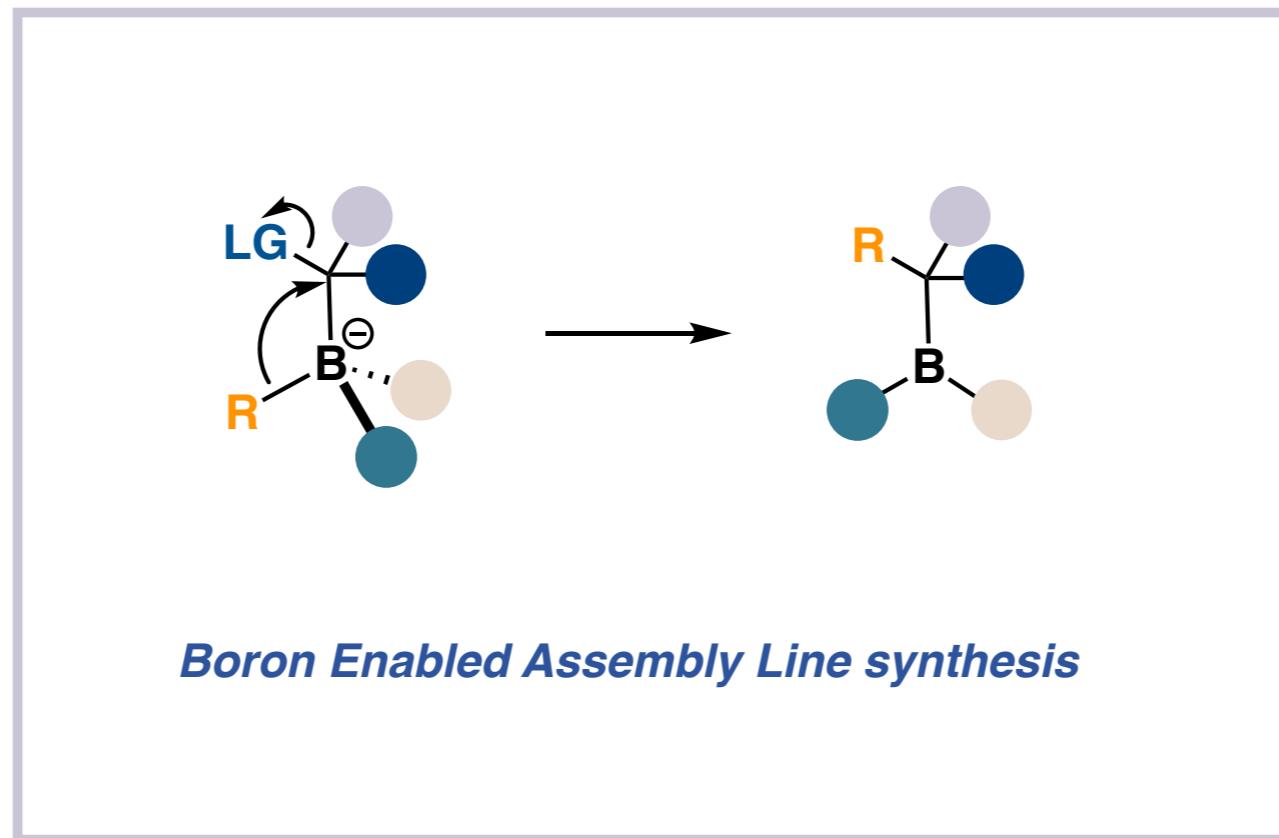
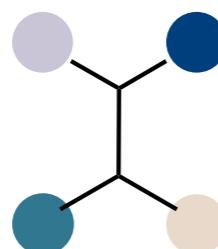


# *Boron Homologation*



**Zhe Dong**  
MacMillan Research Group  
Group Meeting  
Feb 14th, 2019

# *Continuous Carbon Chiral Center Synthesis*



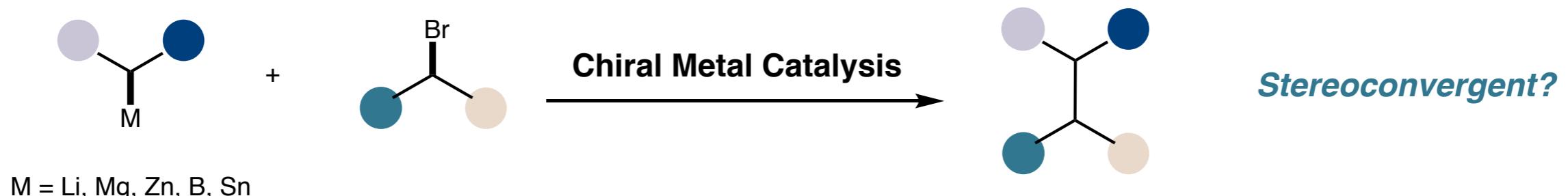
■ *Quick access to starting materials*

■ *Reactivity: overcoming the steric bulk*

**Unsolved Problem**

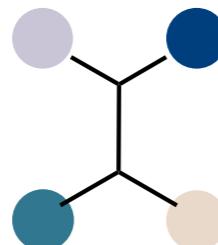
■ *Selectivity: facile synthesis of all 4 enantiomers*

## **Stereospecific or Stereoconvergent?**



*No asymmetric report*  
*No access to all 4 enantiomers*

# *Continuous Carbon Chiral Center Synthesis*



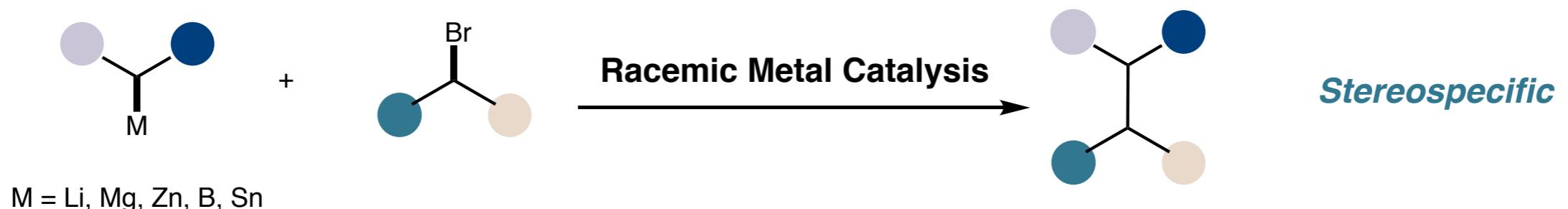
■ *Quick access to starting materials*

■ *Reactivity: overcoming the steric bulk*

**Unsolved Problem**

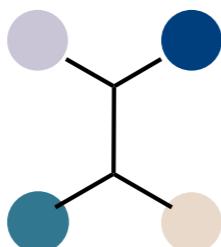
■ *Selectivity: facile synthesis of all 4 enantiomers*

## **Stereospecific or Stereoconvergent?**



*No good way to generate chiral nucleophile  
Match and mismatch issue*

# *Continuous Carbon Chiral Center Synthesis*



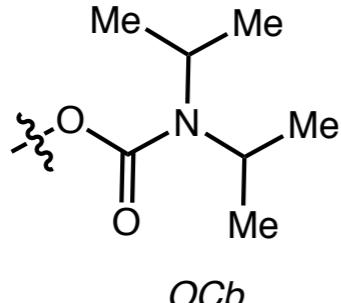
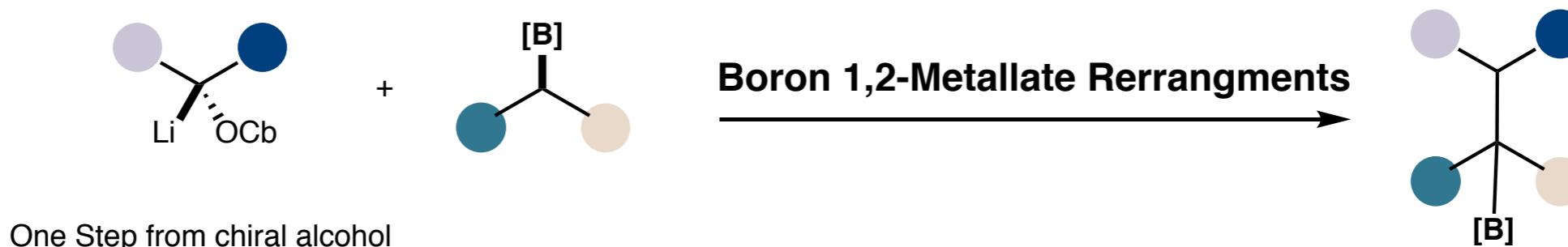
■ *Quick access to starting materials*

■ *Reactivity: overcoming the steric bulk*

**Unsolved Problem**

■ *Selectivity: facile synthesis of all 4 enantiomers*

## ■ *Stereospecific Reaction without Catalysis*



**Solve all the mentioned issues in both selectivity/ selectivity**

**Problem: functional group tolerance/ low temperature**

# *Boron Homologation via 1,2-Metallate Rearrangements*

## **Non-Catalyzed Boron Homologation**

Matteson Homologation

Zweifel Olefination

Aggarwal Homologation

Real-World Application

Assembly Line Synthesis

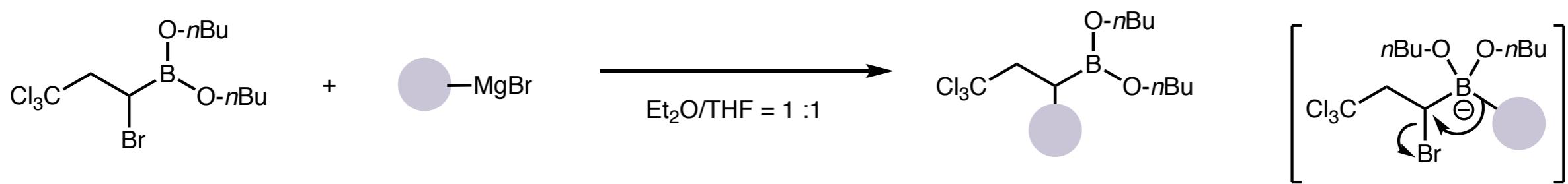
Heteroatom Issue

## **Catalyzed Boron Homologation**

Double Electron Transfer Initiated Reaction

Single Electron Transfer Initiated Reaction

## Matteson Homologation: Reaction Discovery

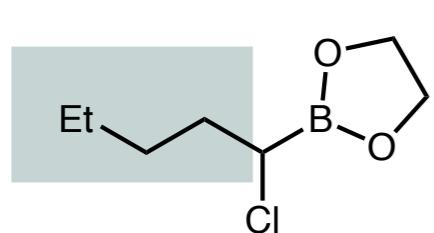
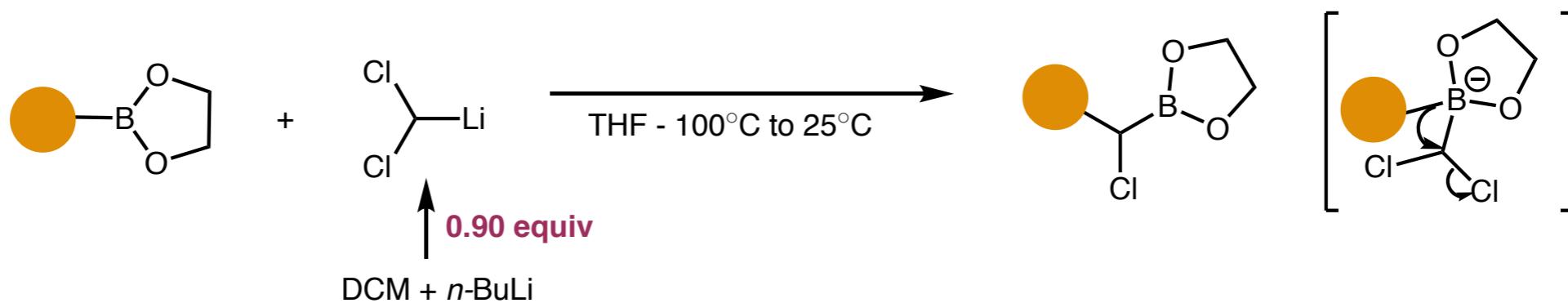


### Scope and Conditions

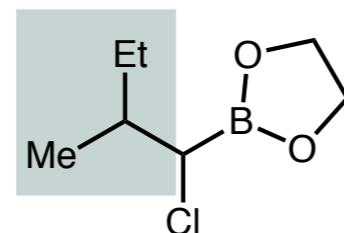
<chem>MgBr</chem>	<chem>MgBr</chem>	<chem>MgBr</chem>	<chem>MgBr</chem>
$25^\circ\text{C}$ 22h 90% IY	$25^\circ\text{C}$ 22h 91% IY	$25^\circ\text{C}$ 2h 80% IY	-70°C 2h 63% IY

More hindered group easier to migrate: **More Bulky, More Reactive**

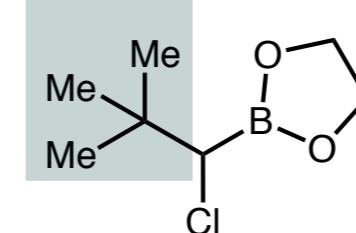
## *α*-Chlorol-alkyl Lithium Reagents for Homologation



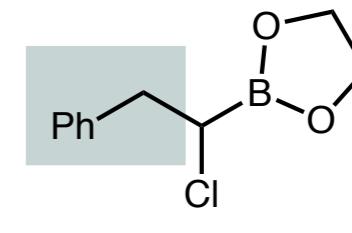
80% IY



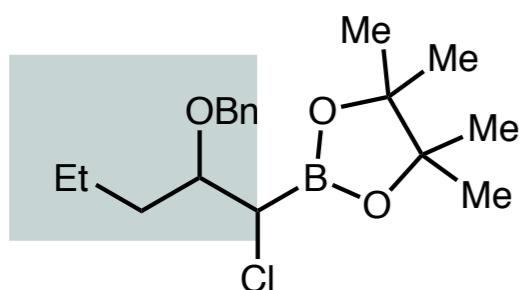
77% IY



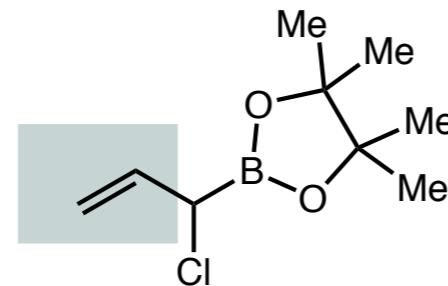
78% IY



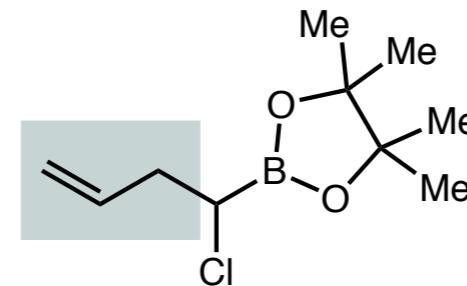
84% IY



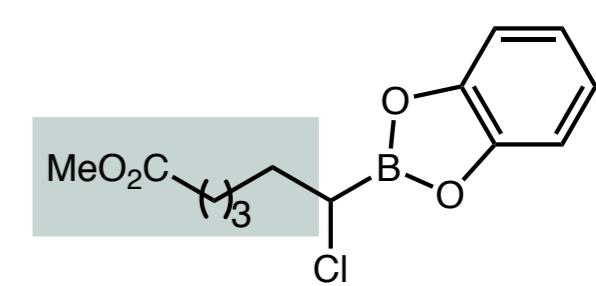
81% IY



90% IY

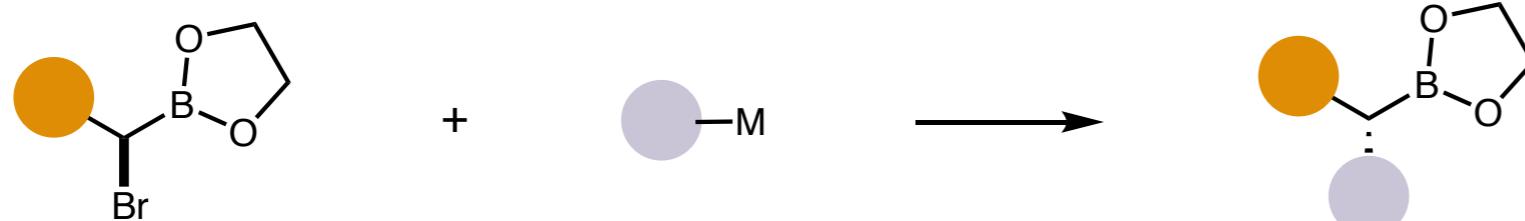


87% IY

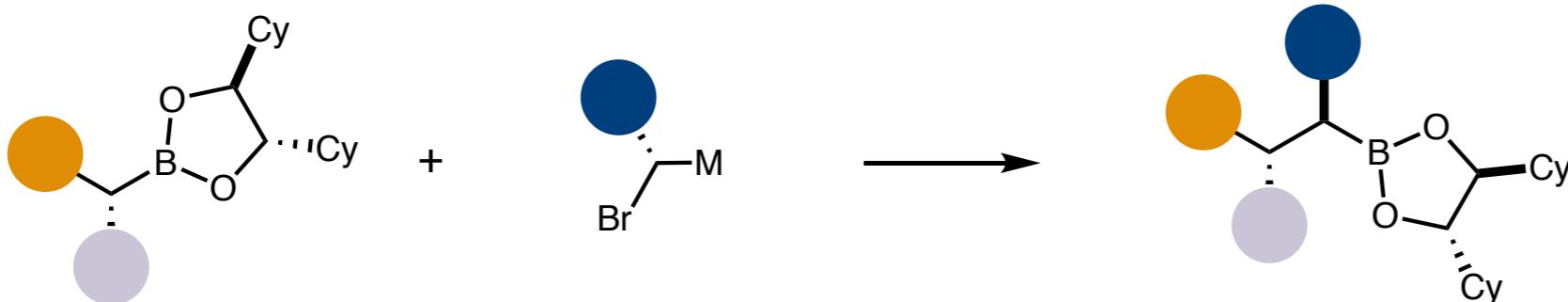


67% IY

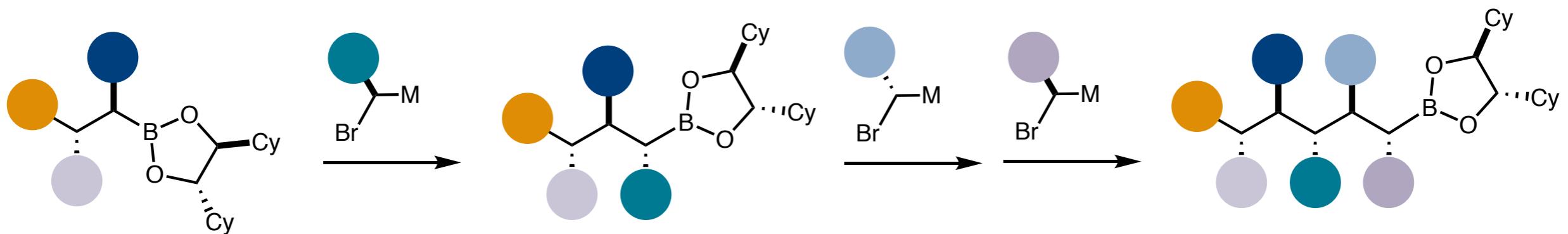
## Matteson Homologation: Iterative Version



Type I : Form S<sub>N</sub>2

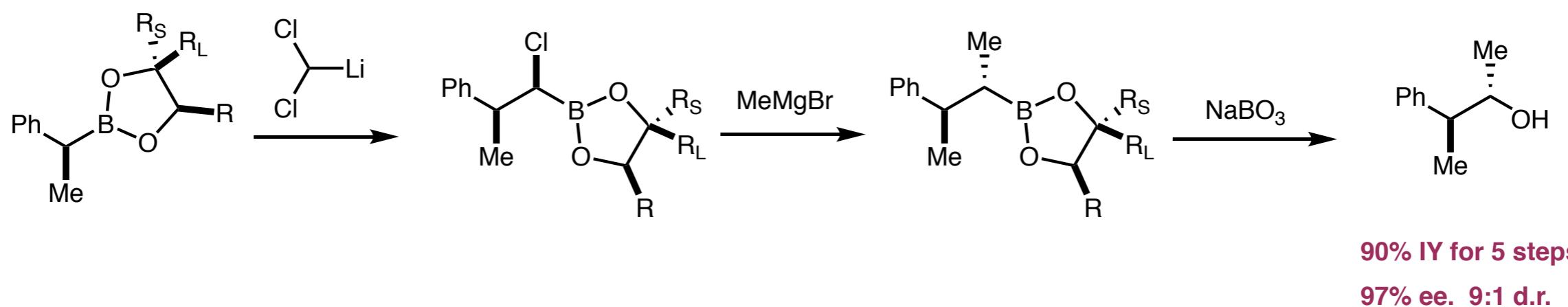
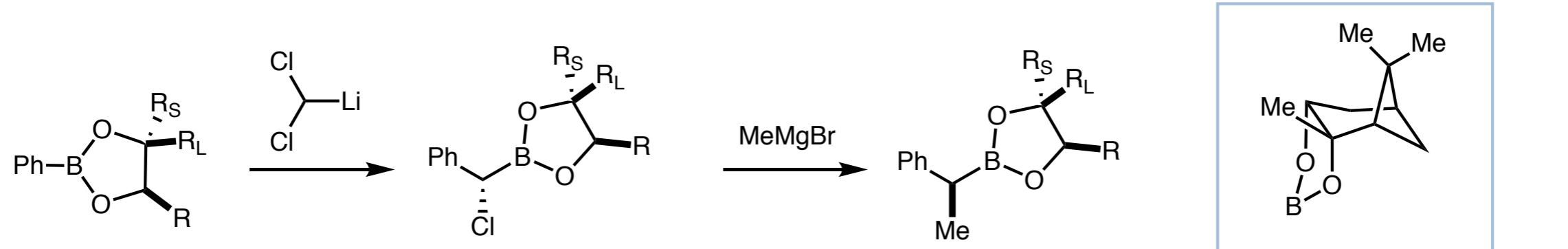
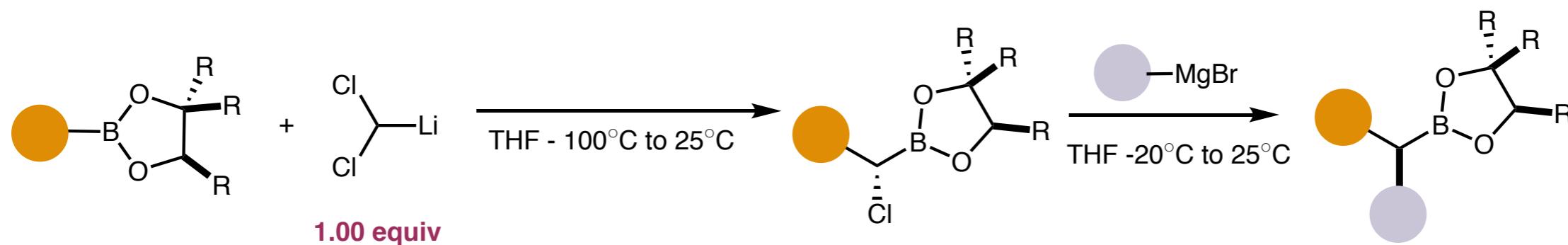


Type II: Homologation

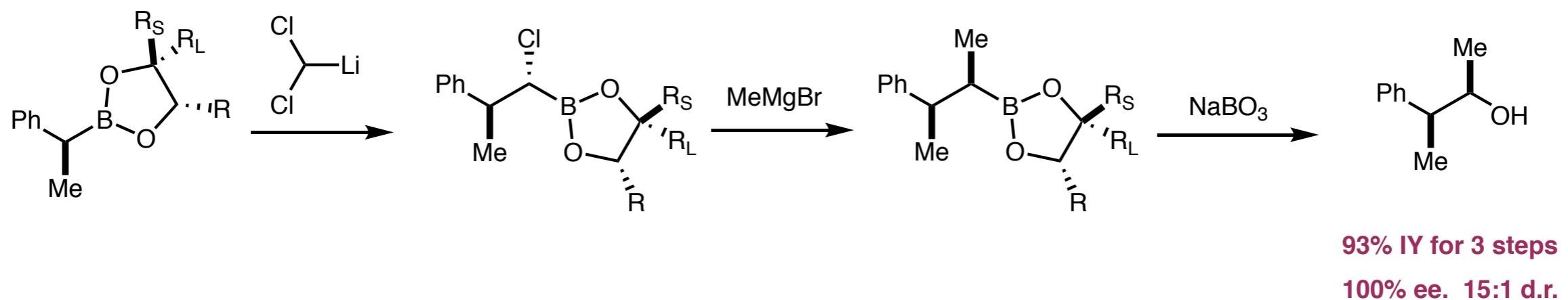
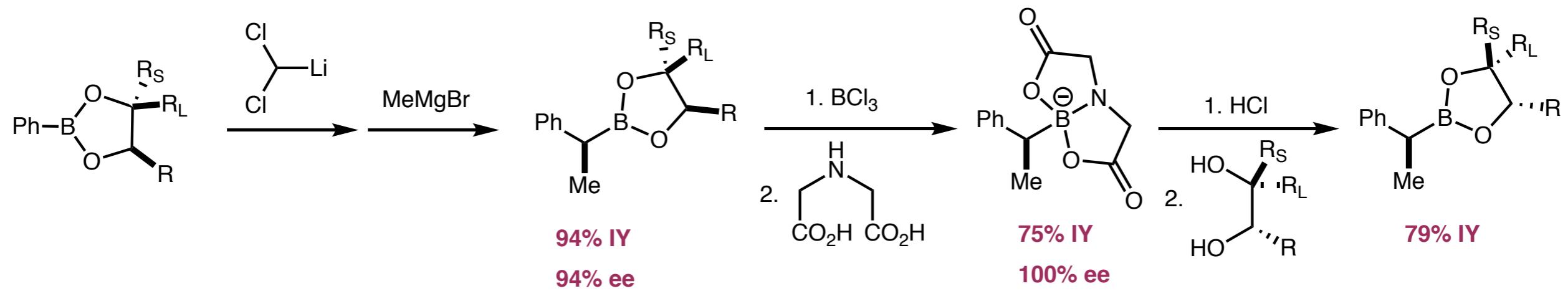
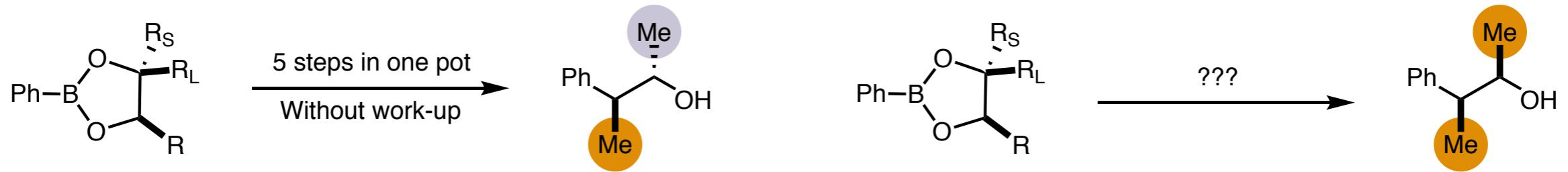


Iterative Synthesis

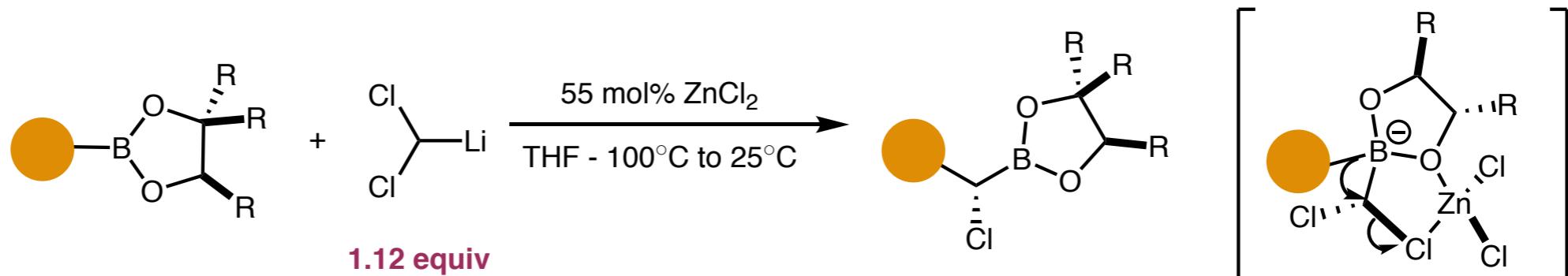
# Chiral Diol as Auxiliary for Diastereoselective Homologation



## Chiral Diol as Auxiliary : Painful Access to Diastereomer



## Chiral Diol as Auxiliary : Improvement on Selectivity

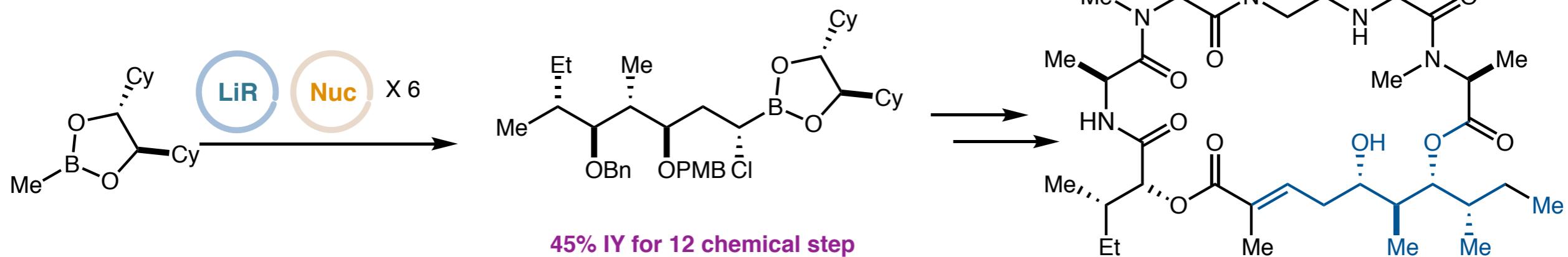
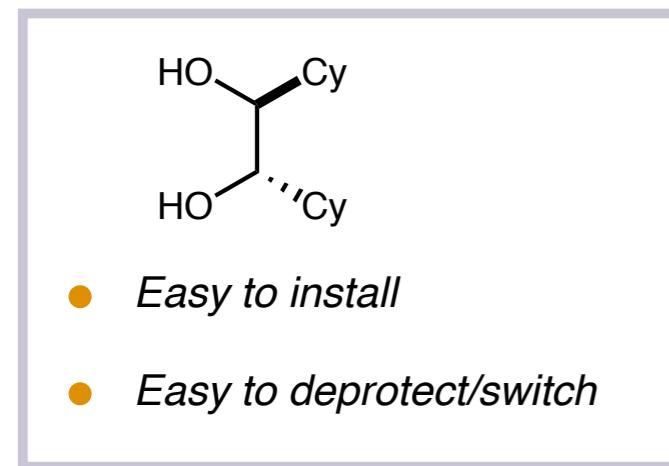
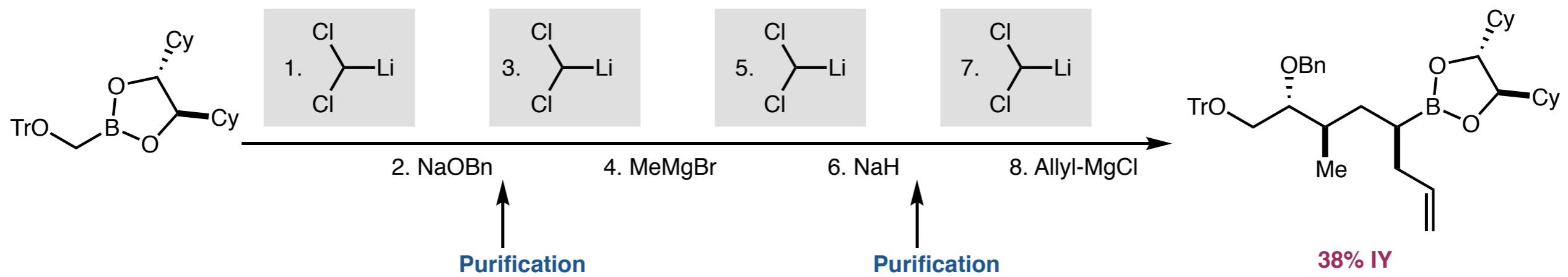


<i>W/O ZnCl<sub>2</sub></i>	57% IY 74% es	61% IY 90% es	30% IY 88% es	75% IY 92.5% es
<i>W ZnCl<sub>2</sub></i>	83% IY 95.7% es	86% IY 98.5% es	89% IY 99.5% es	99% IY 99.5% es

Matteson, D. S.; Sadhu, K. M. *J. Am. Chem. Soc.* **1983**, *105*, 2077.

Matteson, D. S.; Sadhu, K. M.; Peterson, M. L. *J. Am. Chem. Soc.* **1986**, *108*, 810.

# Matteson Homologation: State of Art

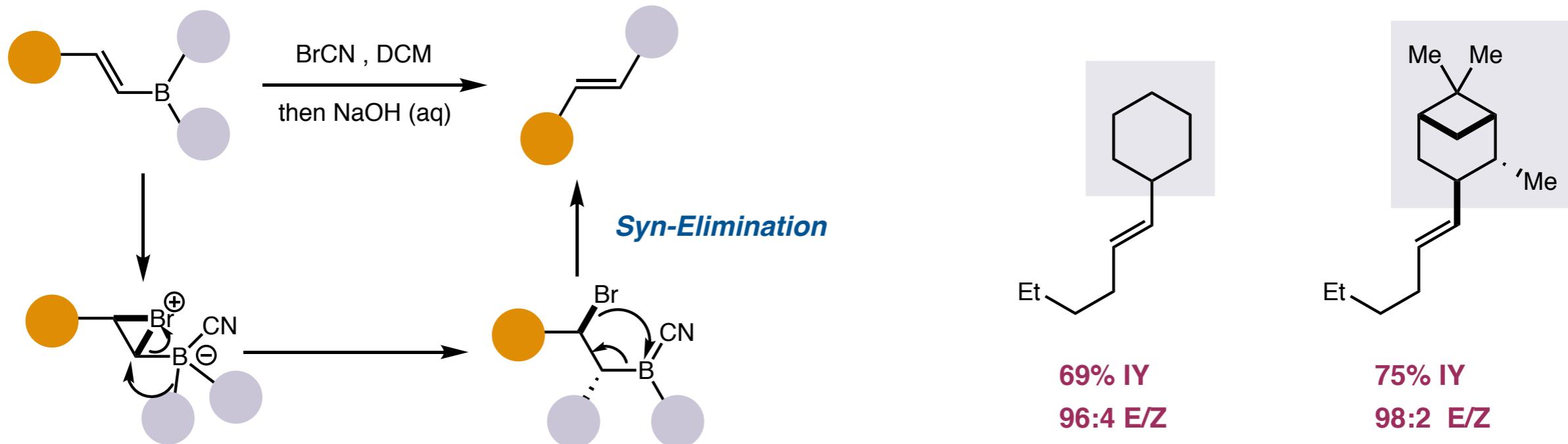
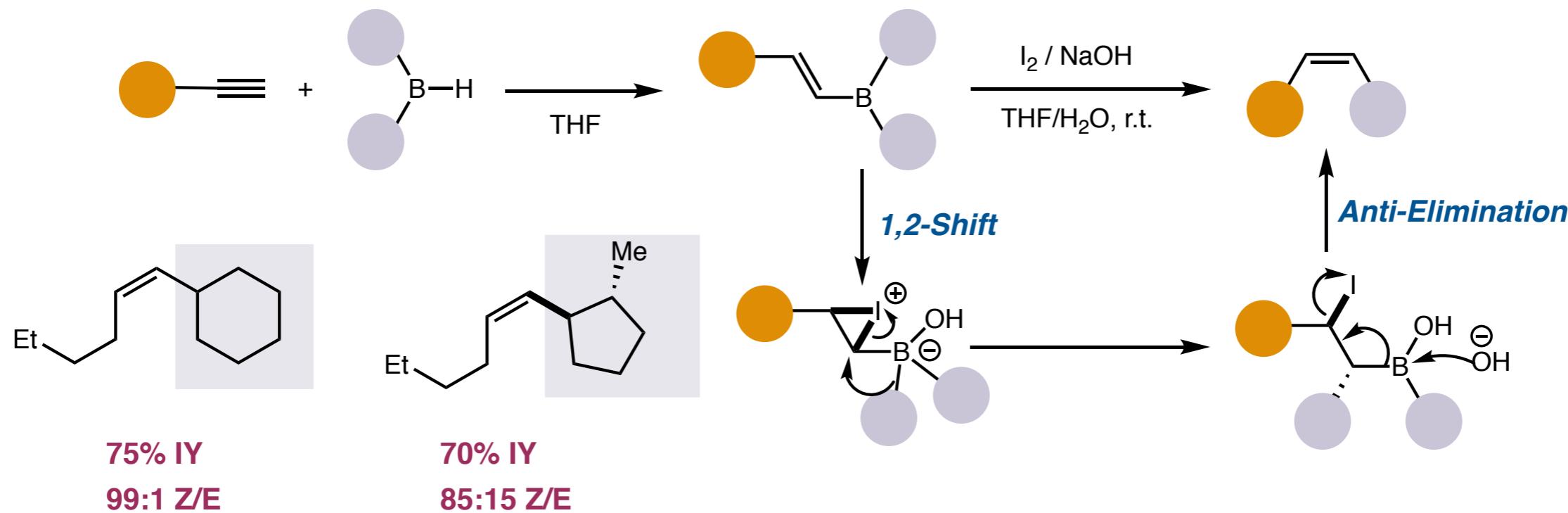


**Lagunamide A**

Matteson, D. S.; Soundararajan, R.; Ho, O. C.; Gatzweiler, W. *Organometallics* **1996**, *15*, 152.

Gorges, J.; Kazmaier, U. *Org. Lett.* **2018**, *20*, 2033.

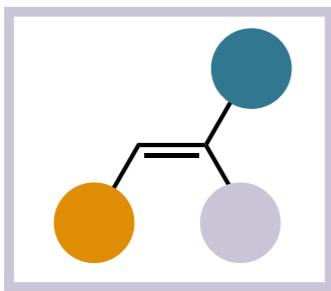
# Zweifel Olefination: Reaction Discovery



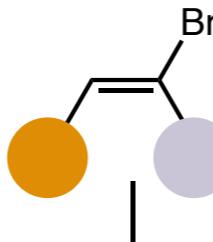
Zweifel, G.; Arzoumanian, H.; Whitney, C. C. *J. Am. Chem. Soc.* **1967**, *89*, 3652.

Zweifel, G.; Fisher, R. P.; Snow, J. T.; Whitney, C. C. *J. Am. Chem. Soc.* **1972**, *94*, 6560.

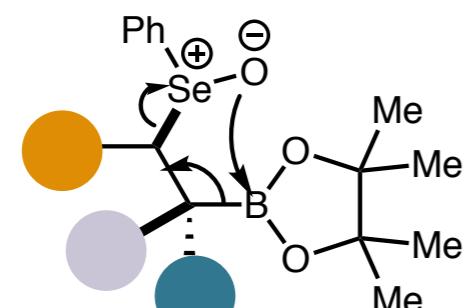
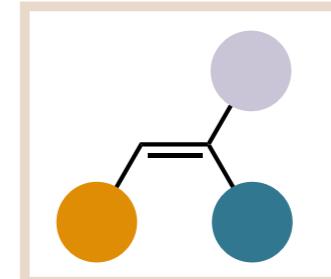
## Zweifel Olefination: State of Art



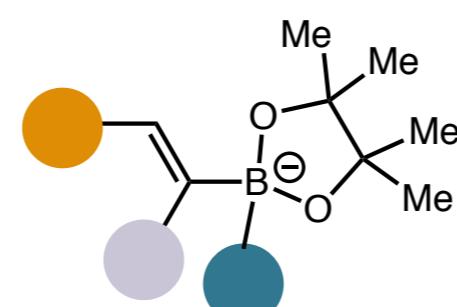
1. *t*-BuLi  
2. Bpin —  
3. PhSeCl  
4. *m*-CPBA



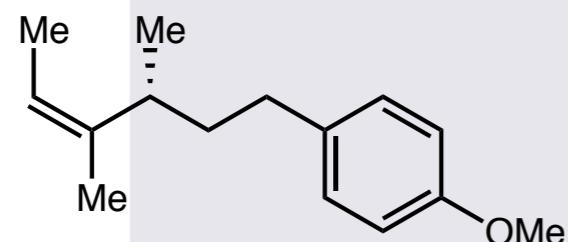
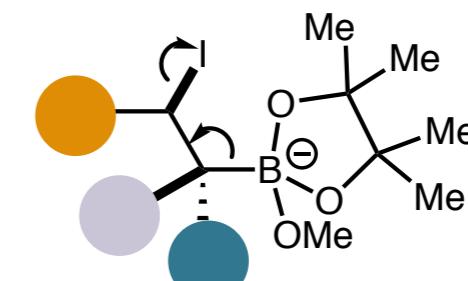
1. *t*-BuLi  
2. Bpin —  
3. I<sub>2</sub>, NaOMe



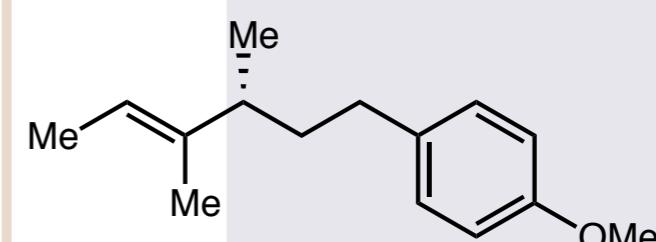
● Excellent Syn Elimination



● High Reactive Olefin

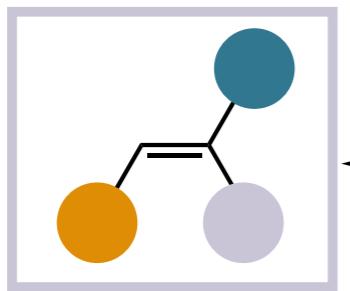


95% IY 100% es  
> 98:2 Z/E

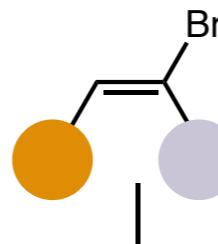


83% IY 100% es  
96:4 E/Z

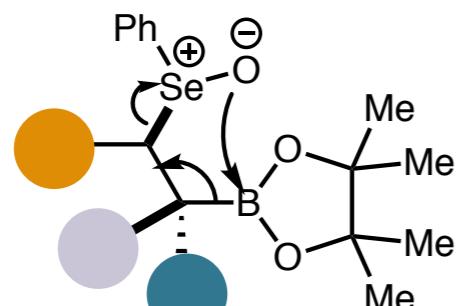
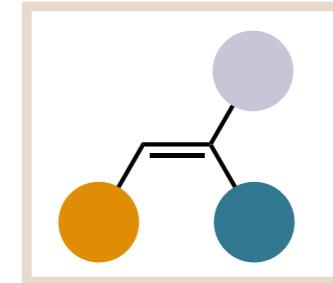
## Zweifel Olefination: State of Art



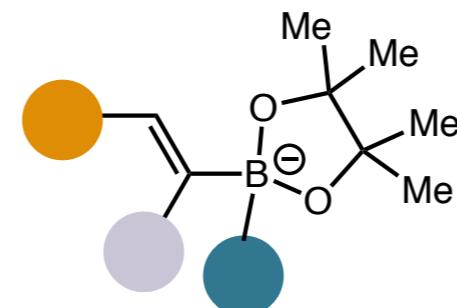
1. *t*-BuLi  
2. Bpin —  
3. PhSeCl  
4. *m*-CPBA



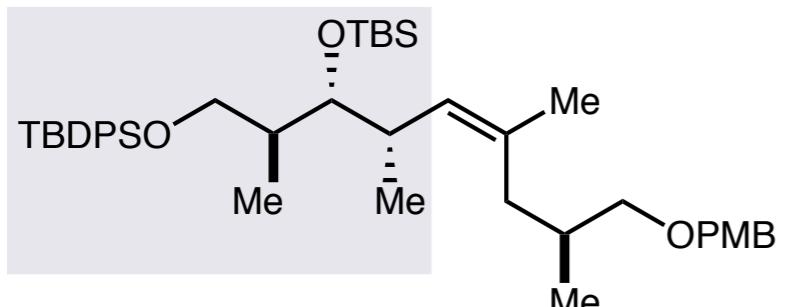
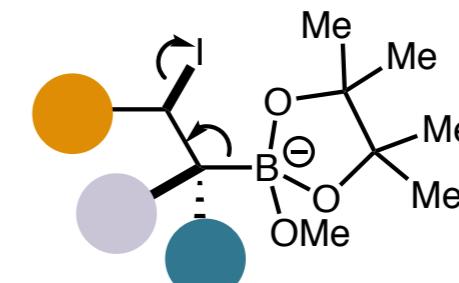
1. *t*-BuLi  
2. Bpin —  
3. I<sub>2</sub>, NaOMe



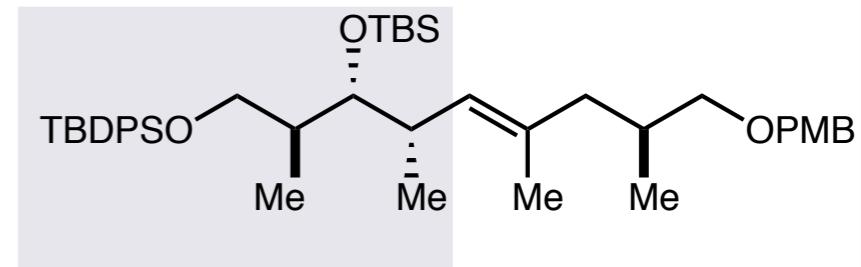
● Excellent Syn Elimination



● High Reactive Olefin

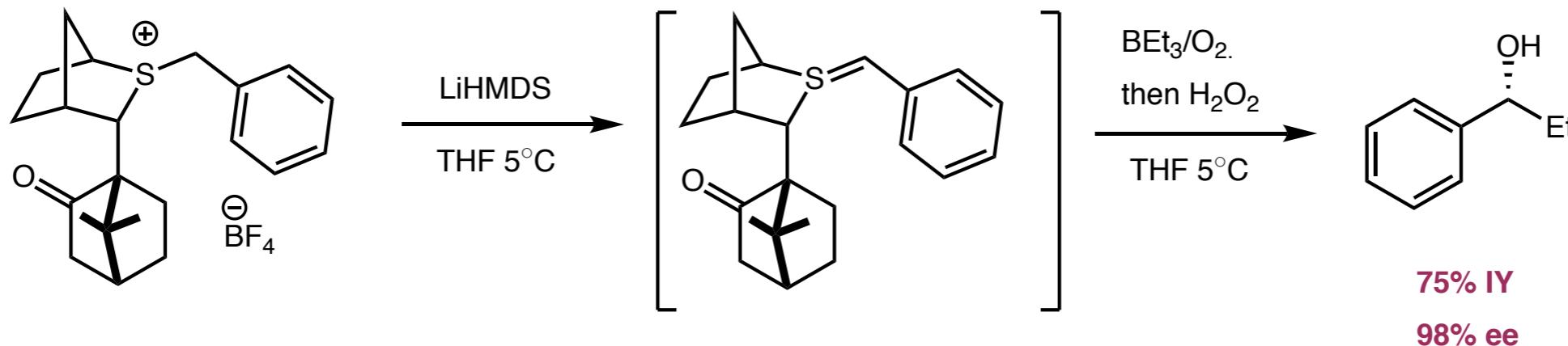


45% IY > 20:1 d.r.  
98:2 Z/E

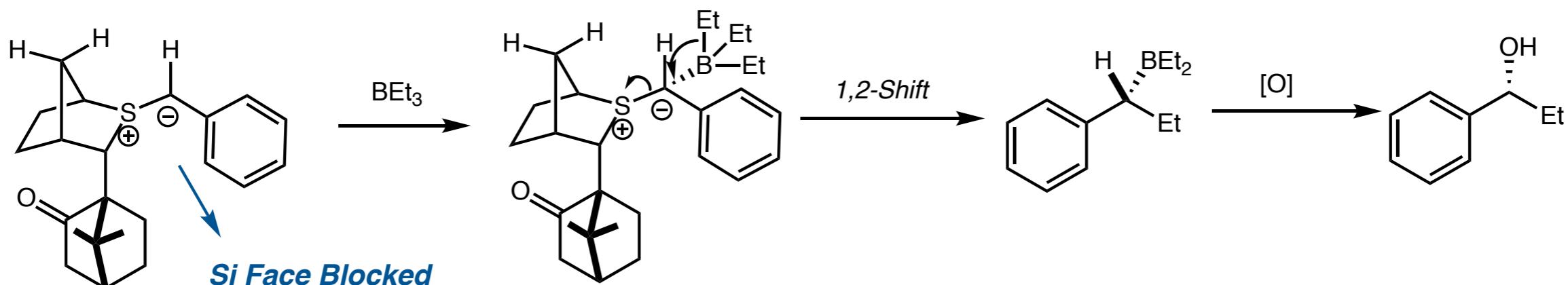


52% IY > 20:1 d.r.  
98:2 Z/E

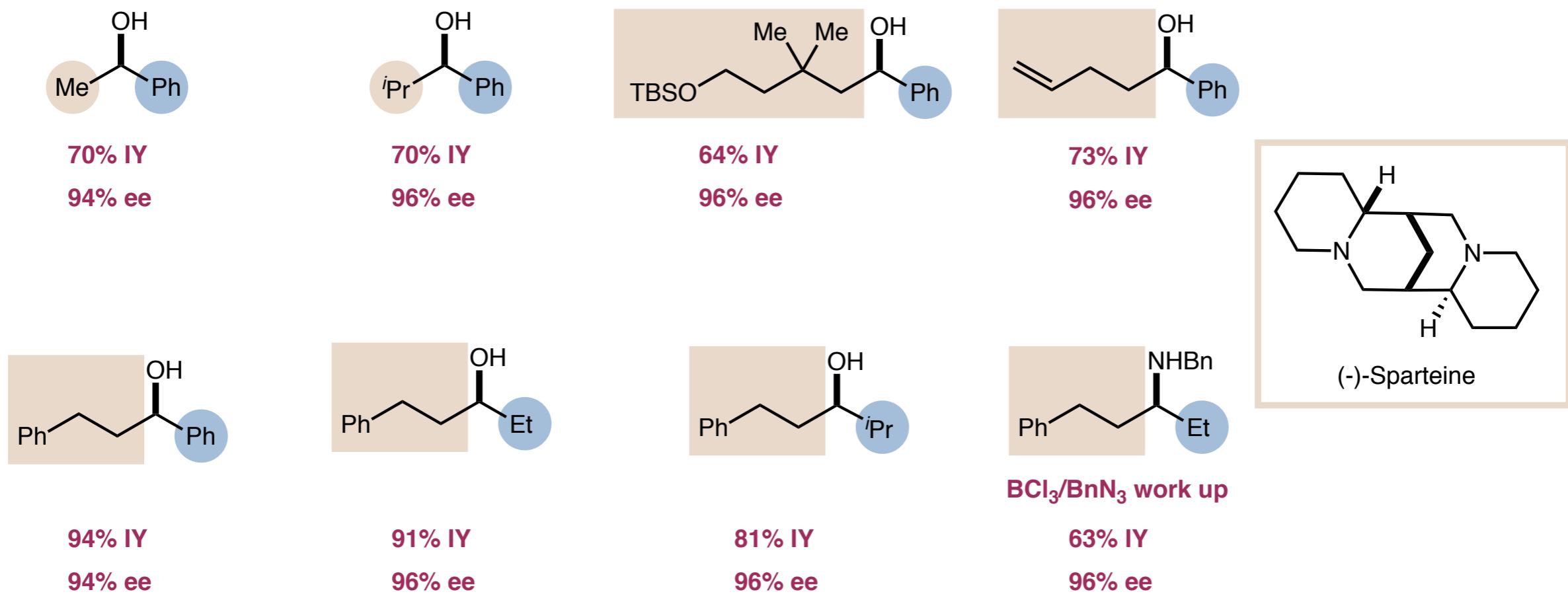
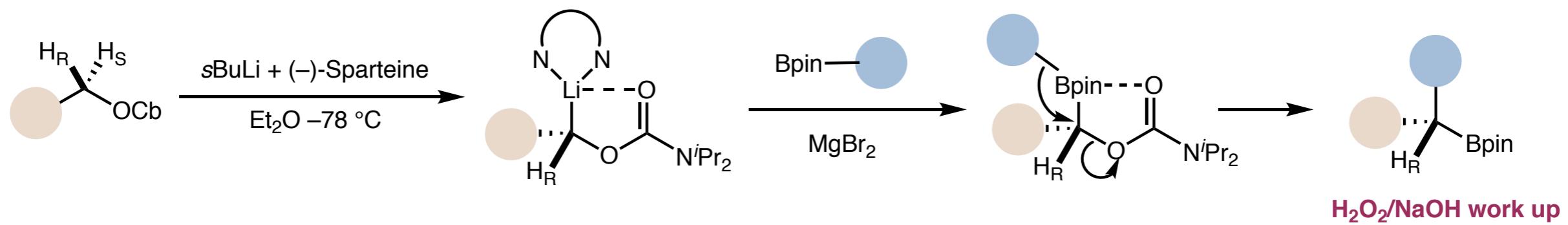
# Aggarwal Homologation: Accidental Discovery?



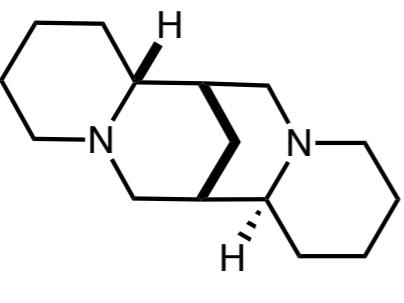
## Proposed Mechanism



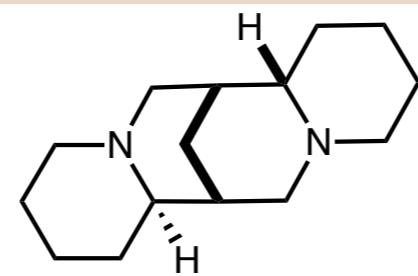
## Aggarwal Homologation: New Leaving Group



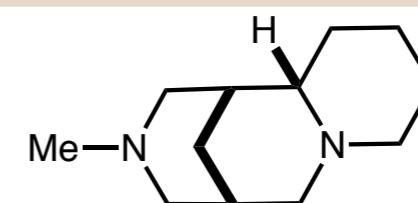
# *Aggarwal Homologation: Reagent Controlled Chirality Transfer*



*Used to be Widely Available*

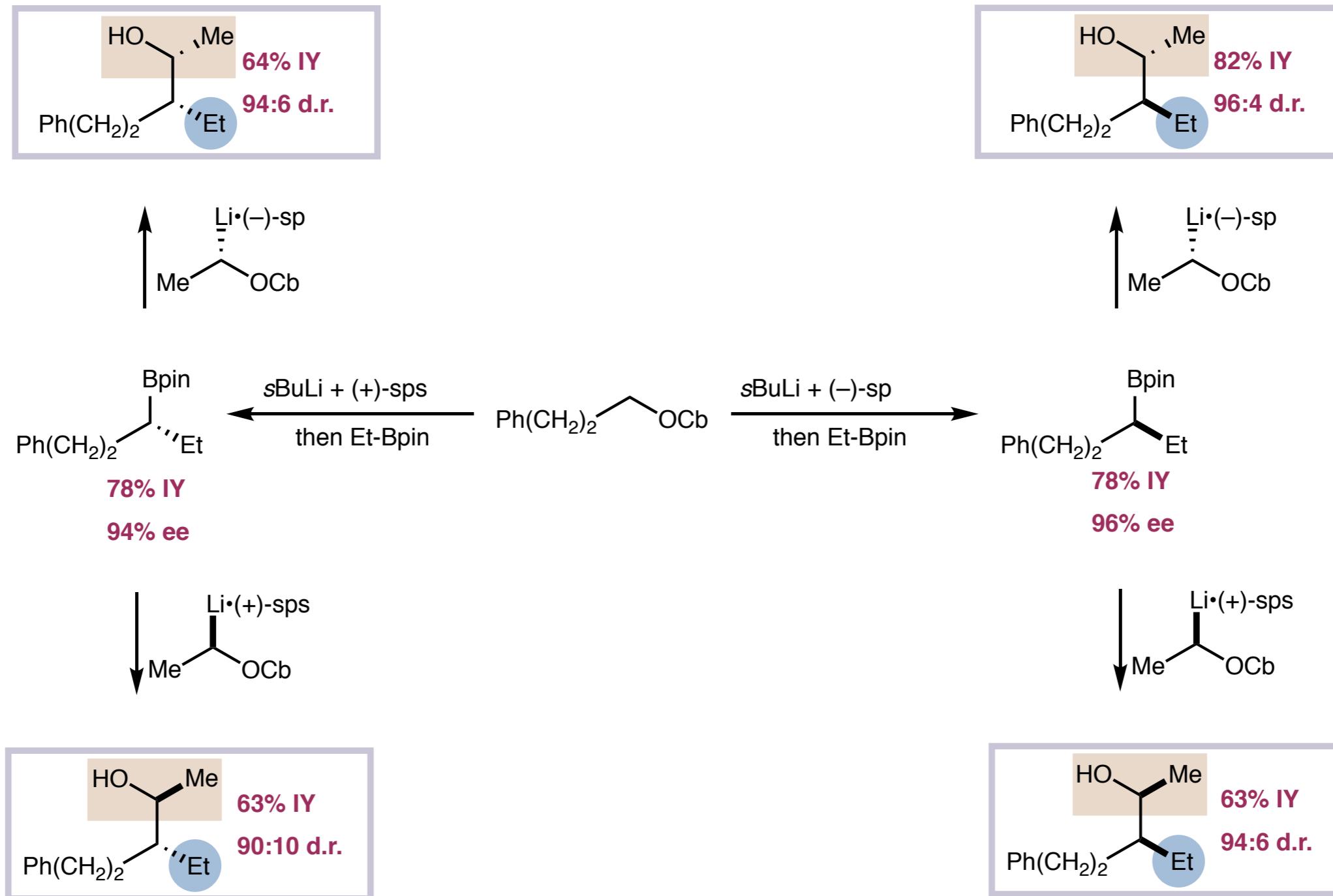


*Not Available*

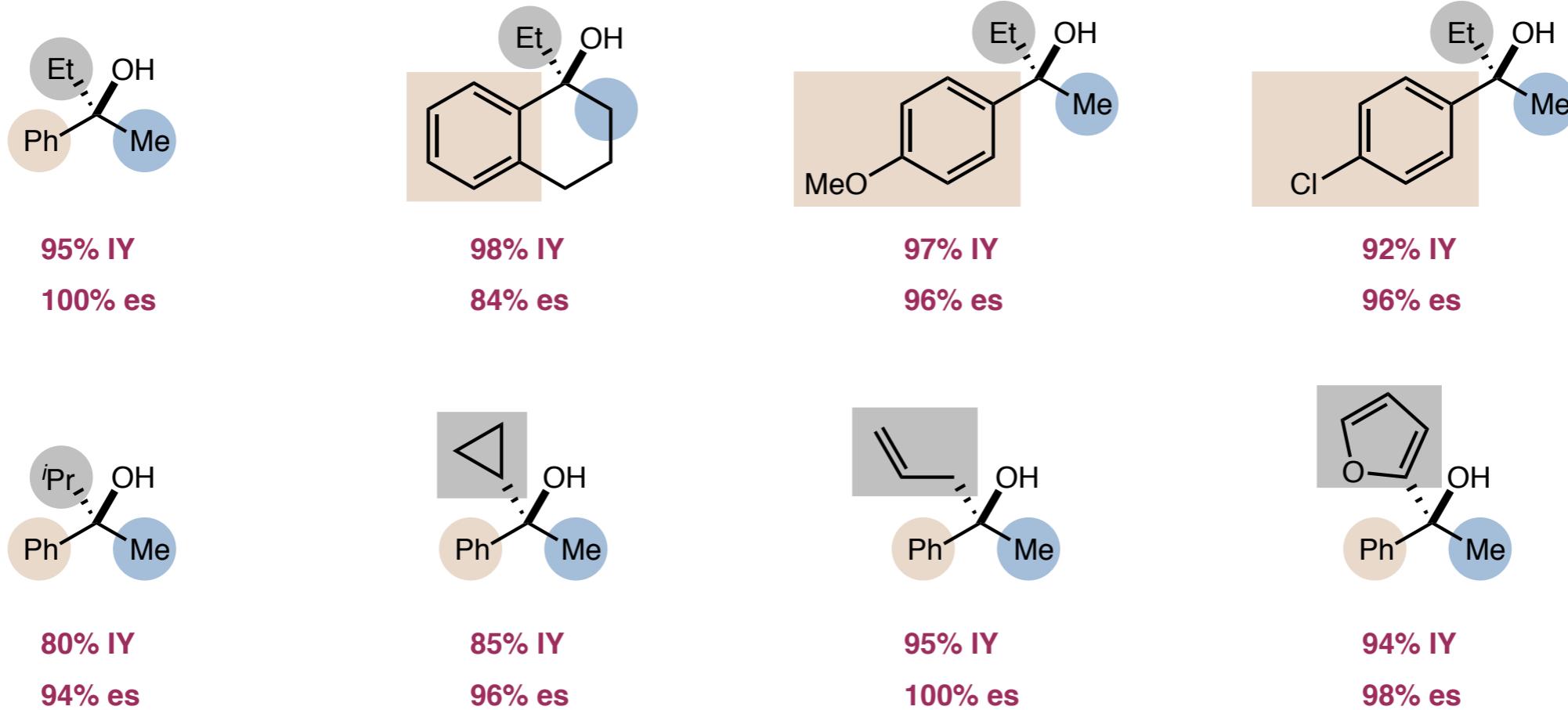
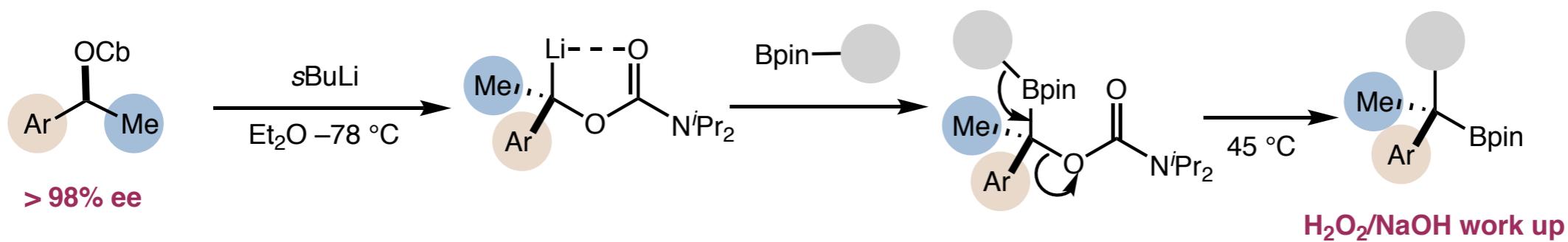


*Not Widely Commercial Available*

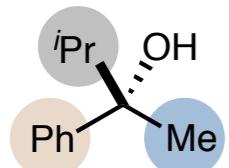
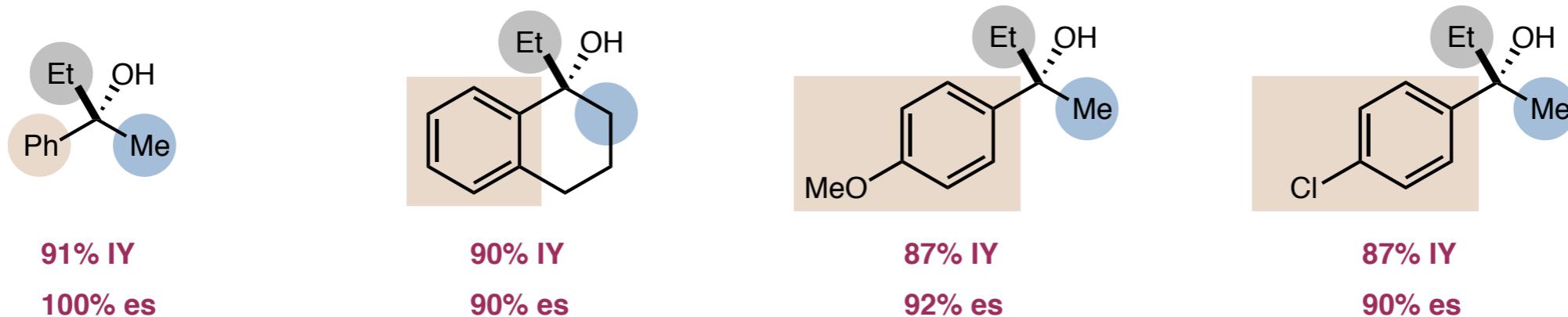
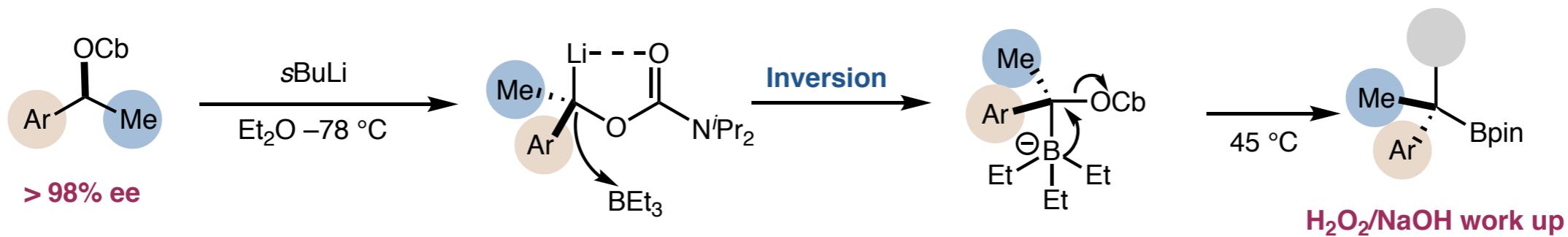
# Aggarwal Homologation: Reagent Controlled Chirality Transfer



## Transfer Alcohol Chirality to C-B bond



## Transfer Alcohol Chirality to C-B bond



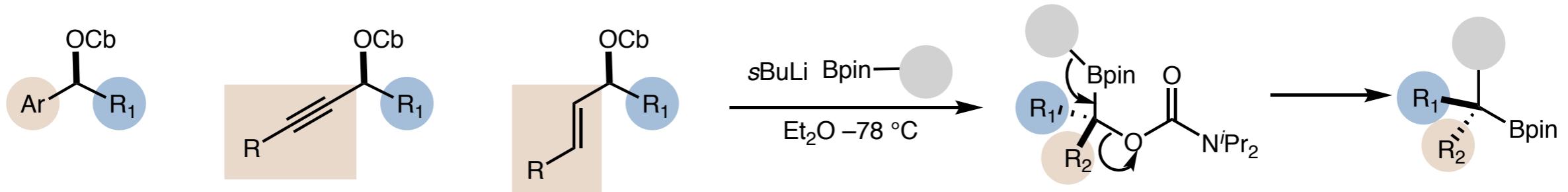
**91% IY**

**96% es**

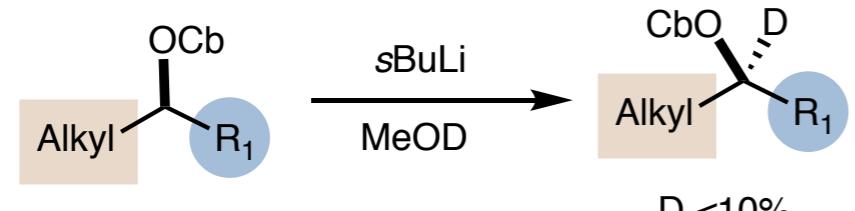
**From *i*Pr-9-BBN**

**Inversion only happen when carbon-anion is benzylic**

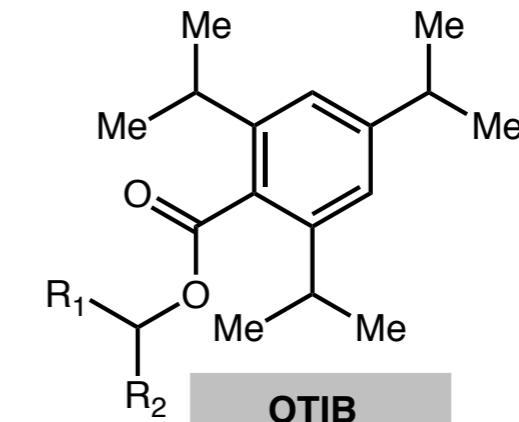
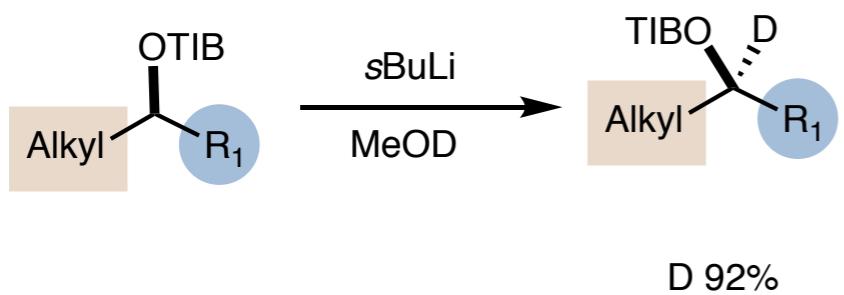
# From Benzylic Alcohol to Normal Secondary Alcohol



## Acidic C-H Bond

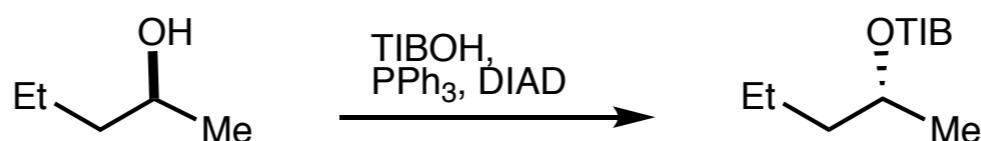


New leaving group required



- More Electron-Withdraw
- Comformationally Stable Anion

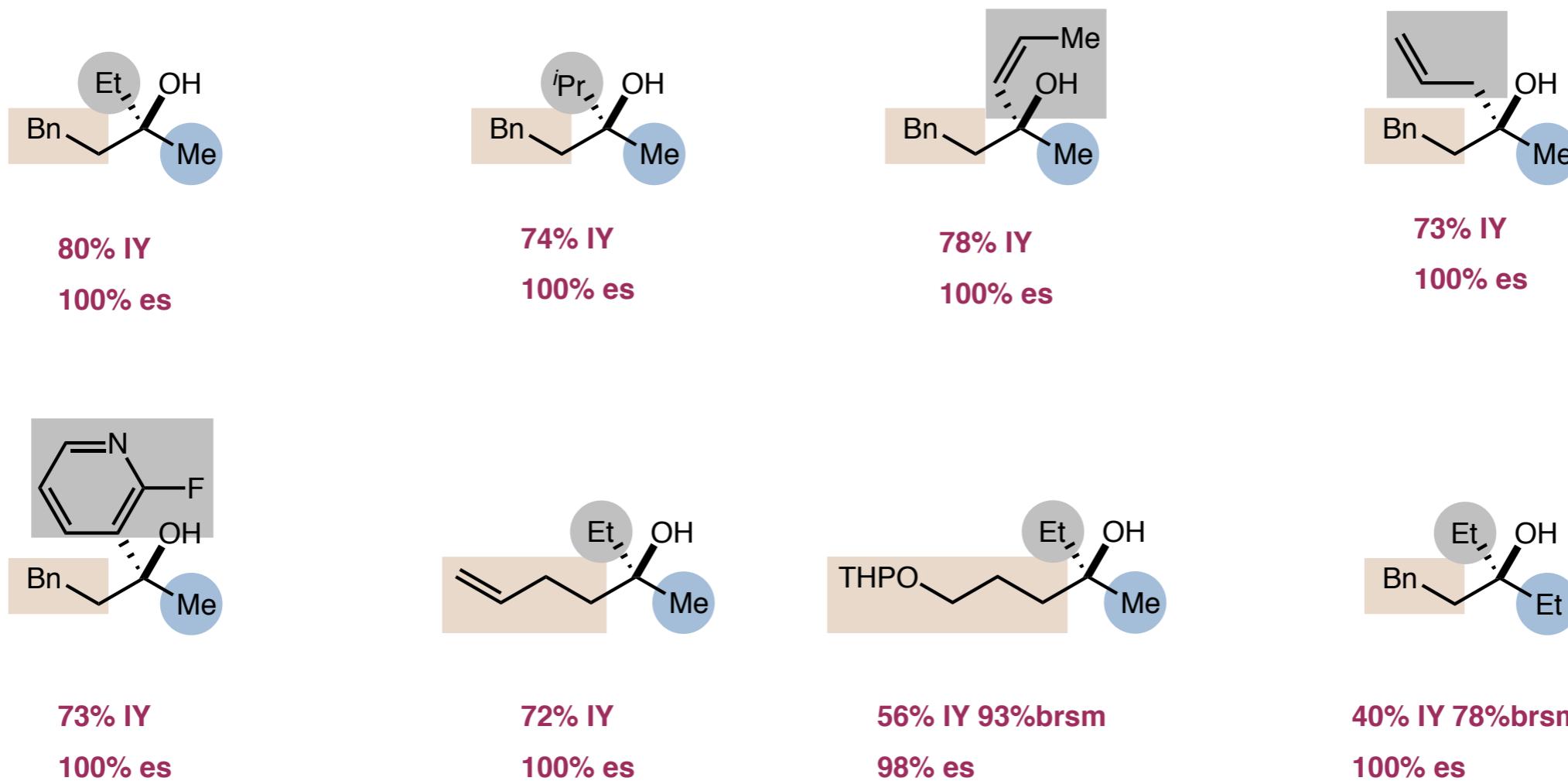
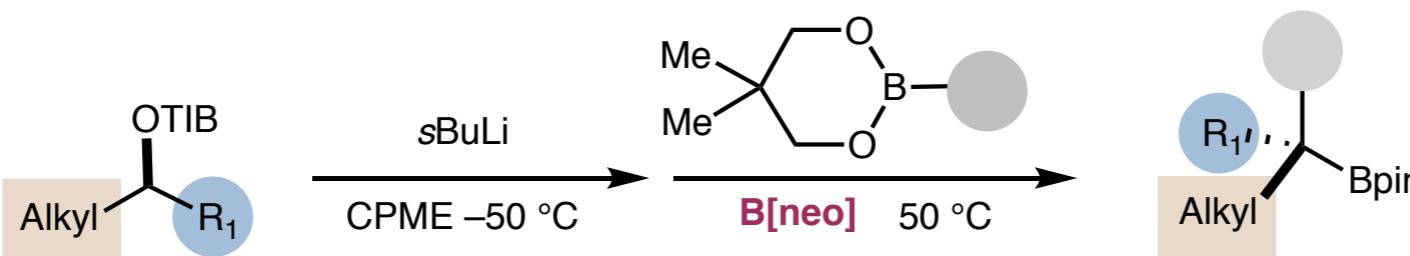
## Installation:



Pulis, A. P.; Aggarwal, V. K. *J. Am. Chem. Soc.* **2012**, *134*, 7570.

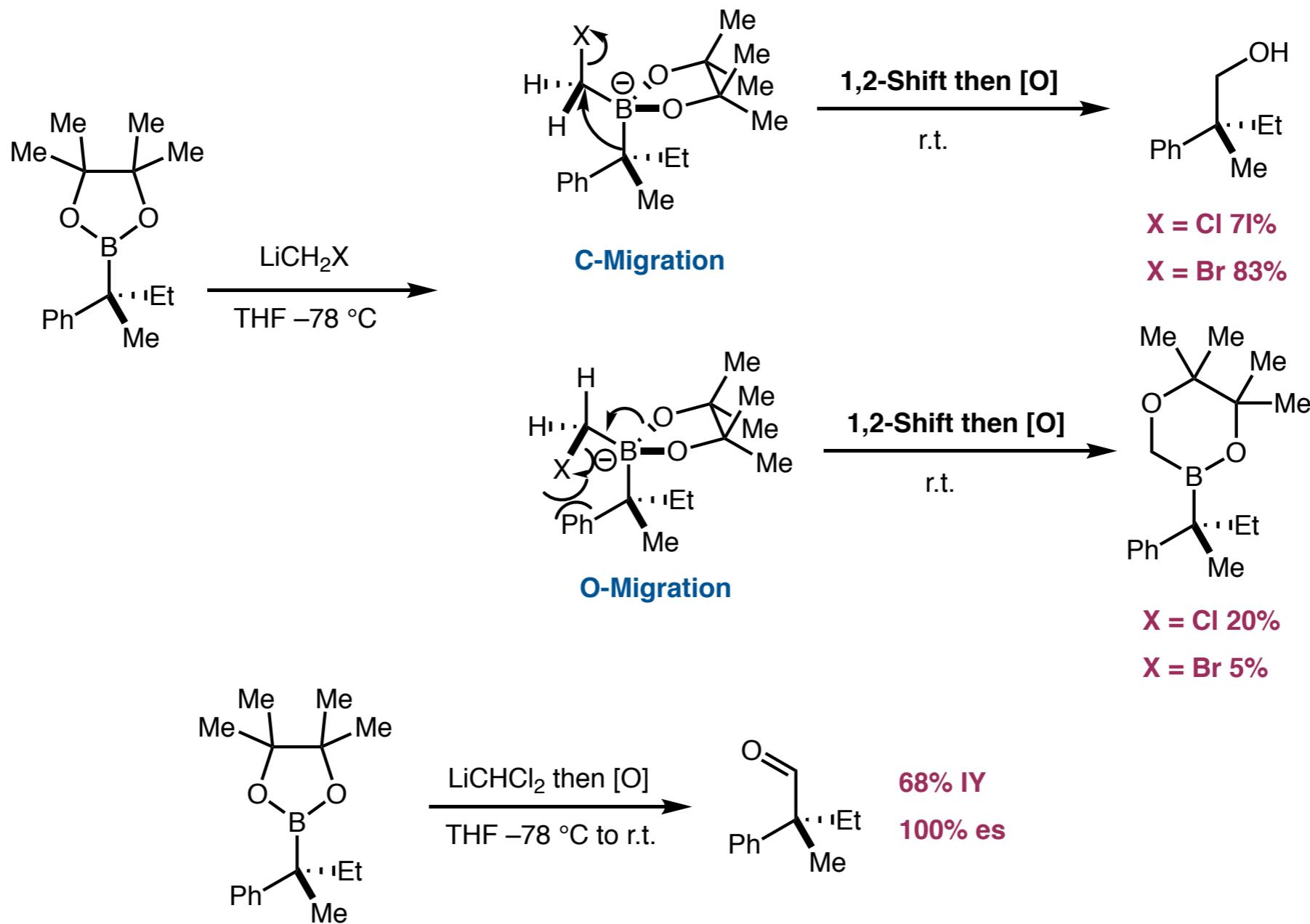
Partridge, B. M.; Chausset-Boissarie, L.; Burns, M.; Pulis, A. P.; Aggarwal, V. K. *Angew. Chem. Int. Ed.* **2012**, *51*, 11795.

## From Benzylic Alcohol to Normal Secondary Alcohol



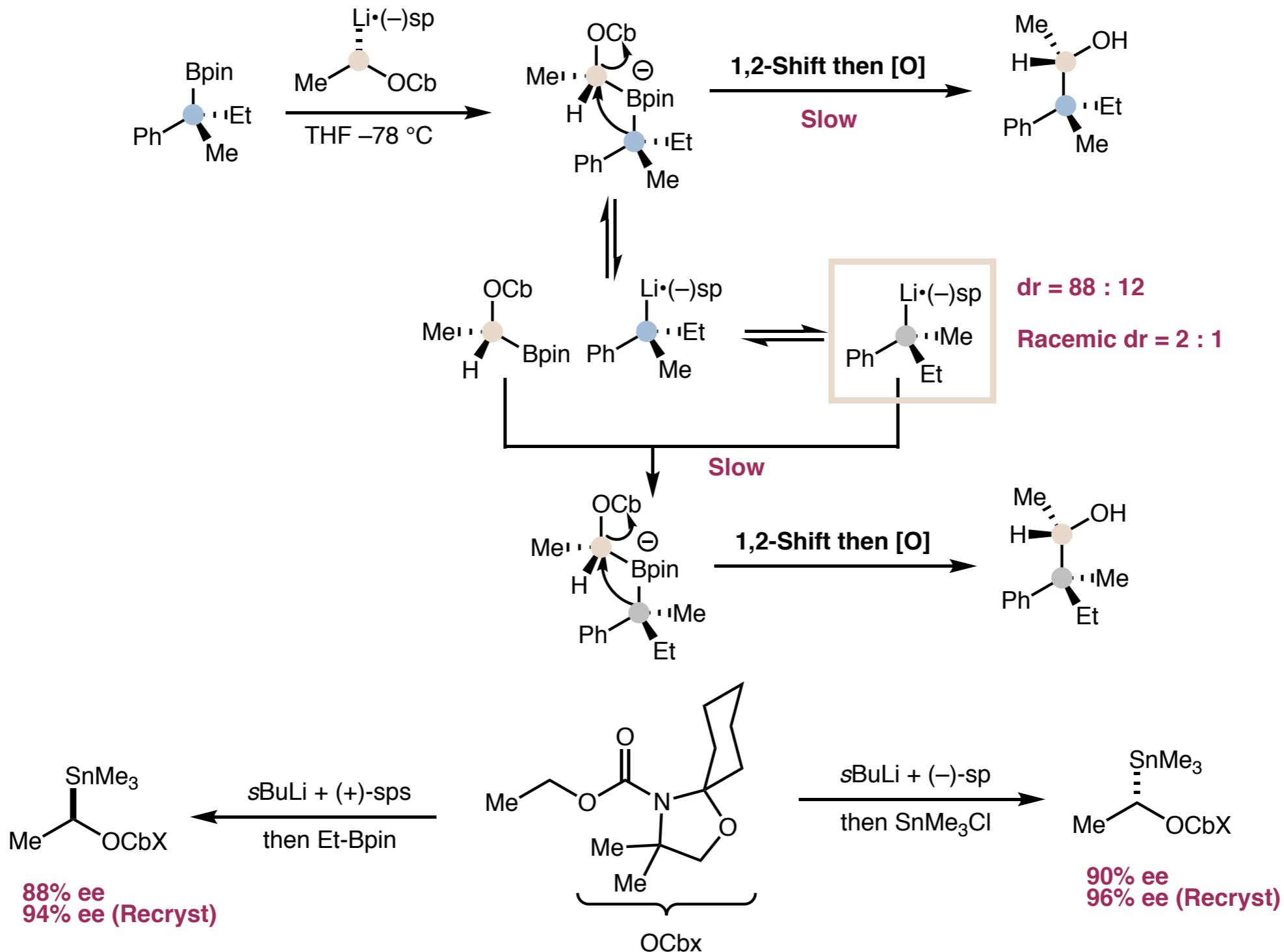
# From Tertiary Boronic Acid to Quaternary Carbon Center

## Quaternary Carbon + Methylenes



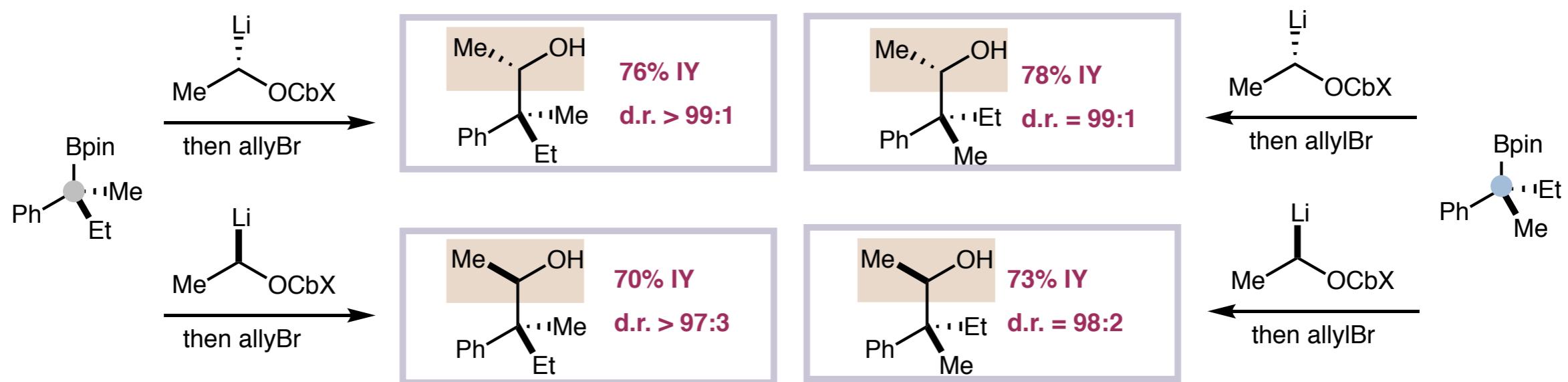
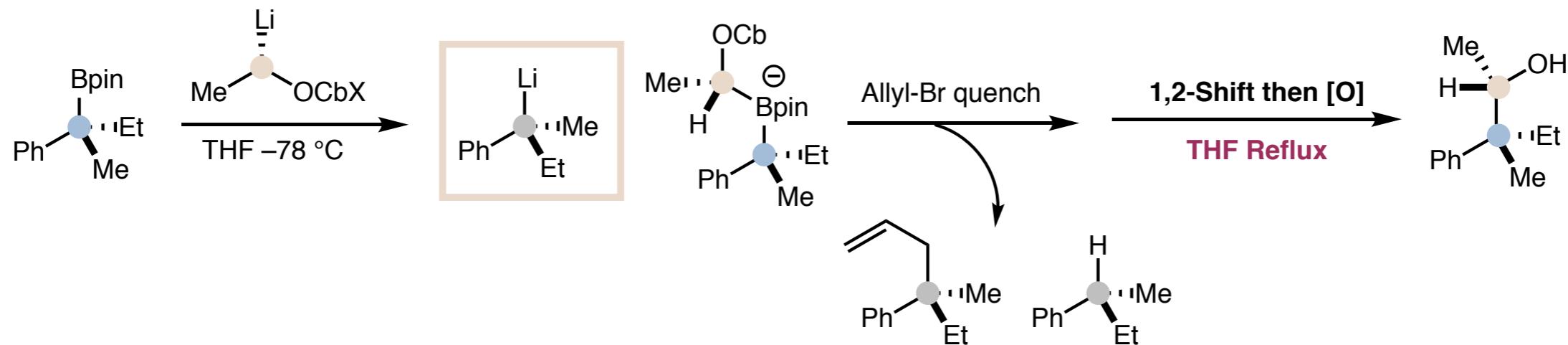
# From Tertiary Boronic Acid to Quaternary Carbon Center

## Quaternary Carbon + Methine



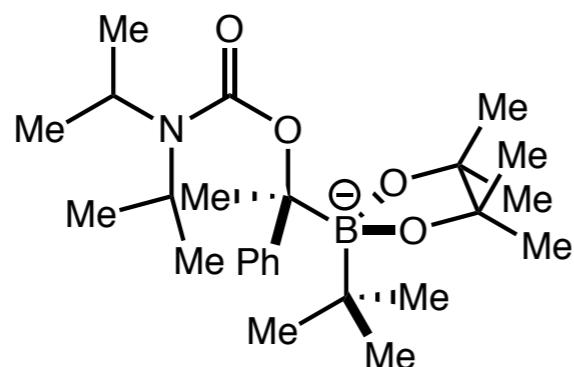
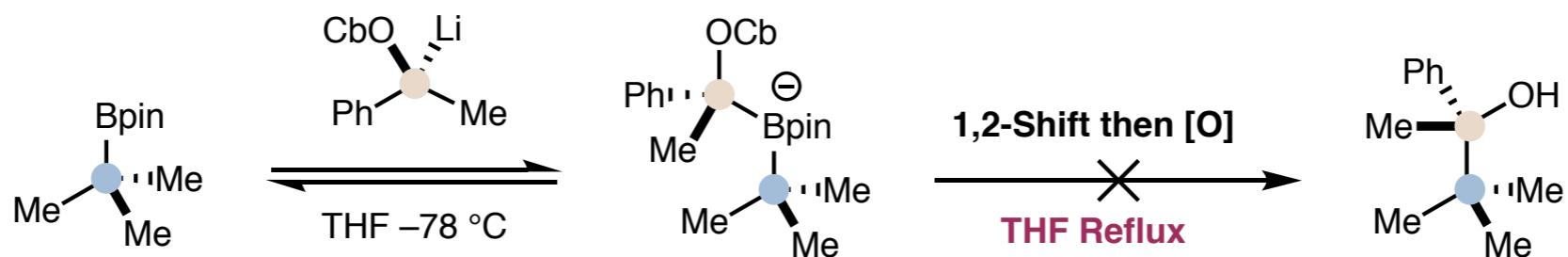
# From Tertiary Boronic Acid to Quaternary Carbon Center

## Quaternary Carbon + Methine

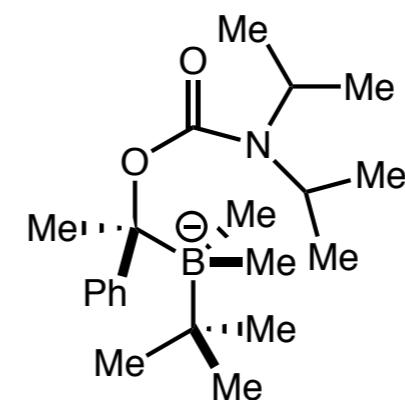


# From Tertiary Boronic Acid to Quaternary Carbon Center

## Quaternary Carbon + Quaternary Carbon

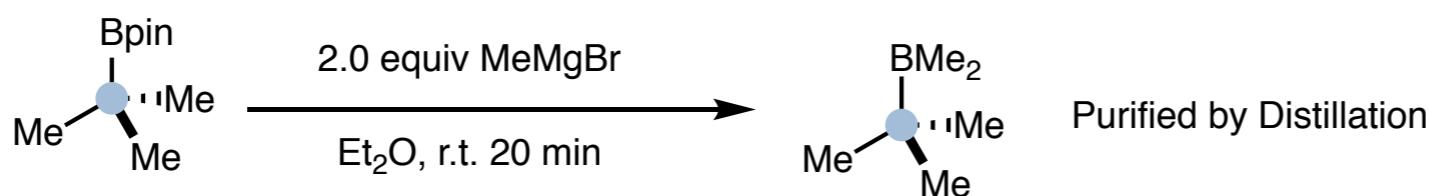


**Too Hindered to form**



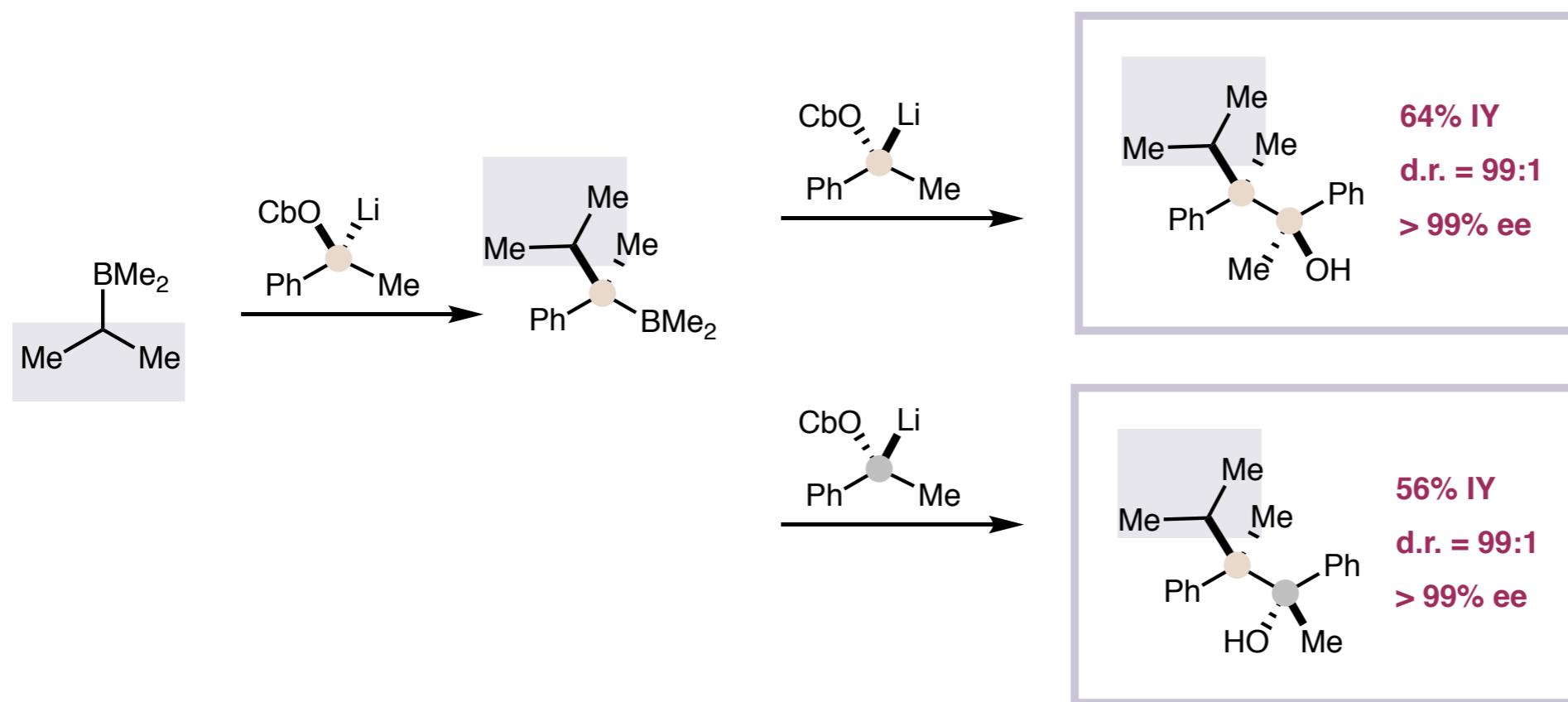
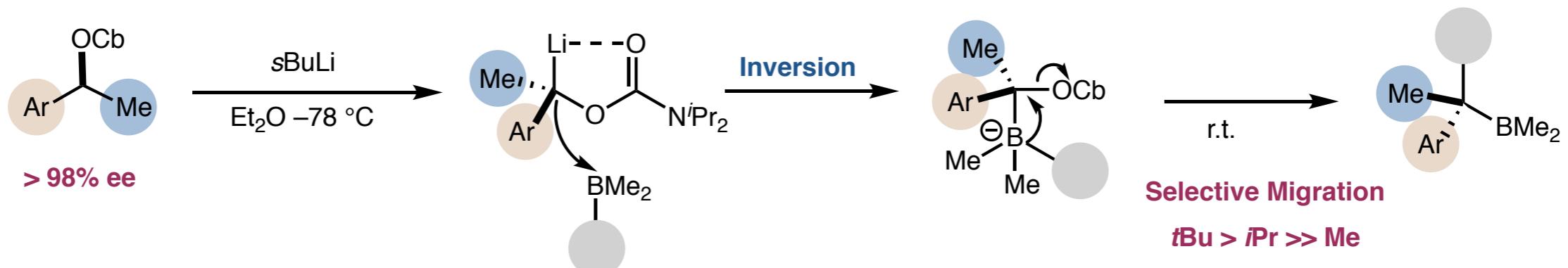
***t*BuBMe<sub>2</sub>**

- *Reduced Steric Hindrance*
- *More Positive Boron Center*



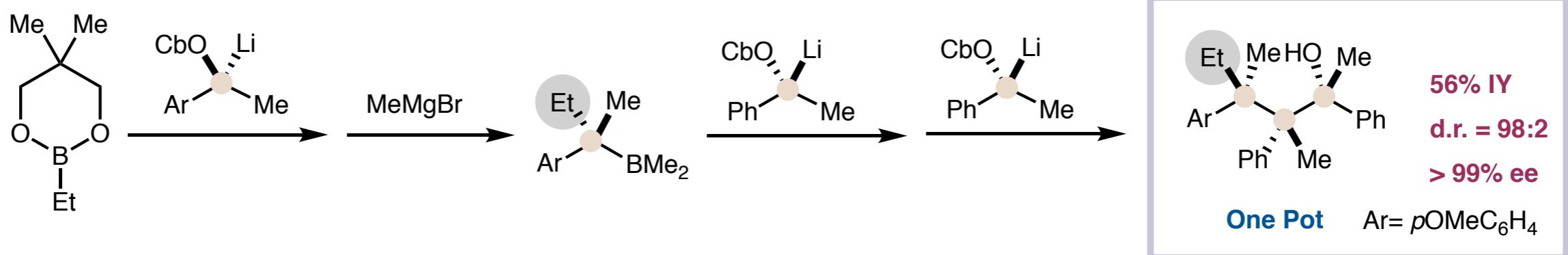
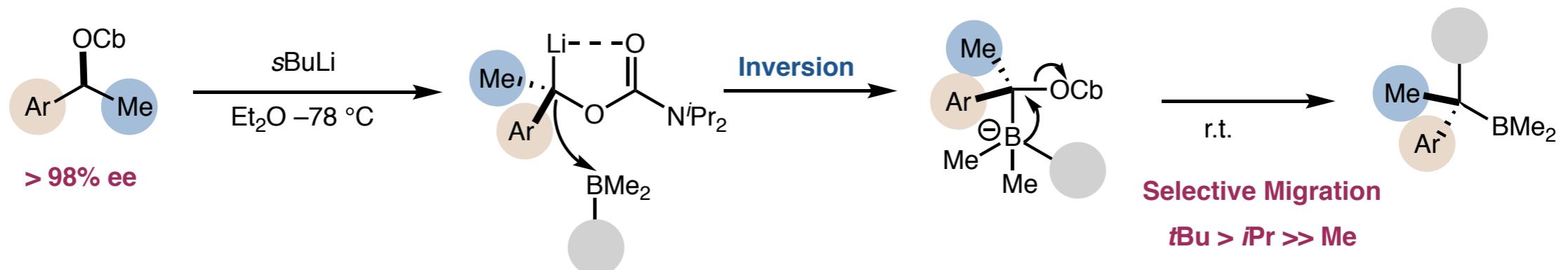
# From Tertiary Boronic Acid to Quaternary Carbon Center

## Quaternary Carbon + Quaternary Carbon

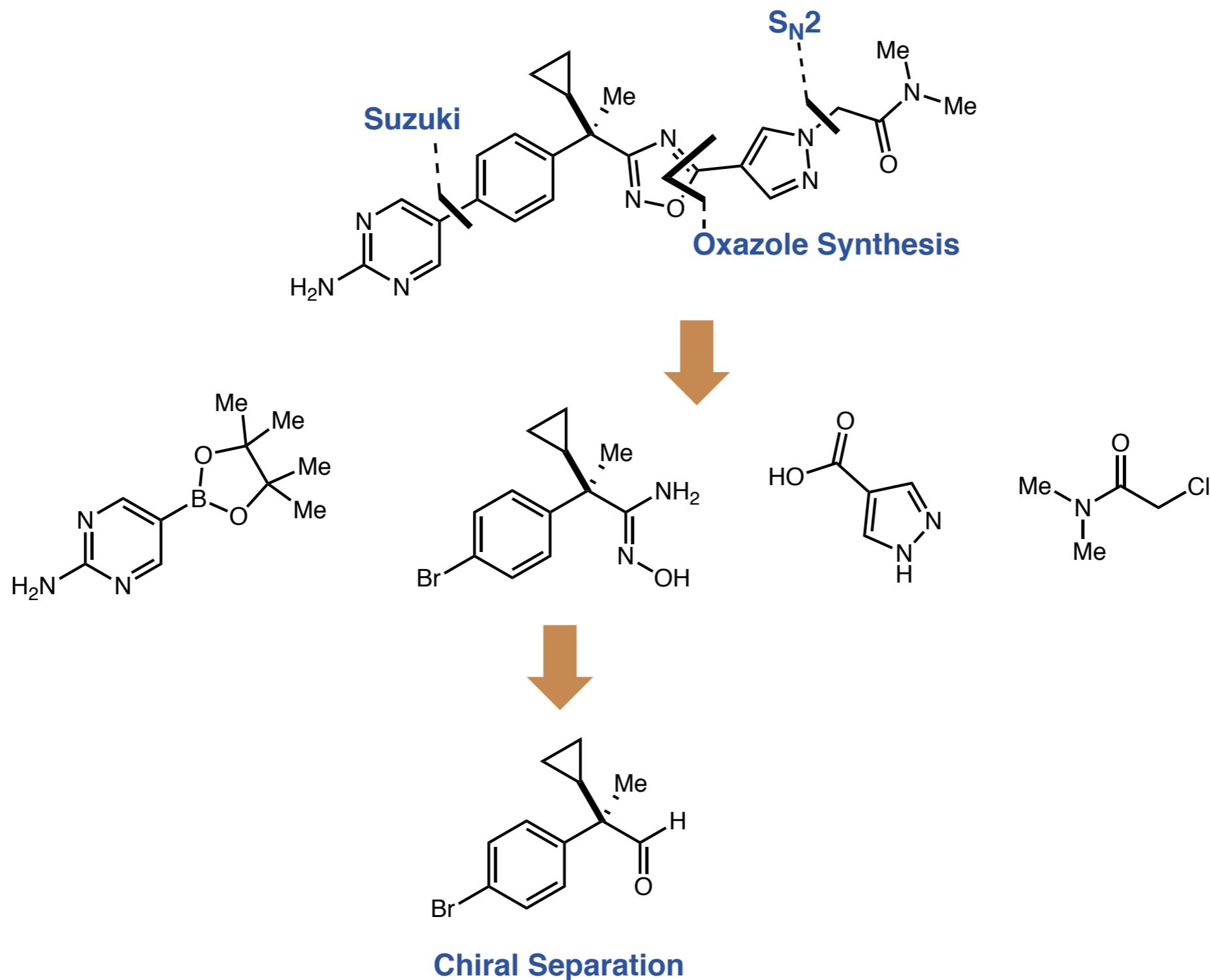


# From Tertiary Boronic Acid to Quaternary Carbon Center

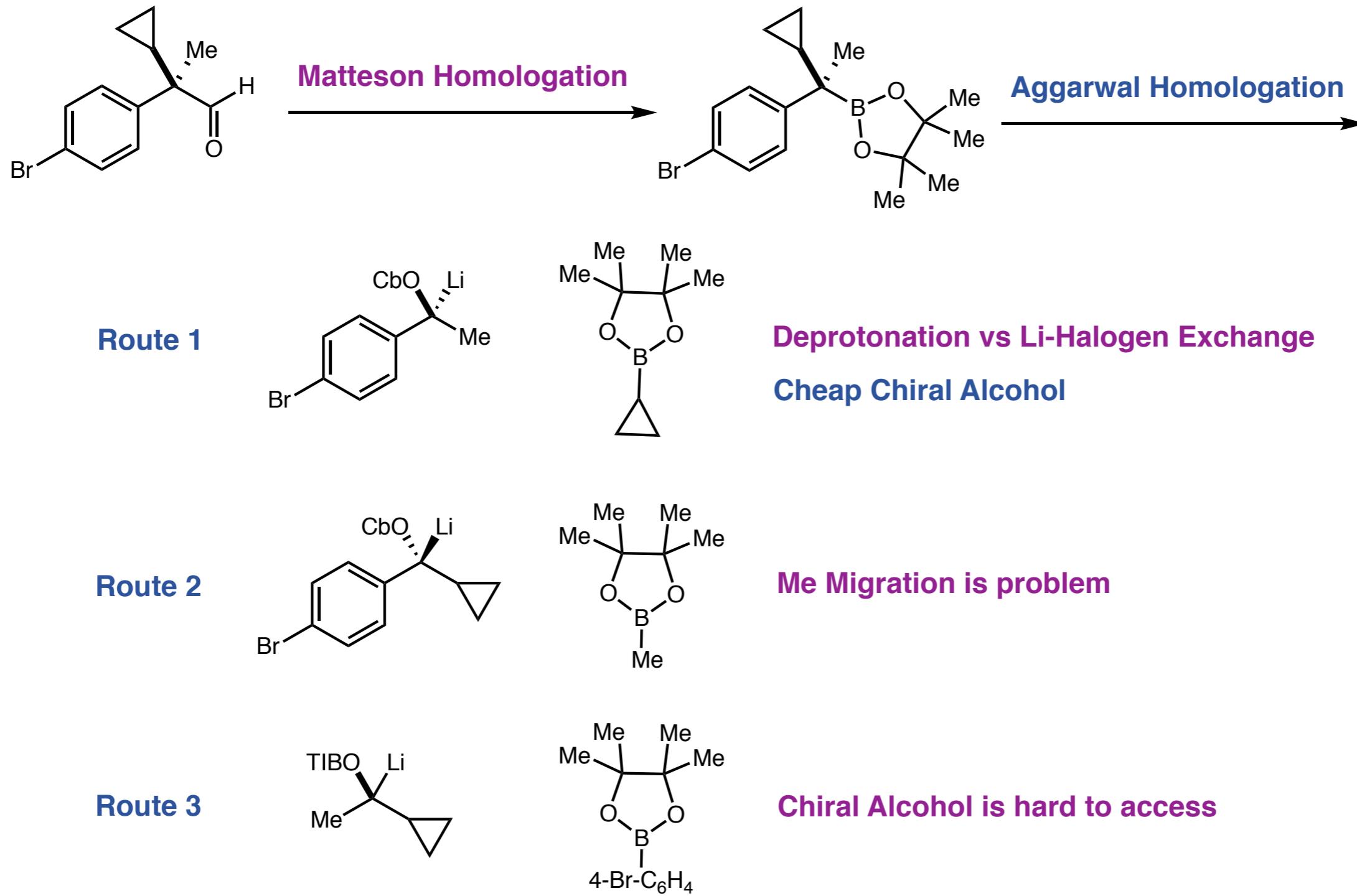
## Quaternary Carbon + Quaternary Carbon



# *Drug Candidate Synthesis*



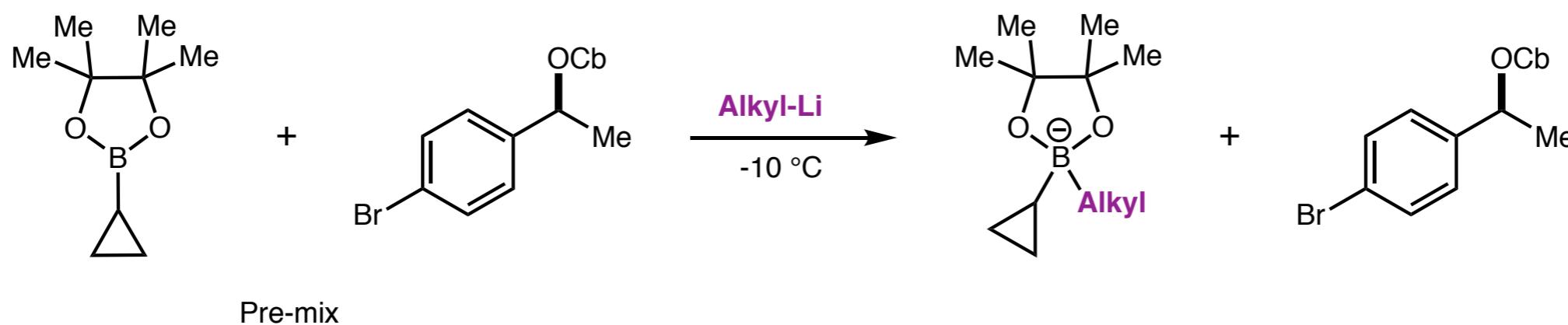
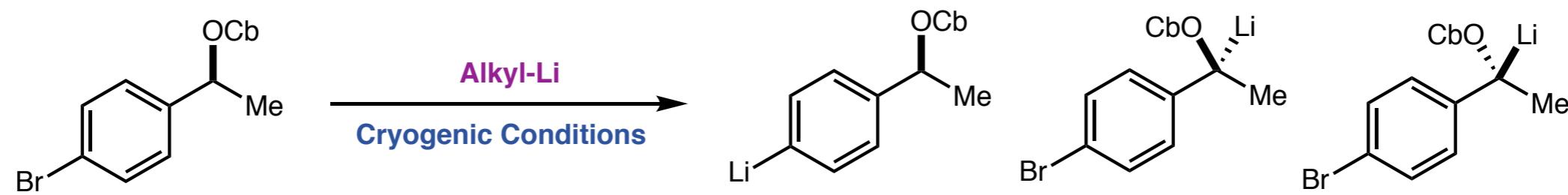
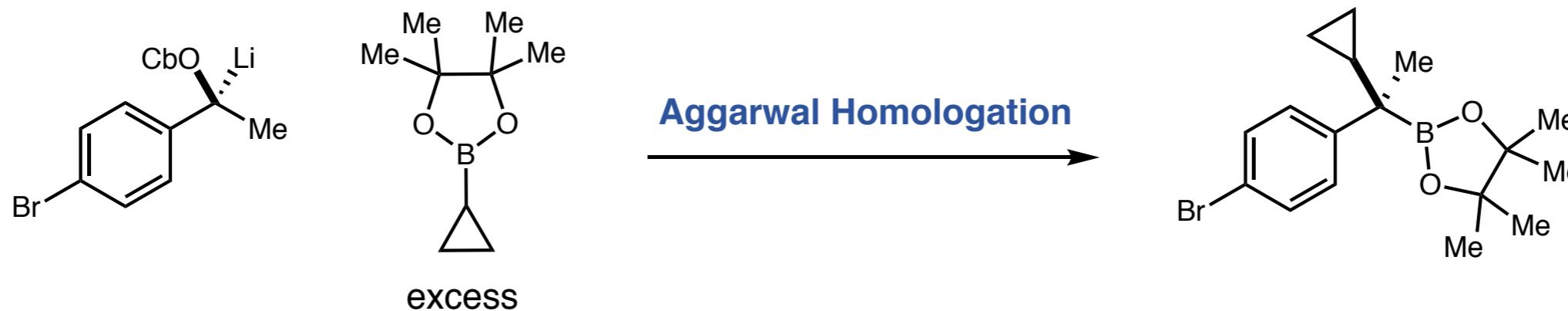
# Drug Candidate Synthesis



Senanayake, C. H. et al. *Org. Lett.* 2014, 16, 4360.

Senanayake, C. H. et al. *J. Org. Chem.* 2015, 80, 1651.

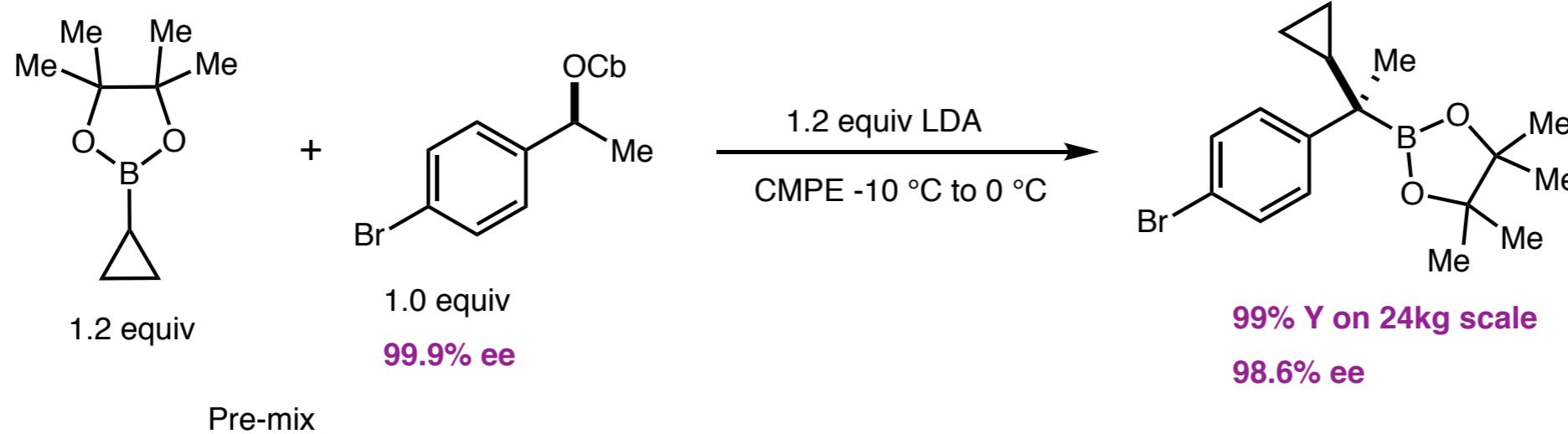
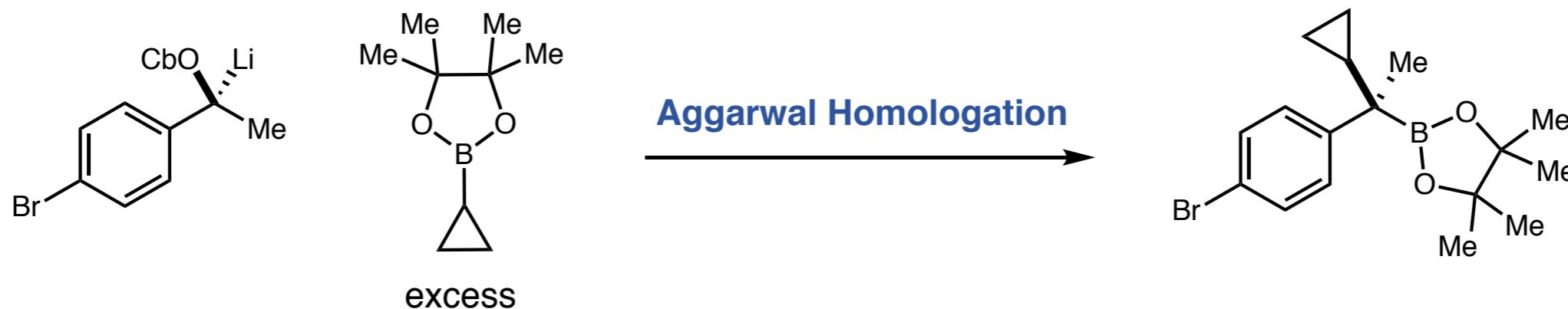
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Senanayake, C. H. et al. *Org. Lett.* 2014, 16, 4360.

Senanayake, C. H. et al. *J. Org. Chem.* 2015, 80, 1651.

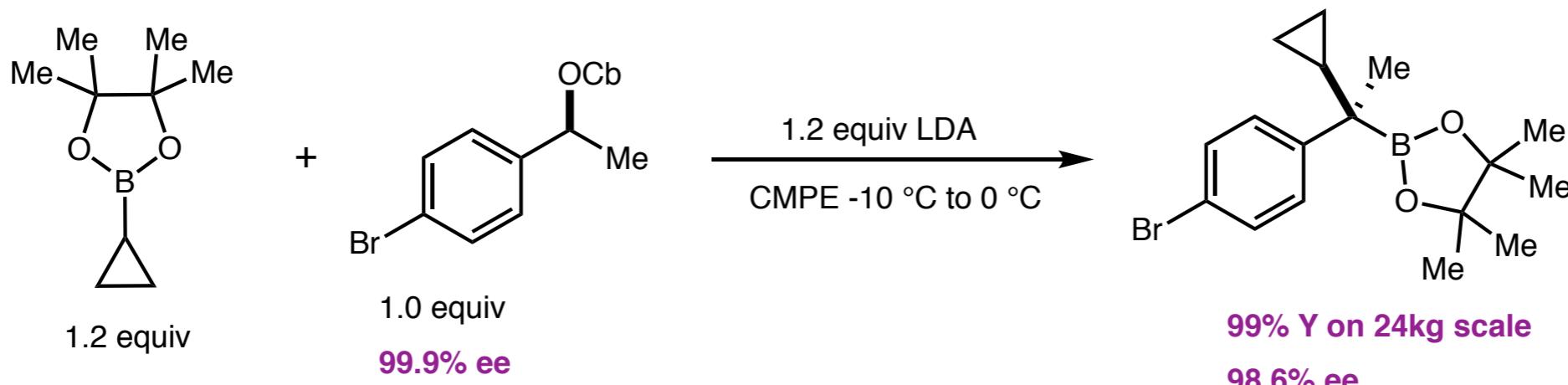
# Drug Candidate Synthesis



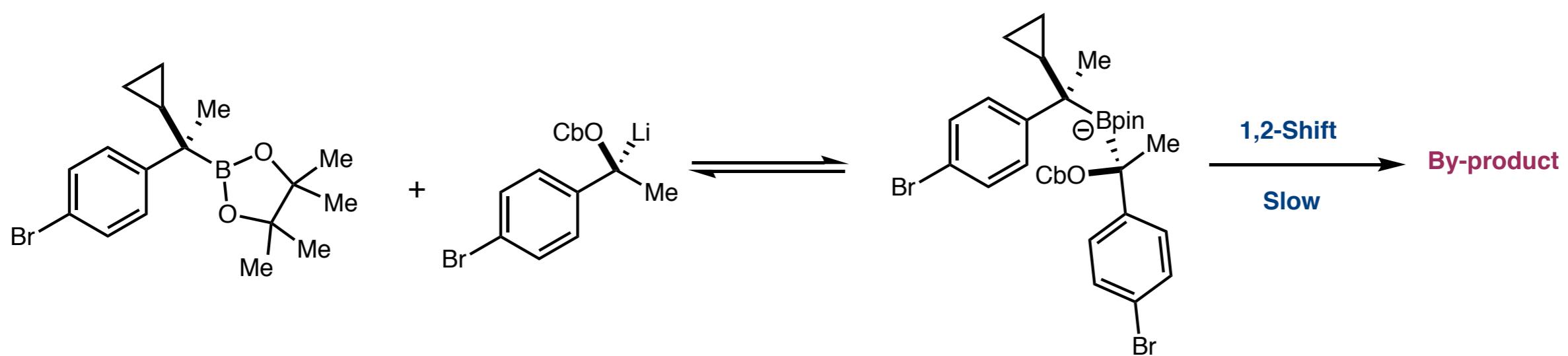
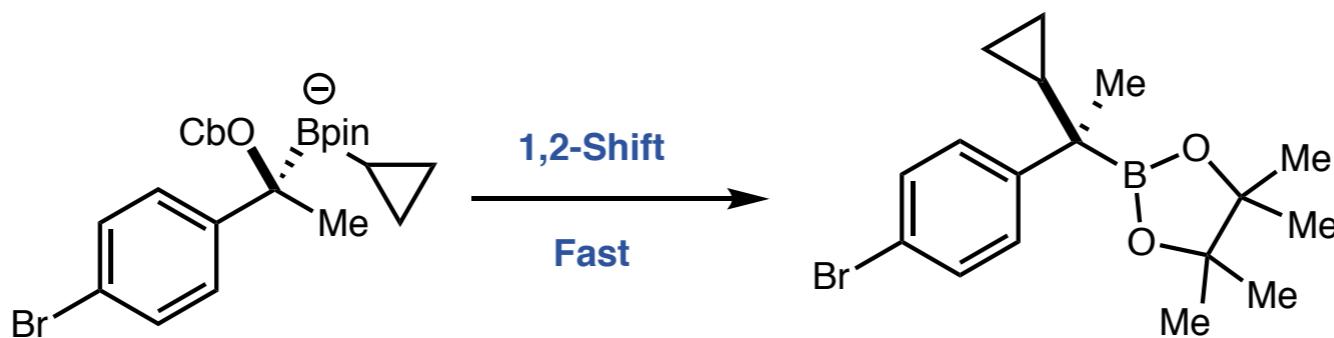
Senanayake, C. H. et al. *Org. Lett.* 2014, 16, 4360.

Senanayake, C. H. et al. *J. Org. Chem.* 2015, 80, 1651.

# Drug Candidate Synthesis



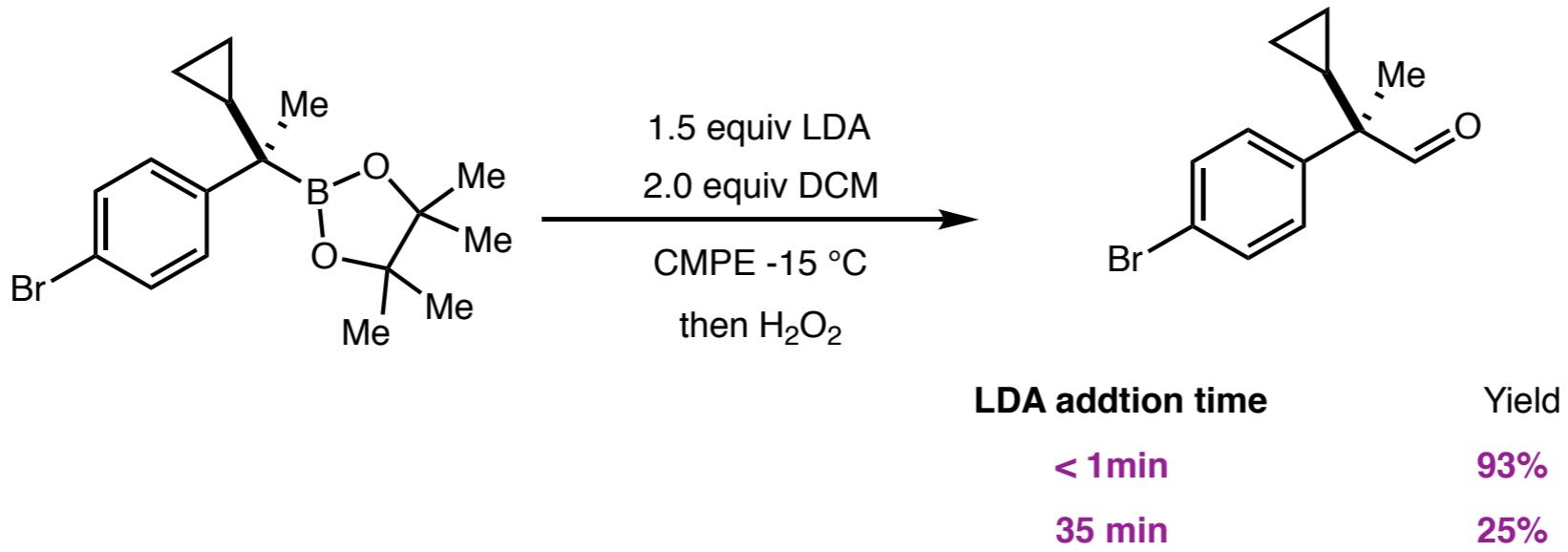
Pre-mix



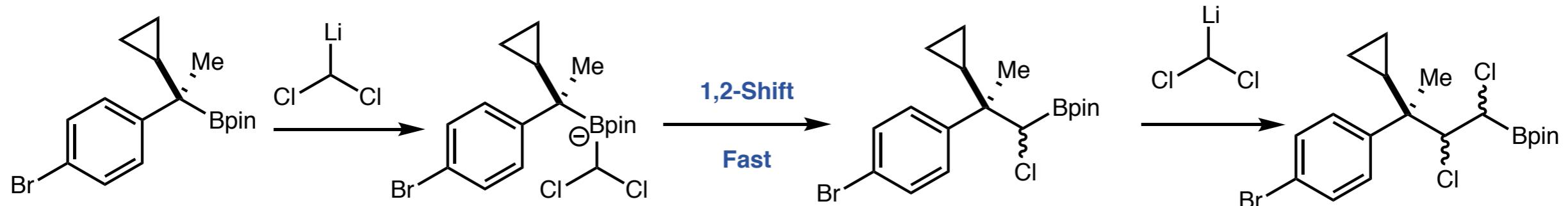
Senanayake, C. H. et al. *Org. Lett.* 2014, 16, 4360.

Senanayake, C. H. et al. *J. Org. Chem.* 2015, 80, 1651.

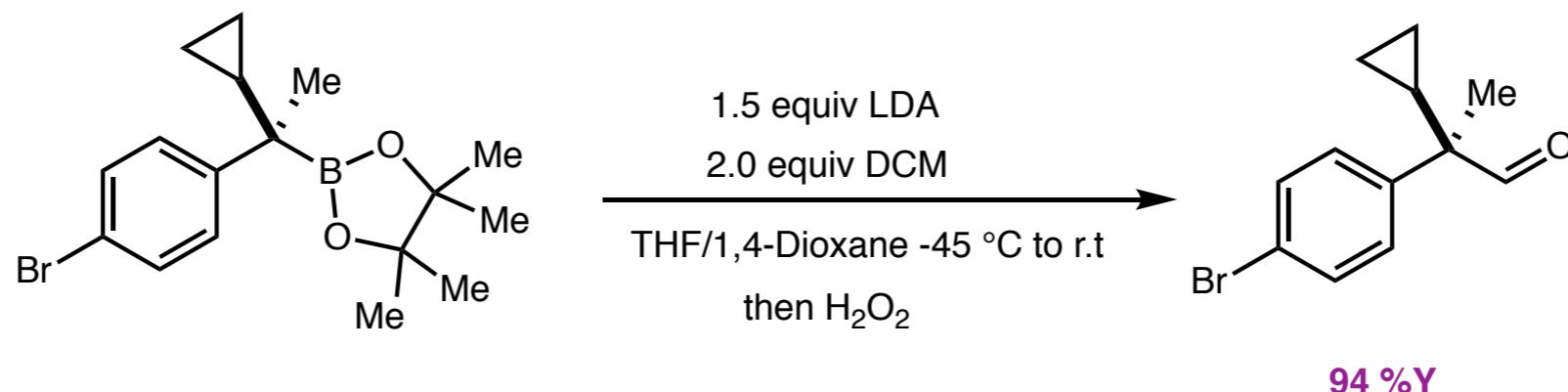
# Drug Candidate Synthesis



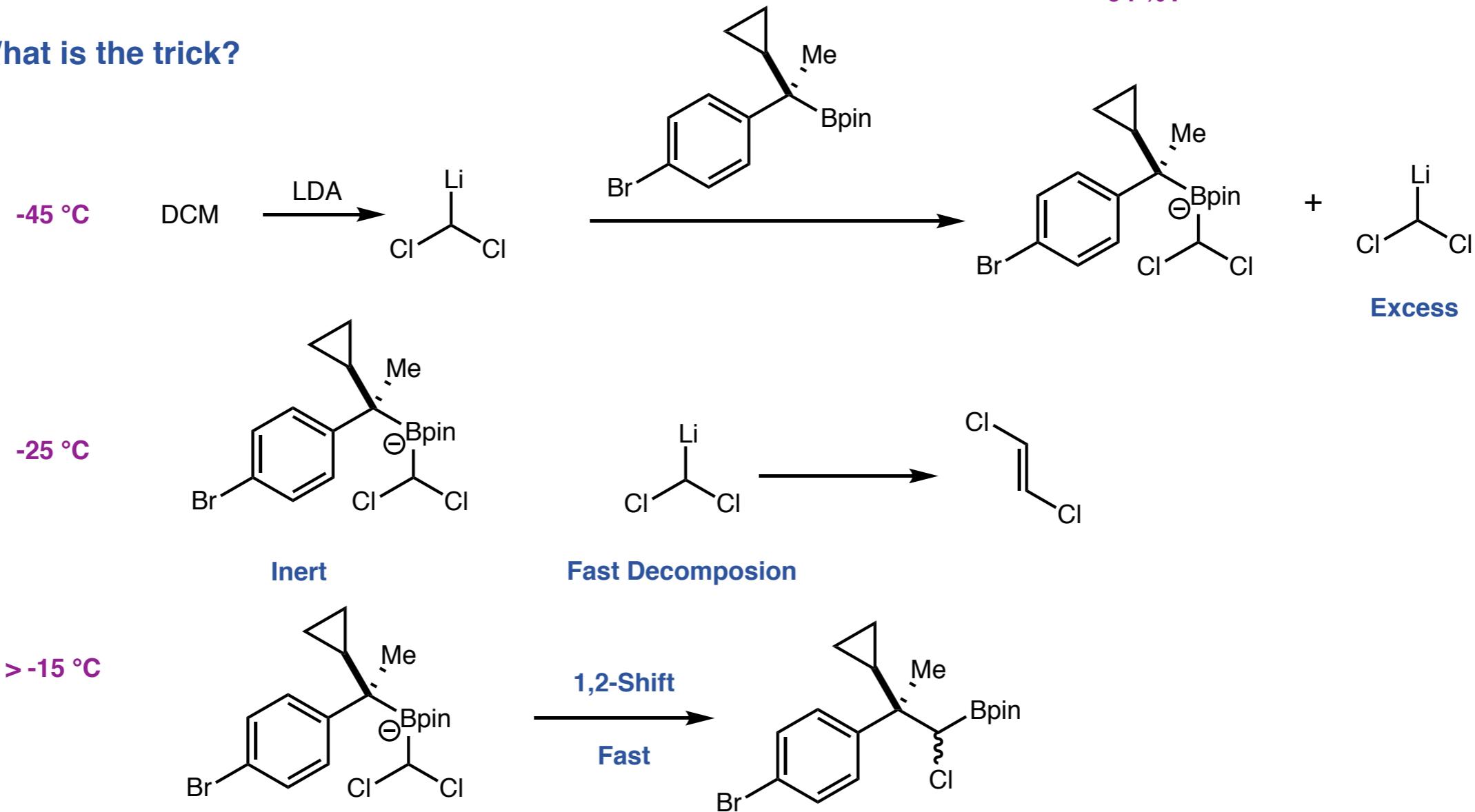
## Problem : Multiple Homologation:



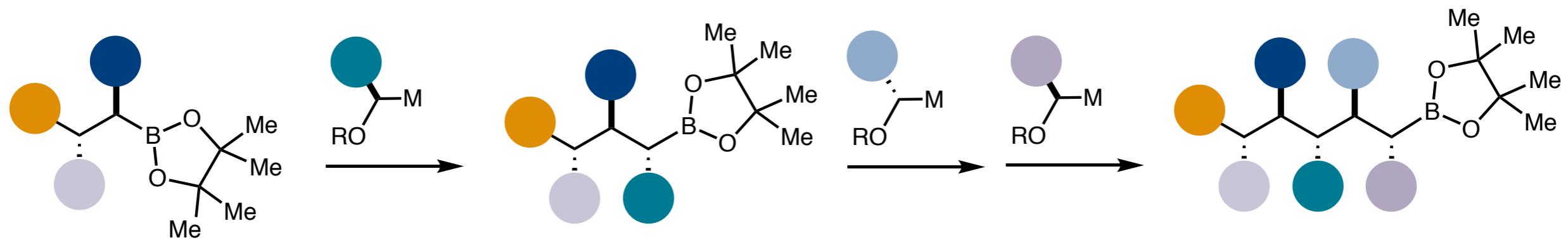
# *Drug Candidate Synthesis*



## What is the trick?

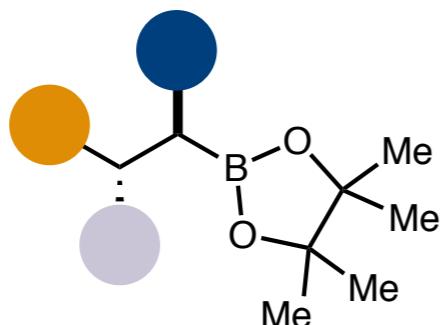


# Programmable Temperature Enabled Assemble Line Synthesis

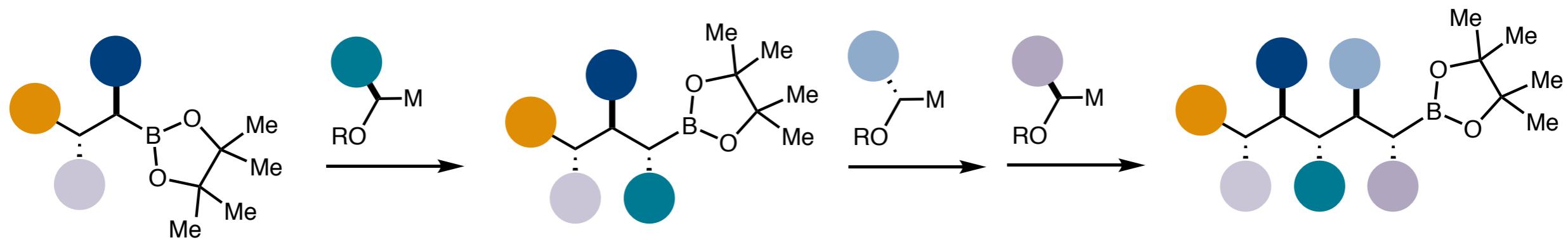


**Auto Synthesis with Reagent Controlled Chirality**

**Working Hypothesis:**

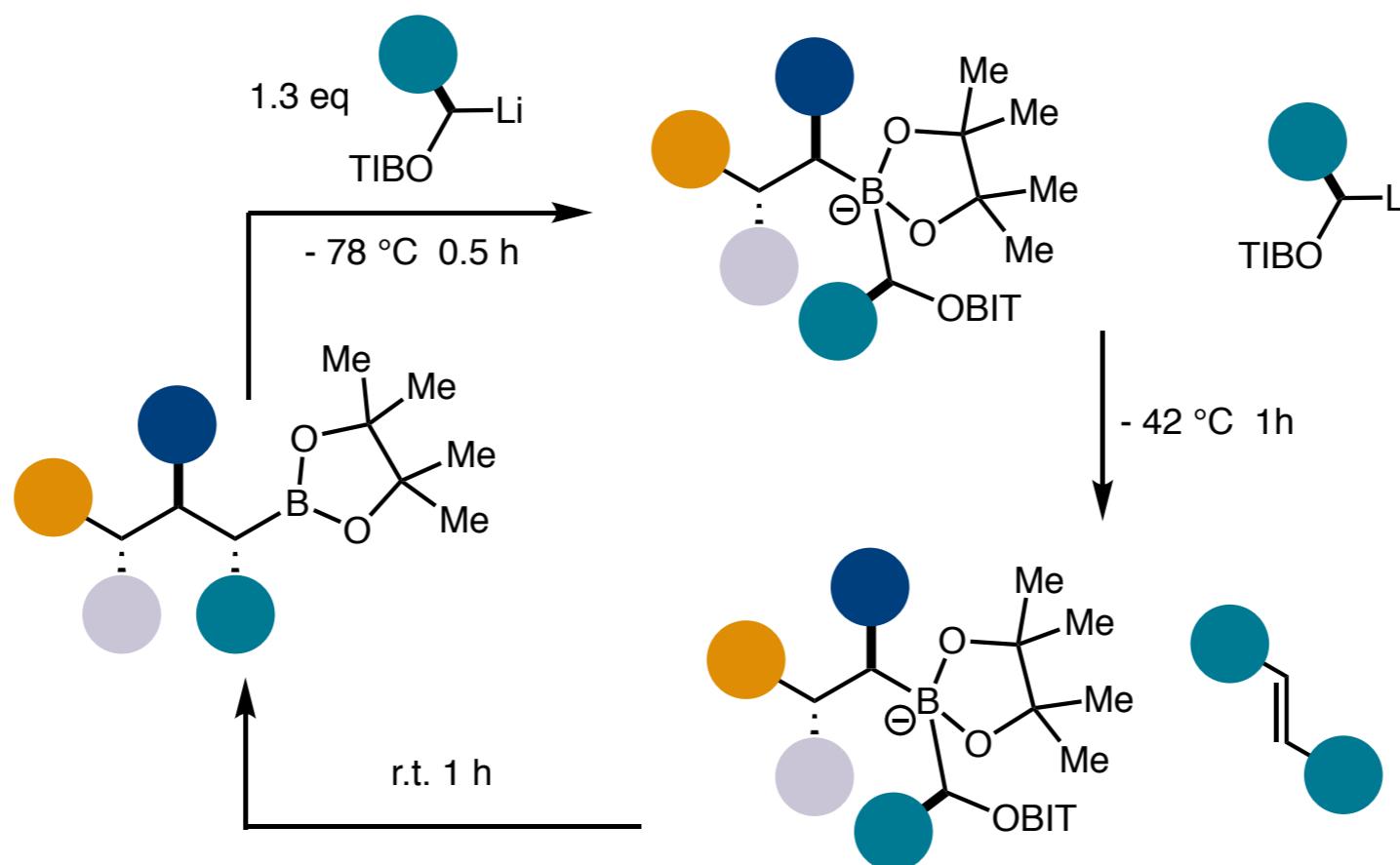


# Programmable Temperature Enabled Assemble Line Synthesis



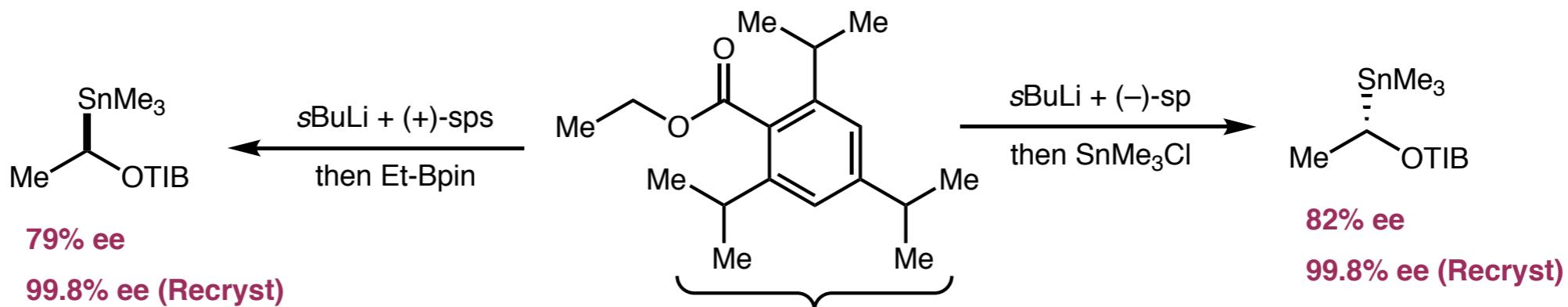
**Auto Synthesis with Reagent Controlled Chirality**

**Working Hypothesis:**

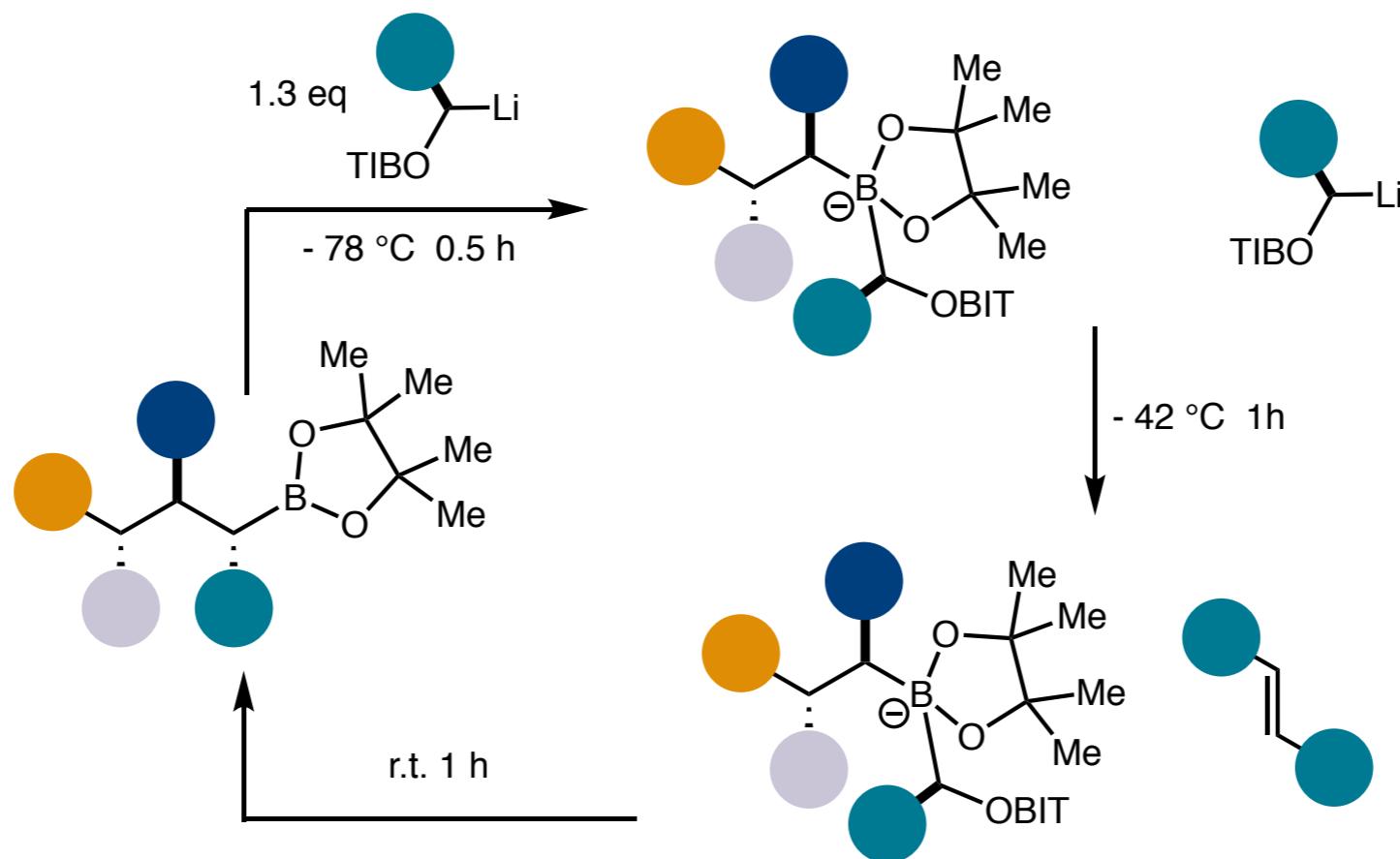


# Programmable Temperature Enabled Assemble Line Synthesis

## High Purity Reagents:

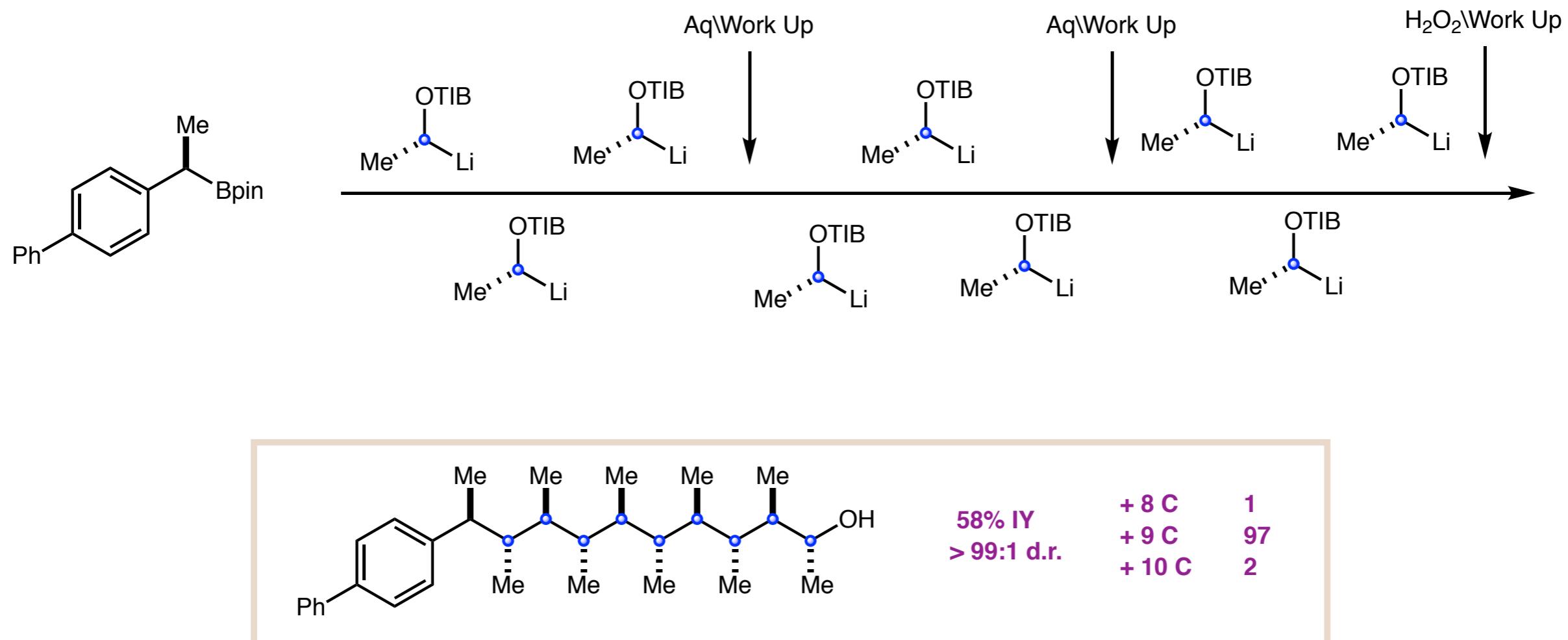


## Working Hypothesis:



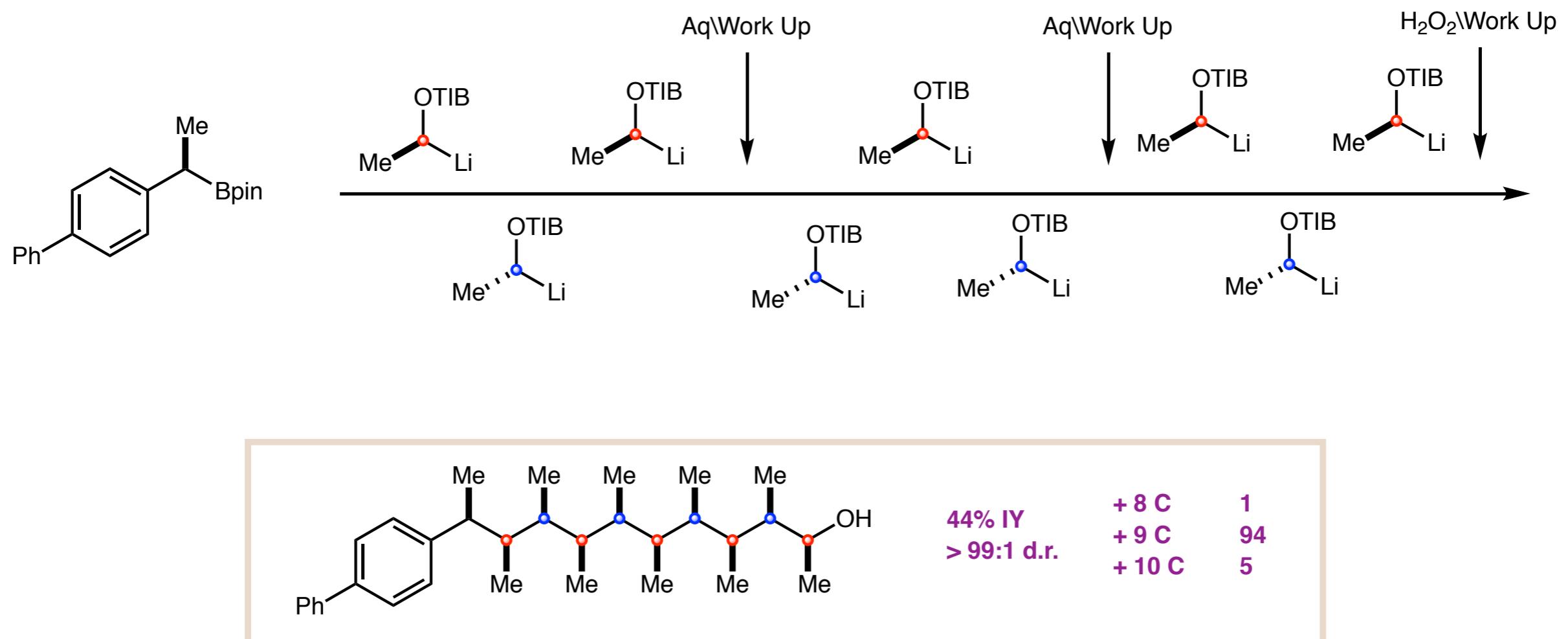
# Programmable Temperature Enabled Assemble Line Synthesis

## State of Art:



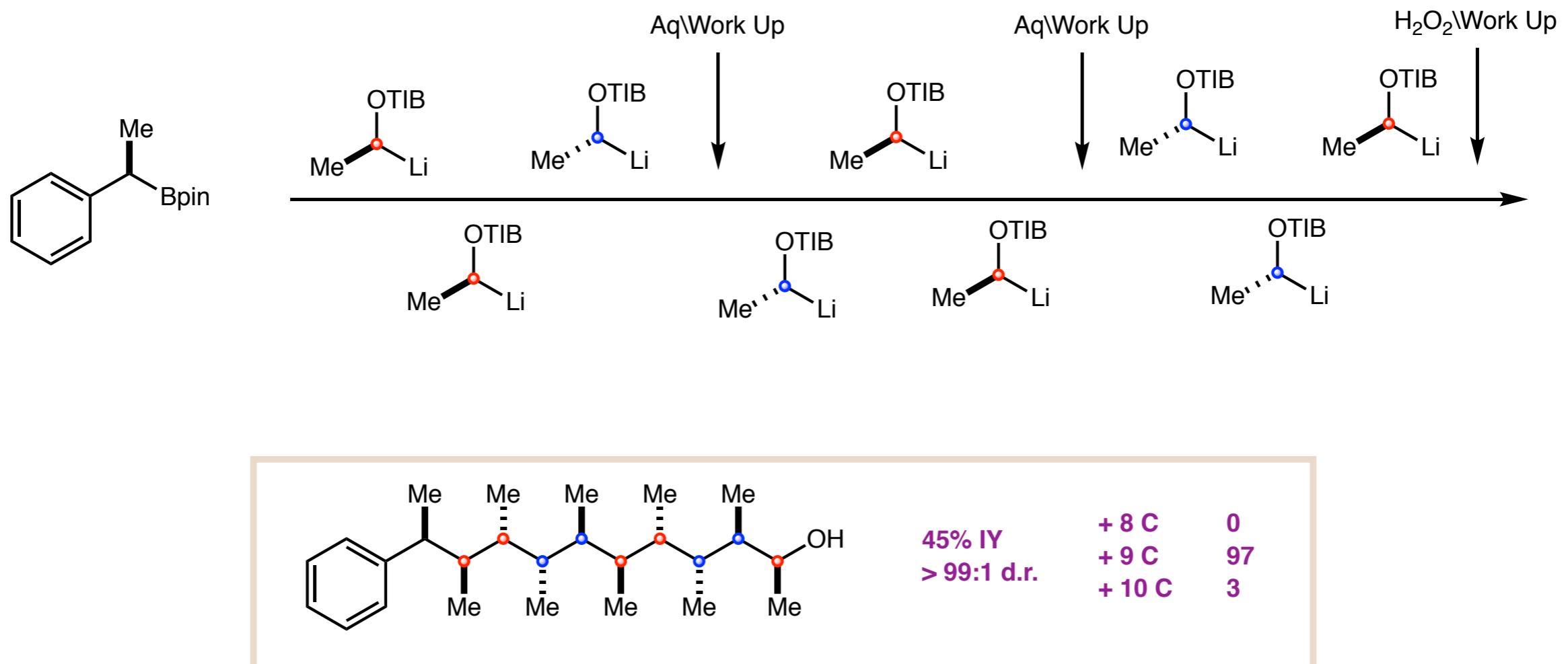
# *Programmable Temperature Enabled Assemble Line Synthesis*

## ***State of Art:***

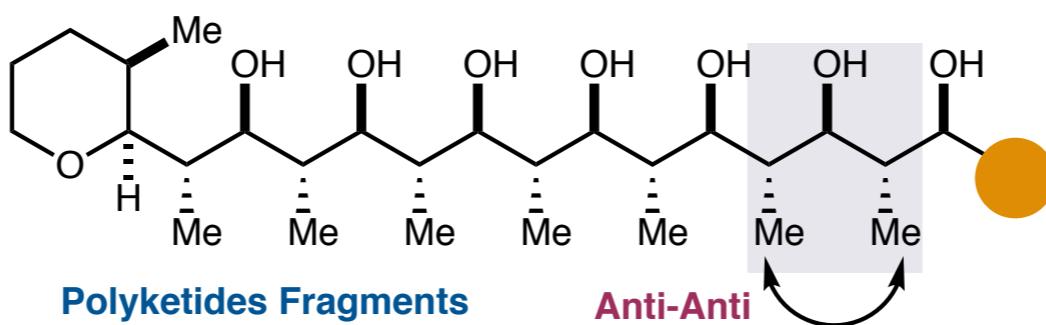
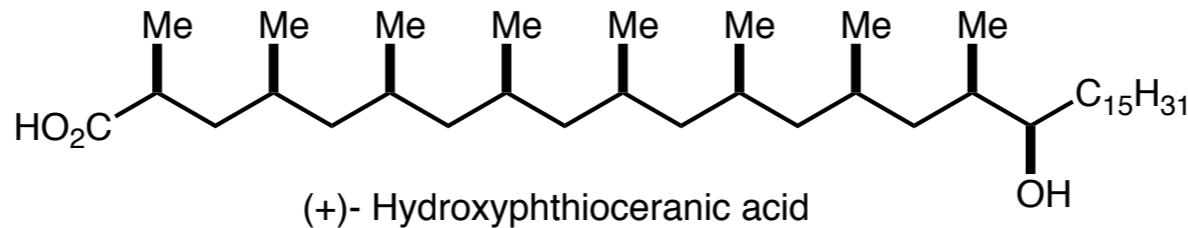


# Programmable Temperature Enabled Assemble Line Synthesis

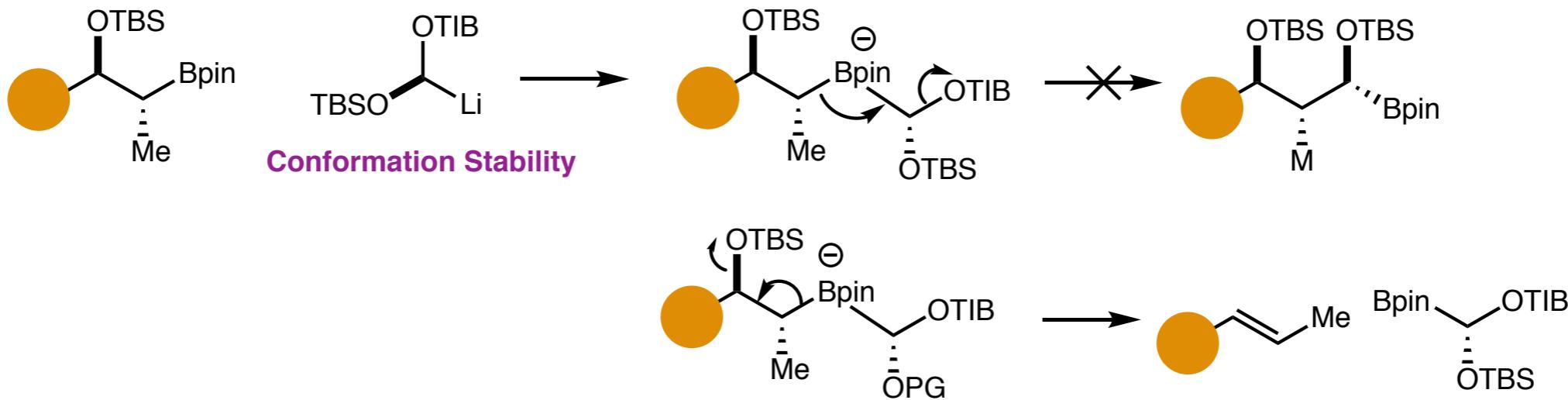
## State of Art:



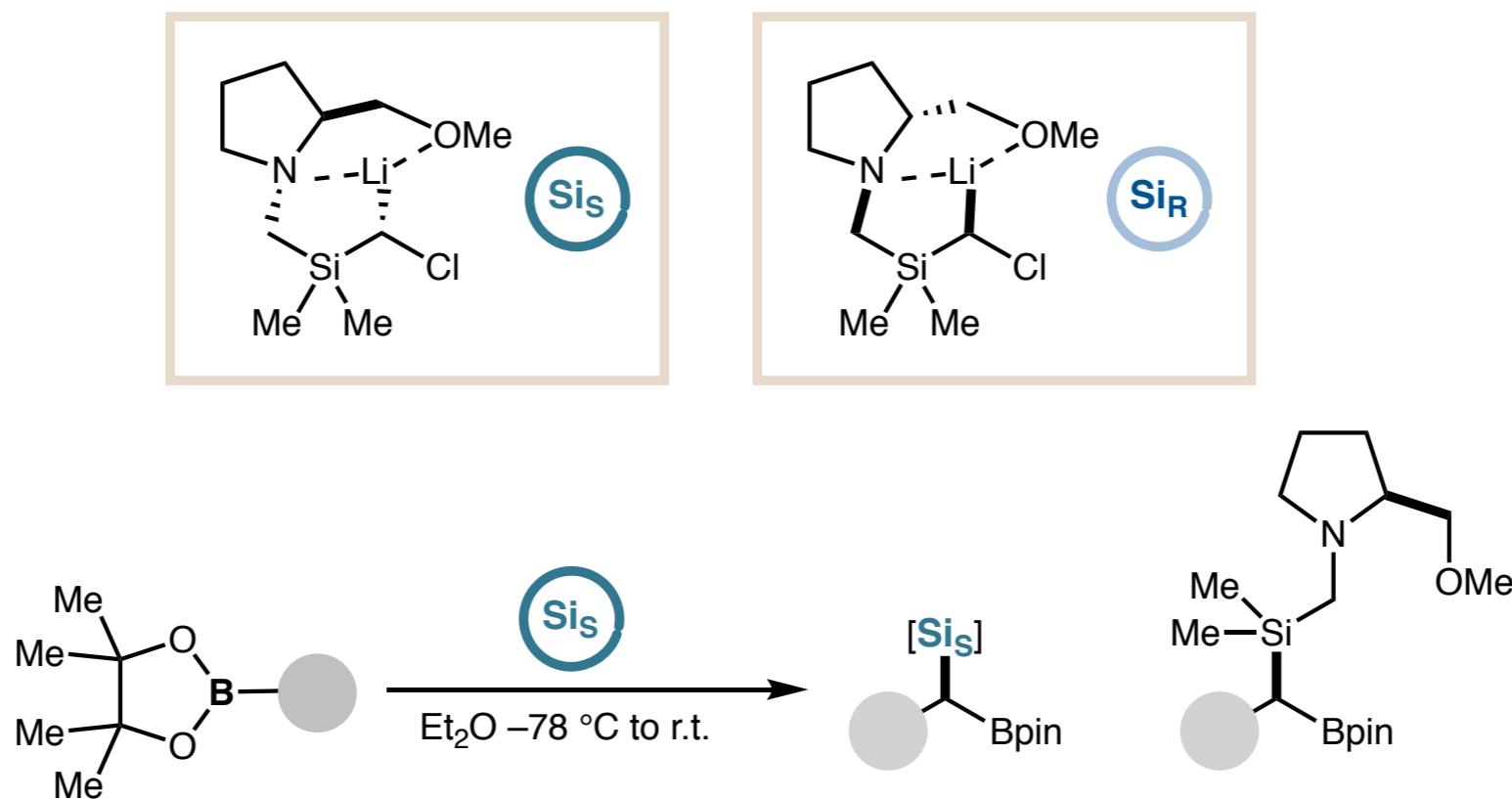
# *From Hydrocarbon to Polypropionates*



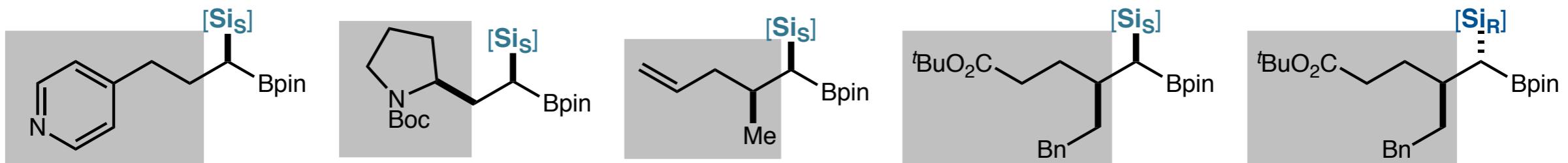
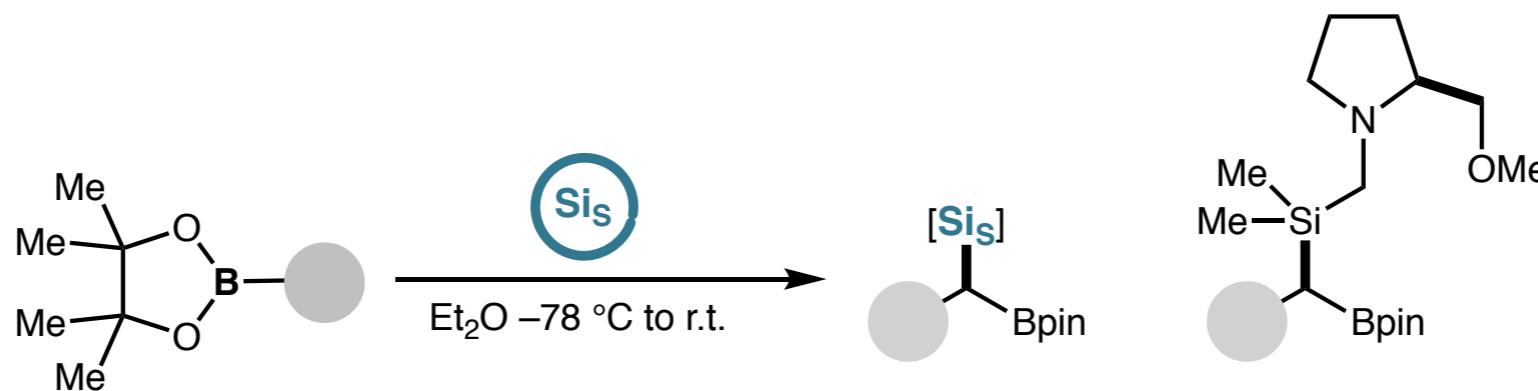
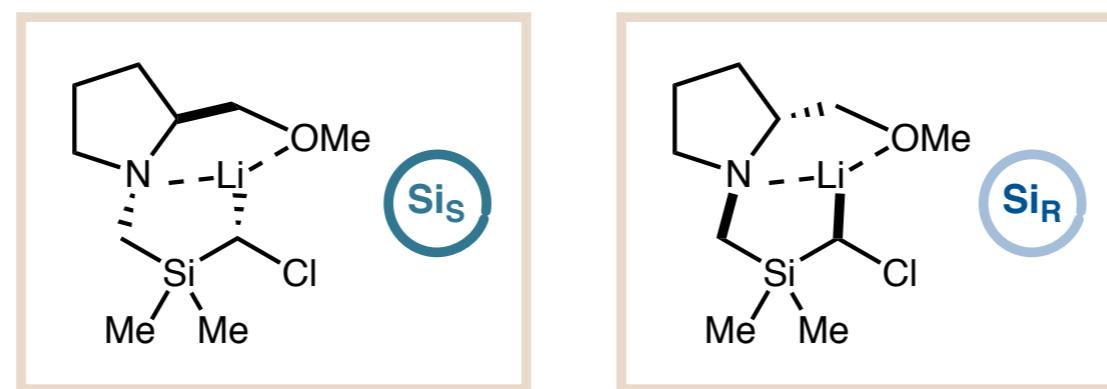
**Why Oxygen is difficult?**



## *Using Si Atom Instead of Oxygen*



## Using Si Atom Instead of Oxygen



79% IY  
92:8 d.r.

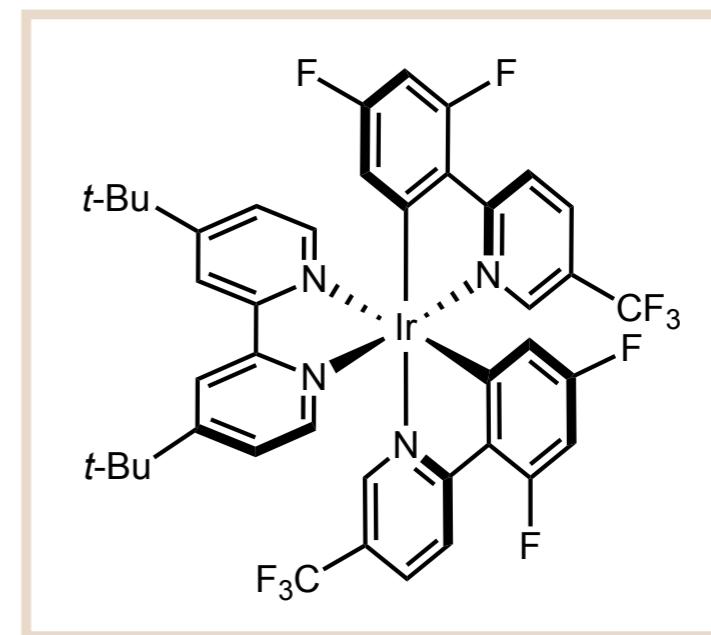
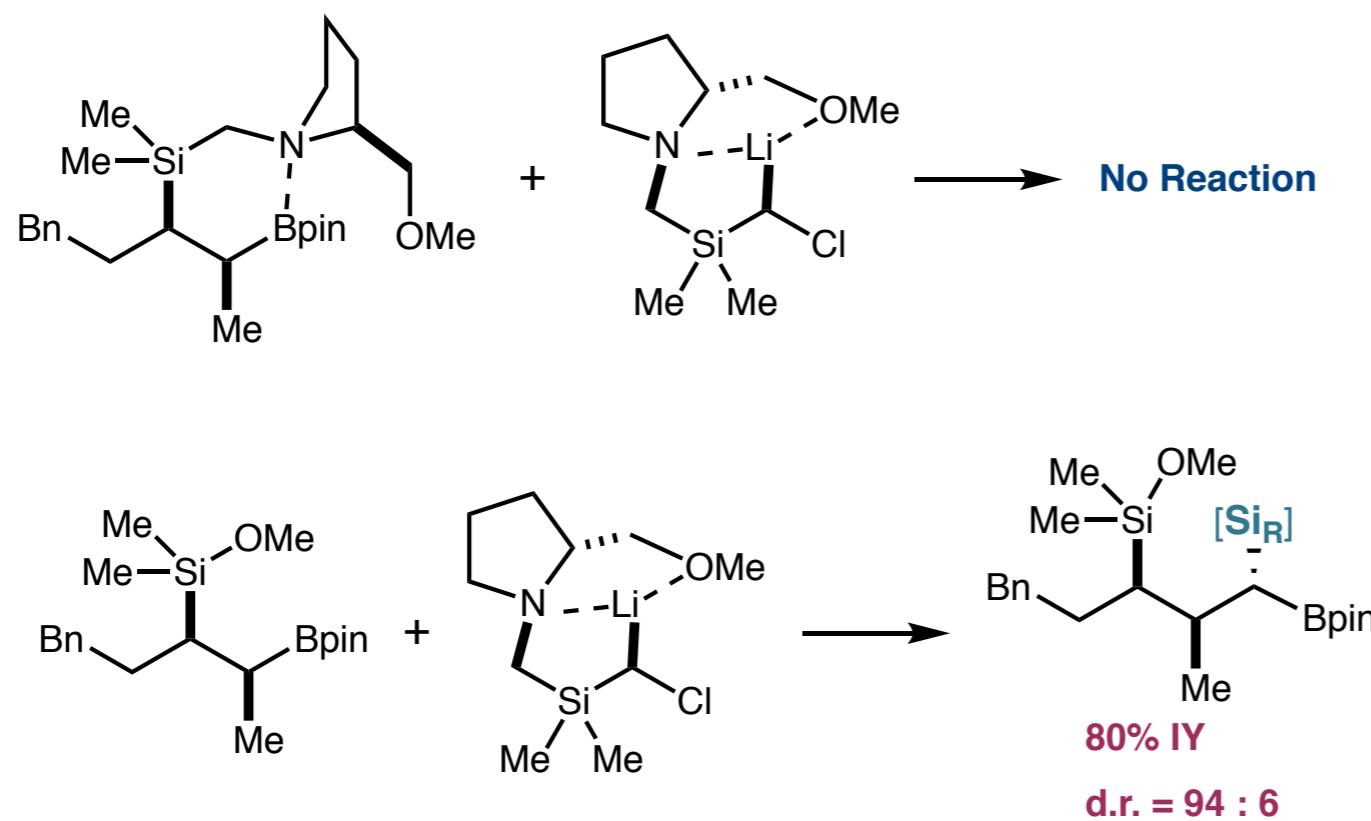
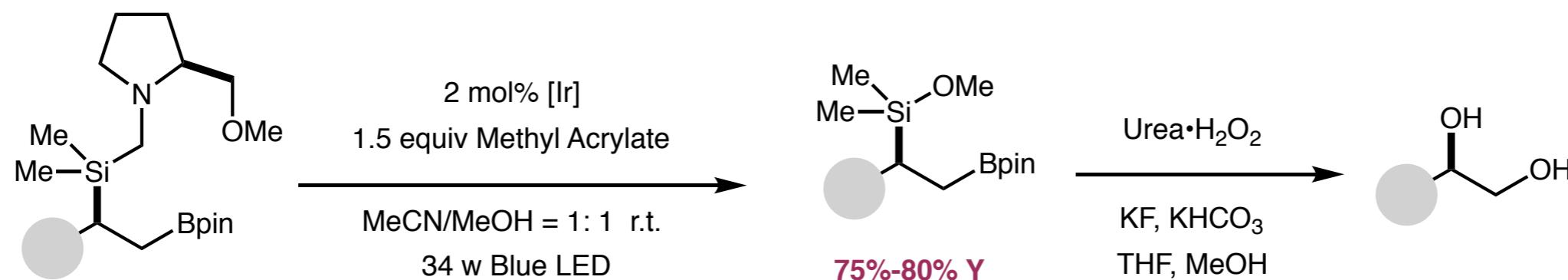
71% IY  
91:9 d.r.

71% IY  
 $\geq 98:2$  d.r.

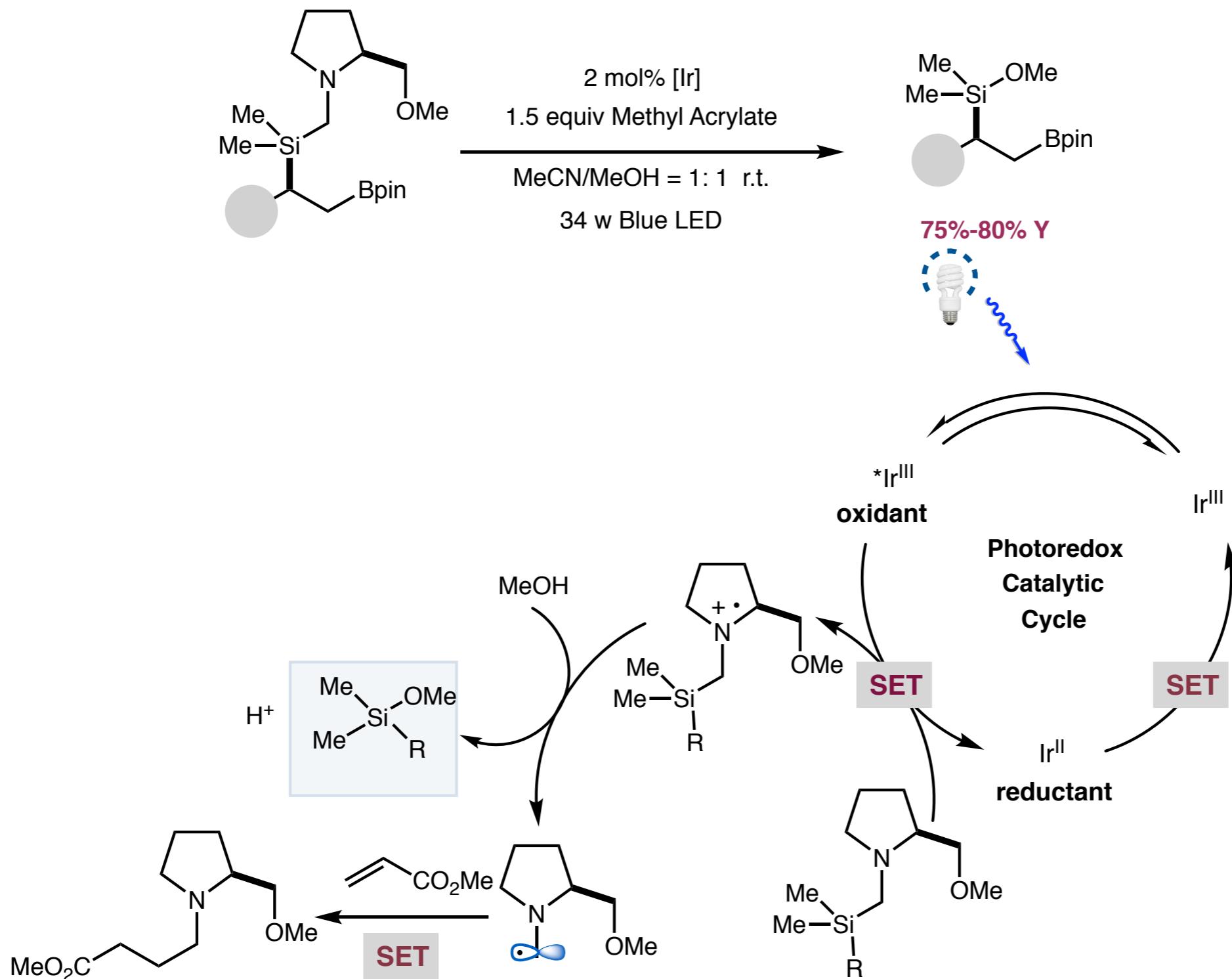
77% IY  
95:5 d.r.

79% IY  
95:5 d.r.

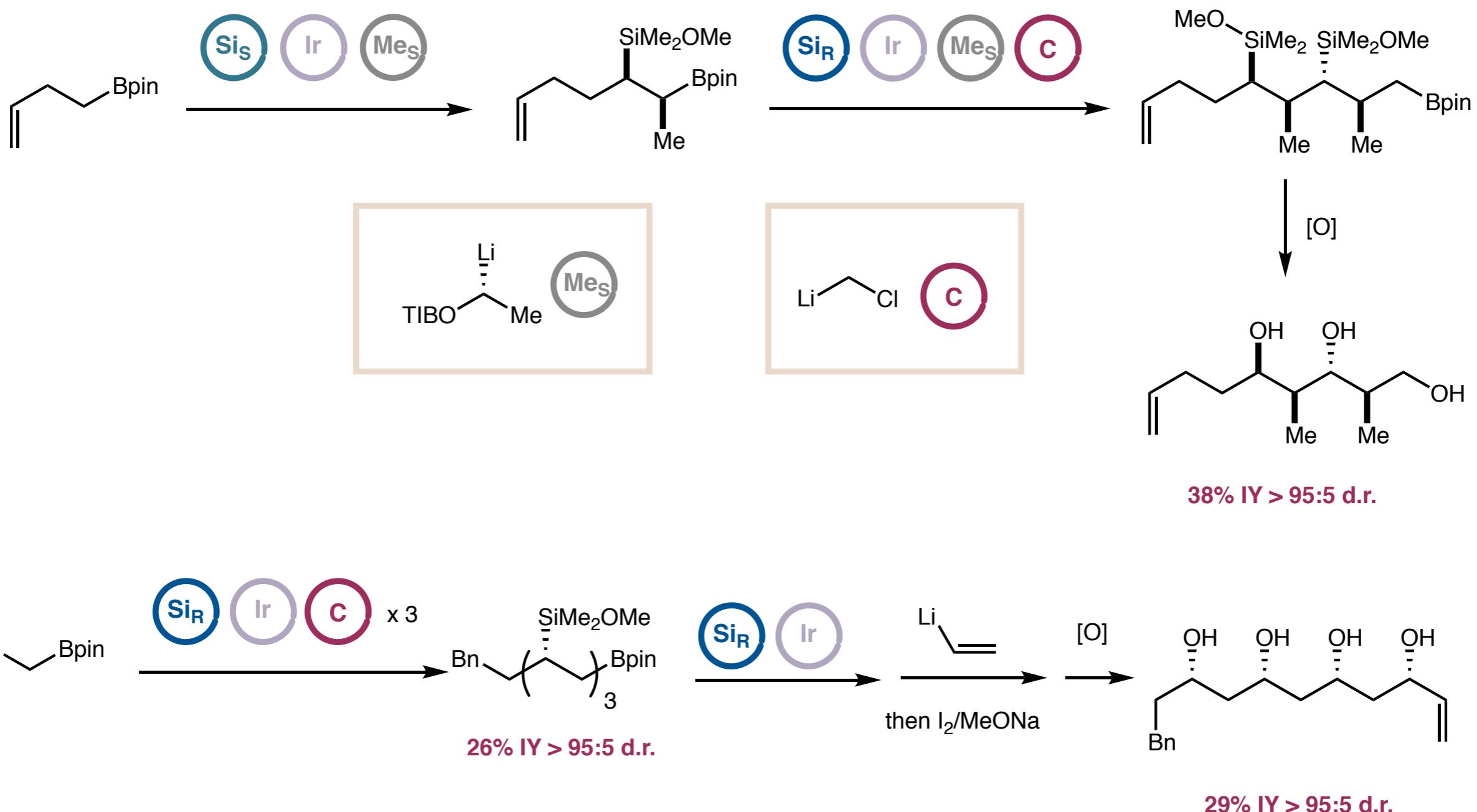
# *Transfer Si Atom to Oxygen*



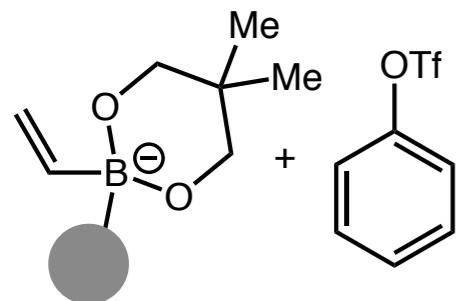
## Transfer Si Atom to Oxygen : Proposed Mechanism



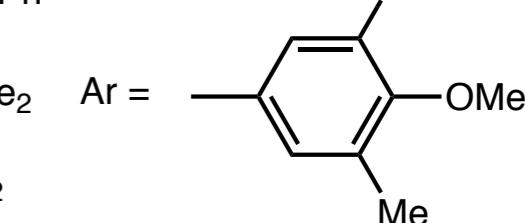
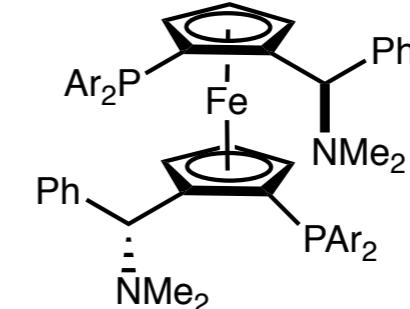
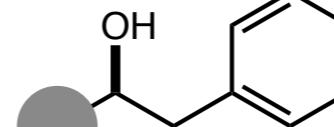
# Iterative Assembly Line Synthesis of Polypropionate



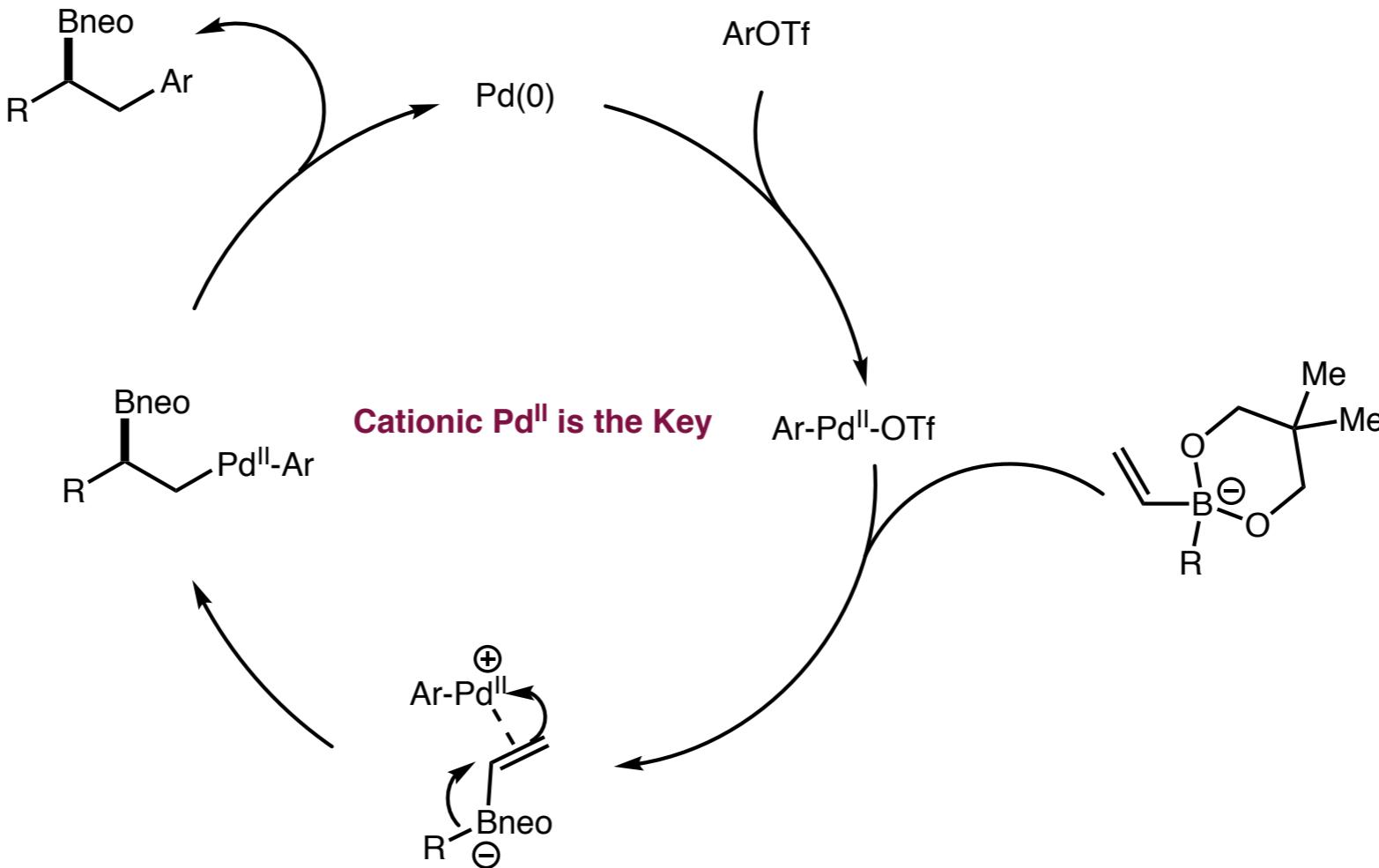
## Merging Pd Catalysis with Boron-1,2-Shift



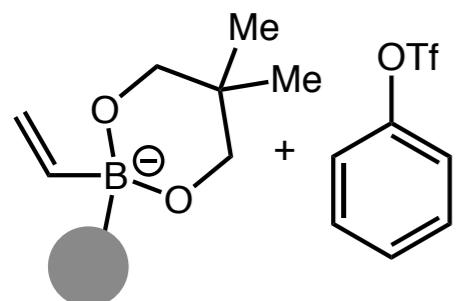
1 mol%  $\text{Pd}(\text{OAc})_2$   
1.5 mol%  $\text{L}^*$   
 $\text{H}_2\text{O}_2/\text{NaOH}$   
THF 60 °C 14 h



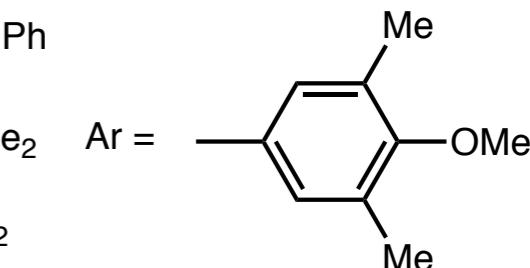
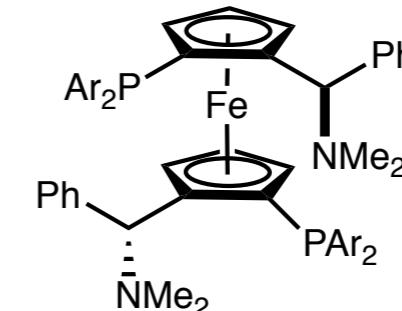
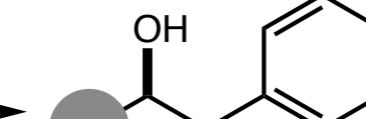
Chiral Ligand



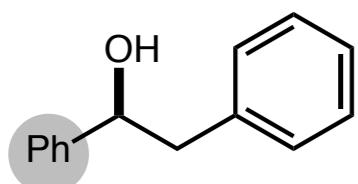
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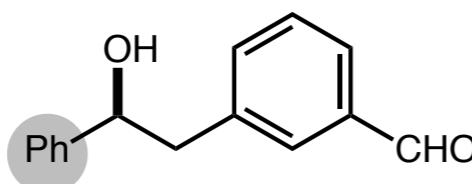


Chiral Ligand



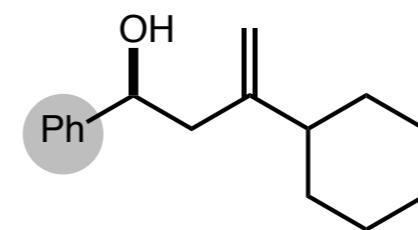
83% IY

94% ee



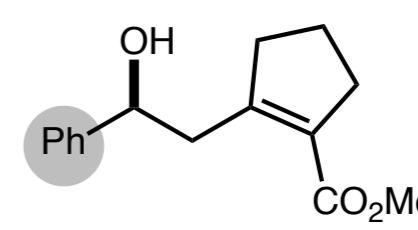
68% IY

92% ee



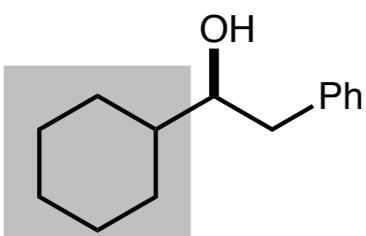
73% IY

90% ee



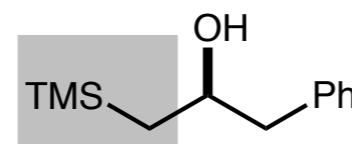
41% IY

80% ee



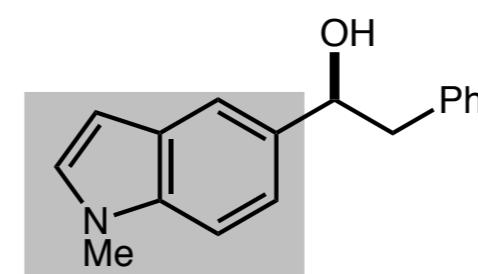
80% IY

90% ee



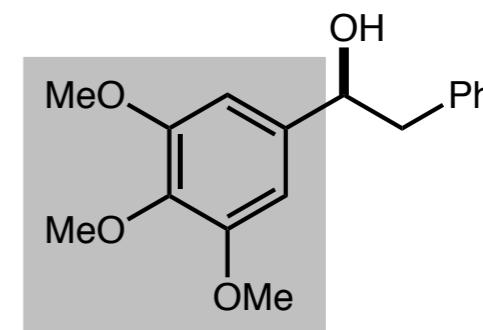
48% IY

76% ee



86% IY

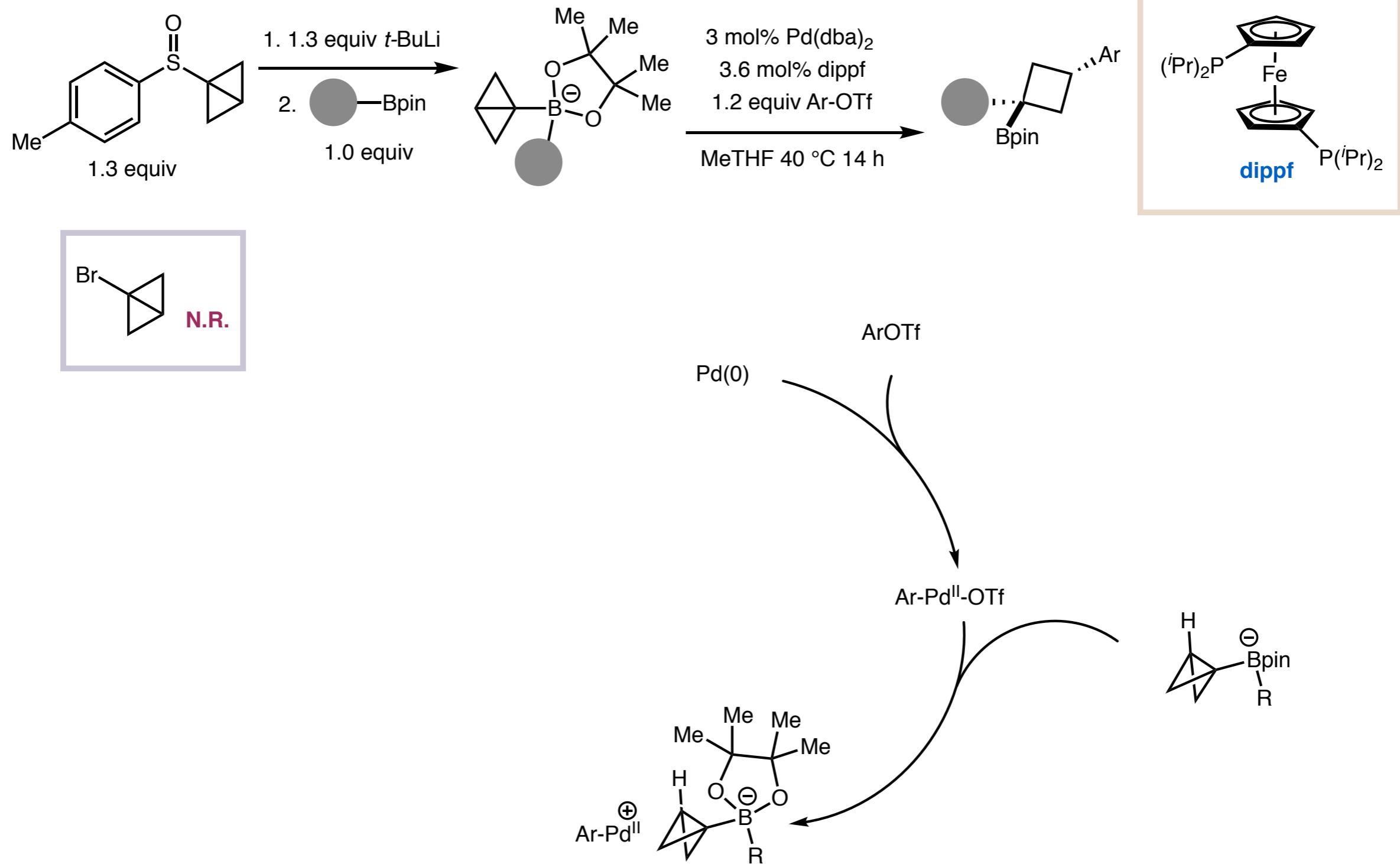
90% ee



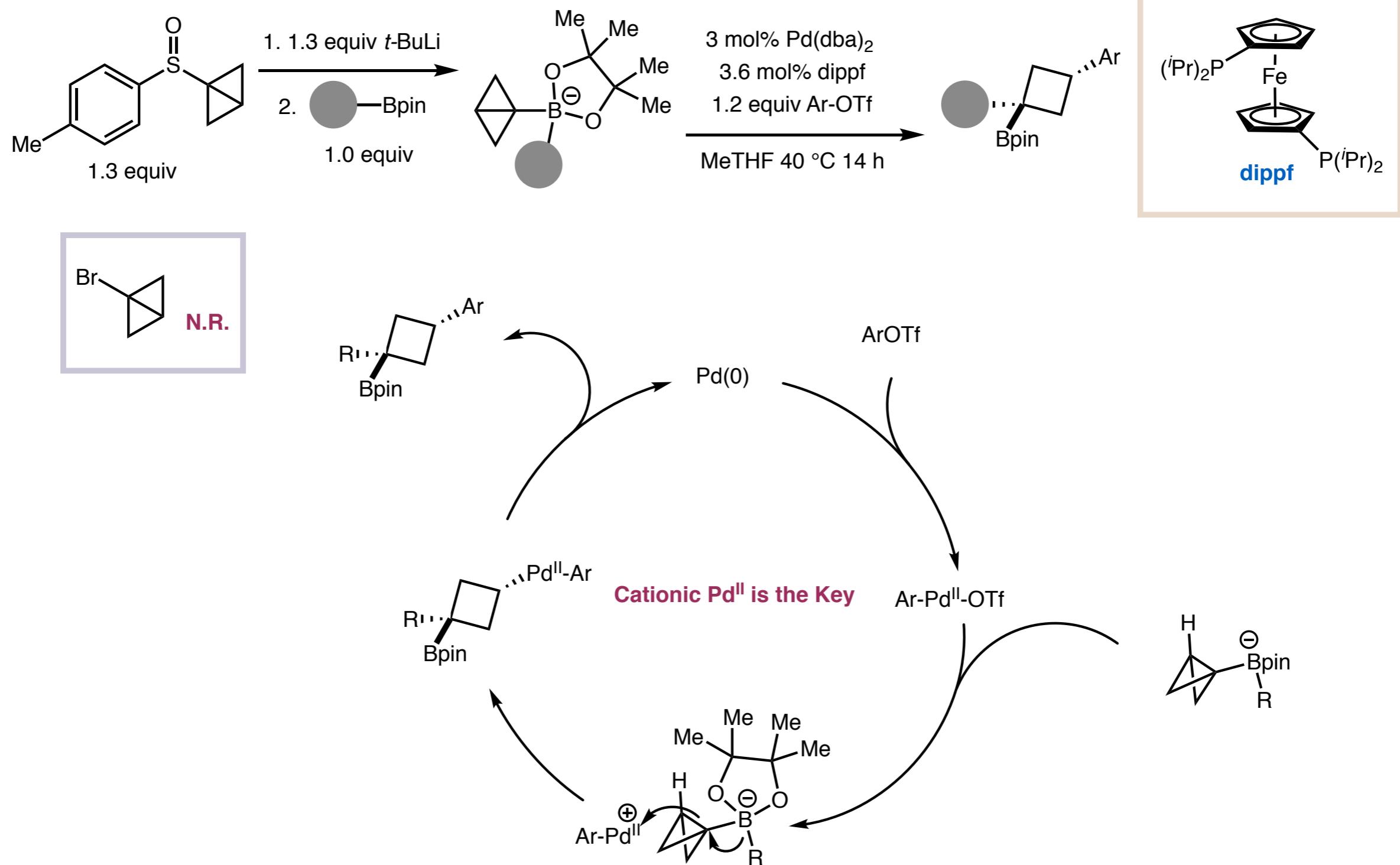
77% IY

88% ee

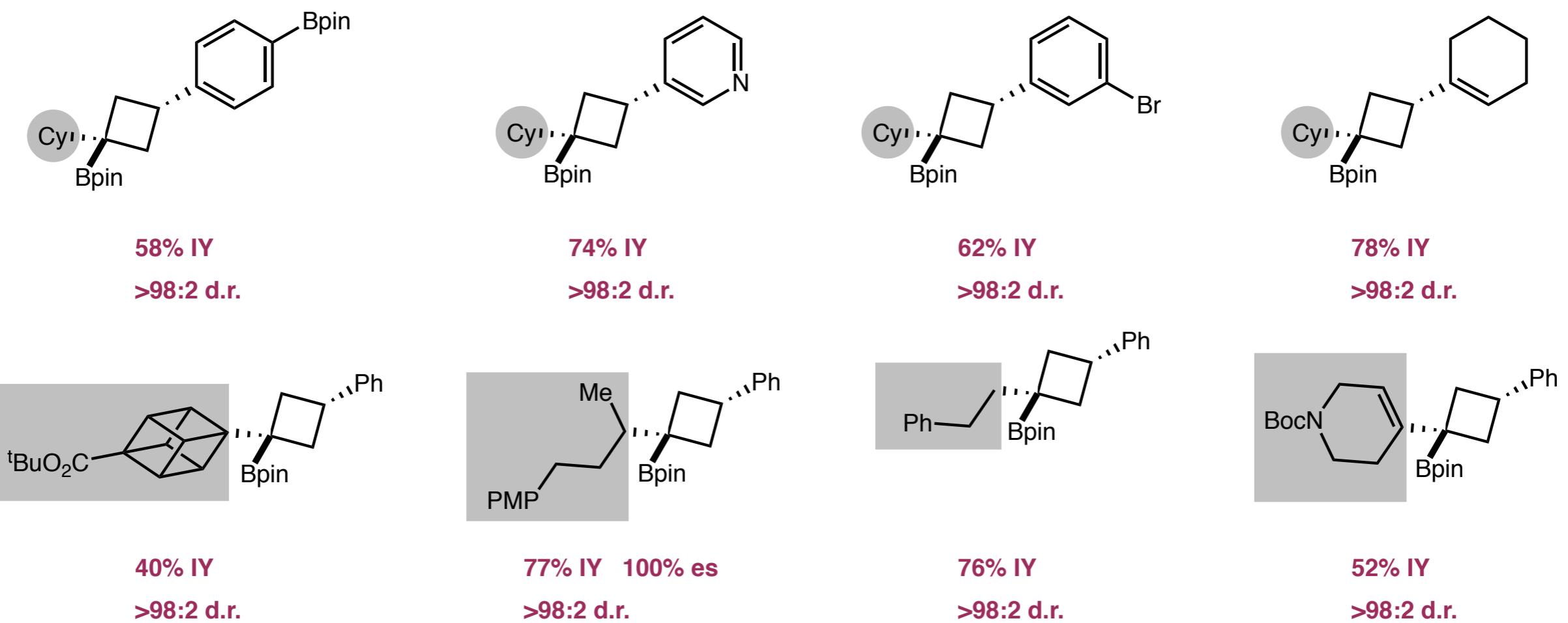
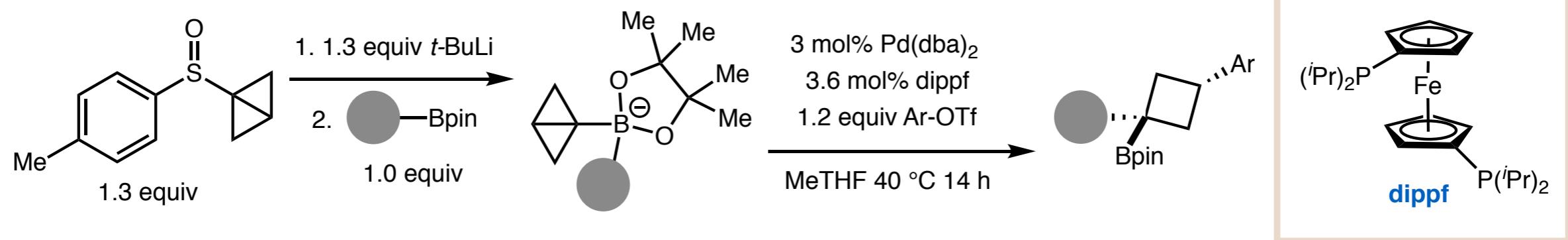
## Merging Pd Catalysis with Boron-1,2-Shift



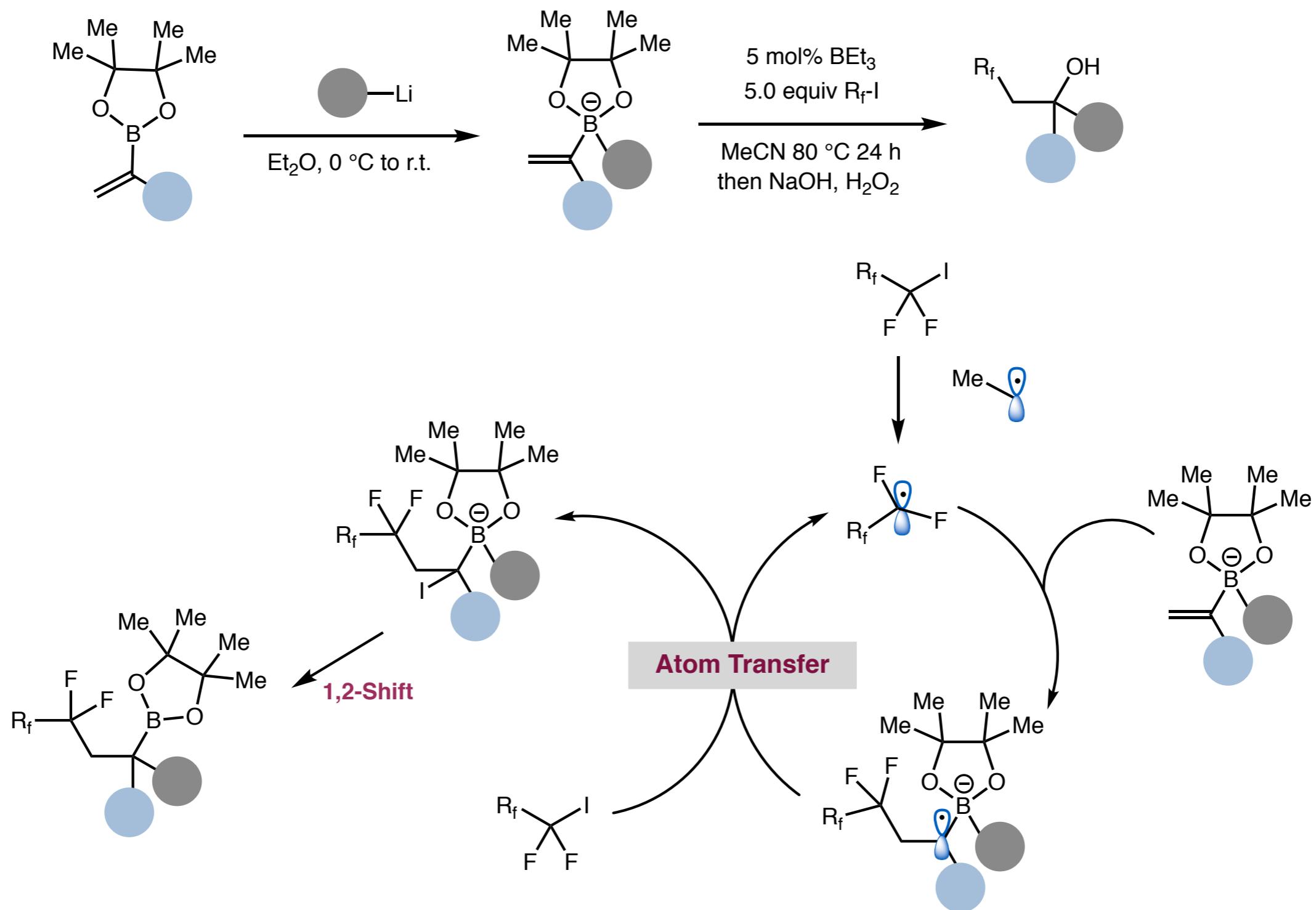
# Merging Pd Catalysis with Boron-1,2-Shift



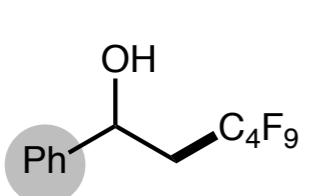
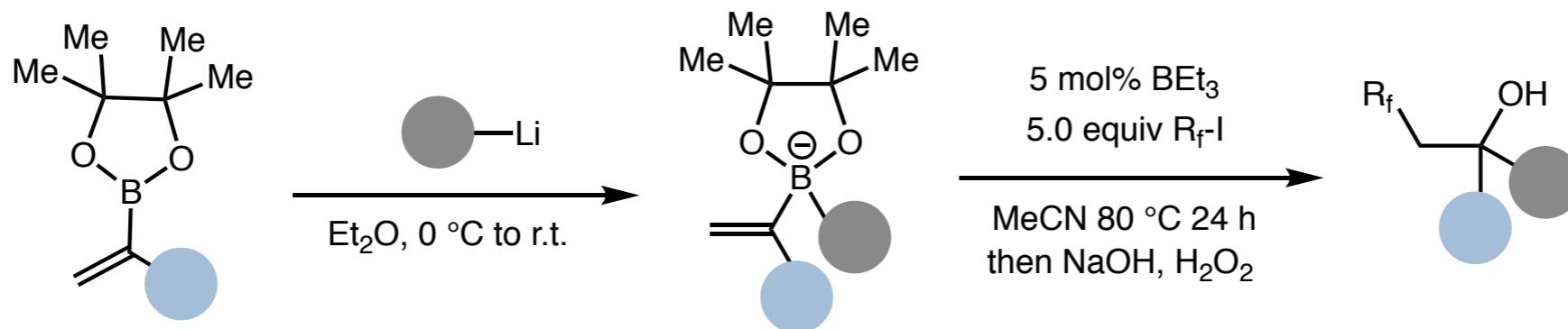
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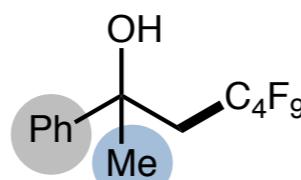
## Merging Radical Reaction with Boron-1,2-Shift



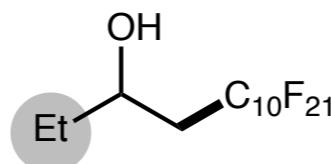
## Merging Radical Reaction with Boron-1,2-Shift



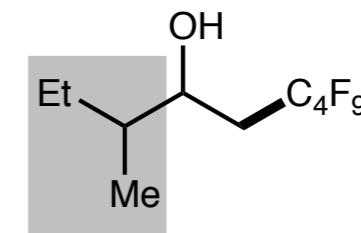
68% IY



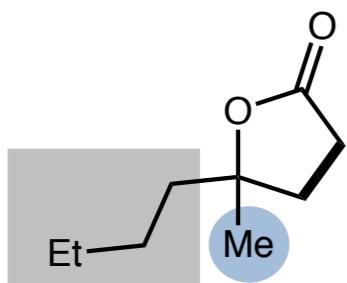
65% IY



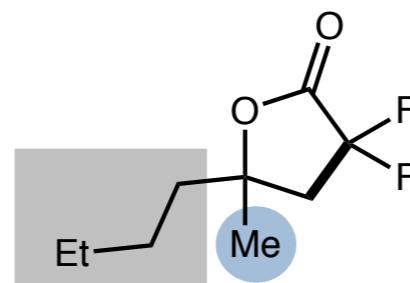
65% IY



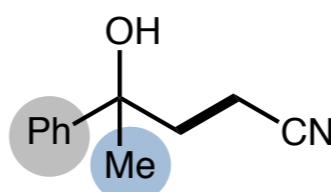
36% IY  
1:1 d.r.



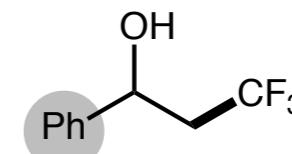
77% IY



58% IY

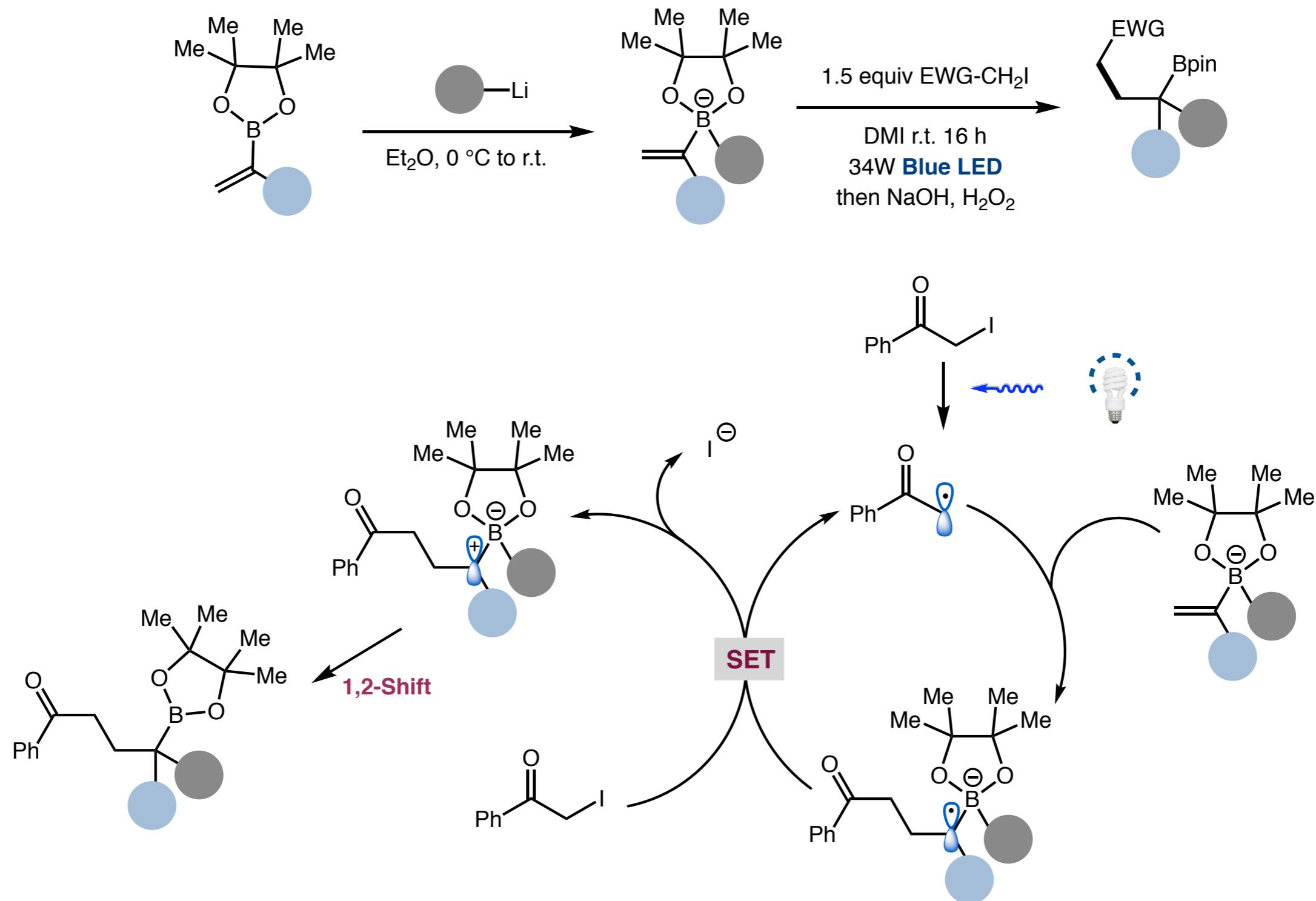


61% IY



41% IY  
With Togni I

## Merging Radical Reaction with Boron-1,2-Shift



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