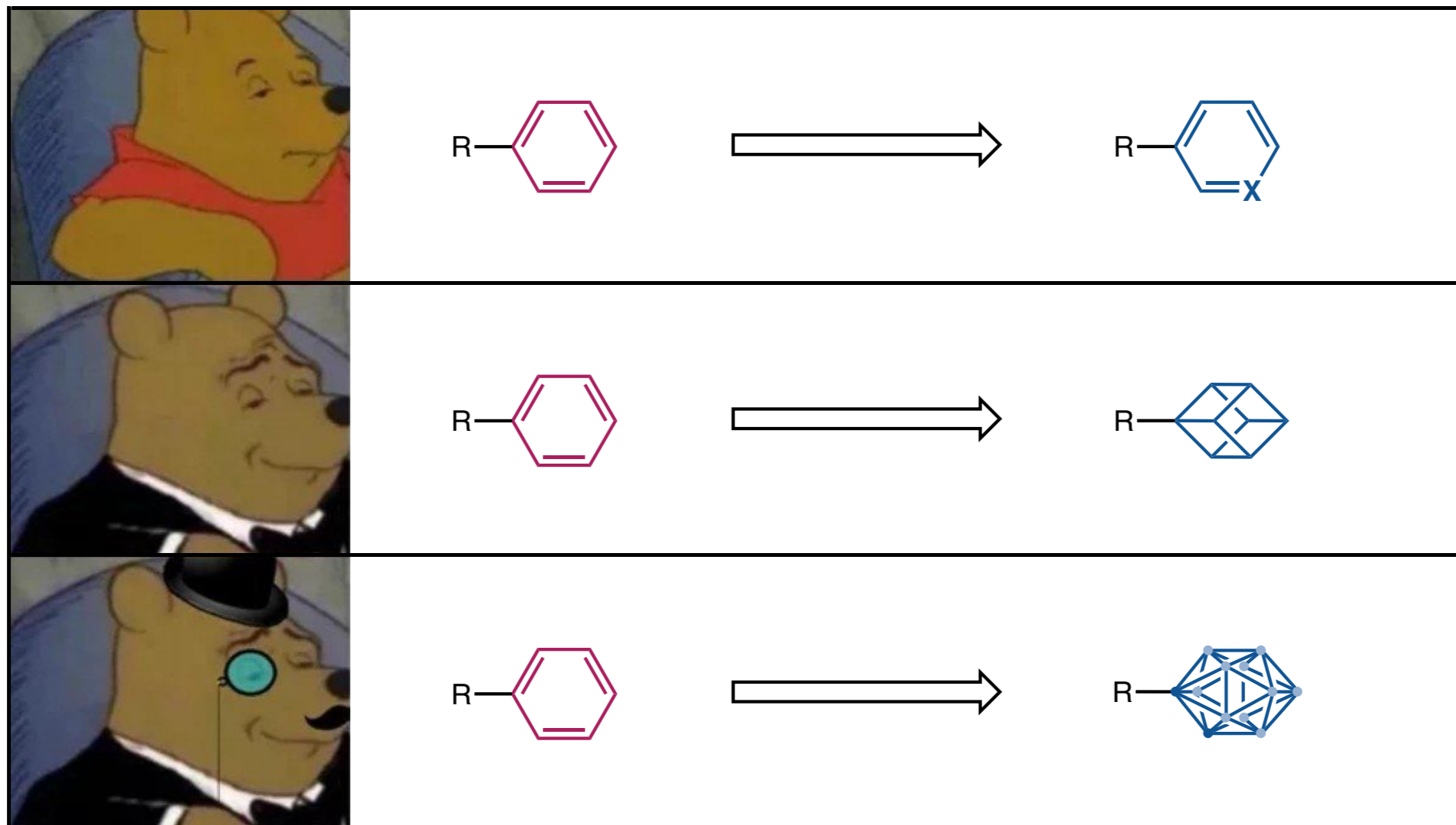
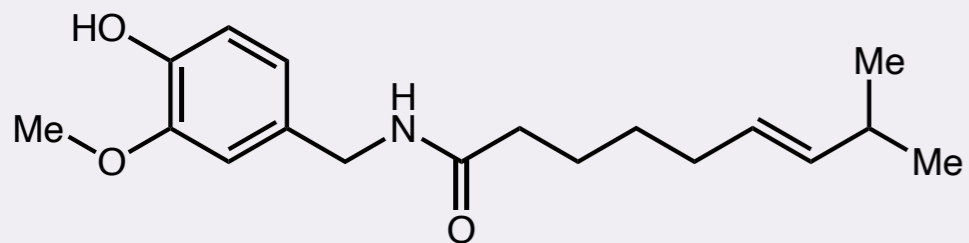


Phenyl Bioisosterism – A Solved Challenge?

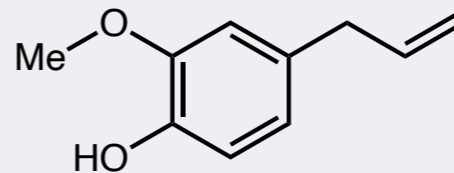


Michael Heilmann
MacMillan Group Meeting
October 20, 2021

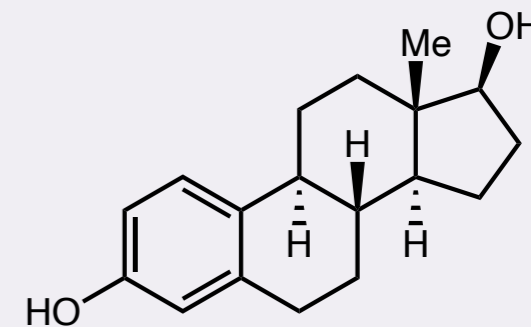
Phenyl Rings as Ubiquitous Building Blocks of Life



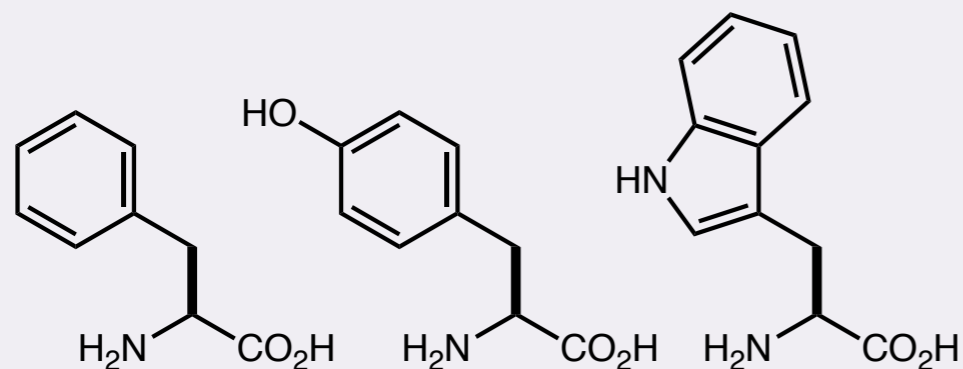
capsaicin
chili pepper alkaloid



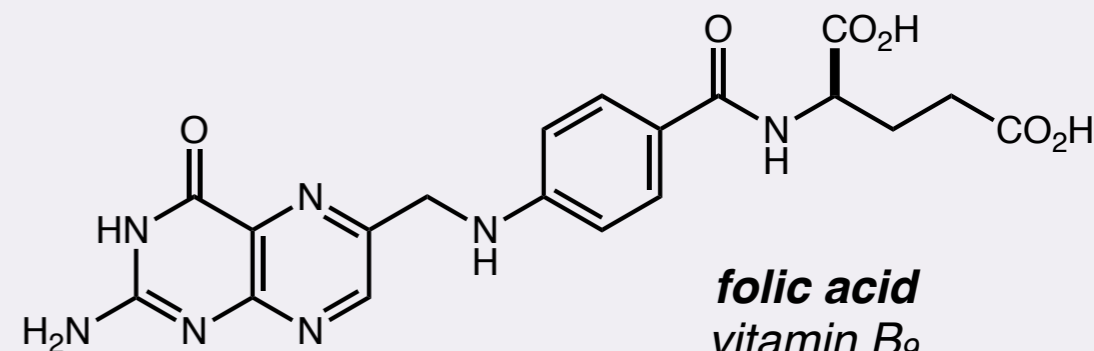
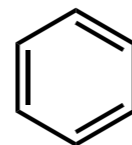
eugenol
natural fragrance



estradiol
hormone



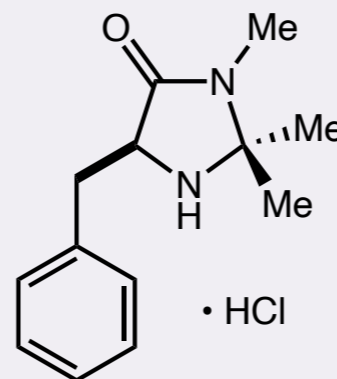
canonical amino acids
building block, catalyst, signal transducer, ...



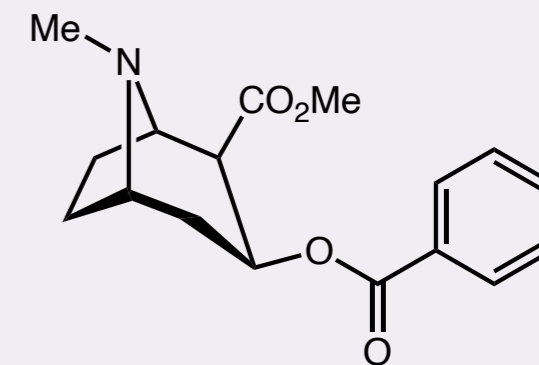
folic acid
vitamin B₉



penicillin G
antibiotic



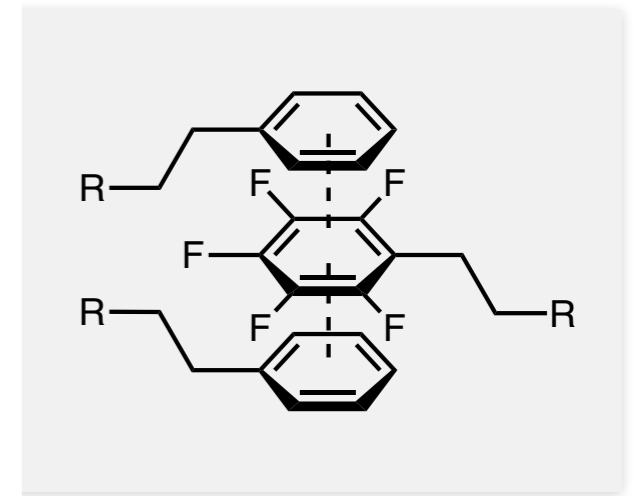
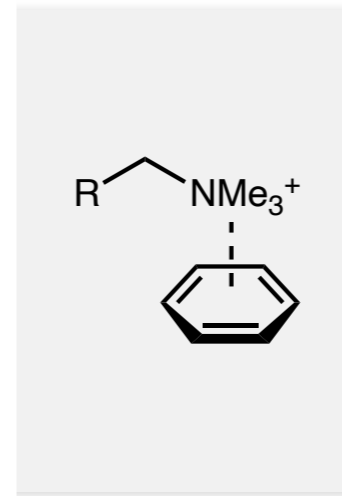
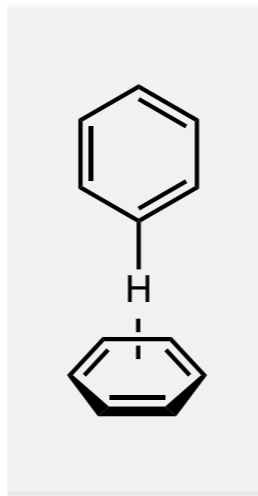
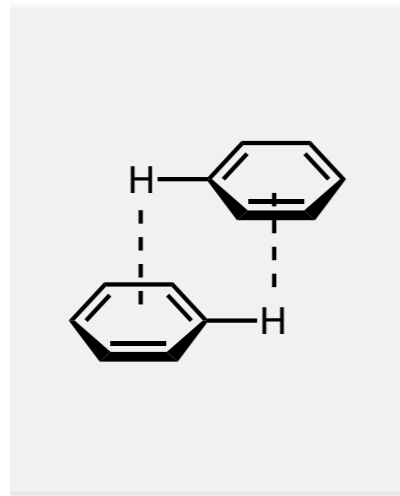
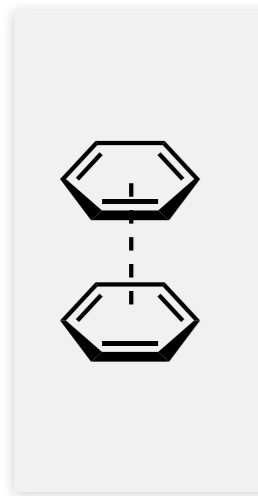
"7"
Nobel Prize winner



cocaine
stimulant

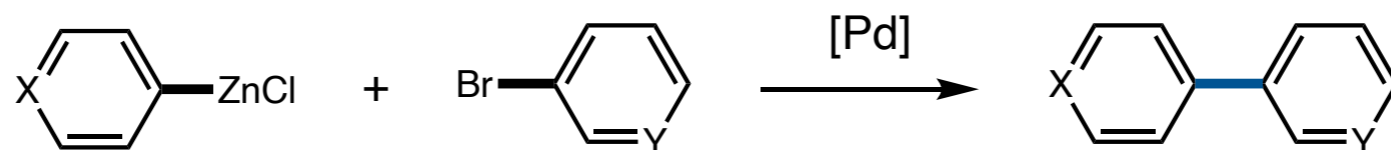
Phenyl Rings as Ubiquitous Building Blocks of Life

- Phenyl rings engage in numerous different modes of intermolecular interaction



Phenyl Rings as Ubiquitous Building Blocks of Life

- We are exceptionally good at forming phenyl–R connections



Richard F. Heck

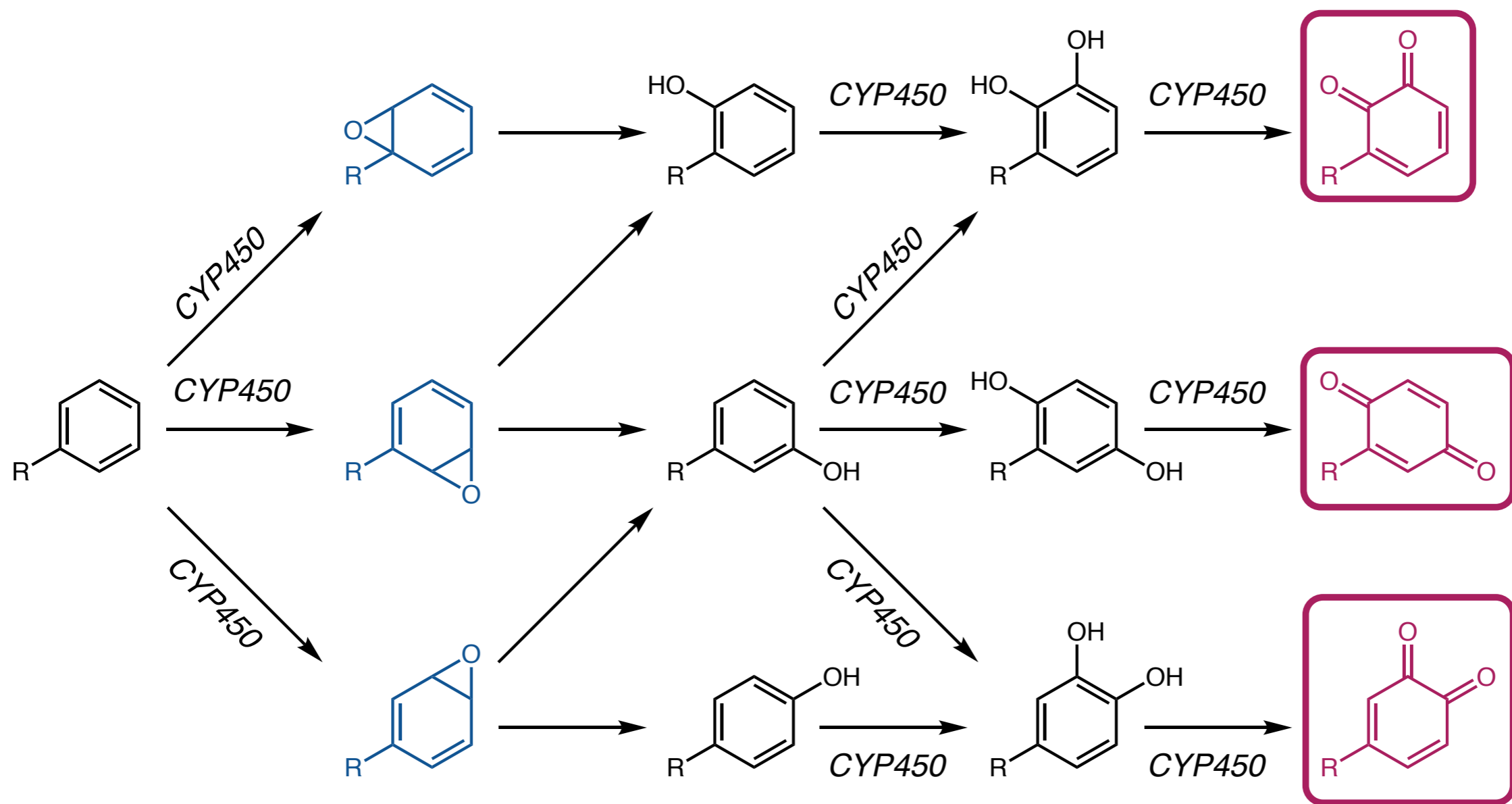


Ei-ichi Negishi



Akira Suzuki

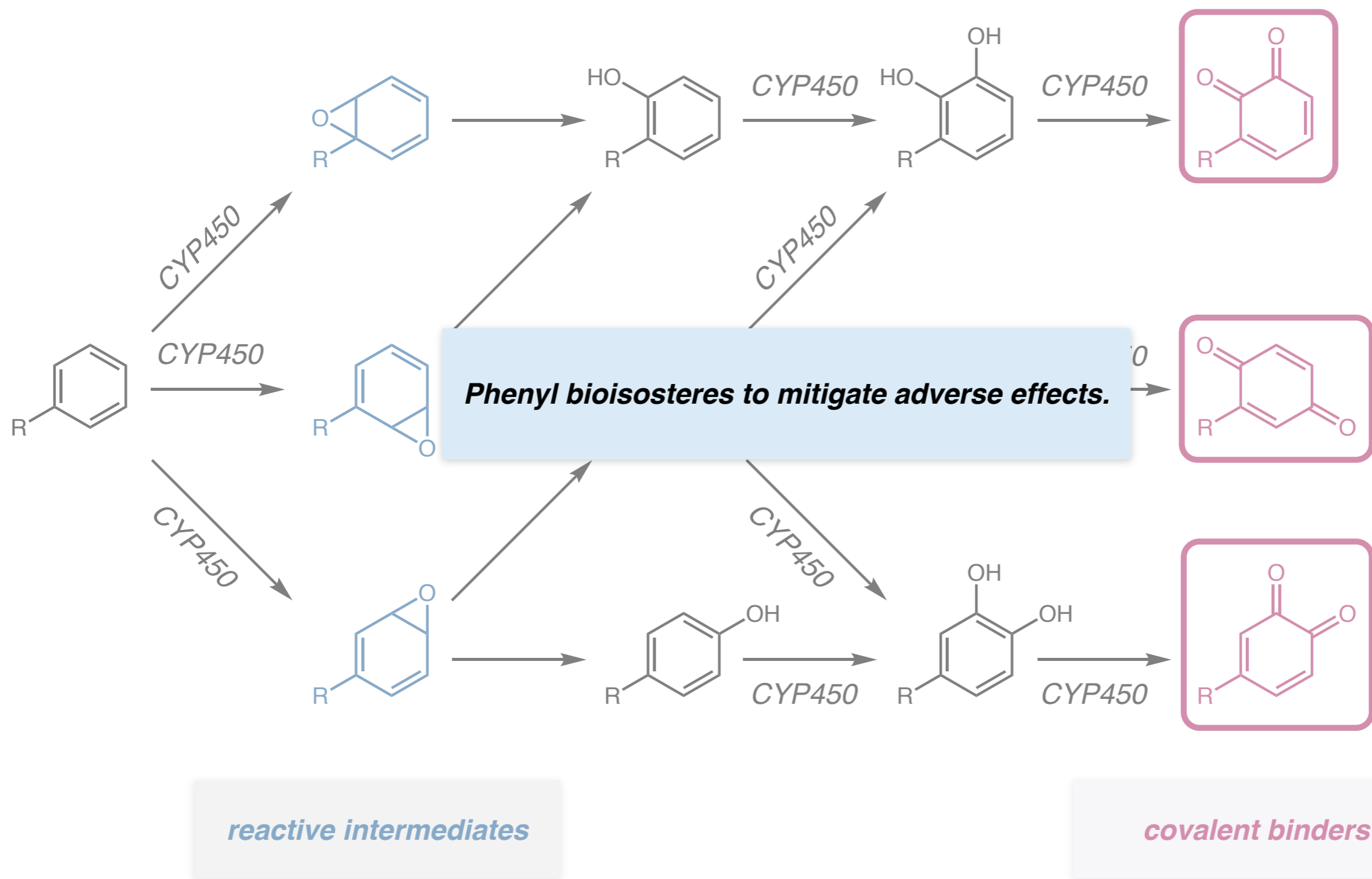
Phenyl Metabolites and Hepatotoxicity



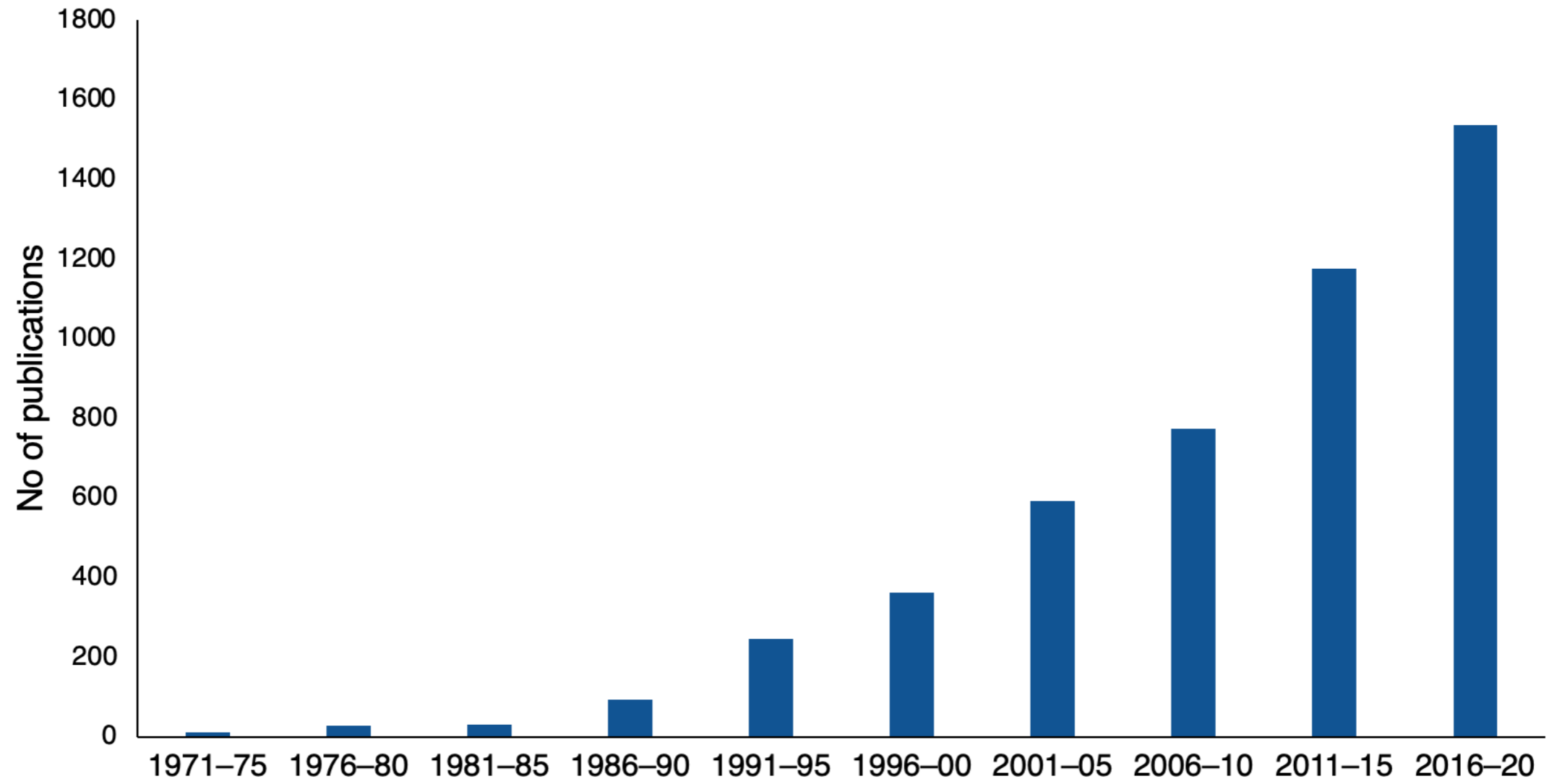
reactive intermediates

covalent binders

Phenyl Metabolites and Hepatotoxicity



Bioisosteres in the Literature



What Does 'Bioisostere' Mean, Really?

Langmuir 1919:
isosteres are "compounds or groups of atoms having the same number of atoms and electrons"

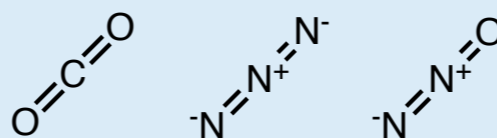
groups	isosteres
1	H ⁻ , He, Li ⁺
2	O ²⁻ , F ⁻ , Ne, Na ⁺ , Mg ²⁺ , Al ³⁺
3	S ²⁻ , Cl ⁻ , Ar, K ⁺ , Ca ²⁺
4	Cu ²⁻ , Zn ²⁺
↓	↓
8	N ₂ , CO, CN ⁻
9	CH ₄ , NH ₄ ⁺
10	CO ₂ , N ₂ O, N ₃ ⁻ , CNO ⁻
↓	↓
20	MnO ₄ ⁻ , CrO ₄ ²⁻
21	SeO ₄ ²⁻ , AsO ₄ ³⁻

1900

1950

2000

2020



Langmuir, I., *JACS* **1919**, 41, 1543.

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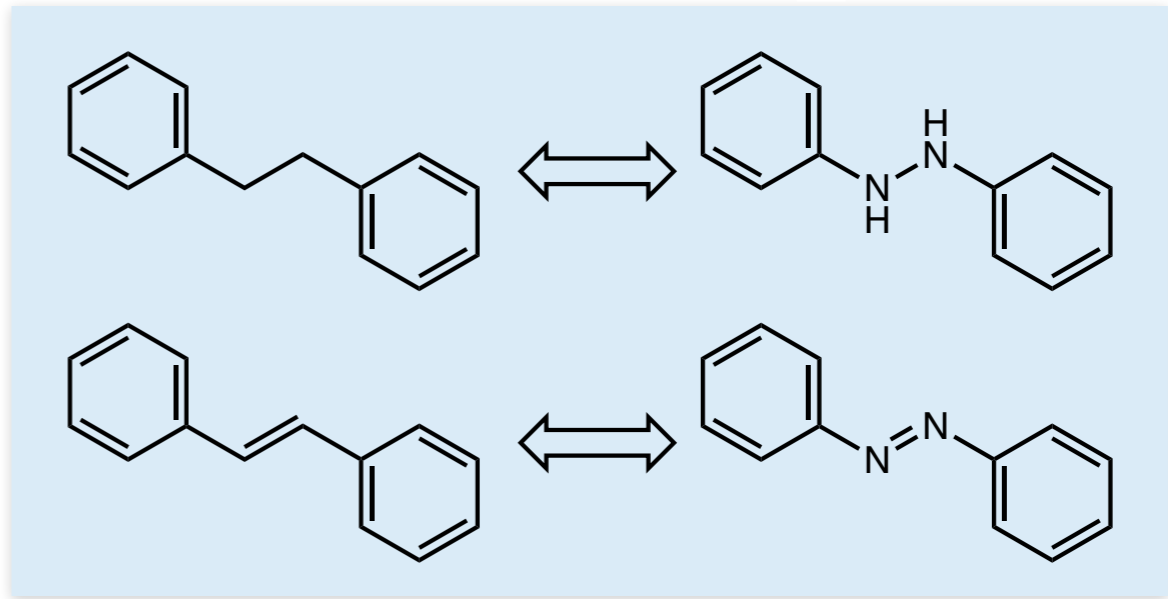
H-Atome	Gruppe					
	IV	V	VI	VII	0	I
0	C	N	O	F	Ne	Na
1		CH	NH	OH	FH	
2			CH ₂	NH ₂	OH ₂	FH ₂ ?
3				CH ₃	NH ₃	OH ₃
4					CH ₄	NH ₄
Valenz	-4	-3	-2	-1	0	+1

↓ Radius

← Radius



Grimm 1925:
 Hydride replacement law for isosterism

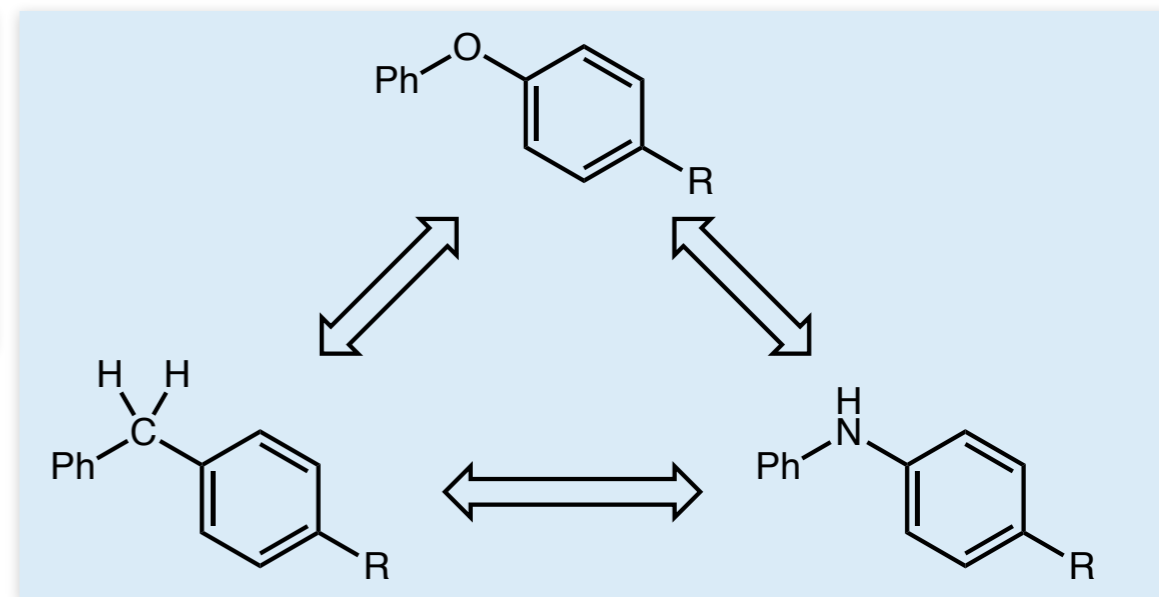


Langmuir, I., *JACS* **1919**, 41, 1543.
 Grimm, H. G., *Z. Electrochem.* **1925**, 31, 474.

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isosteres are "compounds or groups of atoms having the same number of atoms and electrons"

Erlenmeyer 1932:
 "atoms, ions or molecules in which the peripheral layers of electrons can be considered identical"



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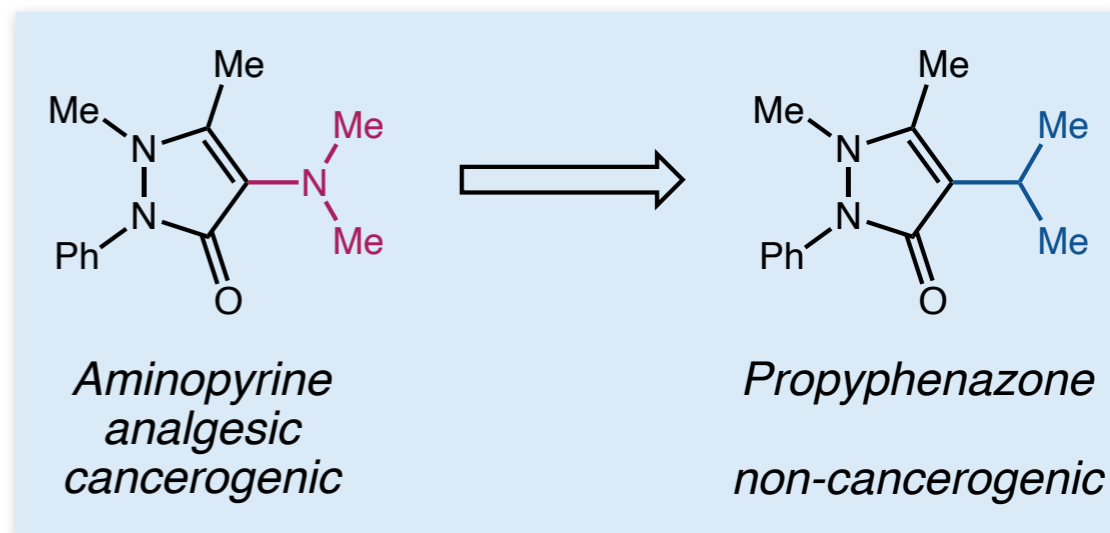
no. of peripheral electrons				
4	5	6	7	8
N ⁺	P	S	Cl	ClH
P ⁺	As	Se	Br	BrH
S ⁺	Sb	Te	I	IH
As ⁺		PH	SH	SH ₂
Sb ⁺			PH ₂	PH ₃

Langmuir, I., *JACS* **1919**, 41, 1543.
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Friedman 1951:
Bioisosteres are "compounds if they fit the broadest definition for isosteres and have the same type of biological activity."



Roche 1933: Saridon

Langmuir, I., *JACS* **1919**, 41, 1543.

Grimm, H. G., *Z. Electrochem.* **1925**, 31, 474.

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Thornber 1979:
“groups or molecules which have **chemical and physical similarities** producing broadly similar biological effects”

1900

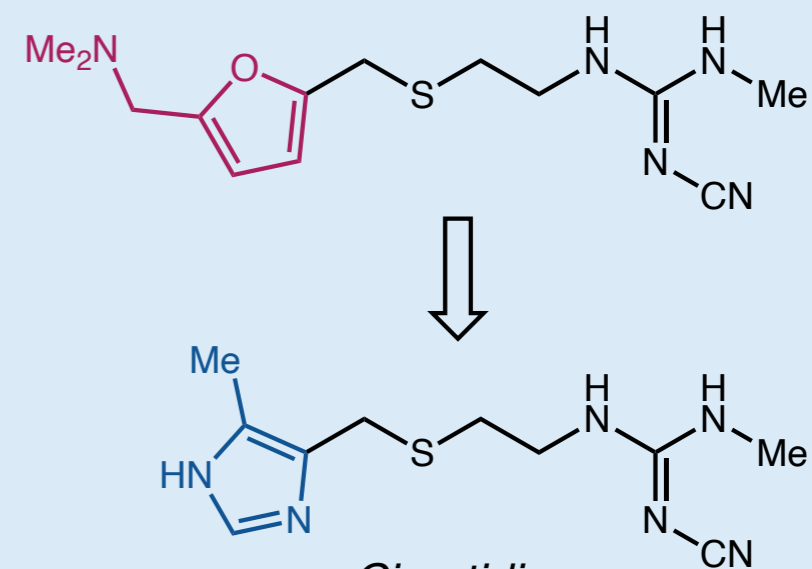
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Meanwell, 2021:
“structural motifs that express similar biological properties or close physicochemical attributes **without the fundamental stipulation that they present a similar shape and size.**”

Langmuir, I., *JACS* **1919**, 41, 1543.

Grimm, H. G., *Z. Electrochem.* **1925**, 31, 474.

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Meanwell, N. A., *J. Med. Chem.* **2021**, 64, 14046.

A Conventional View of Bioisosteres

classical bioisosteres

monovalent bioisosteres

D and H

F and H

NH and OH

RSH and ROH

F, OH, NH₂ and CH₃

Cl, Br, SH and OH

C and Si

bivalent bioisosteres in which two single

bonds are affected

C=C, C=N, C=O, C=S

-CH₂-, -NH-, -O-, -S-

RCOR', RCONHR', RCOOR', RCOSR'

trivalent bioisosteres in which three

bonds are affected

R₃CH, R₃N

R₄C, R₄Si, R₄N⁺

alkene, imine

-CH=CH-, -S-

-CH= and -N=C

nonclassical bioisosteres

are structurally distinct, usually comprise different number of atoms and

exhibit different steric and electronic properties compared to the

functionality being emulated

have been divided into two subgroups:²

1. cyclic and noncyclic isosteres
2. exchangeable group isosterism in which the properties of discrete functional elements are emulated

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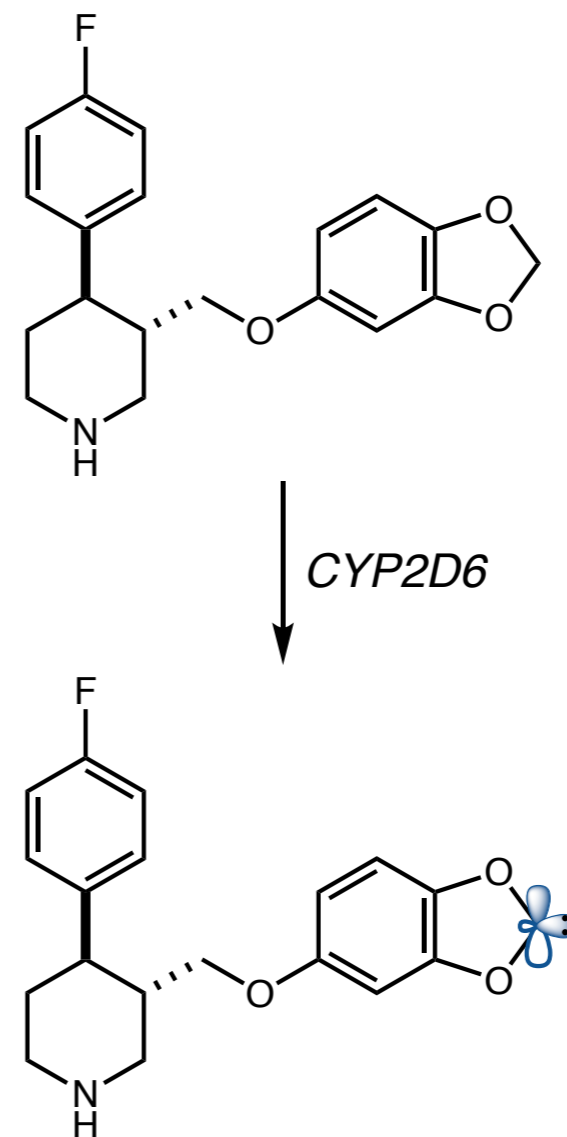
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paroxetine
antidepressant

adverse effects

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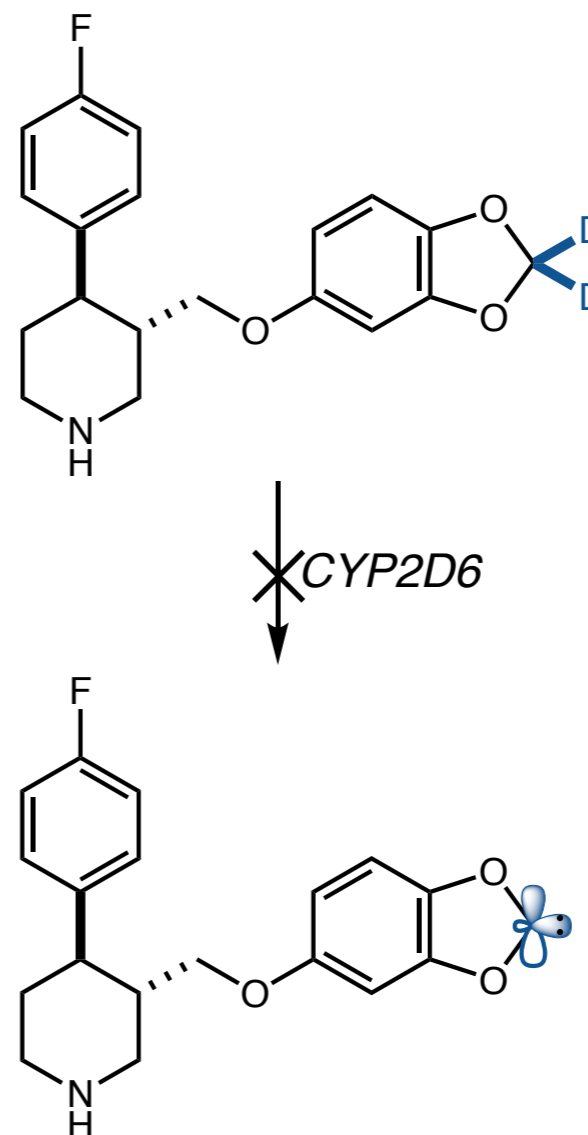
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H → D isostere

**strongly reduced
metabolization**

adverse effects

A Conventional View of Bioisosteres

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classical bioisosteres

monovalent bioisosteres

D and H
 F and H
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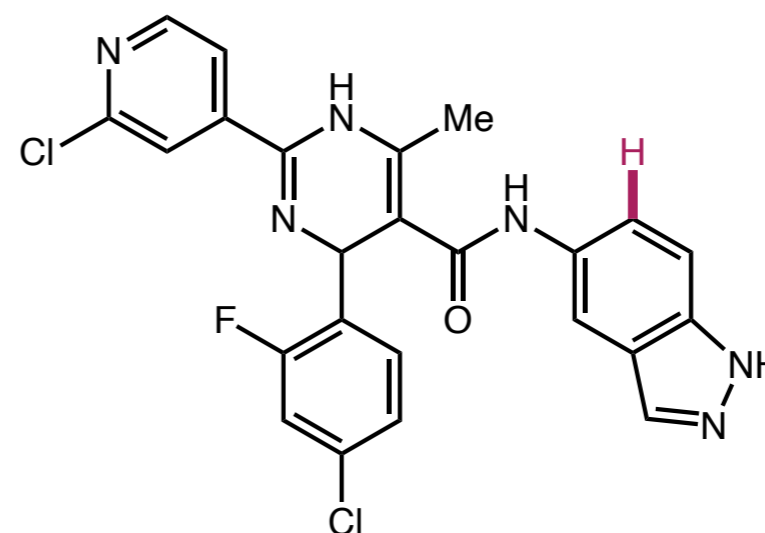
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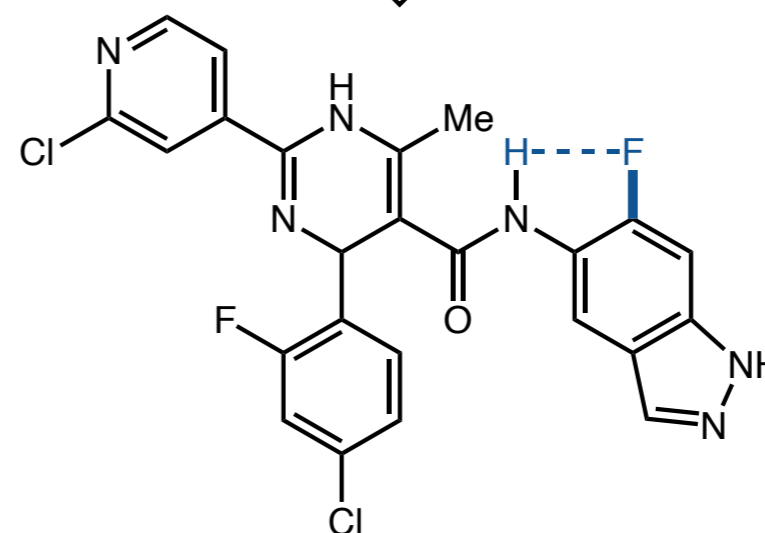
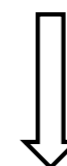
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Rho kinase antagonist

liver toxicity
poor oral bioavailability
(F = 7%)



H → F isostere

equipotent
improved
oral bioavailability
(F = 49%)

LaVoie, E. J., *Chem. Rev.* **1996**, *96*, 3147.
 Meanwell, N. A., *J. Med. Chem.* **2011**, *54*, 2529.
 Lee, D., *J. Med. Chem.* **2008**, *51*, 6631.

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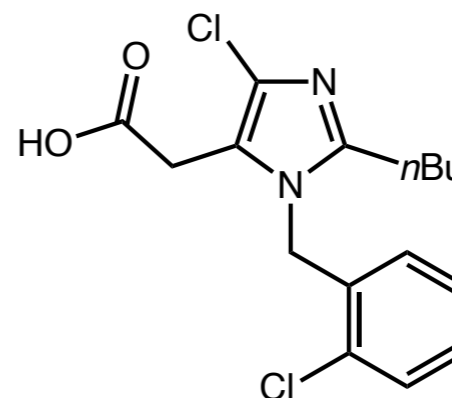
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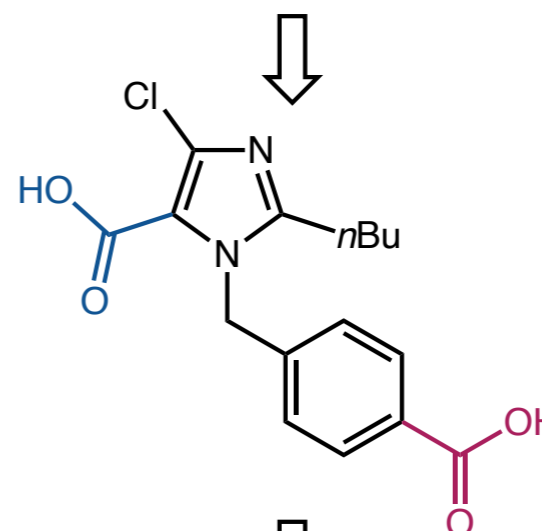
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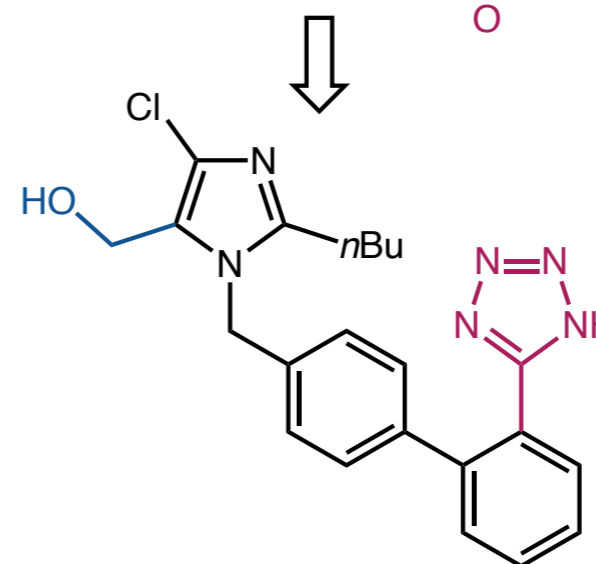
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initial lead
 AT1R antagonist
 IC₅₀ = 40 mM



IC₅₀ = 1.2 mM



Losartan
 Merck

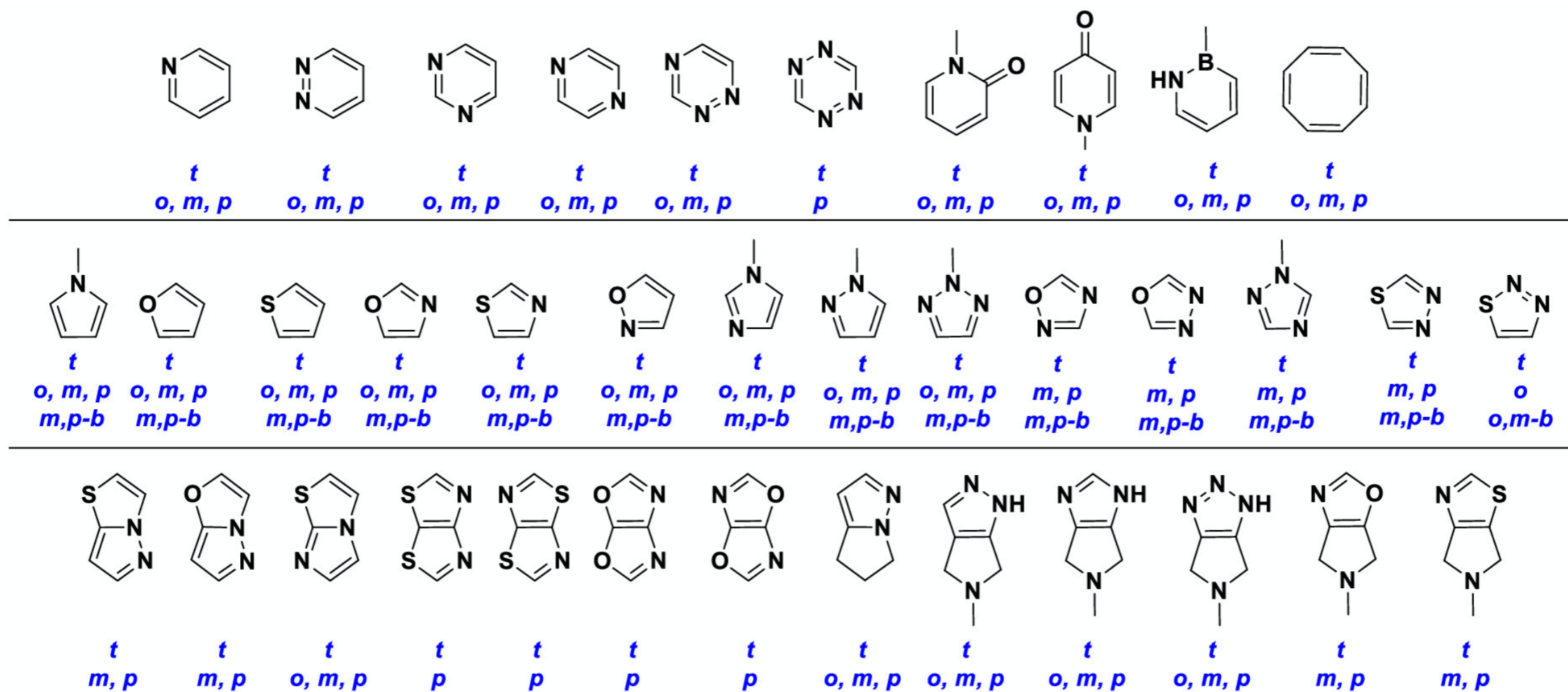
IC₅₀ = 19 μM

60M prescriptions
 (US/2018)

LaVoie, E. J., *Chem. Rev.* **1996**, *96*, 3147.
 Meanwell, N. A., *J. Med. Chem.* **2011**, *54*, 2529.
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 Mavromoustakos, T., *Molecules* **2021**, *26*, 2927.

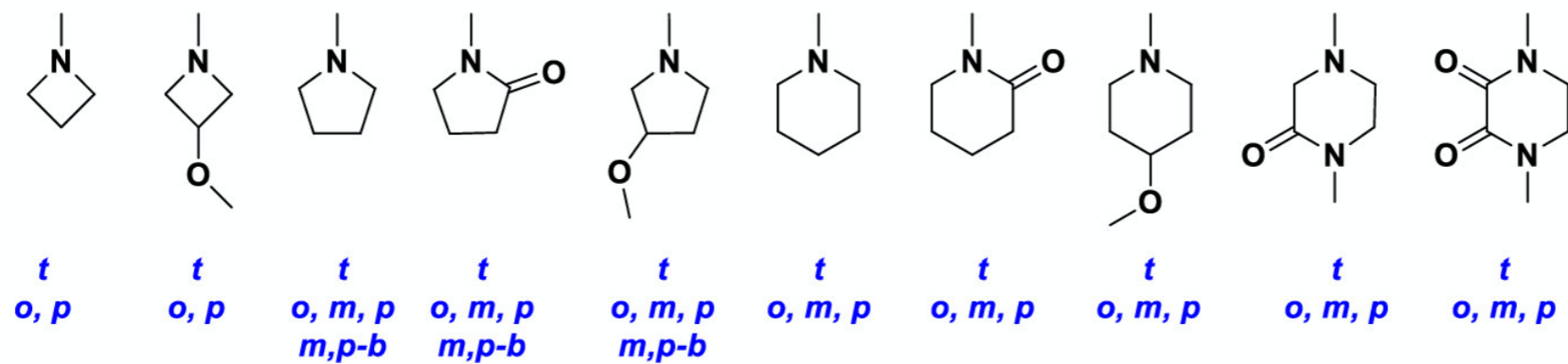
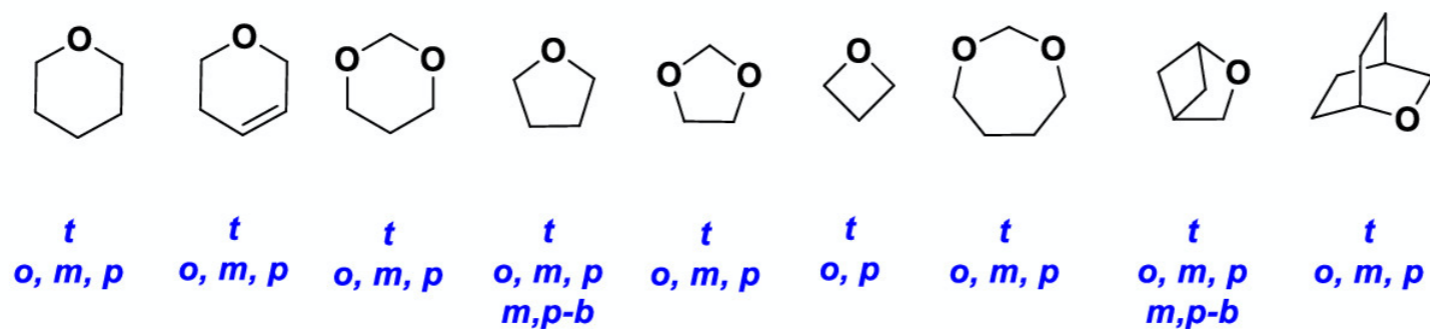
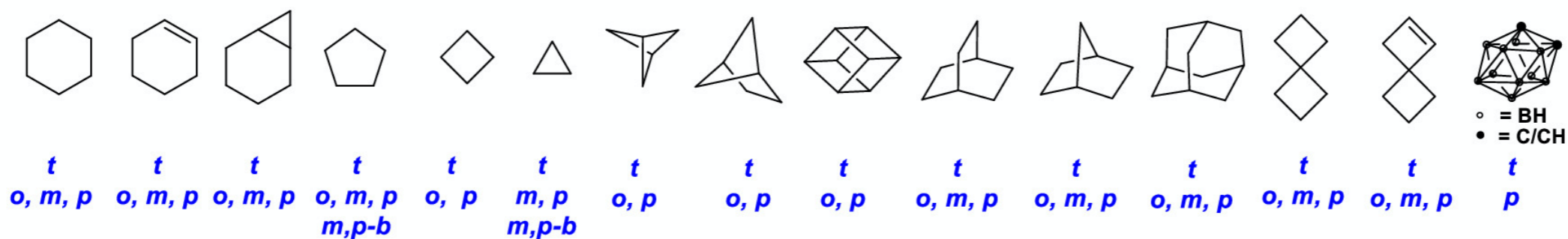
A Most Likely Incomplete List of Proposed Phenyl Bioisosteres

“similar biological properties or close physicochemical attributes without the fundamental stipulation that they present a similar shape and size.”



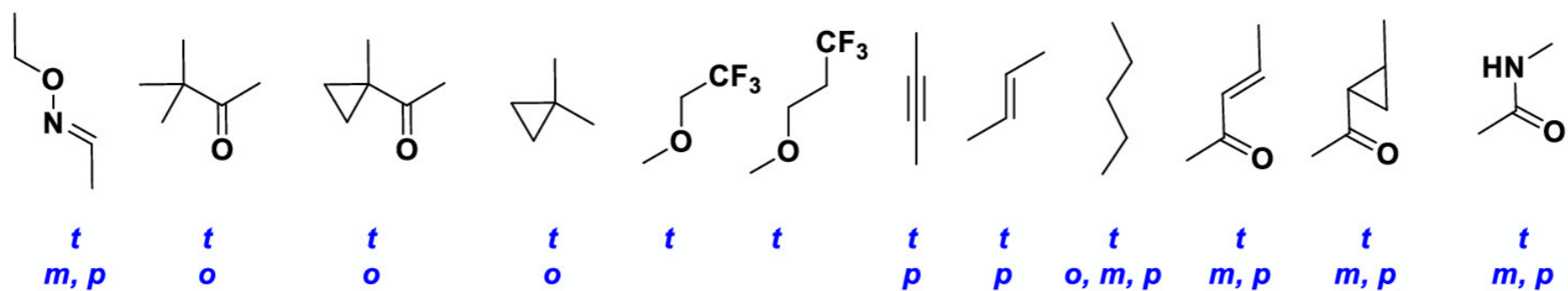
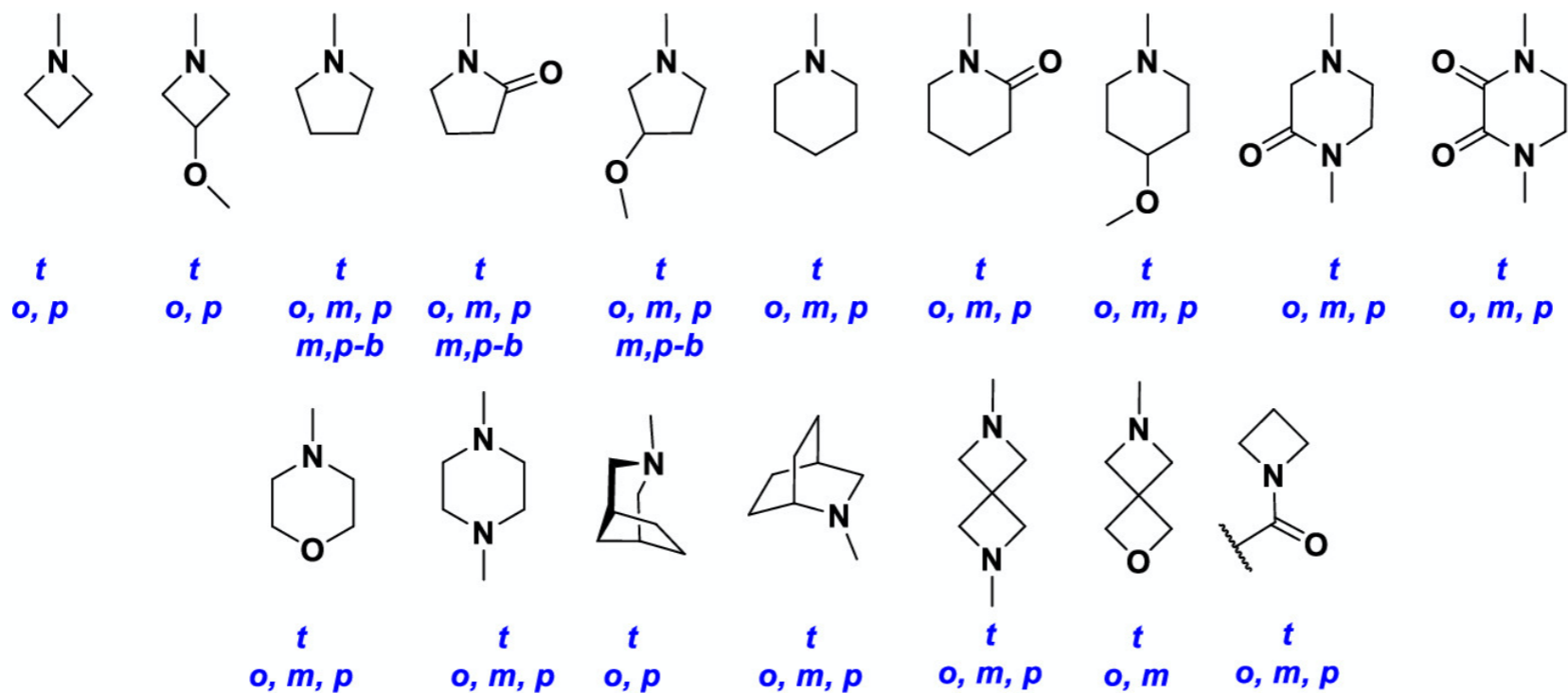
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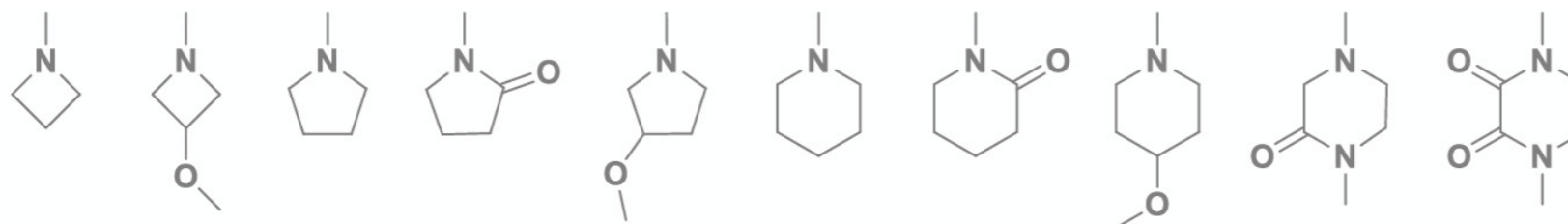
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A Most Likely Incomplete List of Proposed Phenyl Bioisosteres

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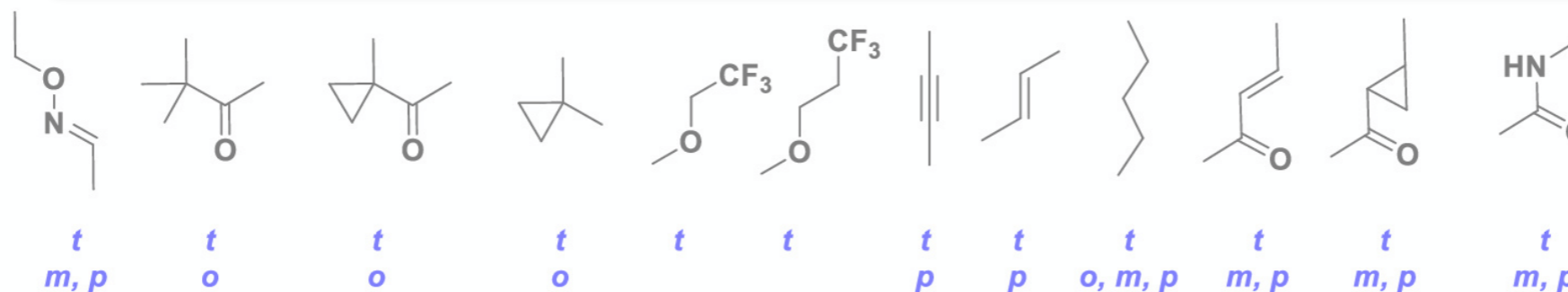


o

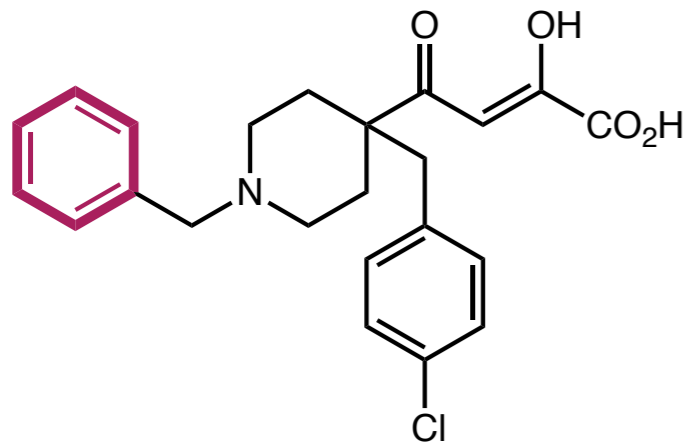
This talk is by no means a complete overview.

Its goal is to give an idea of how challenging it is to predict the the biological consequences and efficacy of bioisosteric replacements.

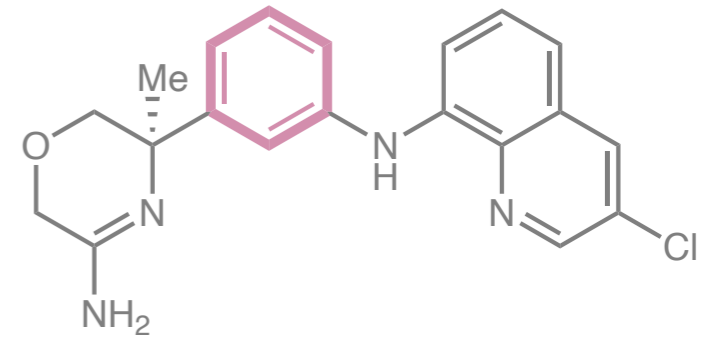
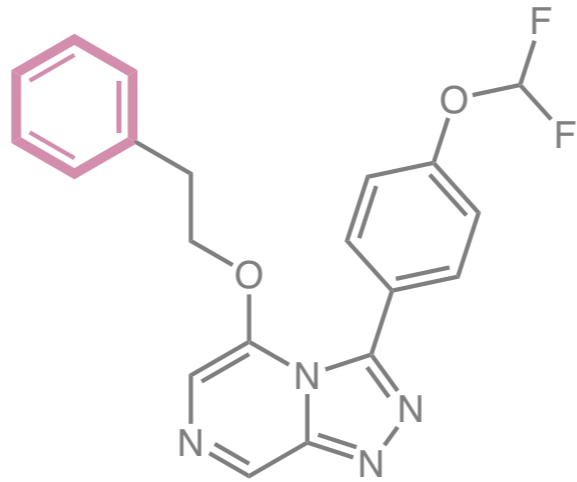
Emphasis will be on a select few more obscure or easily disregarded replacements; replacements the group is well familiar with will only play a minor part.



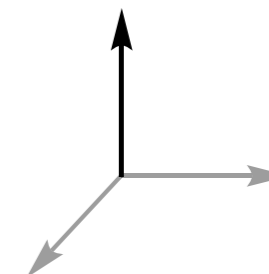
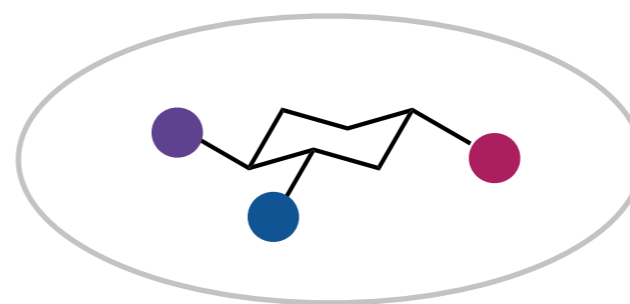
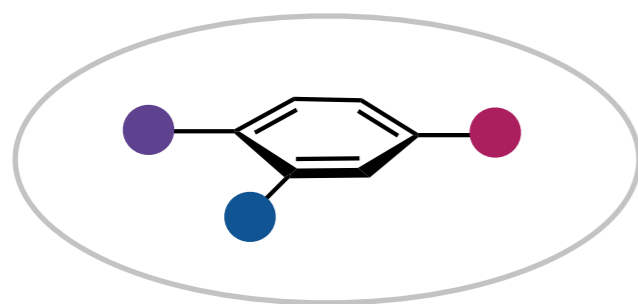
Outline



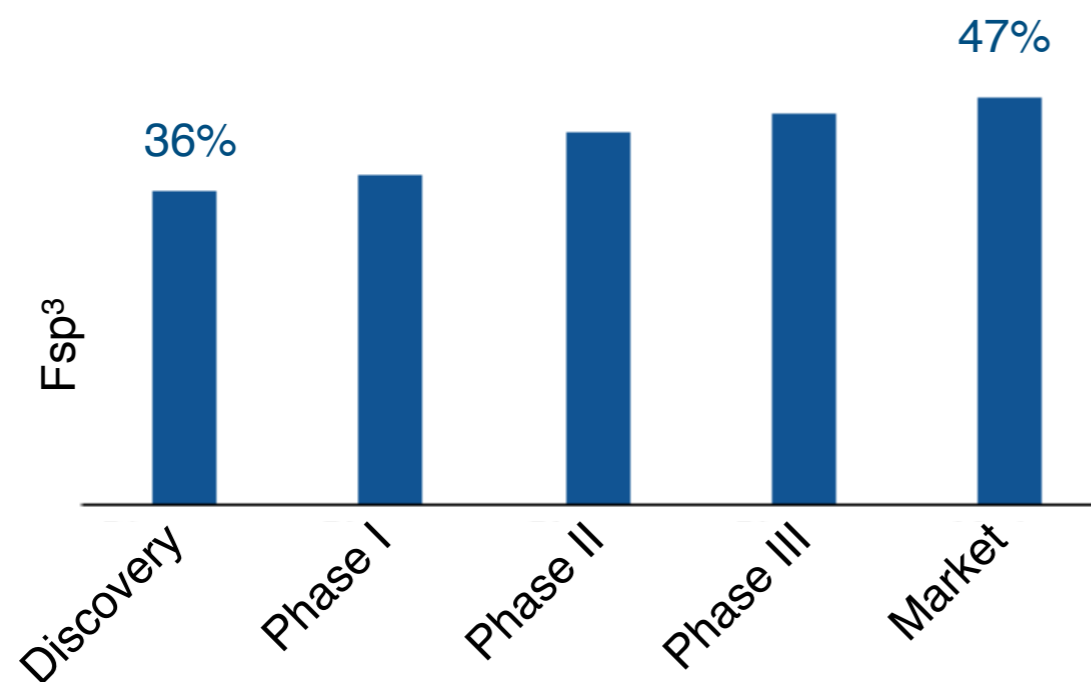
***beneficial effects and
limitations of an often
disregarded bioisostere***



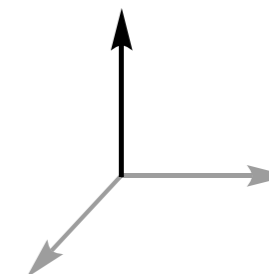
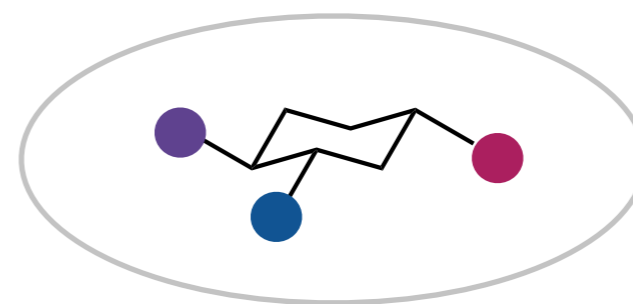
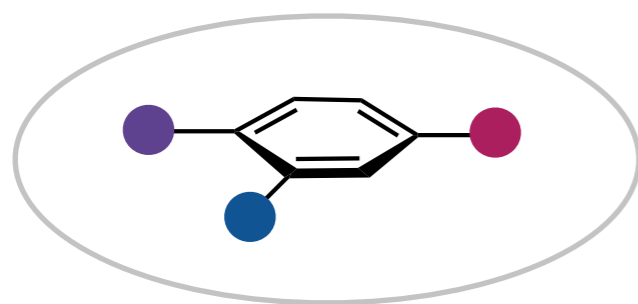
The Escape from Flatland



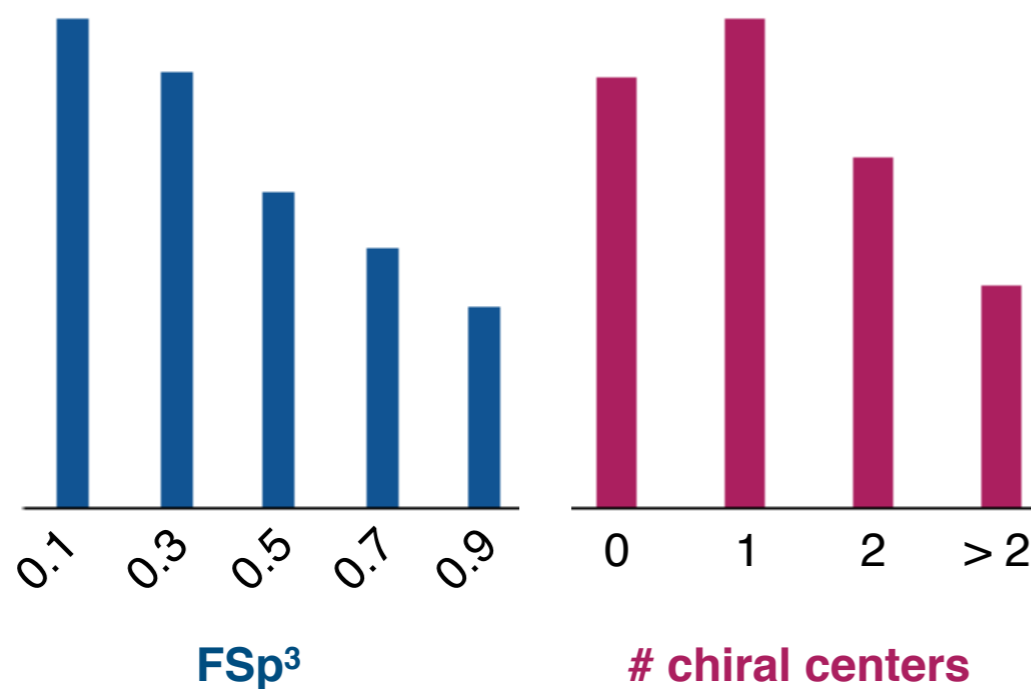
**increasing $C(sp^3)$ content
correlated with clinical
success**



The Escape from Flatland



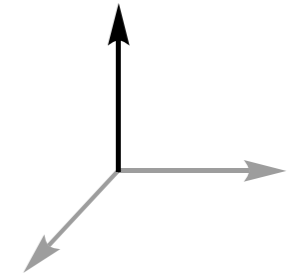
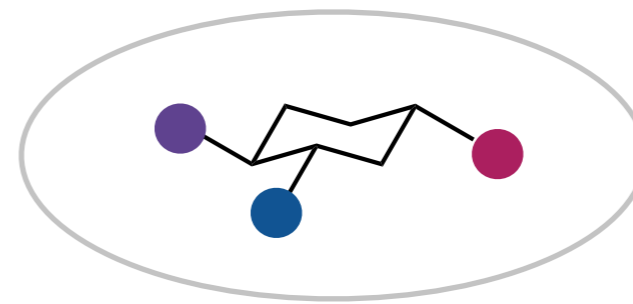
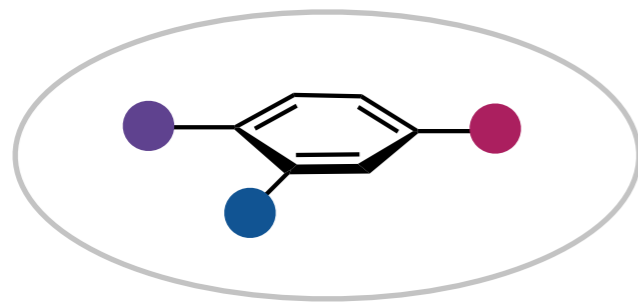
promiscuity



*increasing C(sp³) content
correlated with clinical
success*

*increasing molecular
complexity correlates with
selectivity*

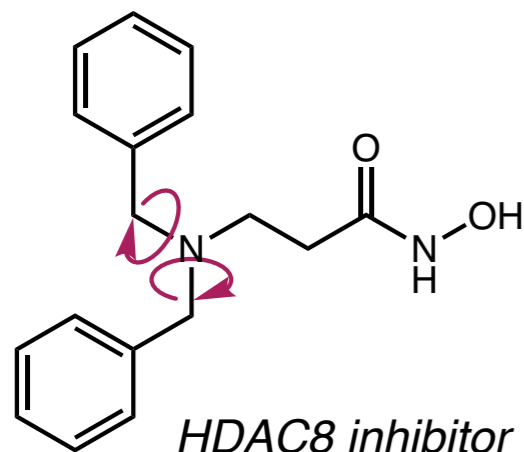
The Escape from Flatland



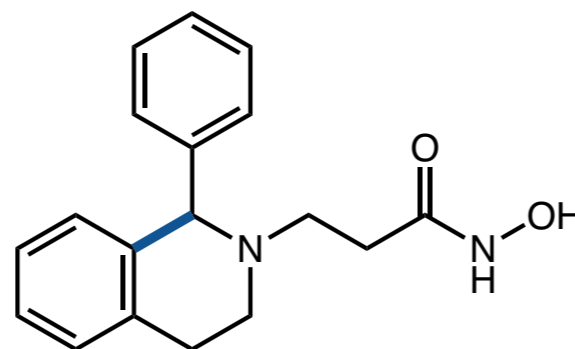
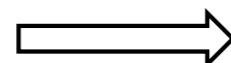
**increasing C(sp³) content
correlated with clinical
success**

**increasing molecular
complexity correlates with
selectivity**

**increasing molecular
rigidity in scaffolds leads
to improved selectivity**



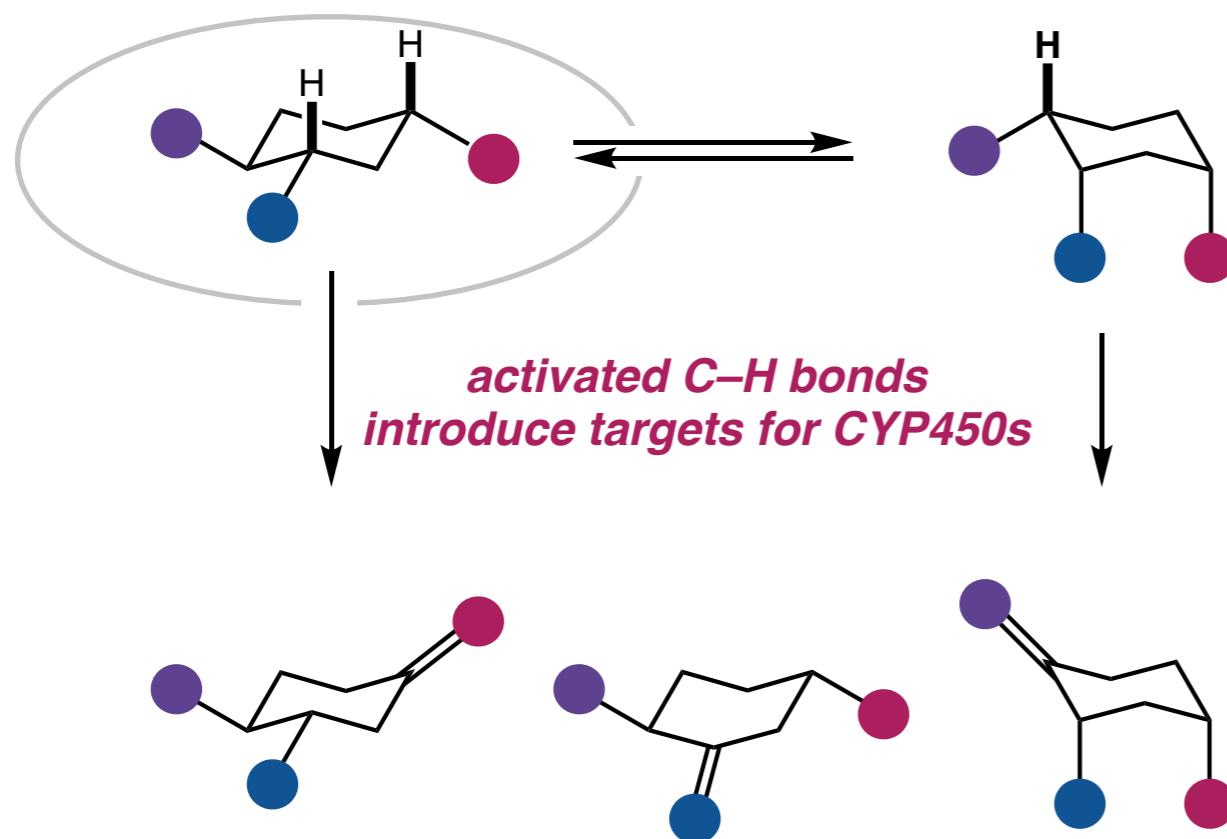
HDAC8 inhibitor
IC₅₀ = 1400 nM



IC₅₀ = 950 nM

The Escape from Flatland

Is cyclohexyl really that useless as a bioisostere?



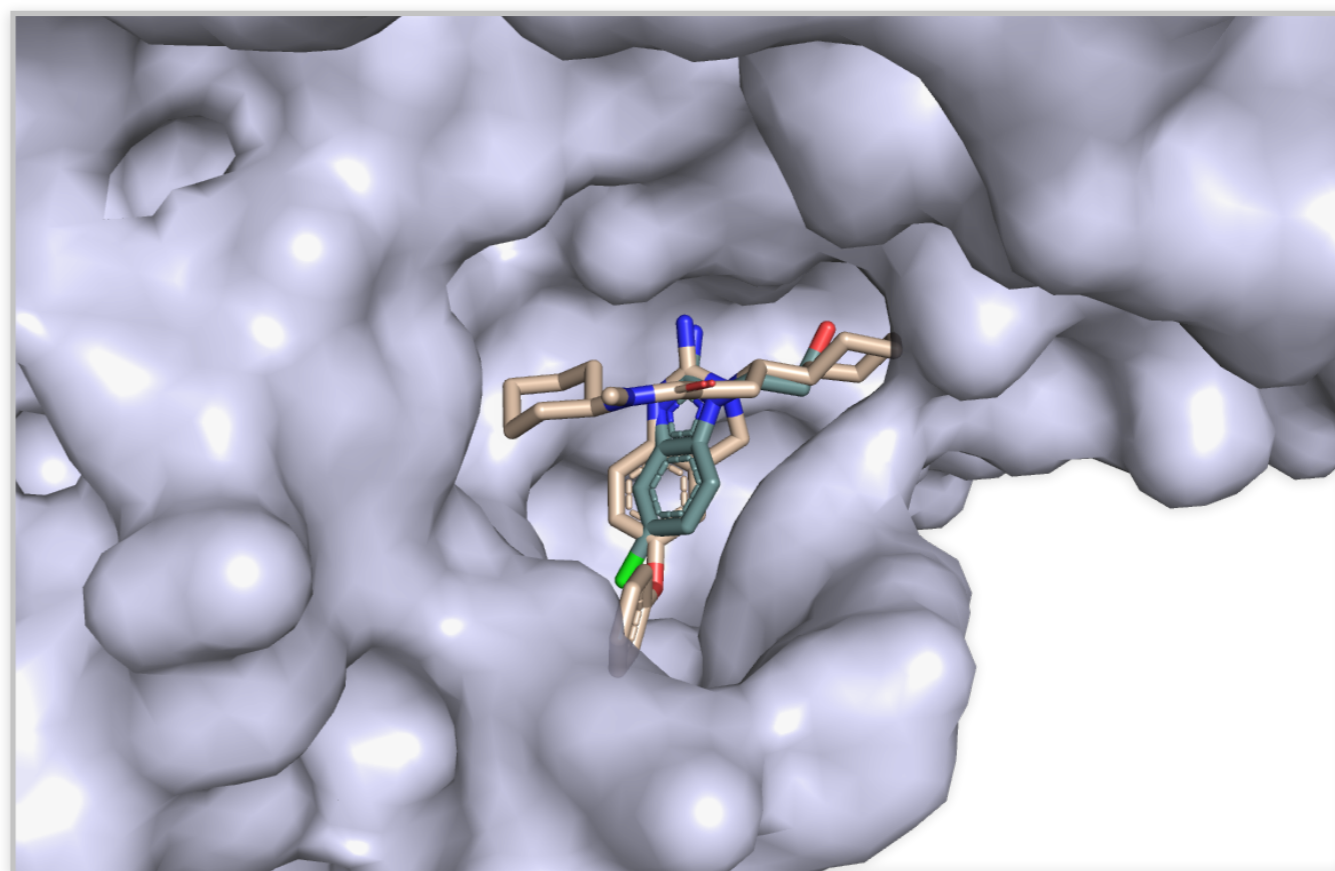
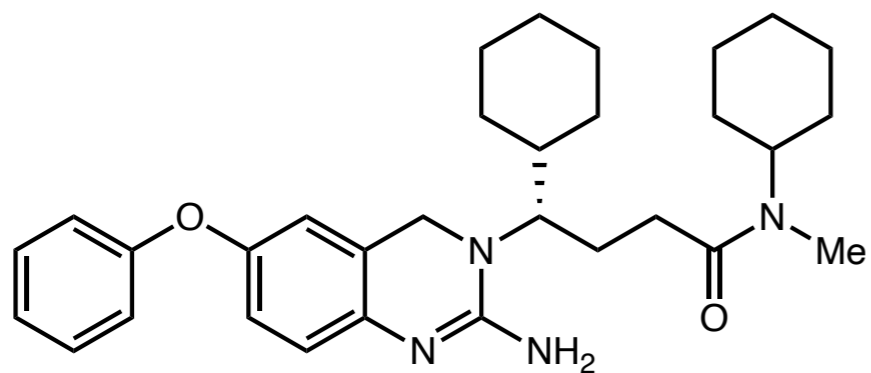
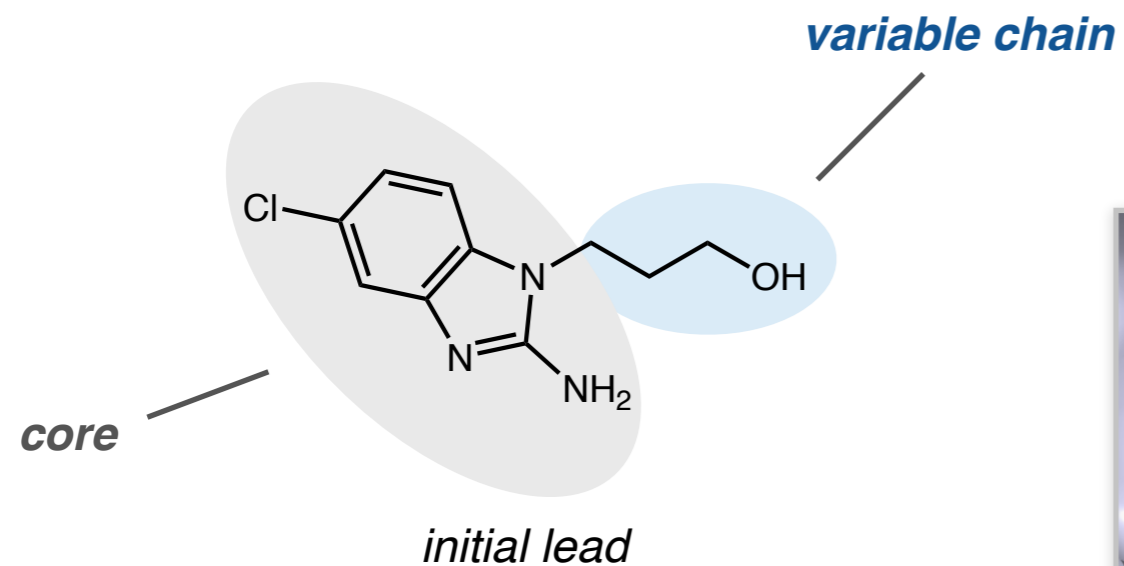
**increasing $C(sp^3)$ content
correlated with clinical
success**

**increasing molecular
complexity correlates with
selectivity**

**increasing molecular
rigidity in scaffolds leads
to improved selectivity**

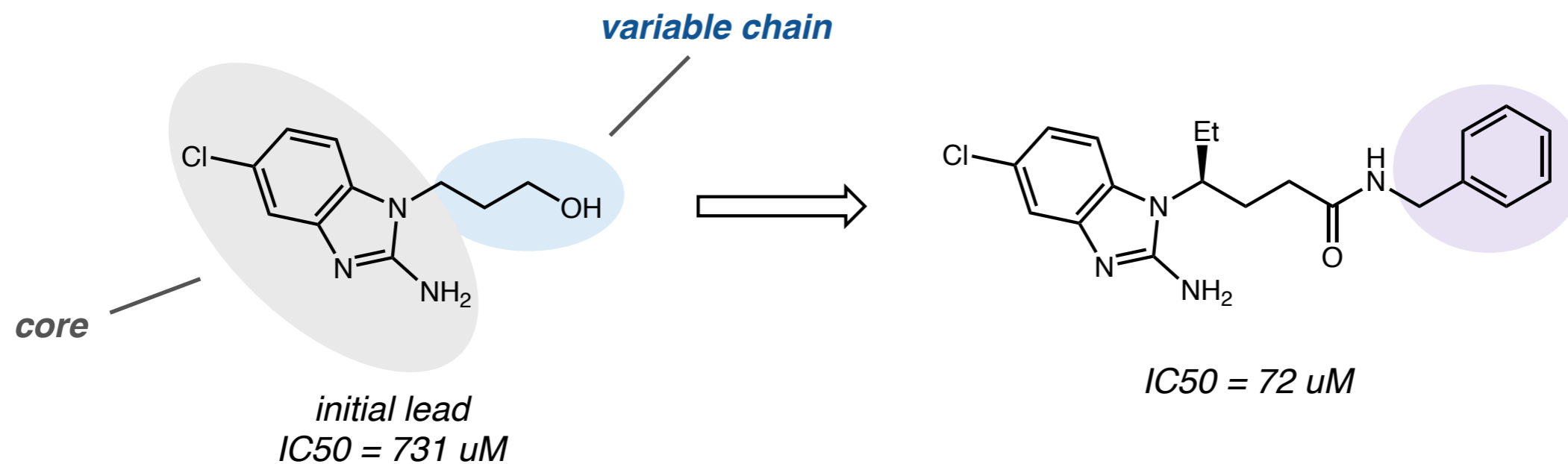
Cyclohexanes as Phenyl Bioisosteres

A Case Study on β -Secretase Inhibitors

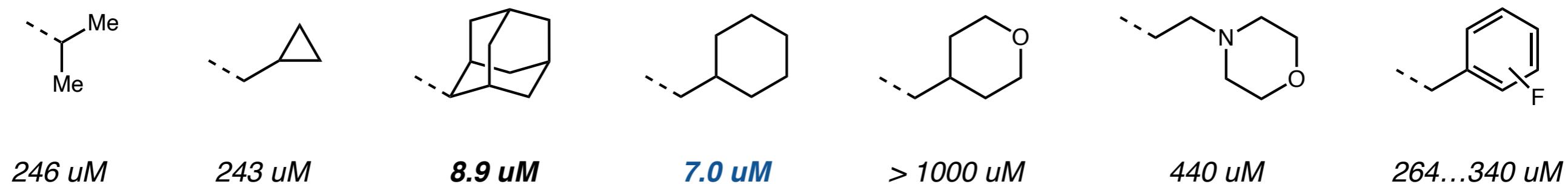


Cyclohexanes as Phenyl Bioisosteres

A Case Study on β -Secretase Inhibitors

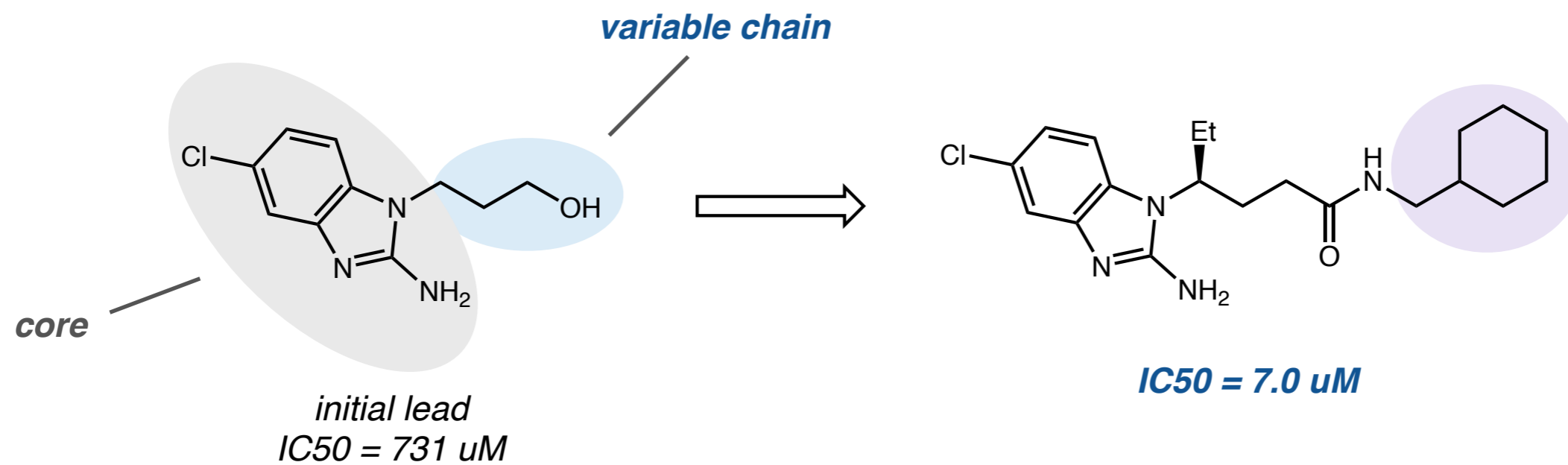


SAR at Eastern amide

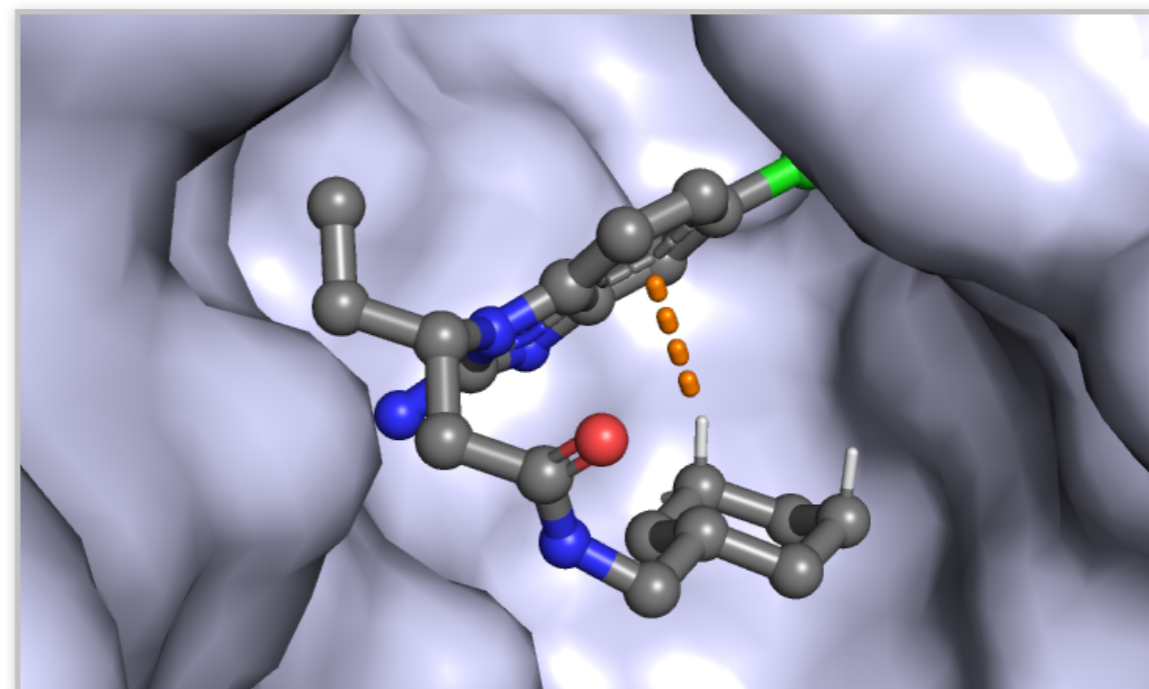
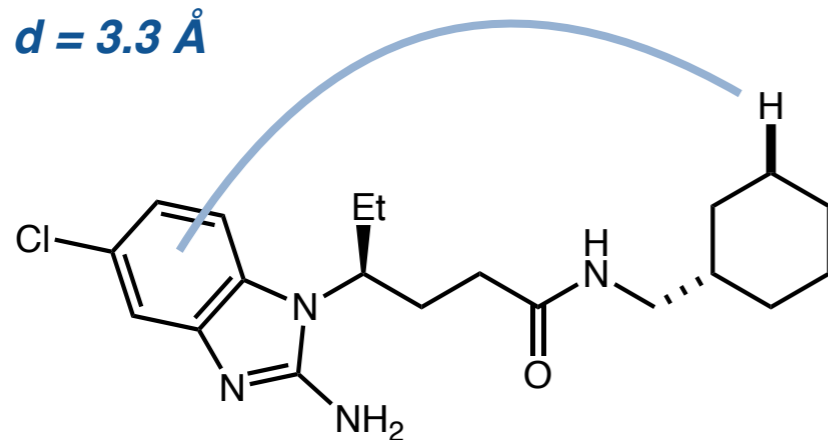


Cyclohexanes as Phenyl Bioisosteres

A Case Study on β -Secretase Inhibitors

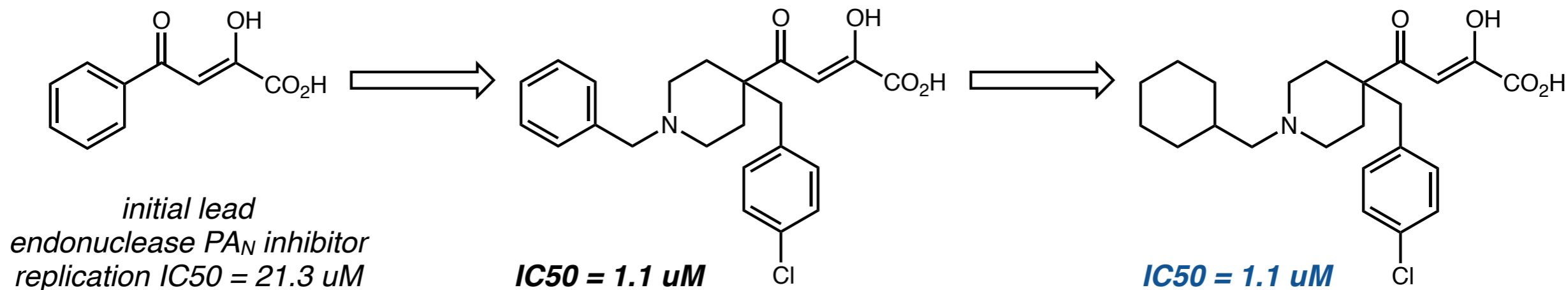


key C–H- π interaction
 $d = 3.3 \text{ \AA}$

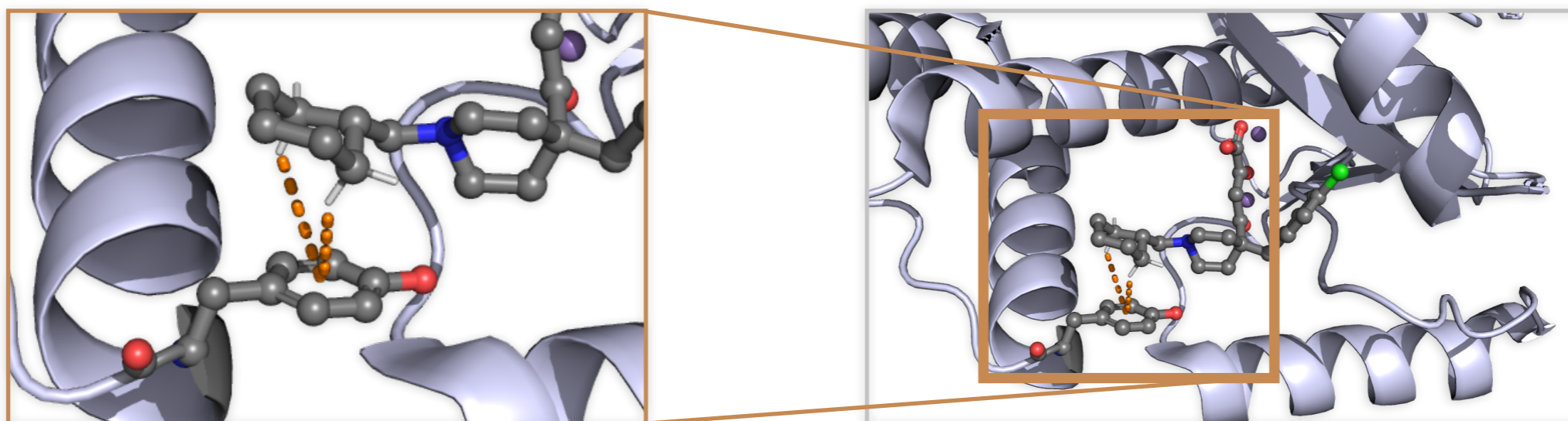


Cyclohexanes as Phenyl Bioisosteres

Intermolecular C–H– π Interactions Can Improve Binding

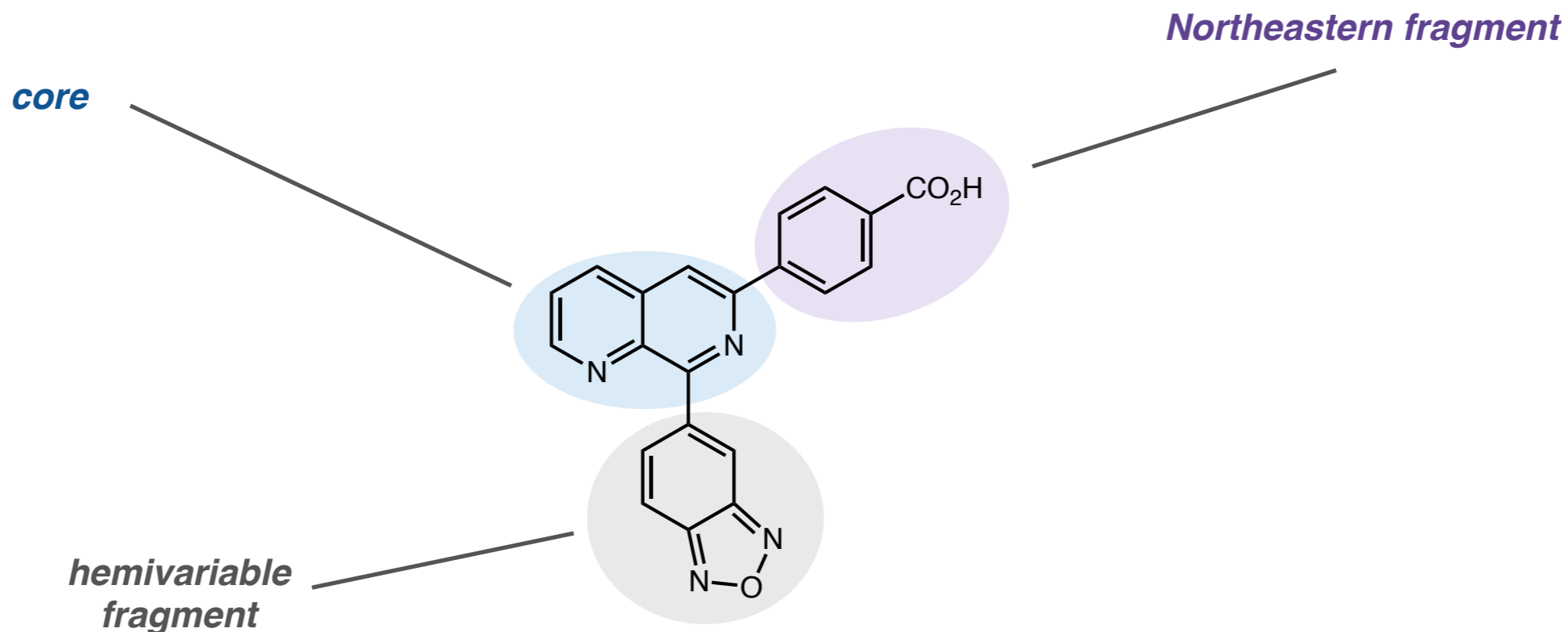


key C–H– π interaction with Tyr24

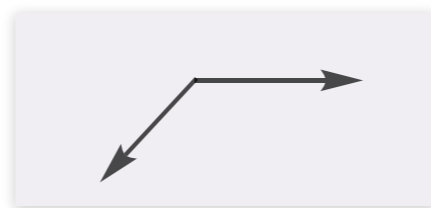


Cyclohexanes as Phenyl Bioisosteres

Viable Scaffolds to Escape Flatland

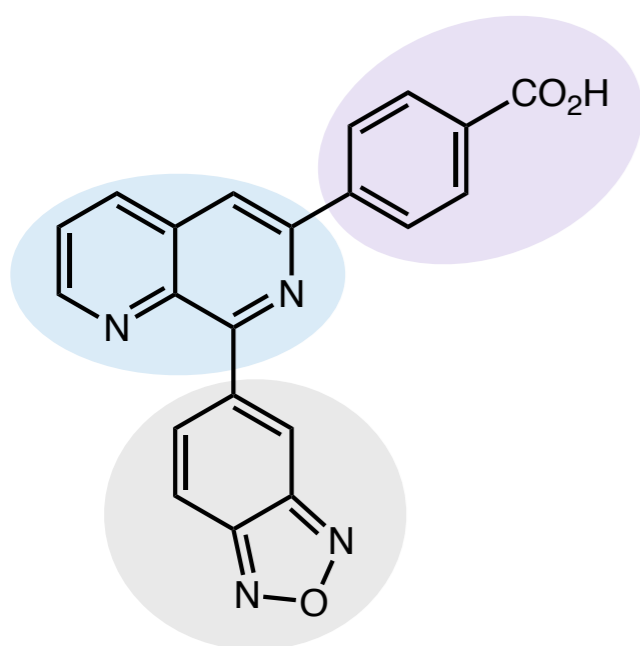


PDE4 inhibitor
initial lead
solubility: 2.3 ug/mL
oral availability: 8%

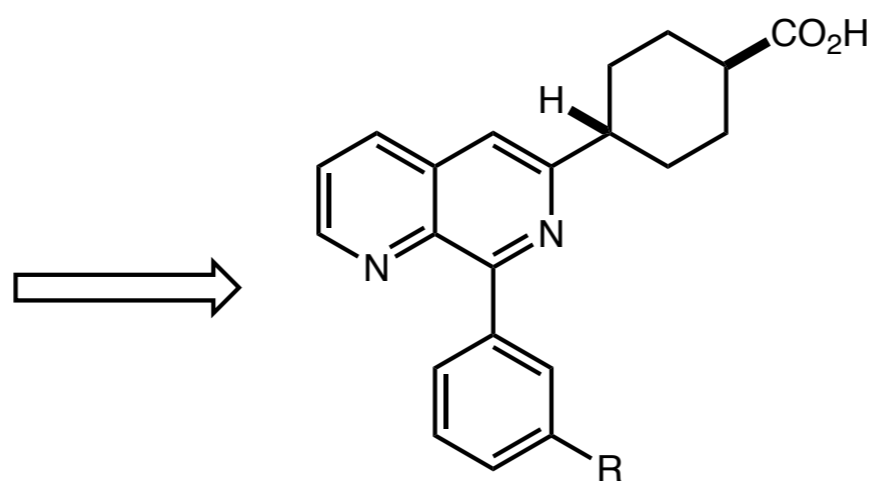
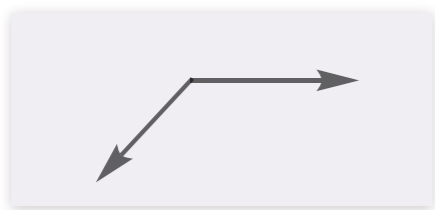


Cyclohexanes as Phenyl Bioisosteres

Viable Scaffolds to Escape Flatland

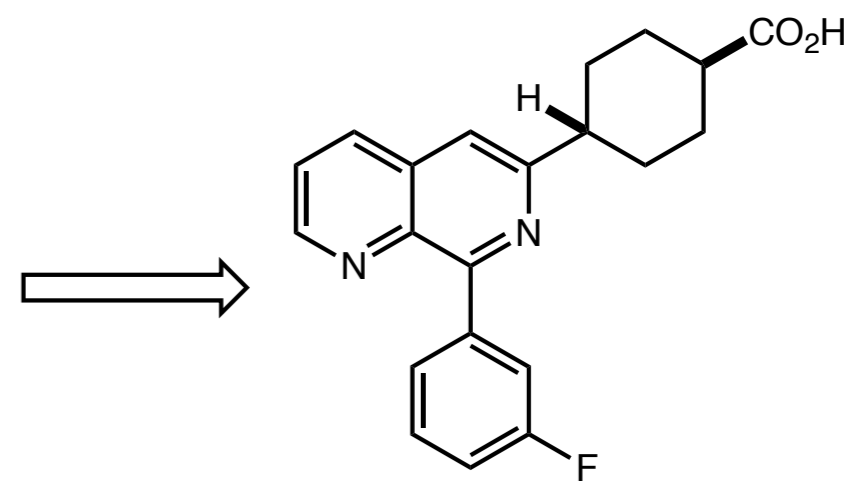


PDE4 inhibitor
initial lead
solubility: 2.3 ug/mL
oral availability: 8%



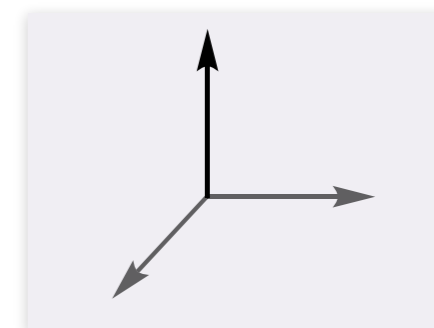
---OCF₃ **solubility: 6.0 ug/mL**
 ---OMe 150 ug/mL
 ---CN 430 ug/mL
 ---SMe **1100 ug/mL**
 adverse C450P?

Trifilieff, A., *J. Pharmacol. Exp. Ther.* **2002**, 301, 241.
 Press, N. J. *J. Med. Chem.* **2012**, 55, 7472.
 Press, N. J., *J. Med. Chem.* **2015**, 58, 6747.



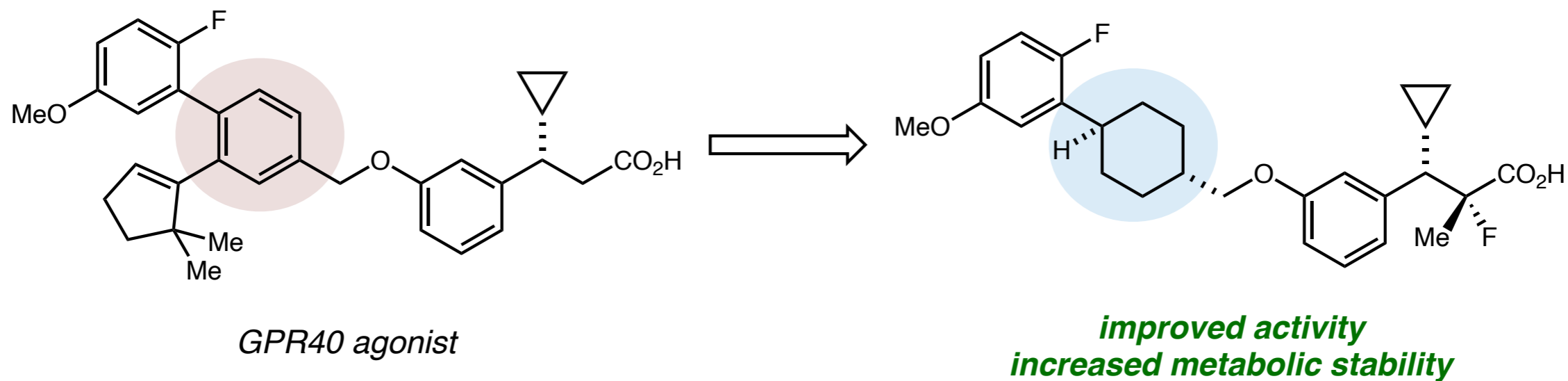
clinical candidate

solubility: 920 ug/mL
oral availability: 52%

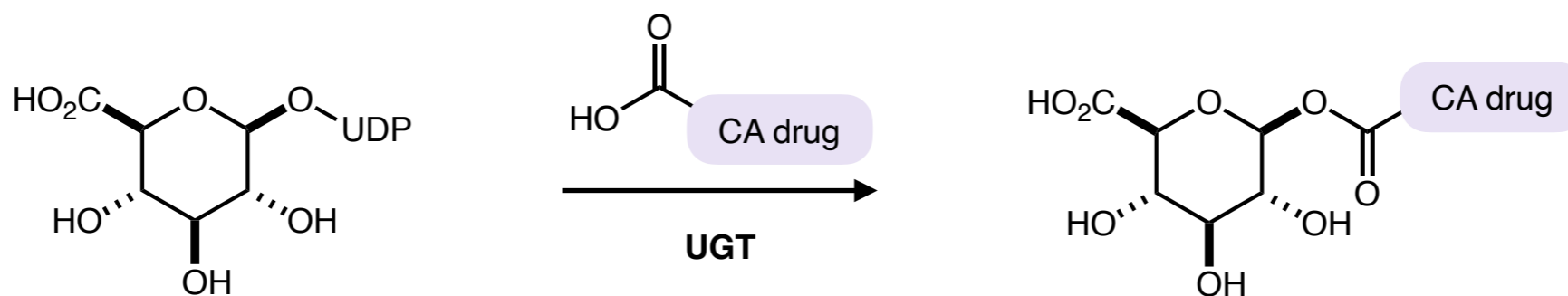


Cyclohexanes as Phenyl Bioisosteres

Viable Scaffolds to Escape Flatland

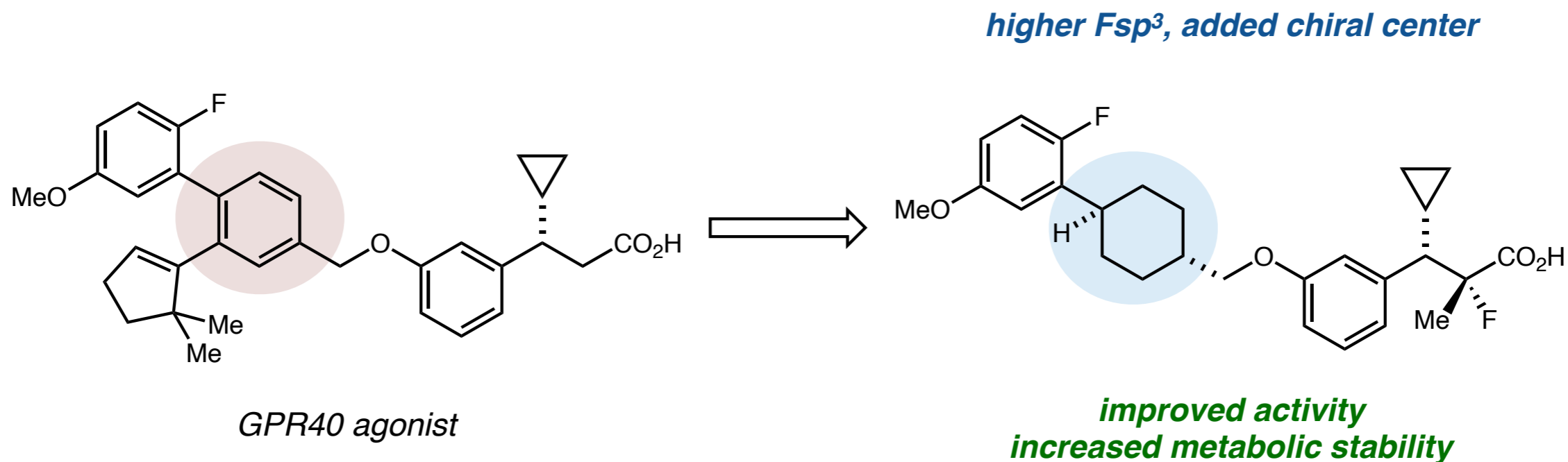


Glucuronide Metabolites

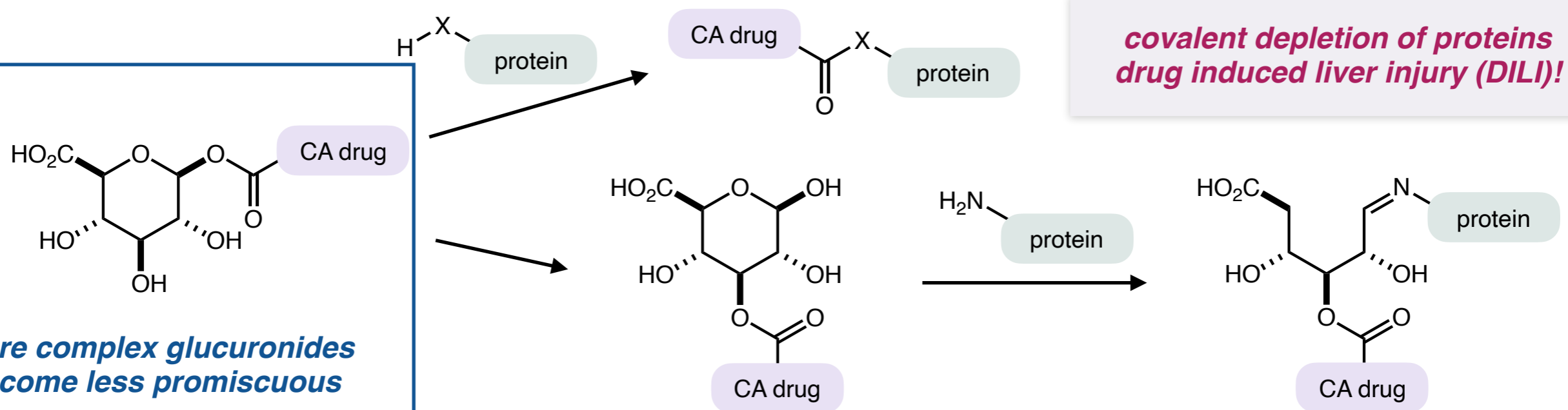


Cyclohexanes as Phenyl Bioisosteres

Viable Scaffolds to Escape Flatland

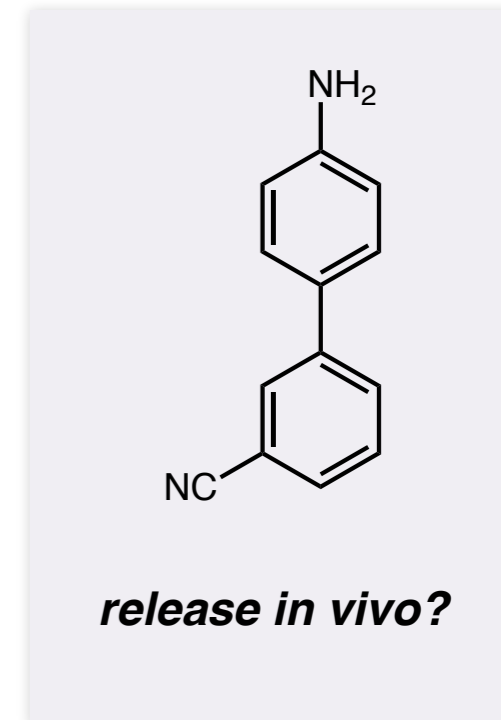
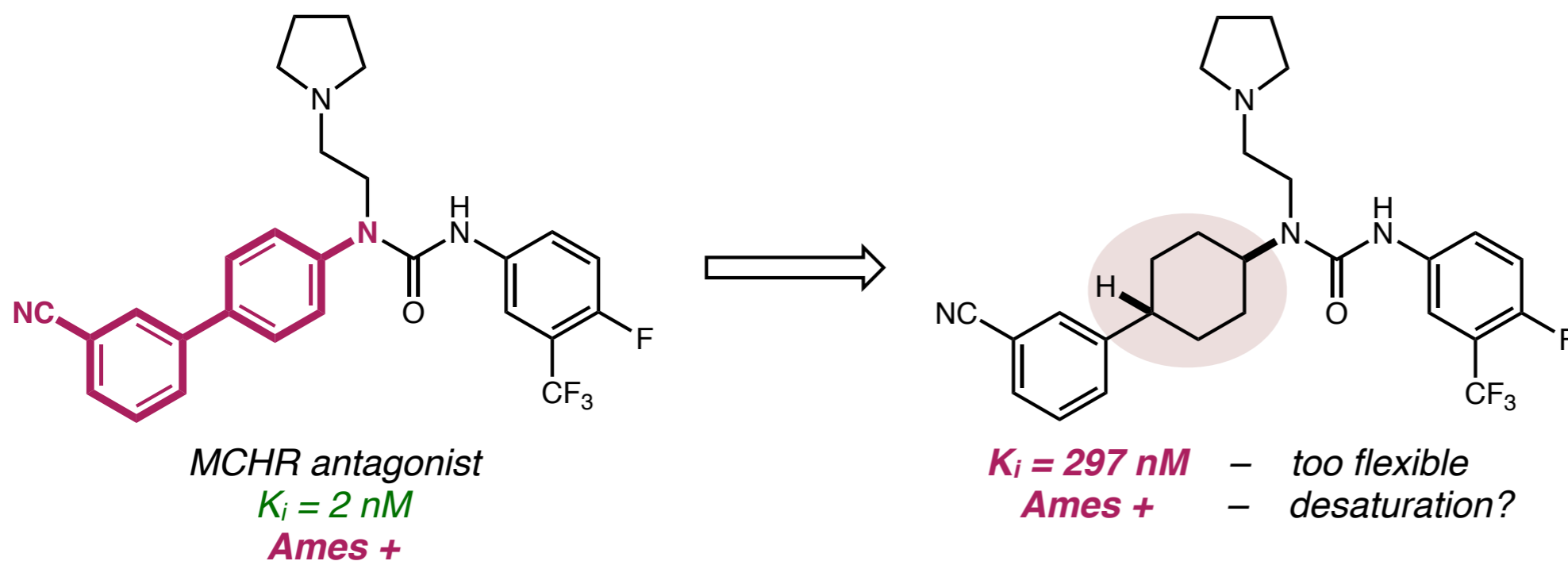
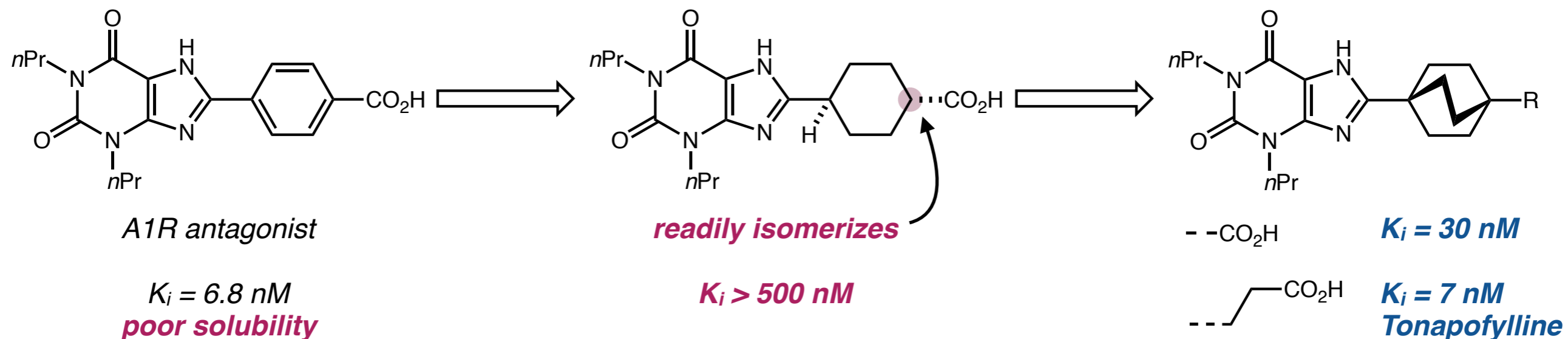


Glucuronide Metabolites



Cyclohexanes as Phenyl Bioisosteres

Limitations & Workarounds



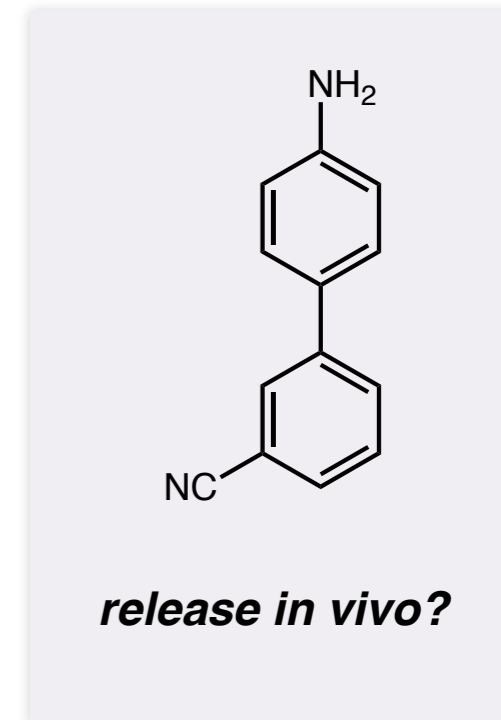
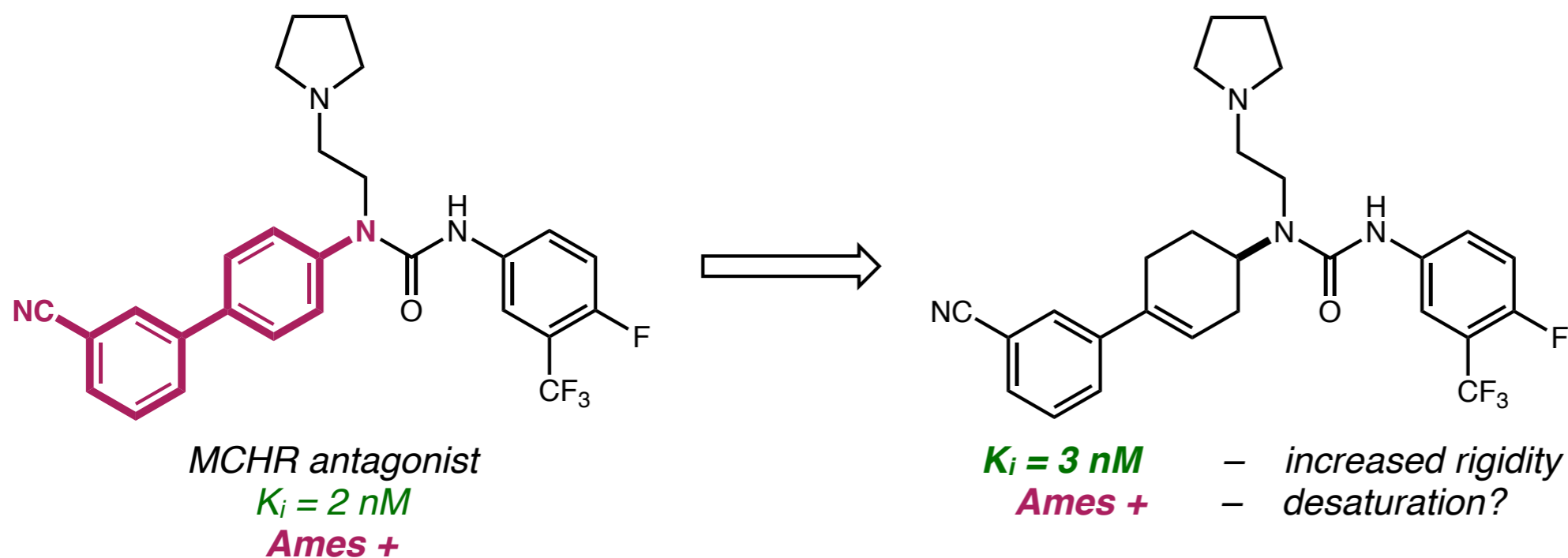
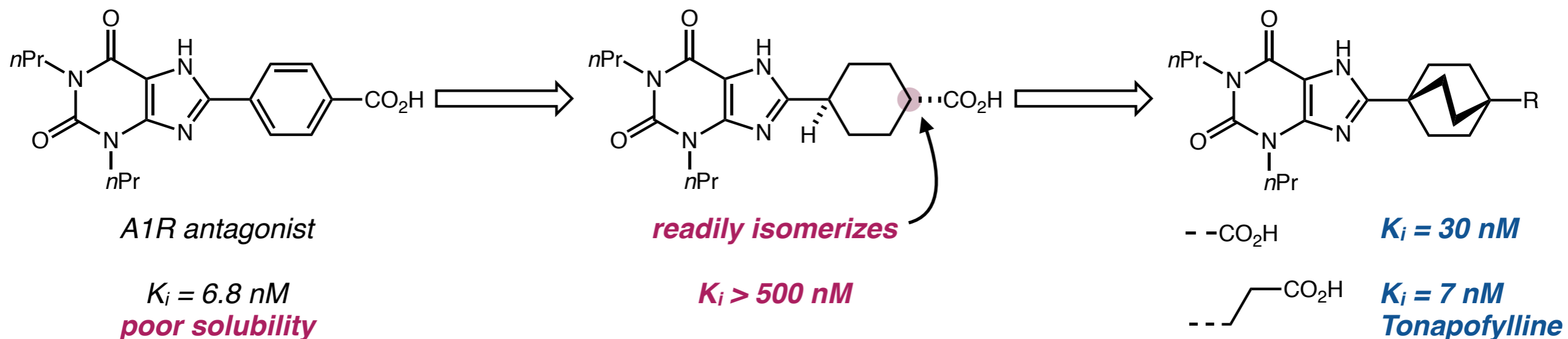
Jacobsen, K. A., *J. Med. Chem.* **2000**, 43, 1165.

Kiesman, W. F., *J. Med. Chem.* **2006**, 49, 7119.

Xu, R., *Bioorg. Med. Chem.* **2006**, 14, 3285.

Cyclohexanes as Phenyl Bioisosteres

Limitations & Workarounds



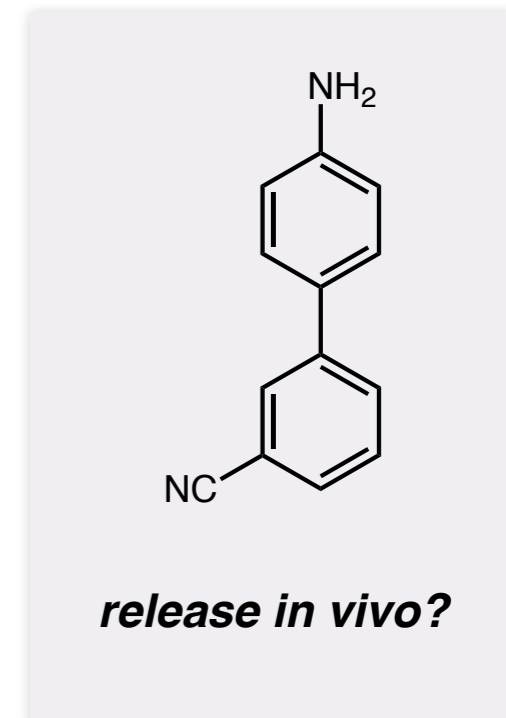
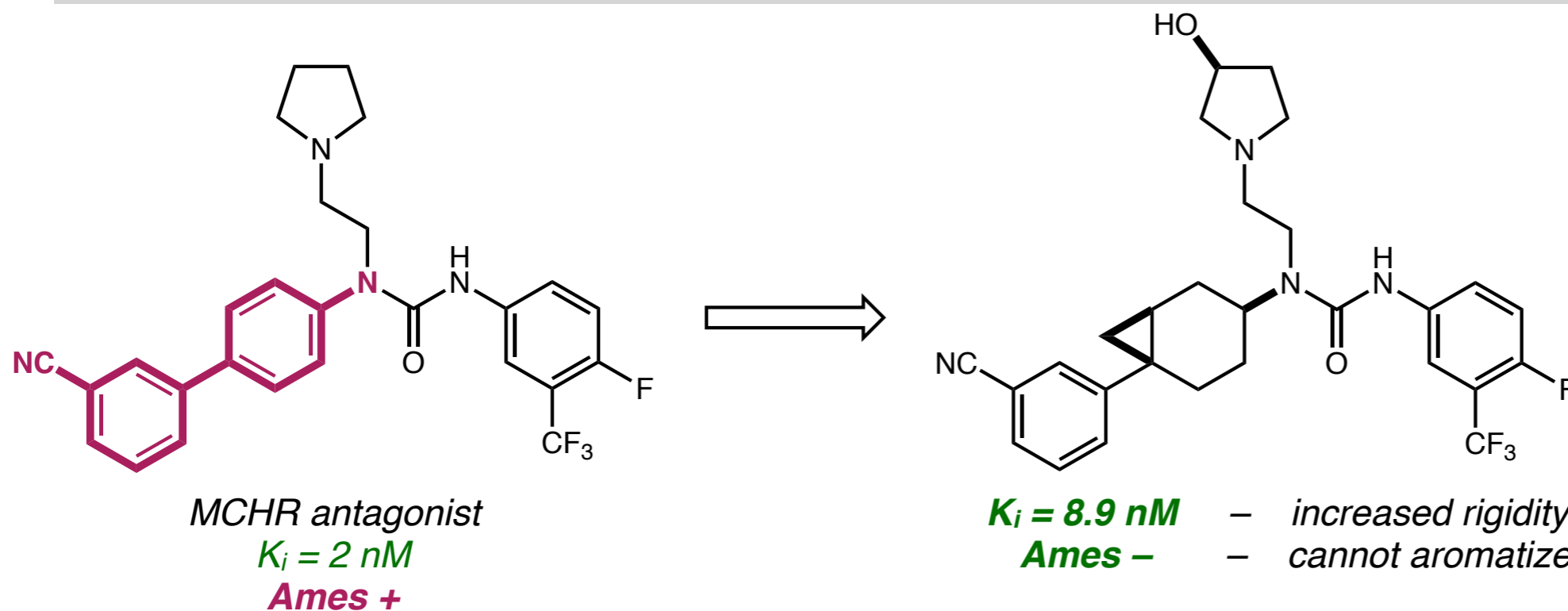
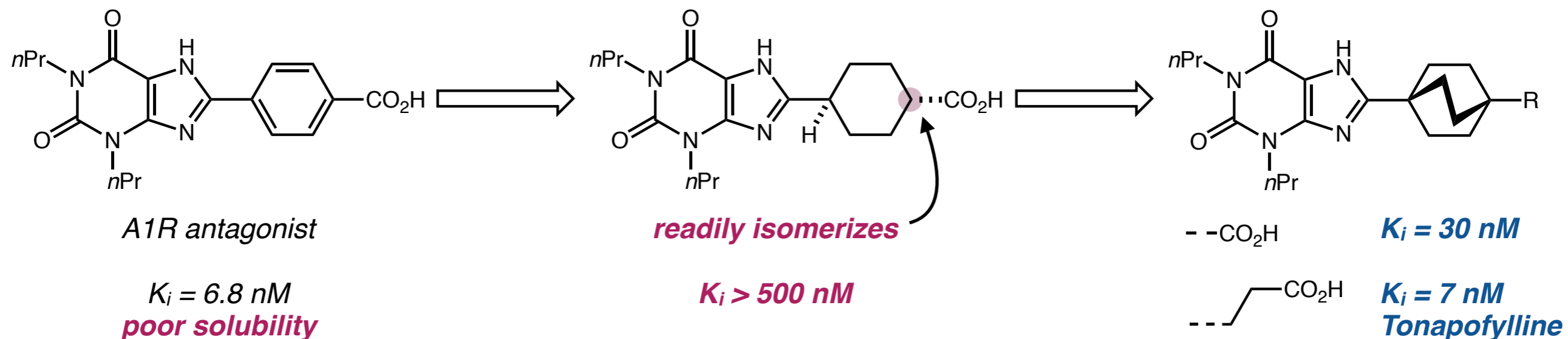
Jacobsen, K. A., *J. Med. Chem.* **2000**, 43, 1165.

Kiesman, W. F., *J. Med. Chem.* **2006**, 49, 7119.

Xu, R., *Bioorg. Med. Chem.* **2006**, 14, 3285.

Cyclohexanes as Phenyl Bioisosteres

Limitations & Workarounds

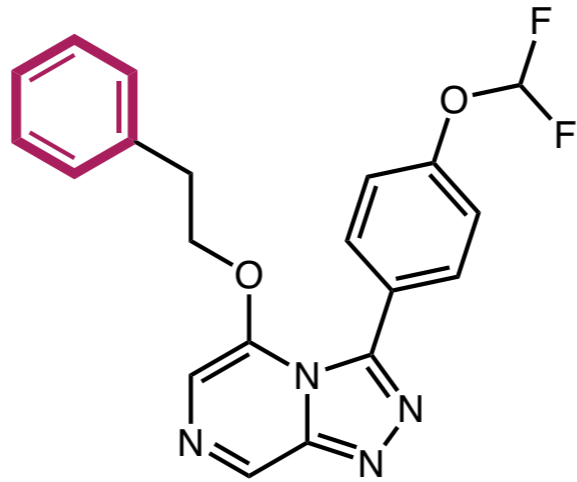
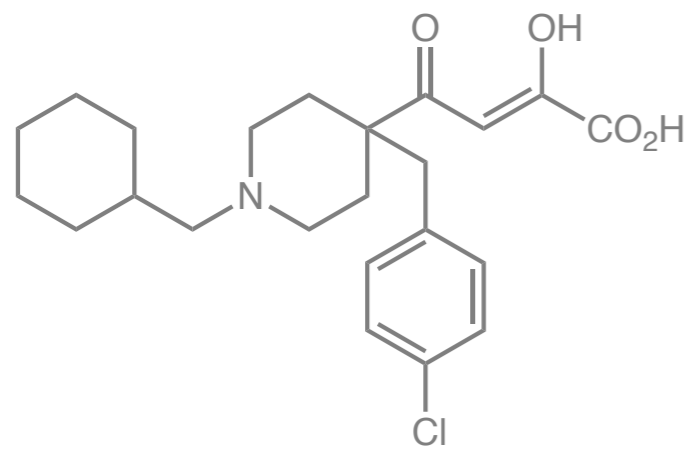
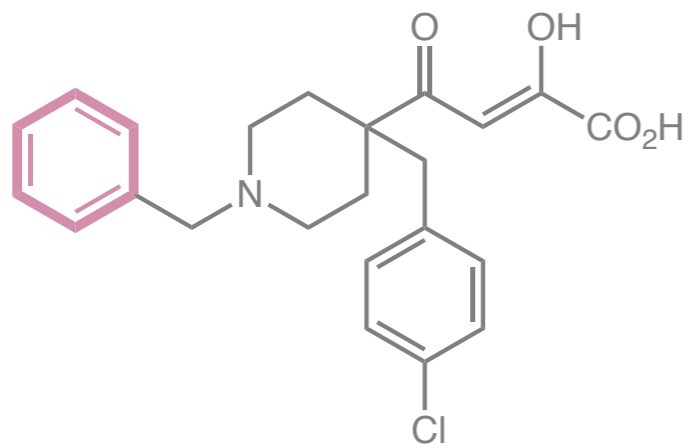


Jacobsen, K. A., *J. Med. Chem.* **2000**, 43, 1165.

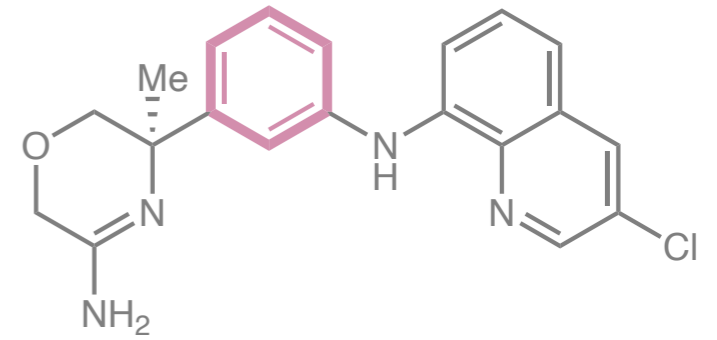
Kiesman, W. F., *J. Med. Chem.* **2006**, 49, 7119.

Xu, R., *Bioorg. Med. Chem.* **2006**, 14, 3285.

Outline



***metabolic stability when
even the strongest C–H
bonds get cleaved***

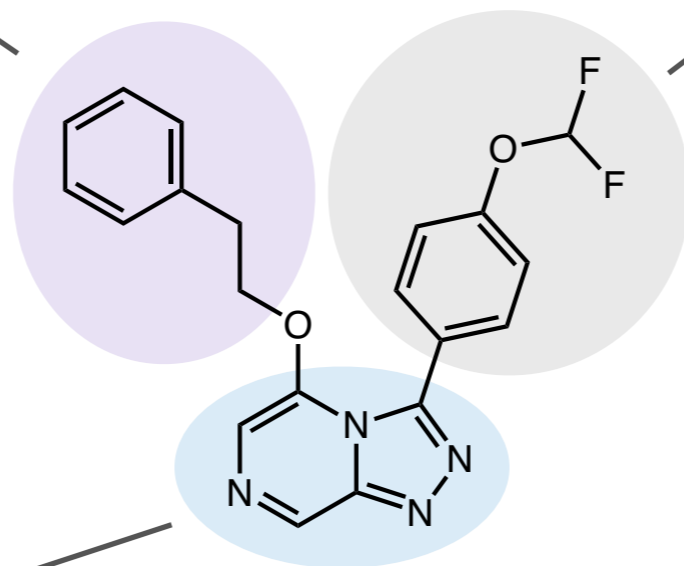


A Case Study in Antimalarial Compounds

Northwestern fragment

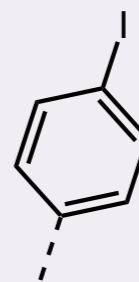
Northeastern fragment

core



Series 4
Open Source Malaria Consortium
PfATP4 inhibitor

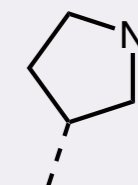
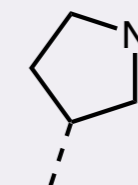
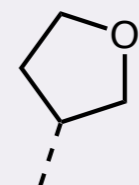
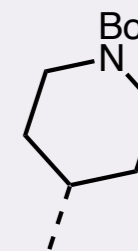
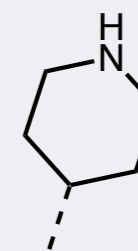
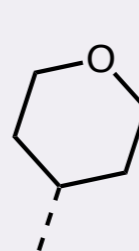
poorly soluble, quickly metabolized



– equipotent
– worse solubility

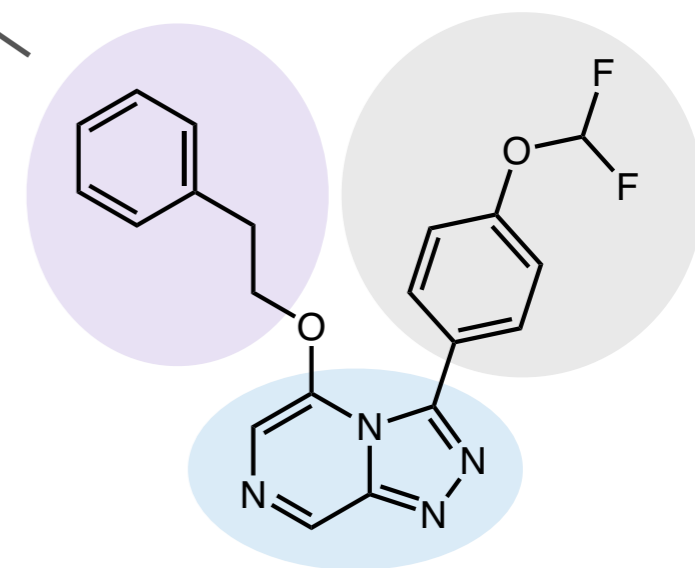
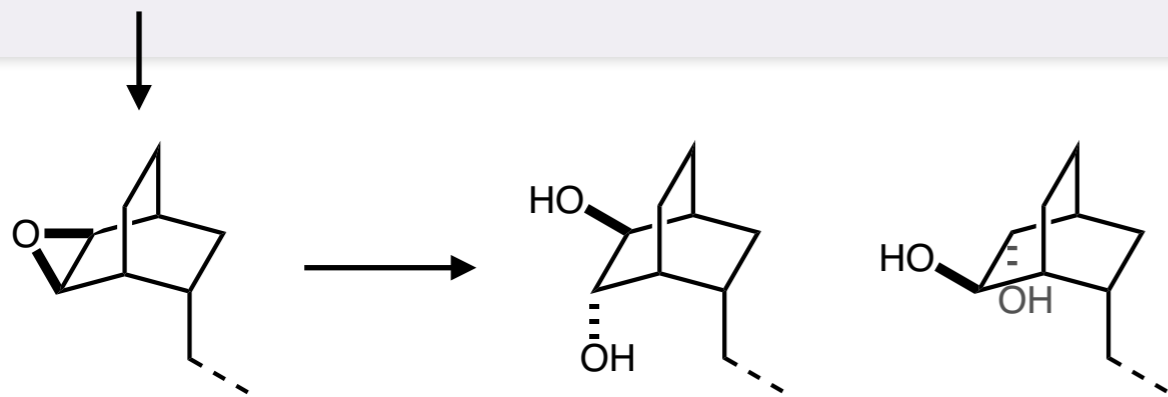
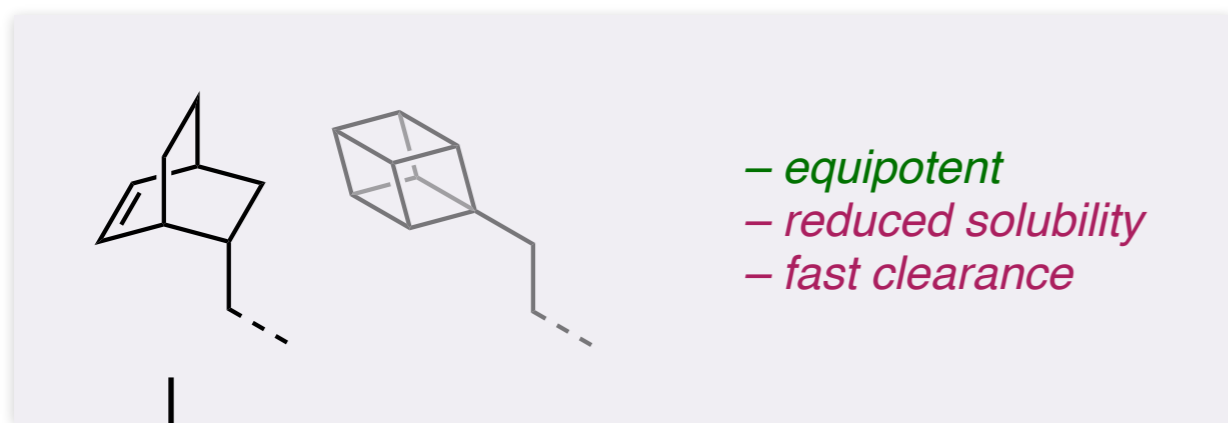
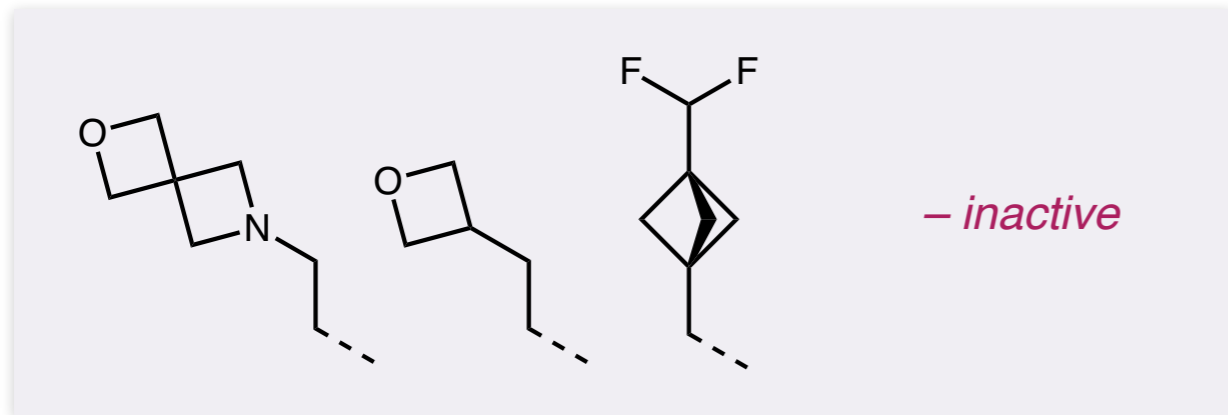


– inactive



A Case Study in Antimalarial Compounds

Northwestern fragment

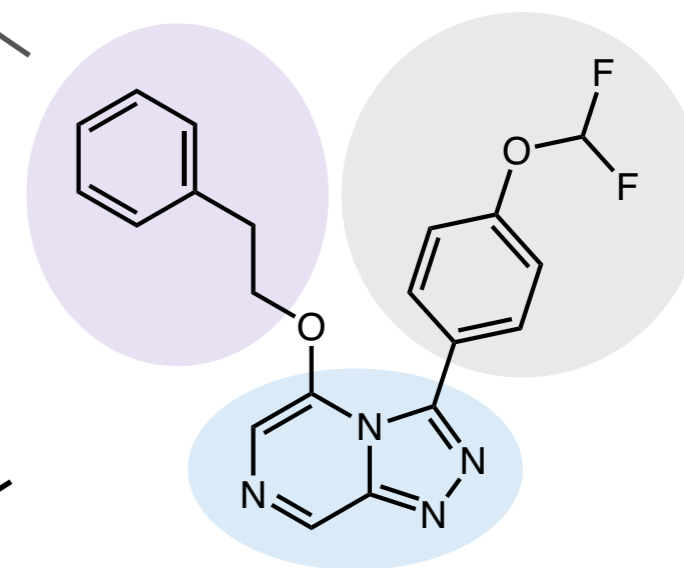
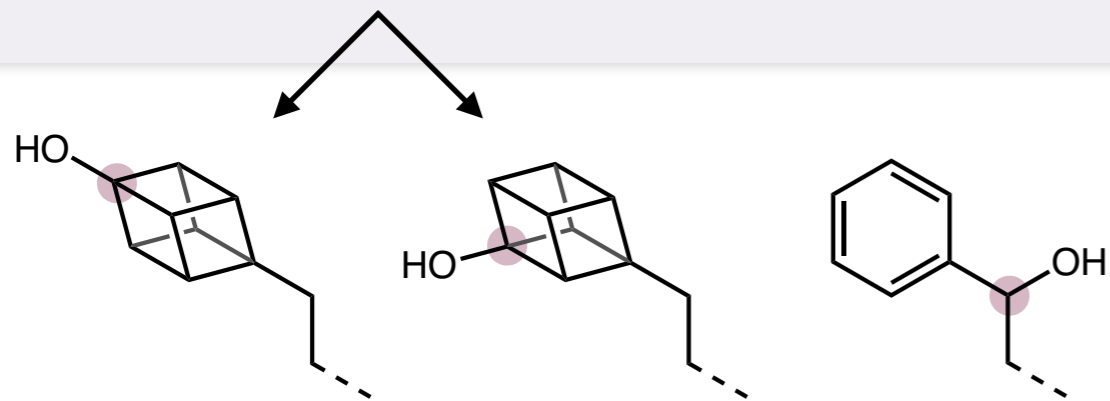
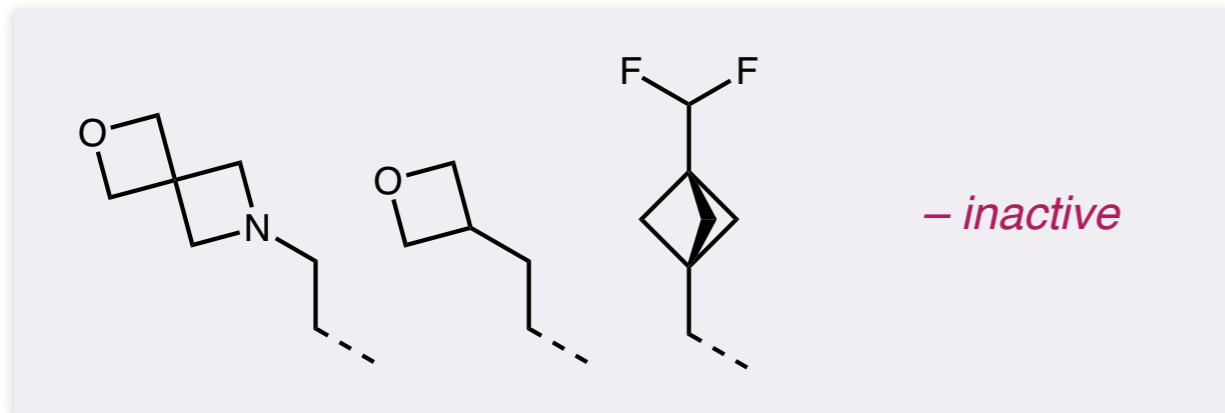


Series 4
Open Source Malaria Consortium
PfATP4 inhibitor

poorly soluble, quickly metabolized

A Case Study in Antimalarial Compounds

Northwestern fragment

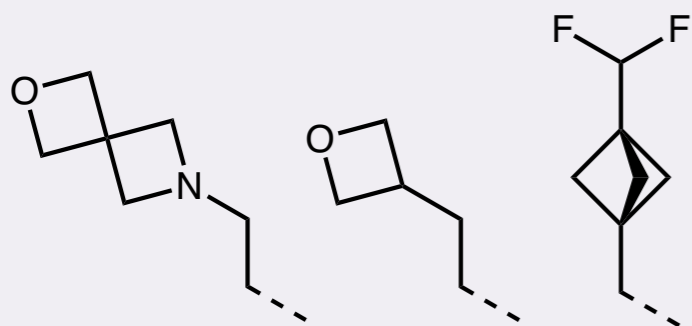


Series 4
Open Source Malaria Consortium
PfATP4 inhibitor

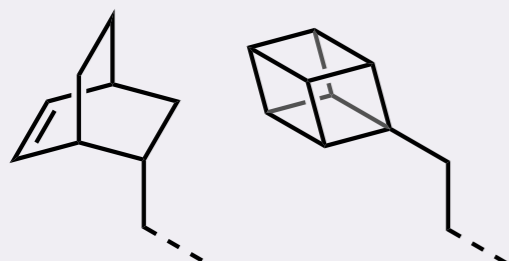
poorly soluble, quickly metabolized

A Case Study in Antimalarial Compounds

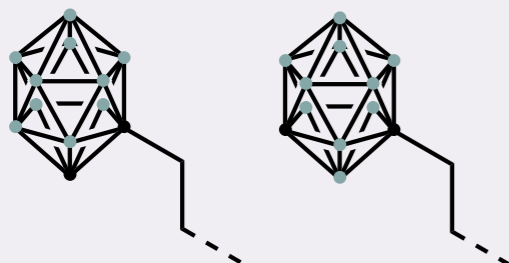
Northwestern fragment



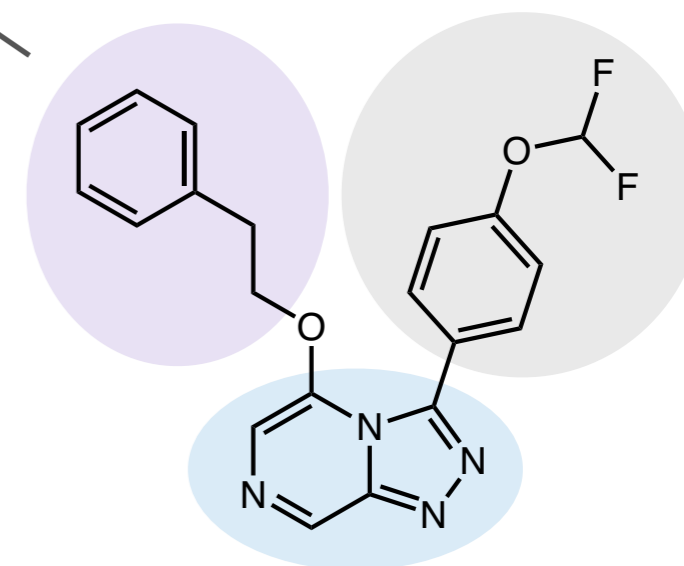
– inactive



– equipotent
– reduced solubility
– fast clearance



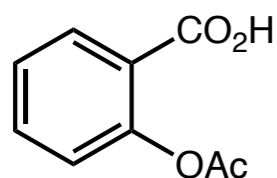
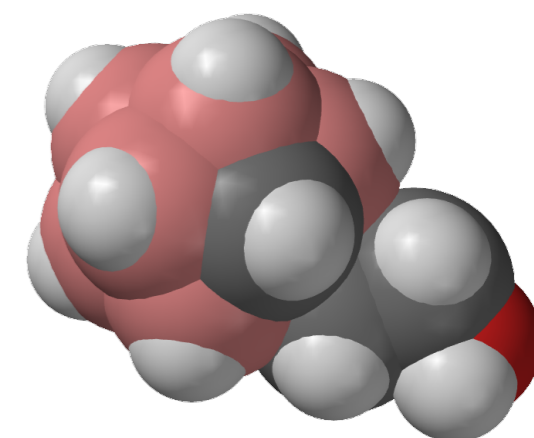
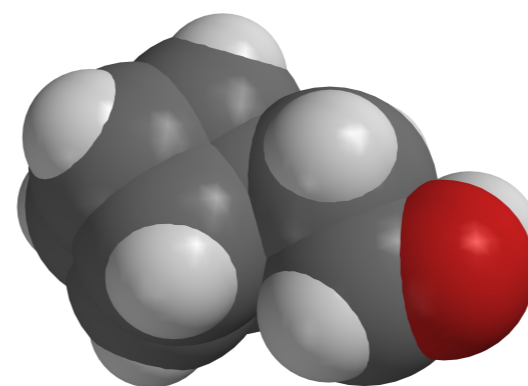
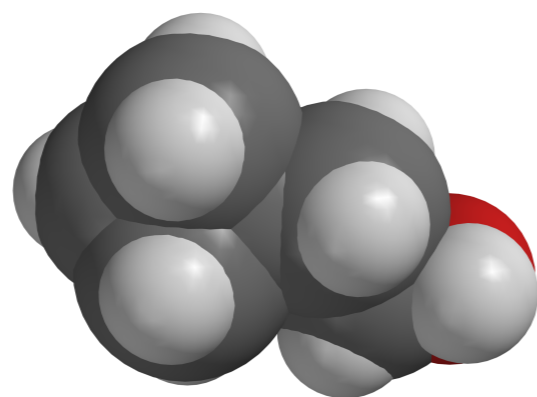
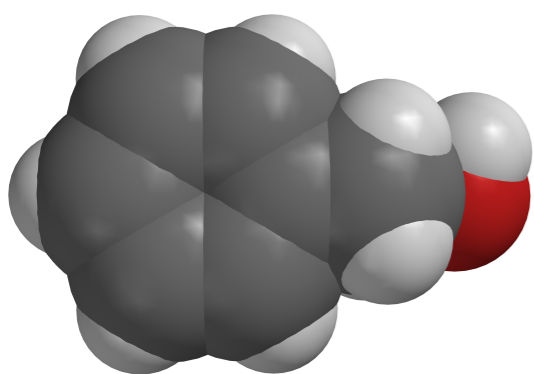
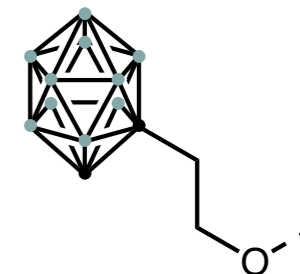
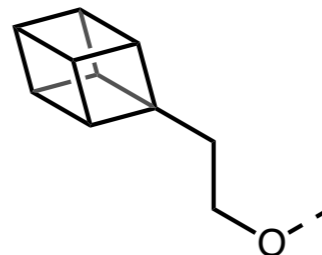
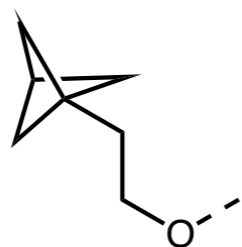
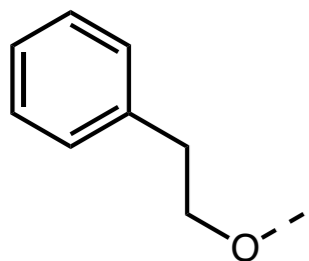
• = C–H/R • = B–H



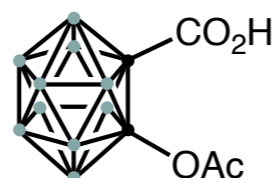
Series 4
Open Source Malaria Consortium
PfATP4 inhibitor

poorly soluble, quickly metabolized

Carboranes as Phenyl Bioisosteres



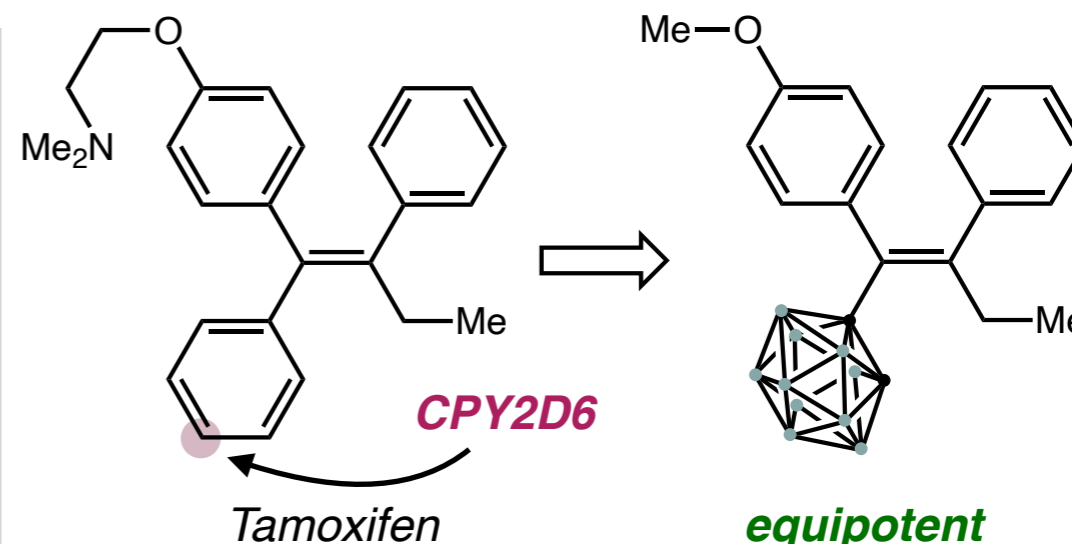
Aspirin



Asborin

IC_{50} (COX-1) = 3.57 μ M
 IC_{50} (AKR-1A1) > 1000 μ M

IC_{50} (COX-1) \approx 500 μ M
 IC_{50} (AKR-1A1) = 1.4 μ M



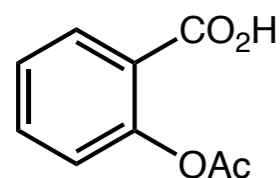
equipotent
much slower metabolism

Hey-Hawkins, E., *Chem. Soc. Rev.* **2019**, 48, 3497.

Ze, S. Q., *J. Med. Chem.* **2012**, 55, 7290.

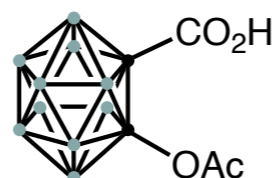
Ye, S. Q., *Drug. Des. Devel. Ther.* **2017**, 11, 629.

Carboranes as Phenyl Bioisosteres



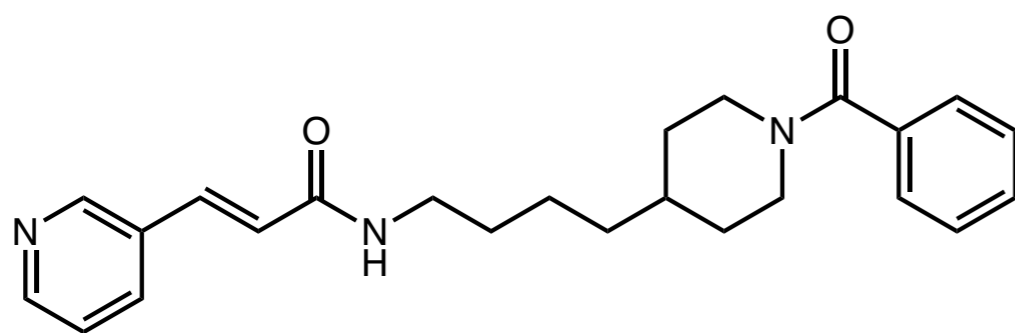
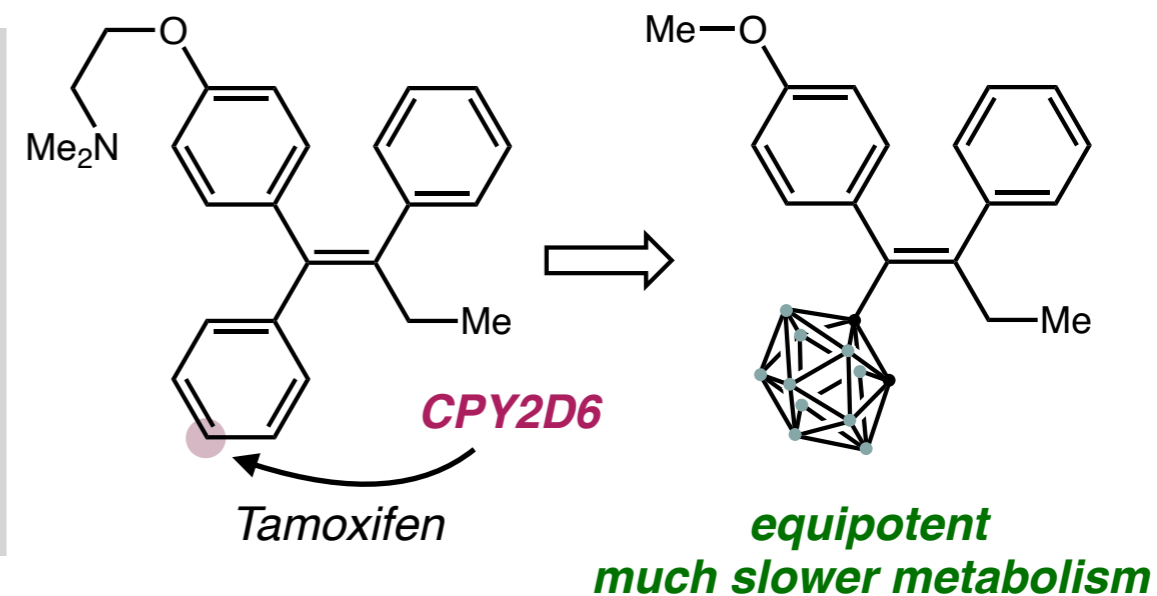
Aspirin

IC50 (COX-1) = 3.57 μ M
IC50 (AKR-1A1) > 1000 μ M



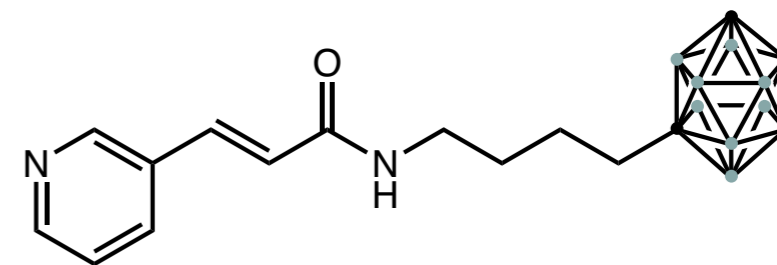
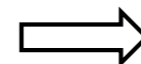
Asborin

IC50 (COX-1) \approx 500 μ M
IC50 (AKR-1A1) = 1.4 μ M



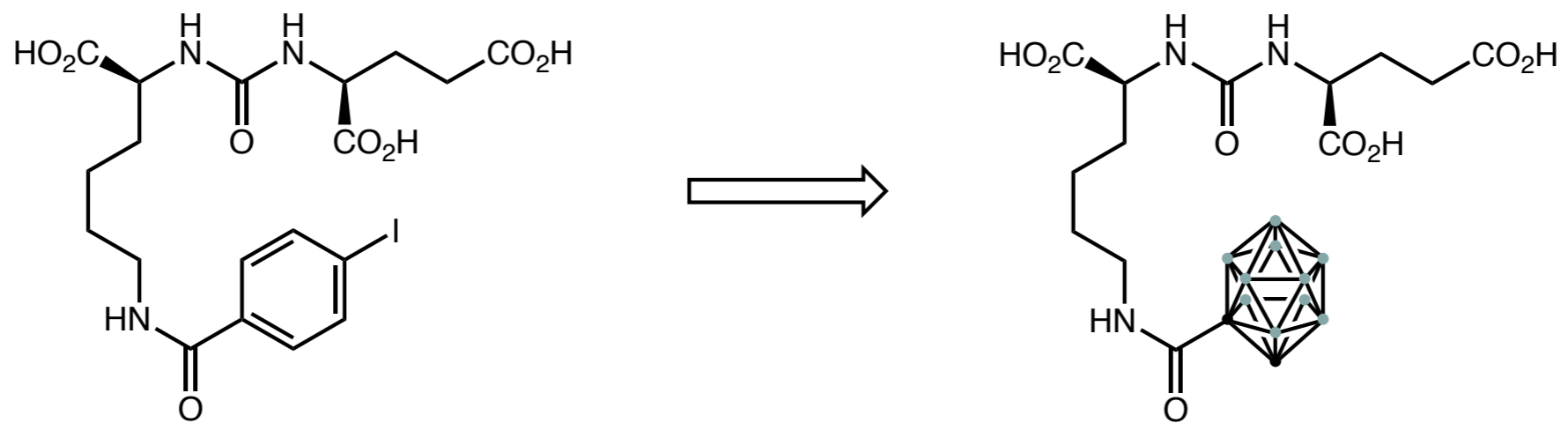
Daporinad
 NAMPT antagonist
IC50 0.09 nM

dosage-related adverse effects in P2



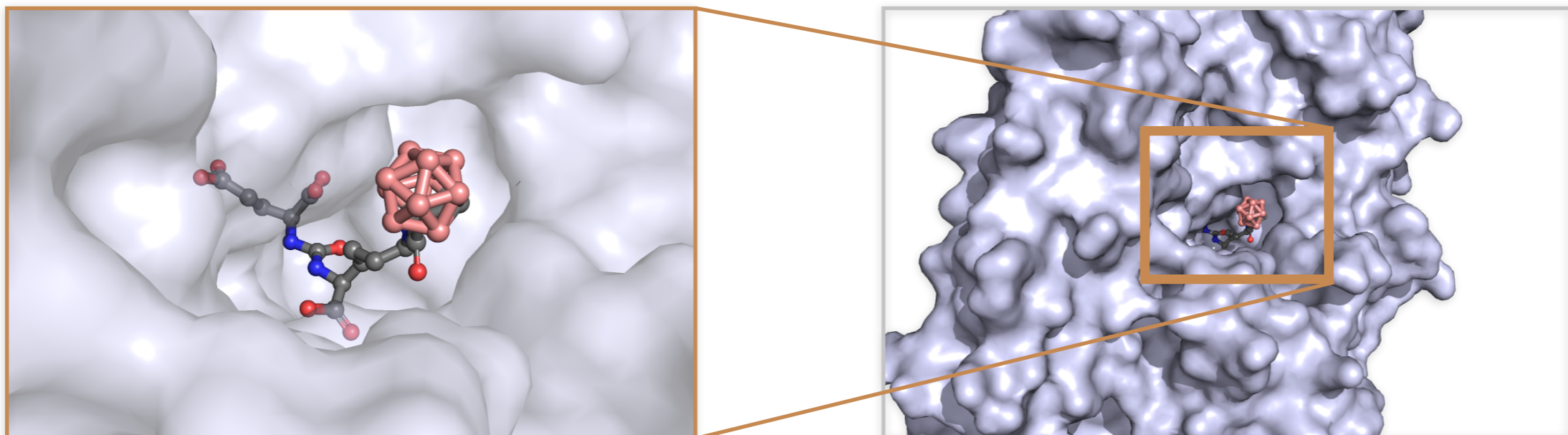
equipotent
slower metabolism
→ lower effective dose

Carboranes as Phenyl Bioisosteres



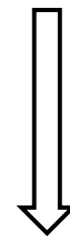
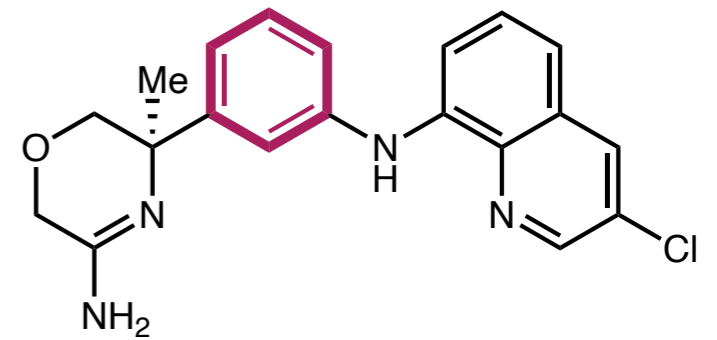
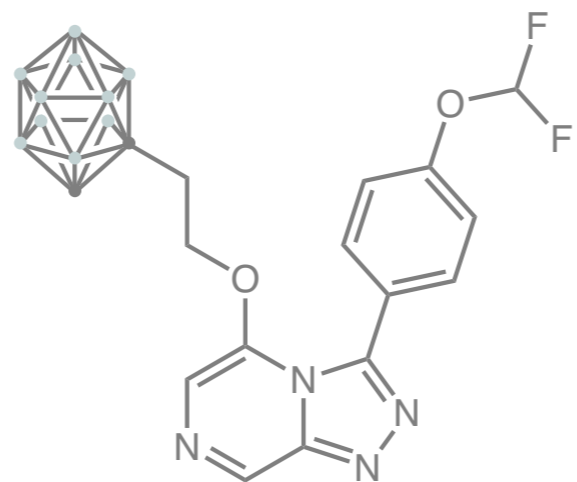
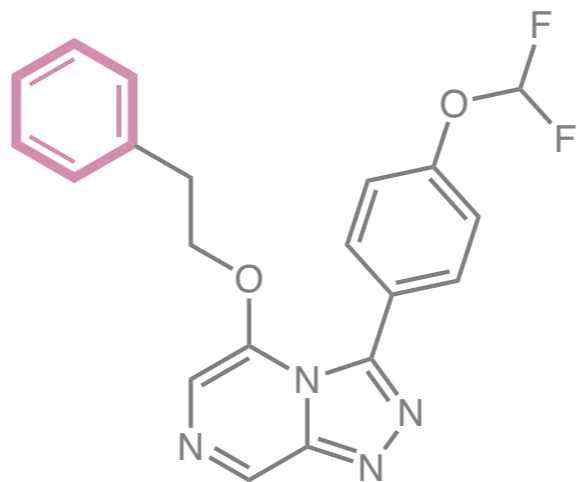
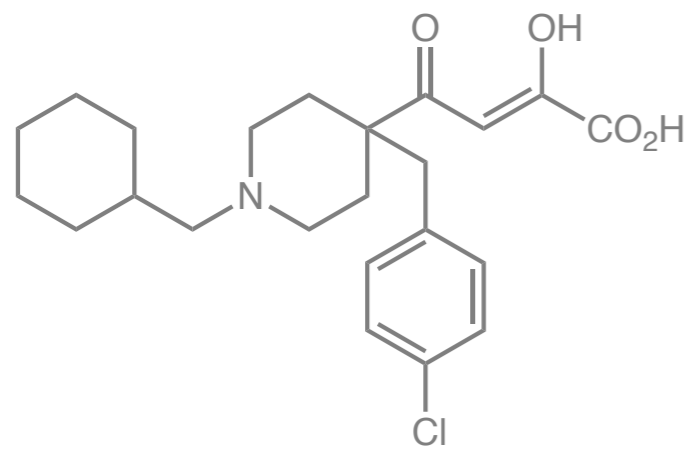
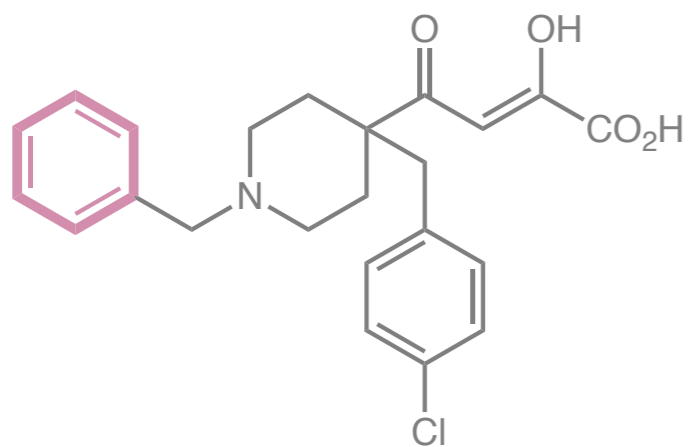
GCP11 inhibitor
 $IC_{50} \approx 1 \text{ nM}$

$IC_{50} = 15.6 \text{ nM}$



Hey-Hawkins, E., *Chem. Soc. Rev.* **2019**, 48, 3497.
Ze, S. Q., *J. Med. Chem.* **2012**, 55, 7290.
Byun, Y., *Bioorg. Med. Chem. Lett.* **2015**, 25, 5232.

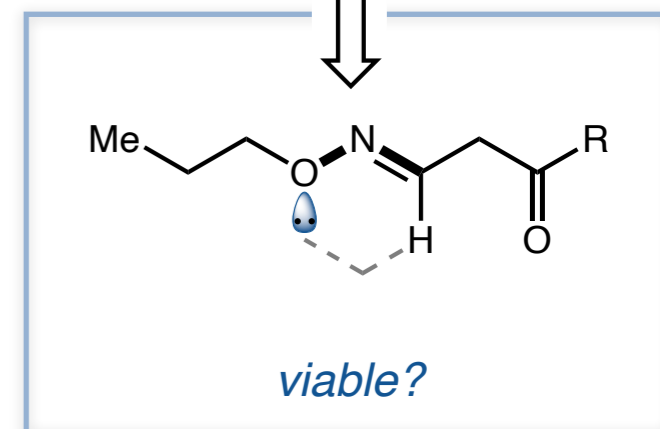
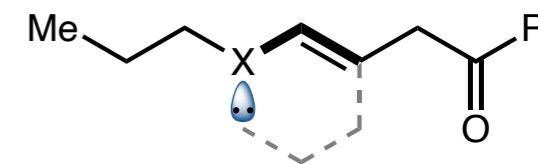
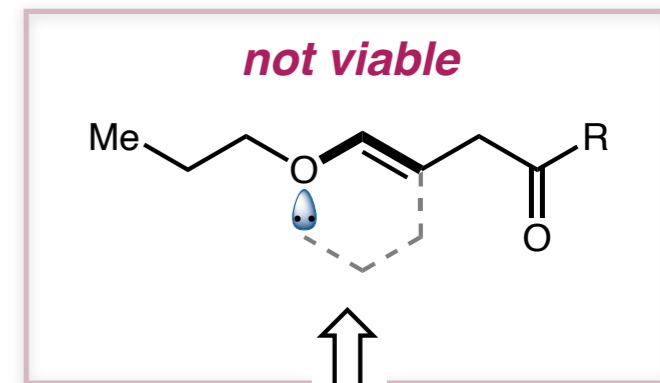
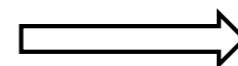
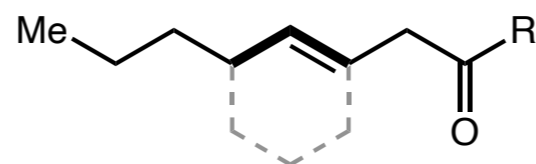
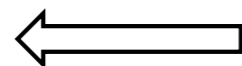
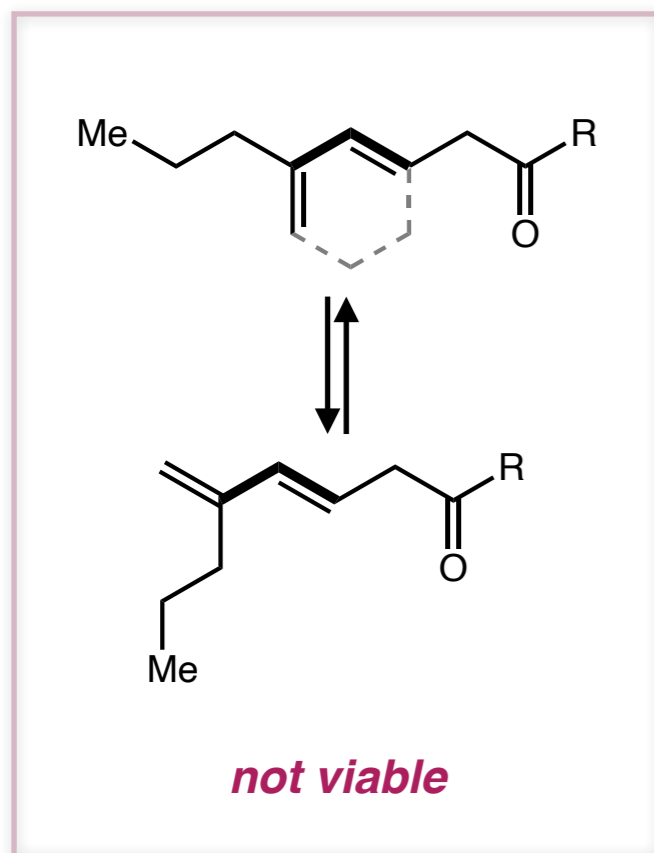
Outline



***acyclic bioisosteres by
reducing open chain
flexibility***

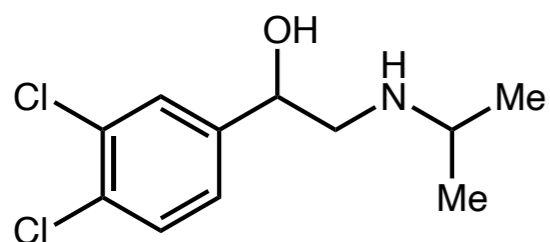
Open Chain Oximes as Phenyl Bioisosteres

increasing molecular rigidity in scaffolds leads to improved selectivity

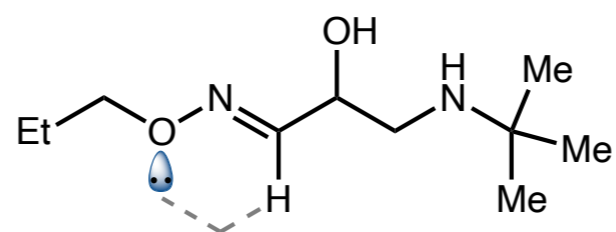
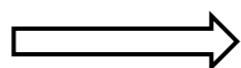


How to introduce rigidity?

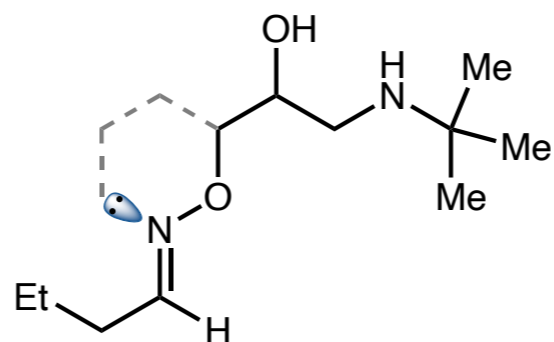
Open Chain Oximes as Phenyl Bioisosteres



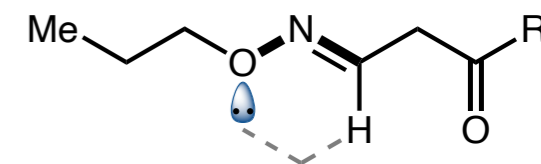
Cl₂-isoprenaline
nonselective β -adrenoceptor agonist
 IC_{50} : $\beta_1 = 0.11 \mu M$ / $\beta_2 = 0.98 \mu M$



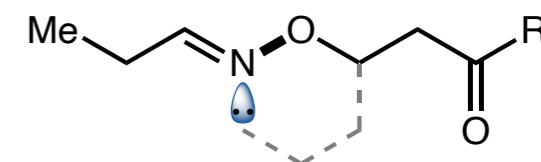
IC_{50} : $\beta_1 = 8.3 \mu M$ / $\beta_2 = 0.27 \mu M$



IC_{50} : $\beta_1 = 26 \mu M$ / $\beta_2 = 1.5 \mu M$

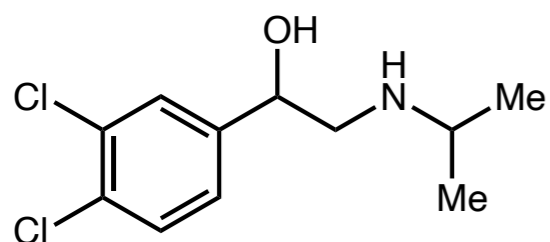


viable

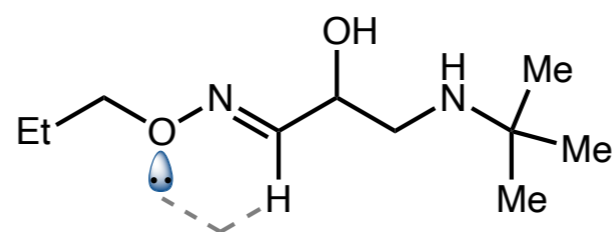
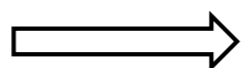


viable, too?

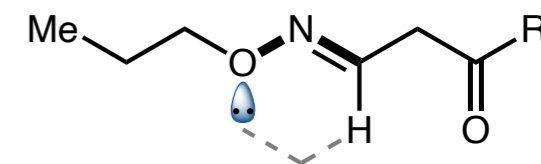
Open Chain Oximes as Phenyl Bioisosteres



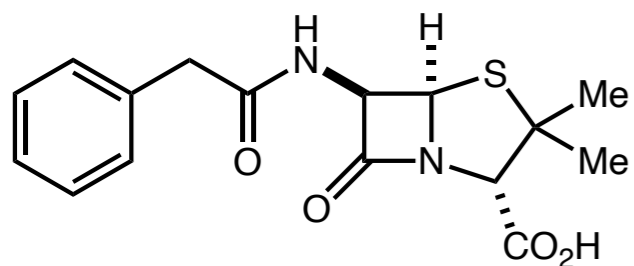
Cl₂-isoprenaline
nonselective β -adrenoceptor agonist
 IC_{50} : $\beta_1 = 0.11 \mu M$ / $\beta_2 = 0.98 \mu M$



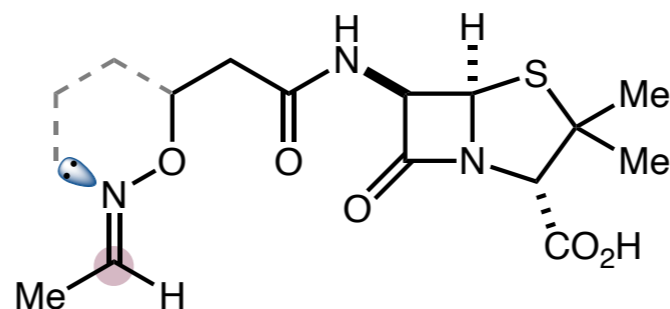
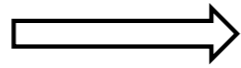
IC_{50} : $\beta_1 = 8.3 \mu M$ / $\beta_2 = 0.27 \mu M$



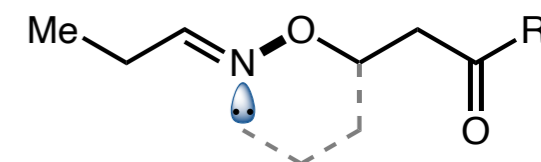
viable



penicillin G
 MIC (gram+): $0.05 \mu g/mL$
 MIC (gram-): $71 \mu g/mL$

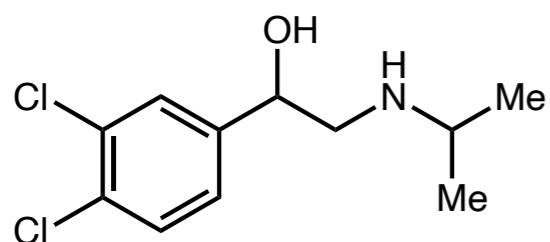


inactive
prone to hydrolysis

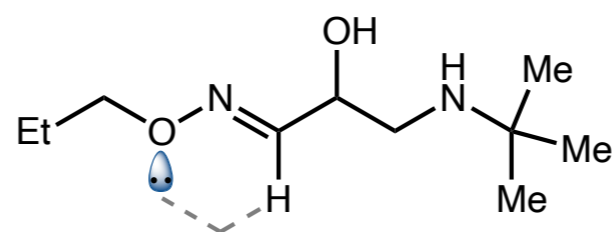
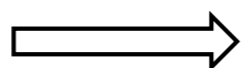


viable, too?

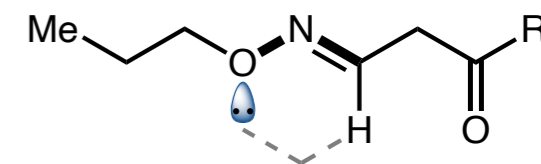
Open Chain Oximes as Phenyl Bioisosteres



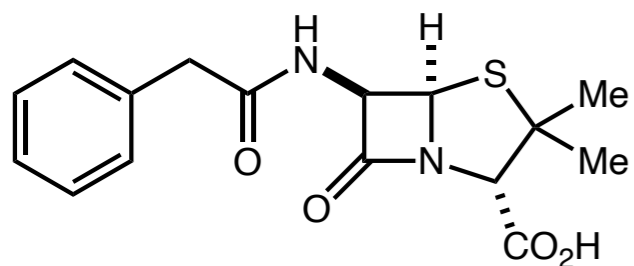
Cl₂-isoprenaline
nonselective β-adrenoceptor agonist
IC₅₀: β₁ = 0.11 μM / β₂ = 0.98 μM



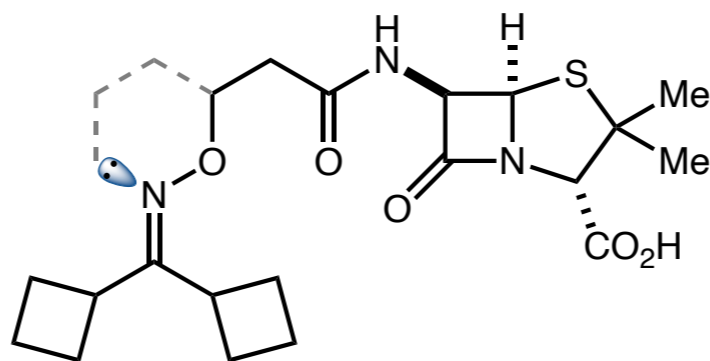
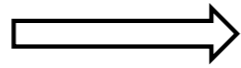
IC₅₀: β₁ = 8.3 μM / β₂ = 0.27 μM



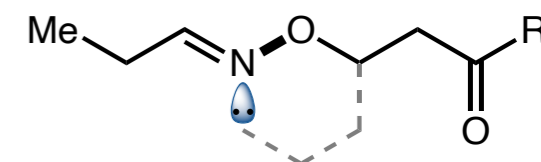
viable



penicillin G
MIC (gram+): 0.05 μg/mL
MIC (gram-): 71 μg/mL

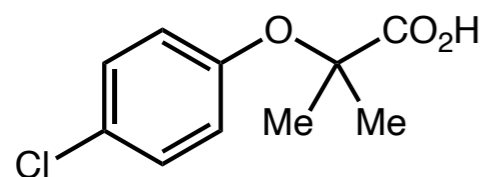


MIC (gram+): 0.06 μg/mL
MIC (gram-): 53 μg/mL



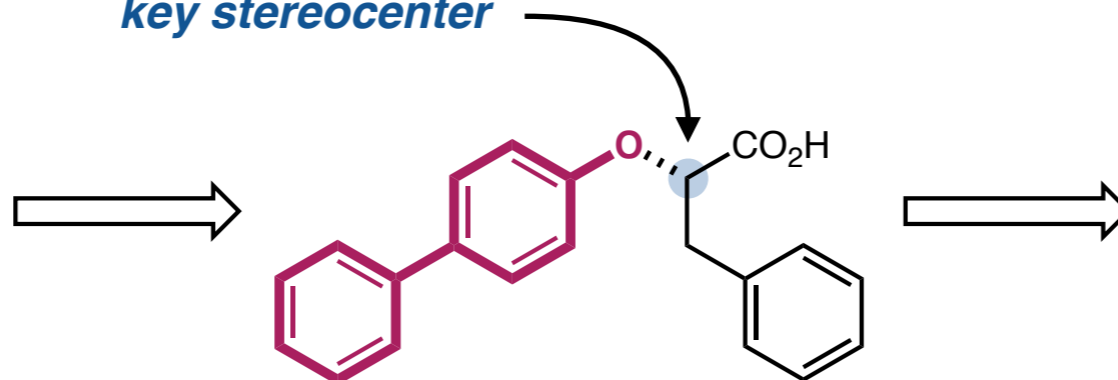
viable

Open Chain Oximes as Phenyl Bioisosteres

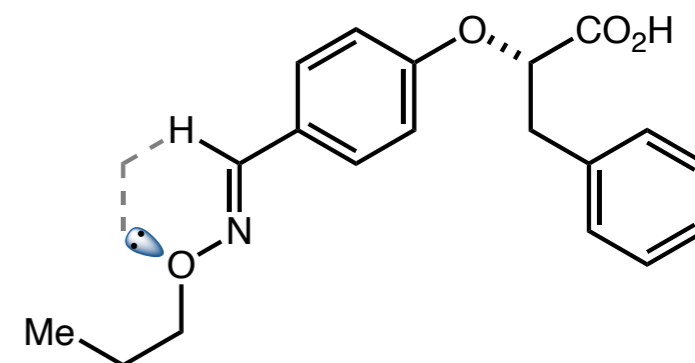


clofibric acid
core of several classes
of PPAR α /PPAR γ agonists

key stereocenter



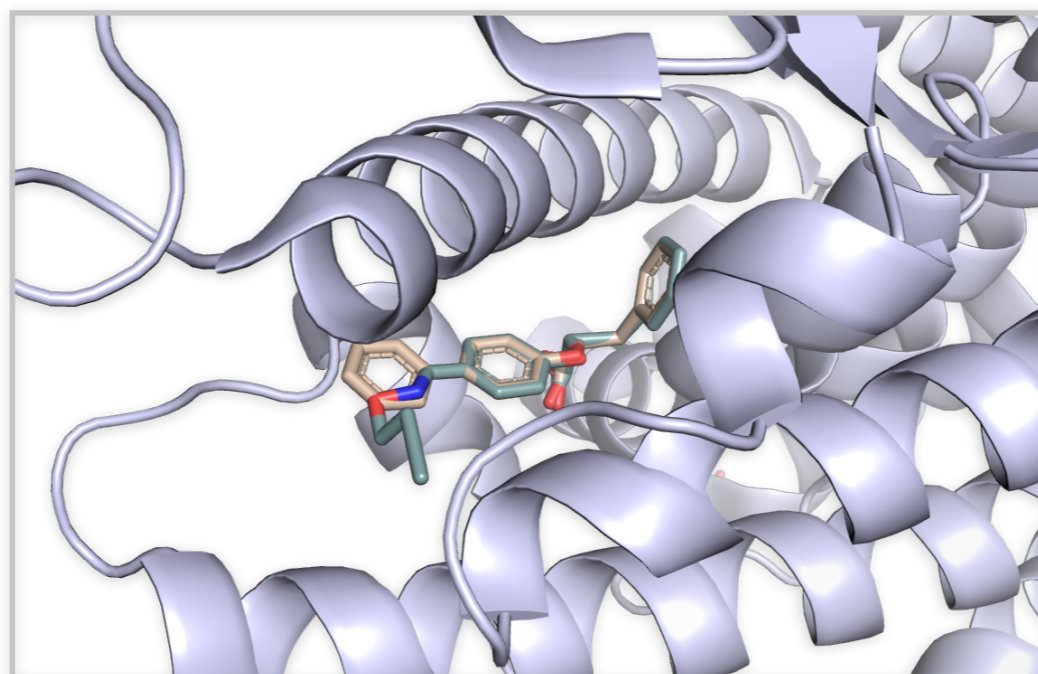
dual agonist
PPAR α EC₅₀ = 0.22 μ M
PPAR γ EC₅₀ = 0.48 μ M
metabolism?



PPAR α EC₅₀ = 3.8 μ M
PPAR γ EC₅₀ = 3.8 μ M

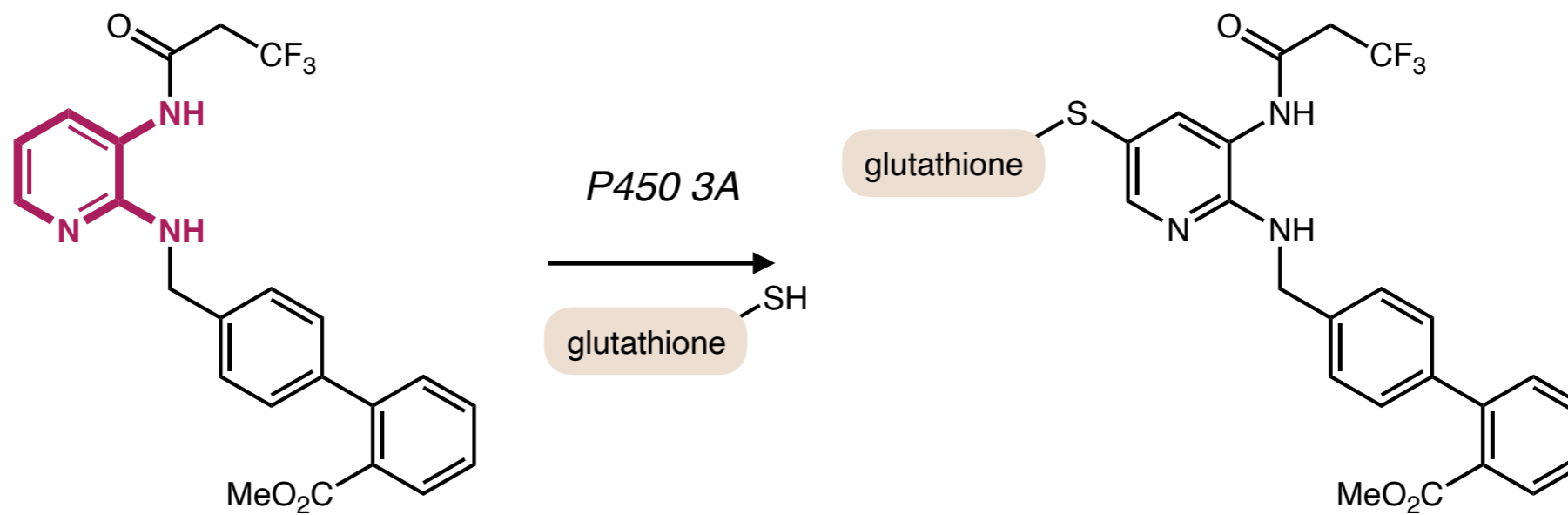
efficacy relative to biphenyl:
PPAR α = 91%
PPAR γ = 154%

*docking of oxime onto
biphenyl co-crystal:
additional H-bond with
Gln277 rigidifies the
complex*



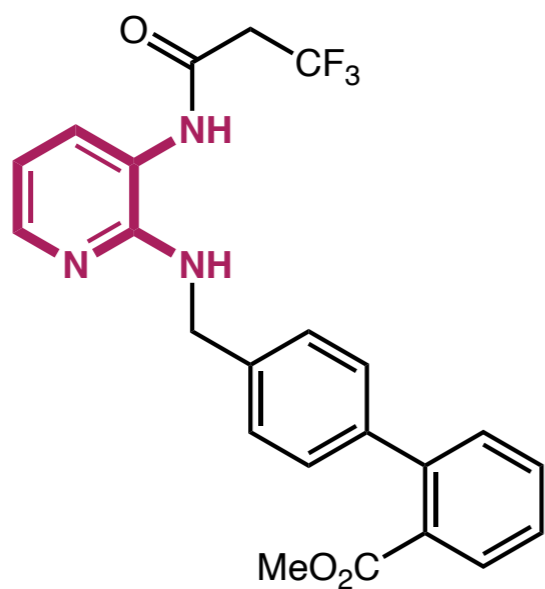
Liodice, F., *J. Med. Chem.* **2005**, 48,5509.
Pochetti, G., *J. Med. Chem.* **2008**, 51, 7768.
Liodice, F., *Eur. J. Med. Chem.* **2015**, 90, 583.

α -Cyclopropyl Carbonyl Compounds

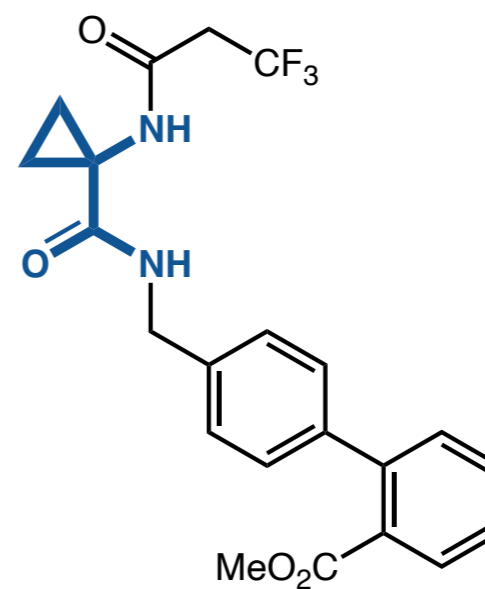
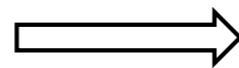


Bradykinin B1 receptor antagonist
 $K_i = 11.8 \text{ nM}$
Liver Toxicity!

α -Cyclopropyl Carbonyl Compounds

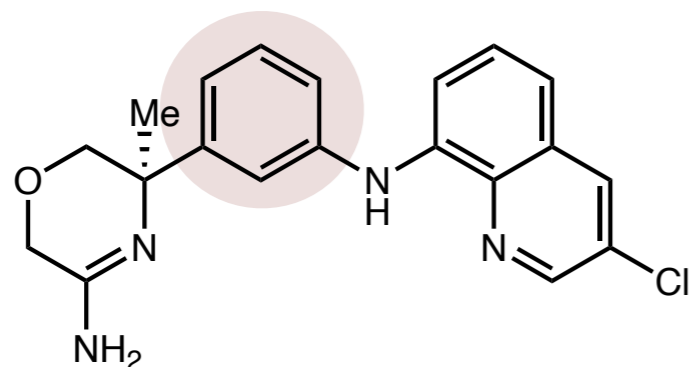
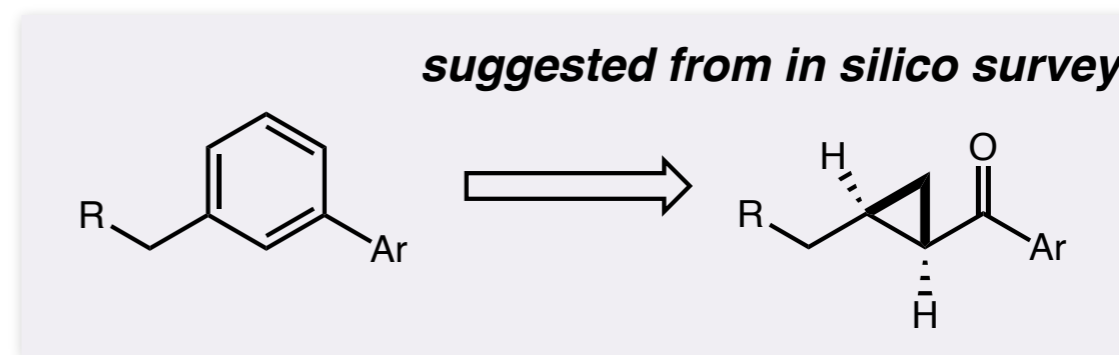
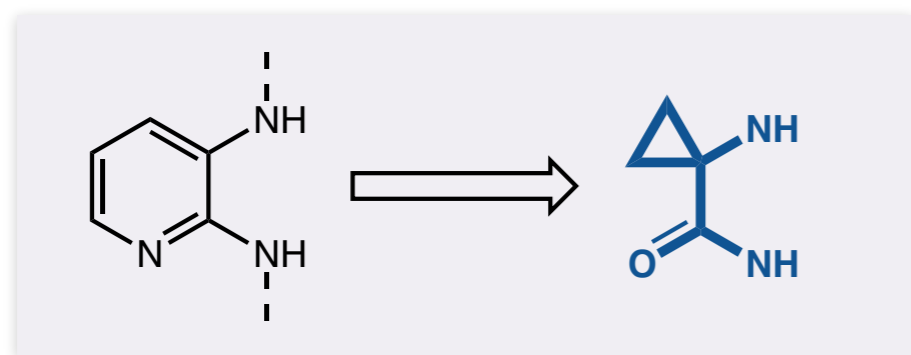


Bradykinin B1 receptor antagonist
 $K_i = 11.8 \text{ nM}$
Liver Toxicity!

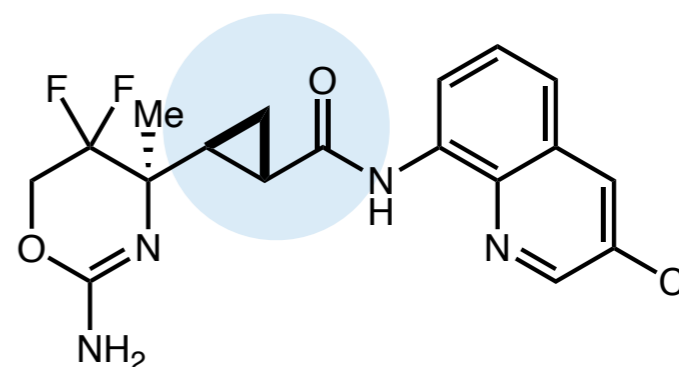
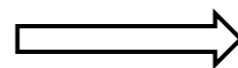


$K_i = 63.0 \text{ nM}$
significantly lower GSH metabolism

α -Cyclopropyl Carbonyl Compounds

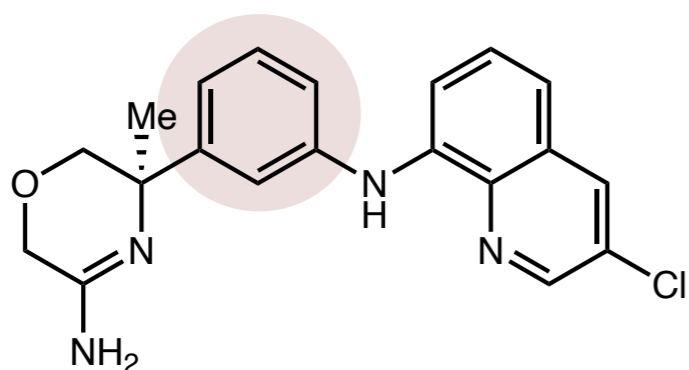


β -secretase antagonist
 $IC_{50} = 250 \text{ nM}$
solubility (pH 6.5) < 1 $\mu\text{g/mL}$

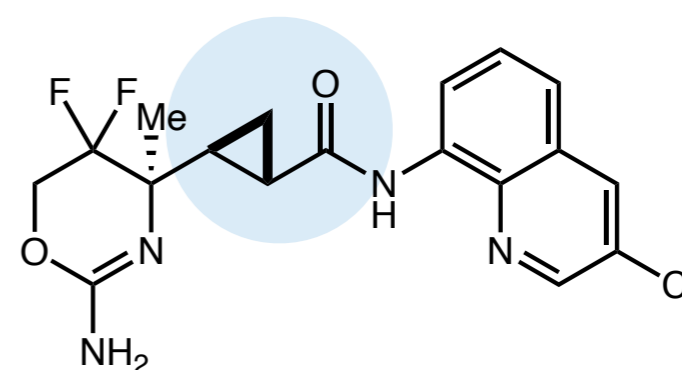
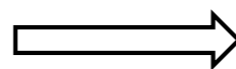


$IC_{50} = 420 \text{ nM}$
solubility (pH 6.5) = 64 $\mu\text{g/mL}$

α -Cyclopropyl Carbonyl Compounds

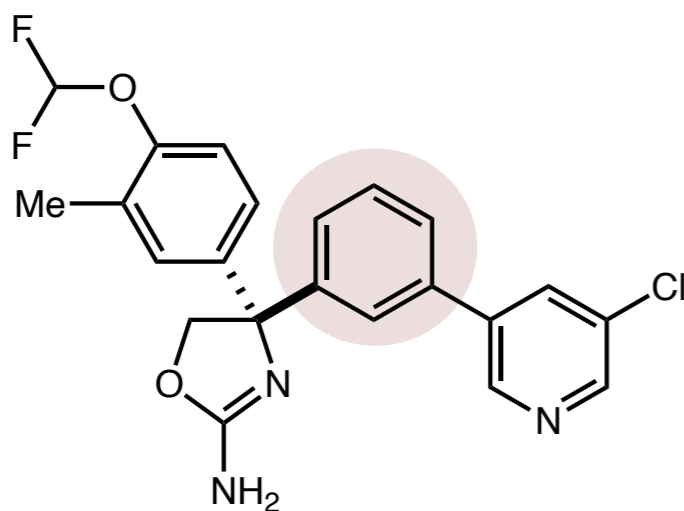


β -secretase antagonist
 $IC_{50} = 250 \text{ nM}$
solubility (pH 6.5) < 1 $\mu\text{g/mL}$

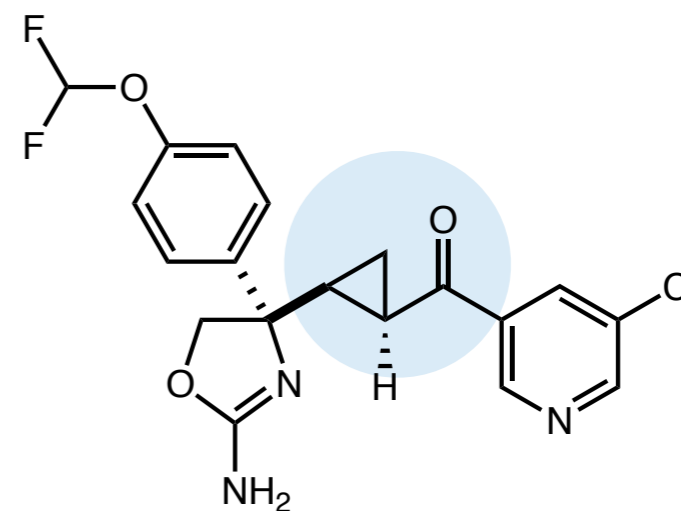
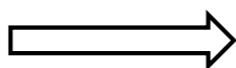


$IC_{50} = 420 \text{ nM}$
solubility (pH 6.5) = 64 $\mu\text{g/mL}$

2nd Series

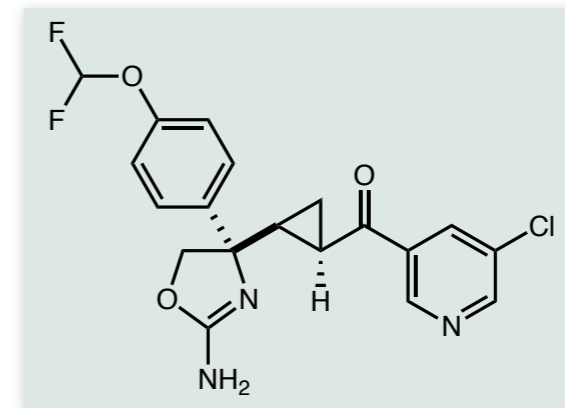
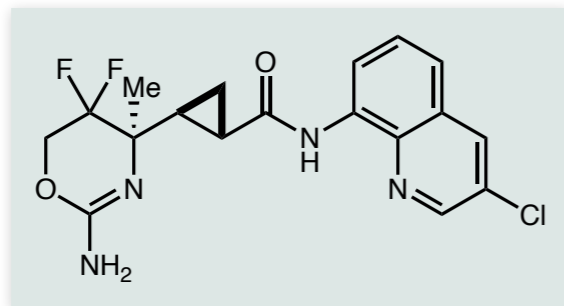
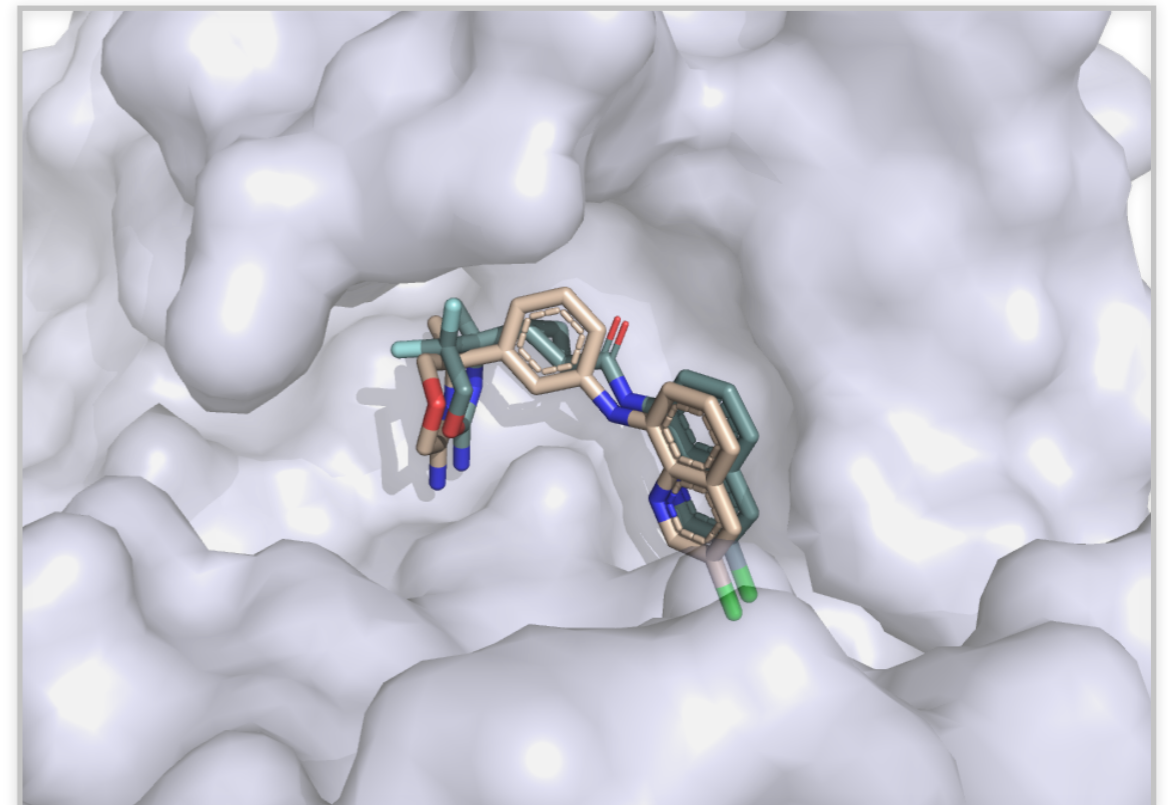
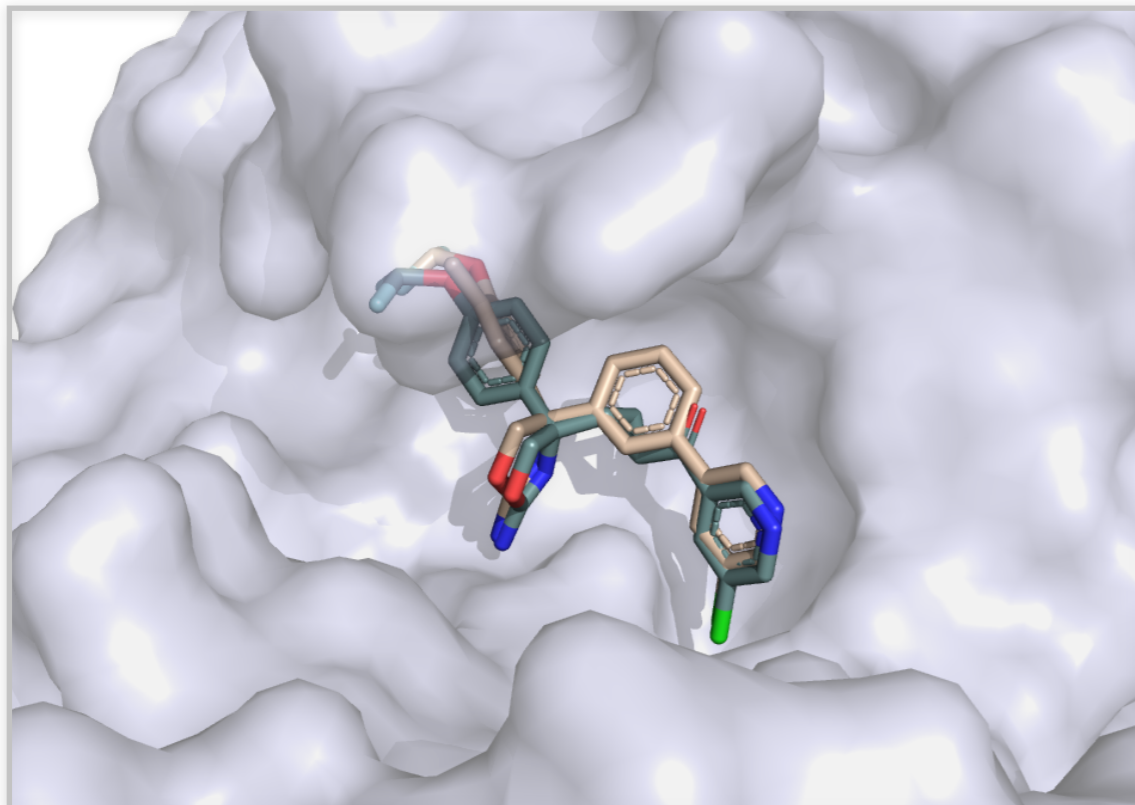
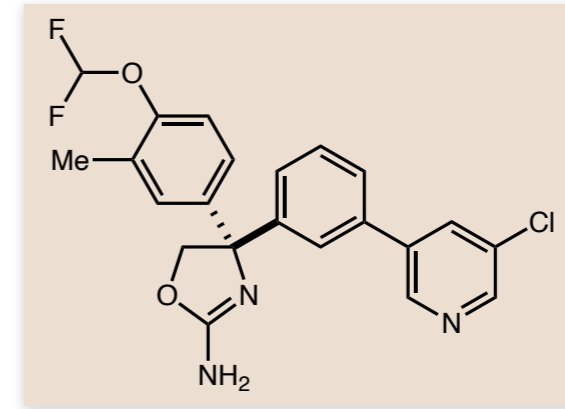
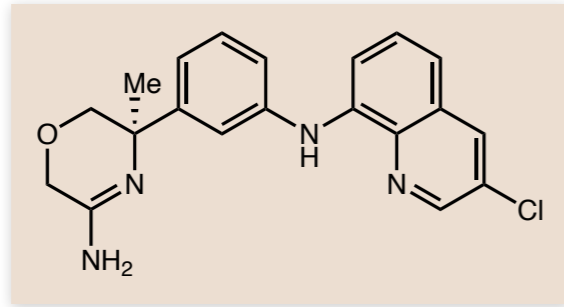


$IC_{50} = 145 \text{ nM}$
solubility (pH 6.5) < 1 $\mu\text{g/mL}$



$IC_{50} = 87 \text{ nM}$
solubility (pH 6.5) = 109 $\mu\text{g/mL}$

α -Cyclopropyl Carbonyl Compounds



An Analgesia Candidate with Synergistic Activity

N-substituent

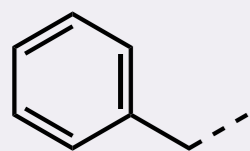
core

Southeastern fragment

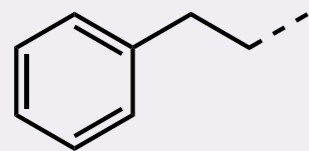
H

loss of activity

hERG IC50

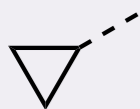


0.1 μ M

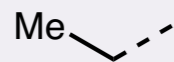


0.2 μ M

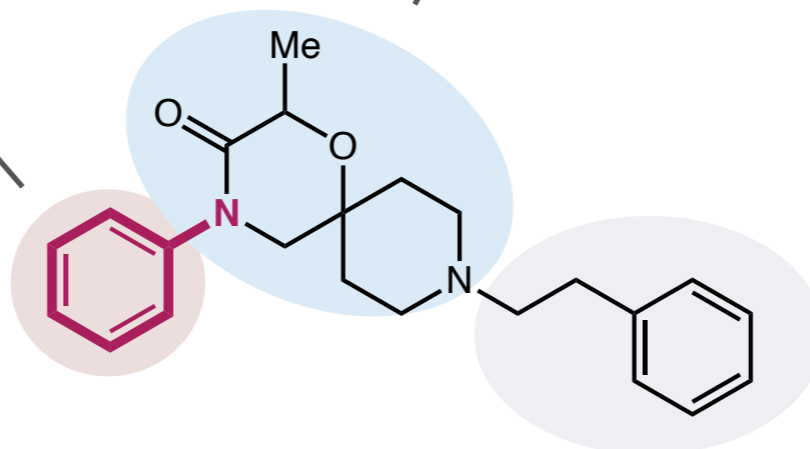
hERG IC50



4.5 μ M



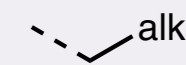
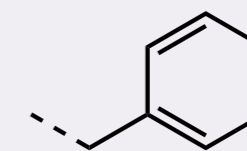
3.1 μ M



initial lead

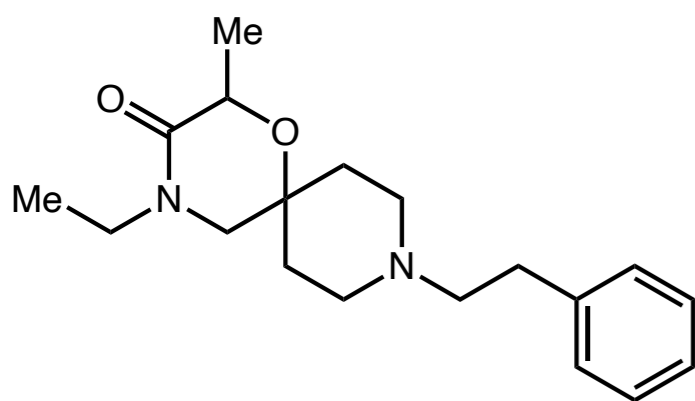
μ opioid receptor (MOR) agonist; $EC_{50} = 14$ nM
 σ_1 receptor antagonist; $K_i = 6$ nM

$cLogP = 4.2$
 $hERG IC_{50} = 0.4$ μ M



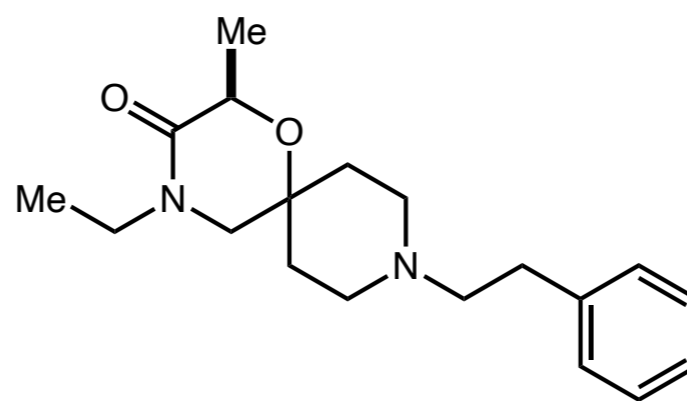
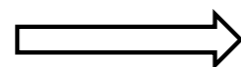
loss of affinity

An Analgesia Candidate with Synergistic Activity



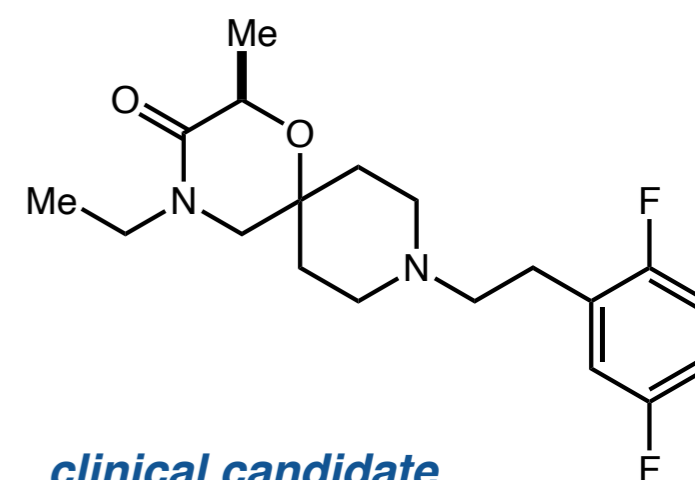
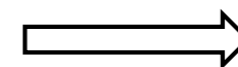
advanced compound
MOR EC₅₀ = 65 nM
 σ_1R K_i = 43 nM

cLogP = 2.9
hERG IC₅₀ = 3.1 μ M



advanced compound
MOR EC₅₀ = 49 nM
 σ_1R K_i = 66 nM

cLogP = 2.9
hERG IC₅₀ = 4.7 μ M

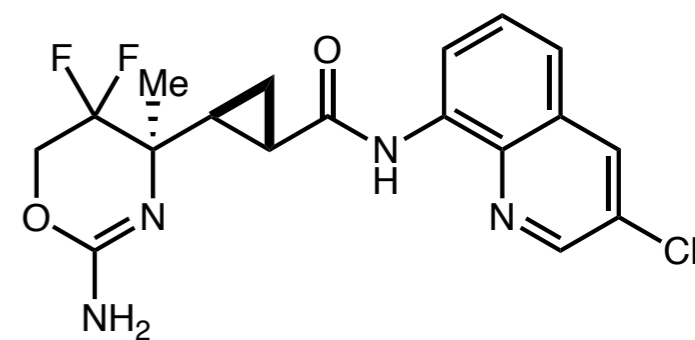
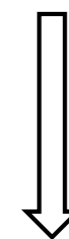
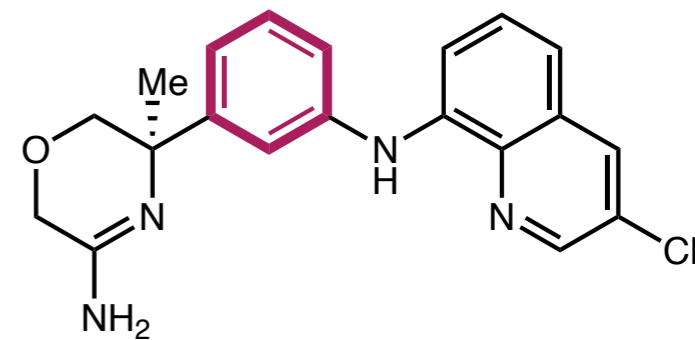
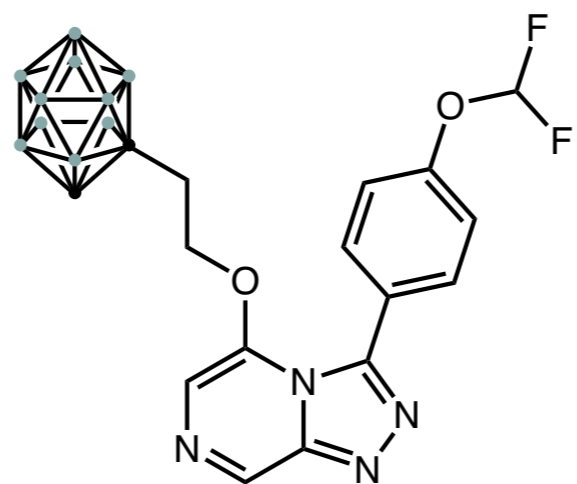
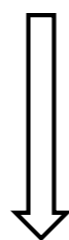
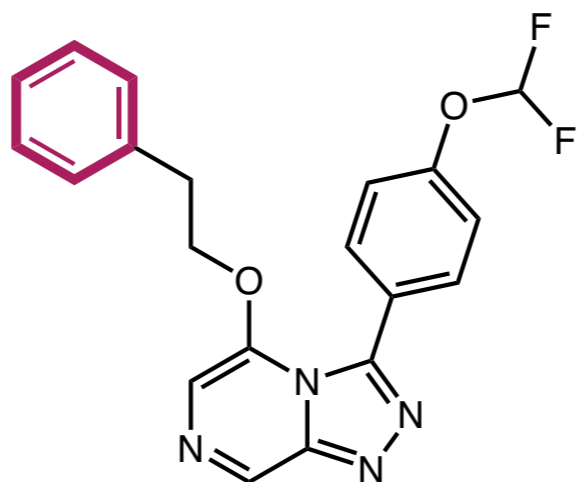
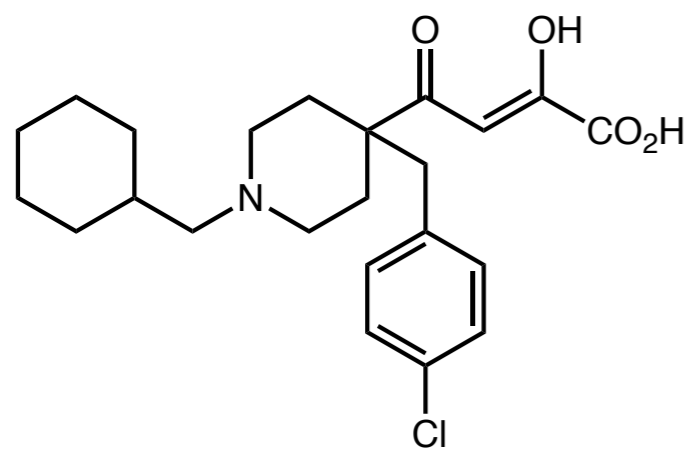
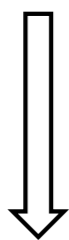
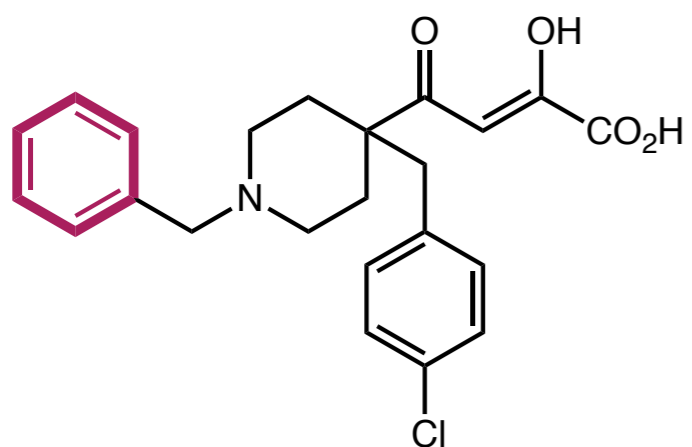


clinical candidate
MOR EC₅₀ = 52 nM
 σ_1R K_i = 118 nM

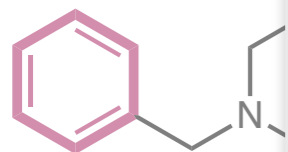
cLogP = 3.2
hERG IC₅₀ > 10 μ M

*“reduced behavioral signs
associated with opioid withdrawal
following repeat drug dosing”*

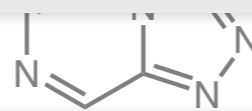
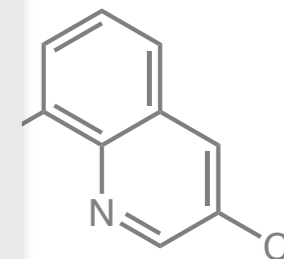
Phenyl Bioisosterism – A Solved Challenge?



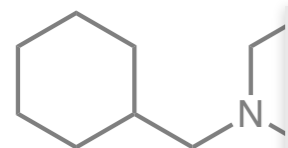
Phenyl Bioisosterism – A Solved Challenge?



Our predictive understanding of SAR and bioisosteric replacement is still in its infancy but the growing amount of data helps both us and machines to improve.



A growing number of scaffolds (and means to synthesize them) does not just lead to better ways to alter physicochemical properties but also to investigate SAR.



There is no “one size fits all” bioisostere. Different bioisosteres may have varying advantages and disadvantages in a given scaffold. An ideal replacement synergizes well with the binding mode.

