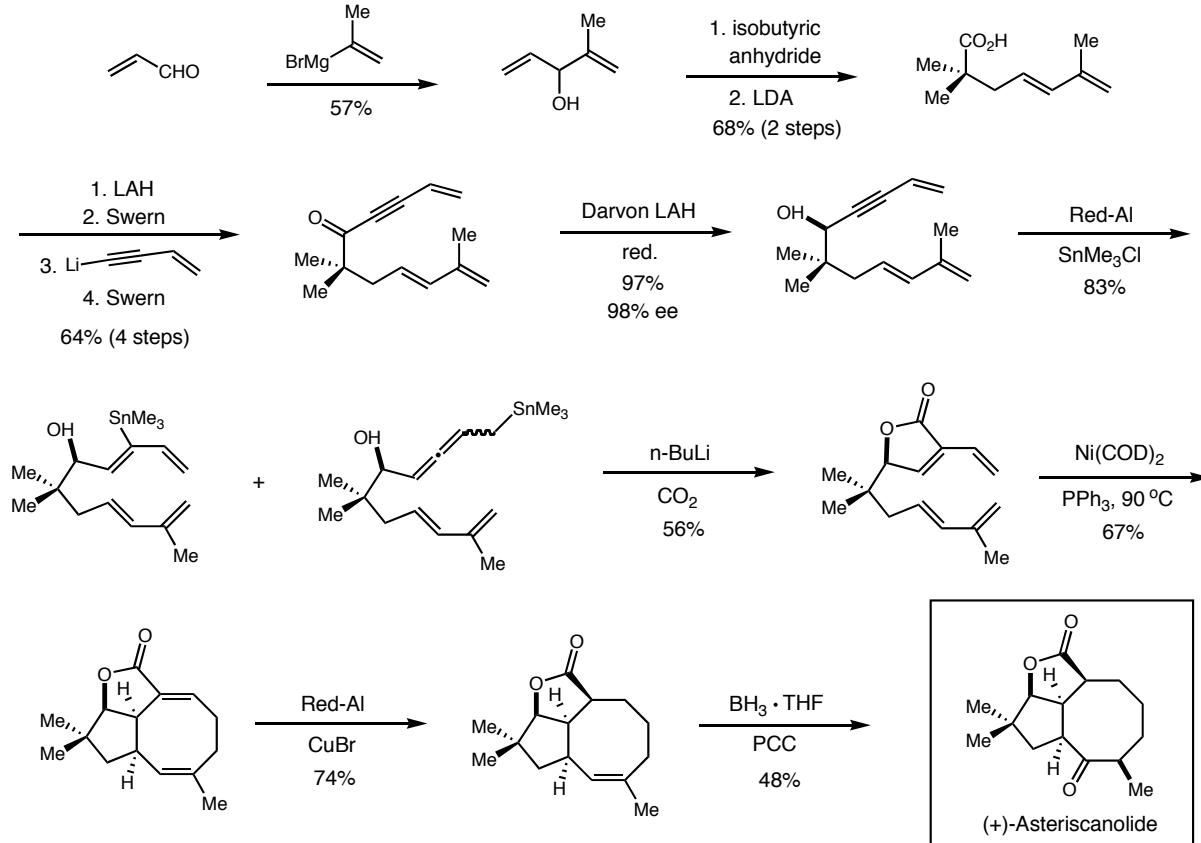
■ Type II



■ Towards taxane skeleton

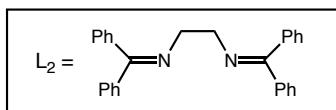
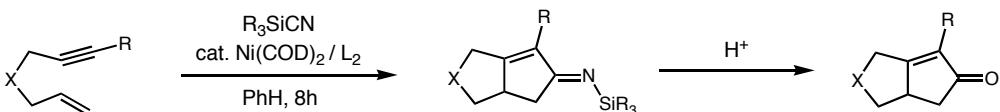
Wender, *Tetrahedron Lett.*, **1987**, *28*, 2221.

[4_{4π} + 4_{4π}] Cycloadditions: Enantioselective Total Synthesis of (+)-Asteriscanolide

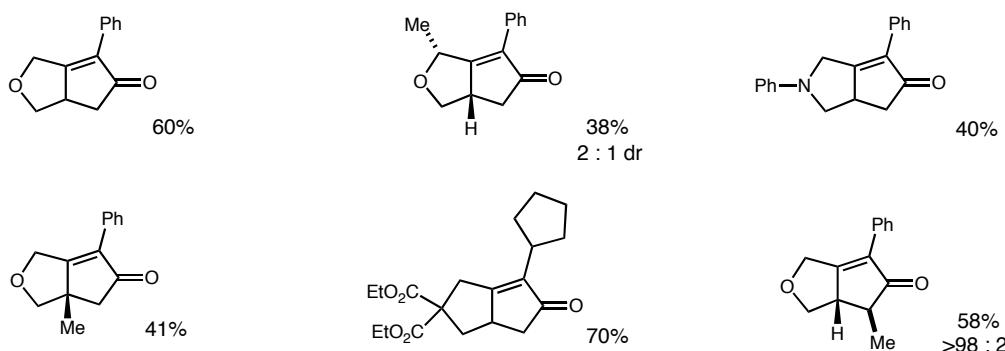


Wender, *J. Am. Chem. Soc.*, **1988**, *110*, 5904.

[2 + 2 + 1] Cycloadditions: Cyclopentenone Synthesis

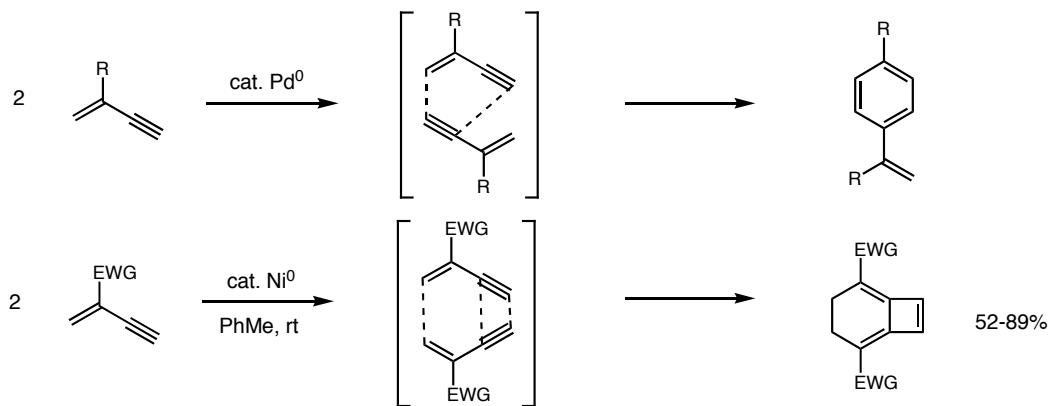


■ Representative examples



Buchwald, J. Org. Chem., 1996, 61, 4498.

Nickel-Catalyzed "Zipper Annulation" of Conjugated Enynes



EWG = n-C₆F₁₃, PhCF₂, n-C₆H₁₃CF₂ Yamamoto, J. Am. Chem. Soc., 2000, 122, 1810.

■ Catalytic cycle

